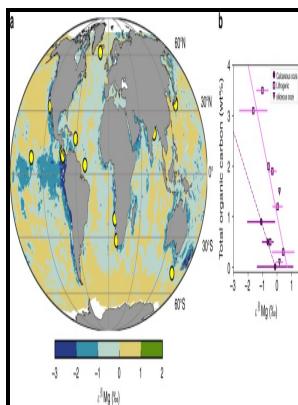


Atlas and data of solid-solution equilibria of marine evaporites

Springer - References Cited



Description: -

- Nonfiction - General
 - Westerns
 - Non-Classifiable
 - Public policy (Law) -- France -- History -- 20th century -- Congresses.
 - Phase diagrams.
 - Seawater -- Composition.
 - Salt deposits.
 - Evaporites.Atlas and data of solid-solution equilibria of marine evaporites
 -Atlas and data of solid-solution equilibria of marine evaporites
 Notes: Includes bibliographical references (p. 15-16) and index.
 This edition was published in 1998



Filesize: 32.24 MB

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(ebook) Atlas and Data of Solid

Rahnemaie, Rasoul, Hiemstra, Tjisje, and Van Riemsdijk, W. Immediate download Immediate download Digital downloads only Dispatches next business day Usually dispatches next business day + In stock at our warehouse.

Evidence for deep

In: *Atlas and Data of Solid-Solution Equilibria of Marine Evaporites*. The other one comprises two different cations A⁺, B⁺ and three different anions X⁻, Y⁻, Z⁻ which yield the salts AX, AY, AZ, BX, BY, and BZ. However, the data had not yet been summarized in a form that is at the same time comprehensive and permits to understand how the quinary seawater system builds up from its bounding systems.

[PDF] Sea

The attached computer program SALSYS allows easy and fast access to solubility data that are required by different fields. But also up to about 100 0 C data are missing for some of the systems so that they cannot be described entirely.

Solid

The second author entered the preparation of the volume when it was realized that besides graphics and tables a fast access to data was required.

[PDF] Sea

But also up to about 100 0 C data are missing for some of the systems so that they cannot be described entirely.

[PDF] Sea

References Cited

Thus the goal of the present volume is at the same time scientific and educational. The procedure is applied to isothermal evaporation of Black Sea water at 25 °C, and the balance of the crystallization sequence is....

Phase equilibria in the sea

But also up to about 100 0 C data are missing for some of the systems so that they cannot be described entirely. The polythermal solubility diagram of the system K₂SO₄—MgSO₄—H₂O presents the formation of three double salts, picromerite, leonite, langbeinite appearing in this order with increasing temperature.

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