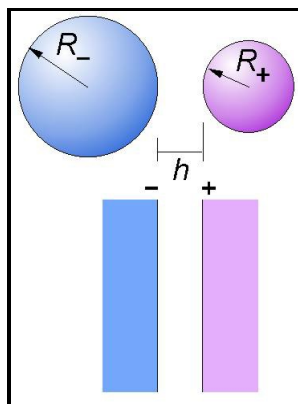


Surface chemistry and colloids

Butterworth - CHAPTER 5



Description: -

-Surface chemistry and colloids

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Physical chemistry series 1 -- Vol 7Surface chemistry and colloids

Notes: MTP International review of science.

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SCI

For example: the solutions of metals like Ag and Au, hydroxides Fe OH 3, metal sulphides like AS 2S 3 etc.

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It is an actinide element, the sixth transuranium element to be synthesized, and has the second-highest atomic mass of all the elements that have been produced in amounts large enough to see with the unaided eye after einsteinium. Give one example of each type.

Colloid and Surface Science

The term has been used to refer to clusters of that arise via liquid-liquid or liquid-solid within cells.

Class 12 Chemistry Chapter 5 Surface Chemistry Notes

When dissolved in water, it dissociates into RCOO — and Na⁺ ions. Macromolecular Colloids — Those colloidal solutions which are formed by the dispersion of macromolecules in the suitable dispersion medium are called macromolecular colloids.

Surface Chemistry: Colloids, Emulsions, Adsorption, Catalysis, Video, Q&A

What are the dispersed phase and dispersion medium in milk? Magnesium is the ninth most abundant element in the universe. The substance whose minute particles are suspended in another substance is called the dispersed phase while the substance in which it is suspended is called dispersion medium.

Class 12 Chemistry Chapter 5 Surface Chemistry Notes

Due to increase in the viscosity of the medium, globules of dispersed phase remain dispersed in dispersion medium.

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If a small quantity of an electrolyte is added to these types of sols, then dispersed phase and dispersion medium are easily get separated. Give one example for them.

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