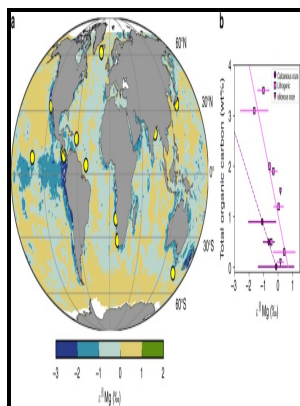


# Atlas and data of solid-solution equilibria of marine evaporites

Springer - References Cited



Description: -

-  
Nonfiction - General  
Westerns  
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Public policy (Law) -- France -- History -- 20th century --  
Congresses.  
Phase diagrams.  
Seawater -- Composition.  
Salt deposits.  
Evaporites. Atlas and data of solid-solution equilibria of marine  
evaporites  
-Atlas and data of solid-solution equilibria of marine evaporites  
Notes: Includes bibliographical references (p. 15-16) and index.  
This edition was published in 1998



Filesize: 32.24 MB

Tags: #Evidence #for #deep

## (ebook) Atlas and Data of Solid

Rahnemaie, Rasoul, Hiemstra, Tjisse, and Van Riemsdijk, W. Immediate download Immediate download Digital downloads only Dispatches next business day Usually dispatches next business day + In stock at our warehouse.

## Evidence for deep

In: Atlas and Data of Solid-Solution Equilibria of Marine Evaporites. The other one comprises two different cations A +, B + and three different anions X -, Y -, Z - which yield the salts AX, AY, AZ, BX, BY, and BZ. However, the data had not yet been summarized in a form that is at the same time comprehensive and permits to understand how the quinary seawater system builds up from its bounding systems.

## [PDF] Sea

The attached computer program SALSYS allows easy and fast access to solubility data that are required by different fields. But also up to about 100 0 C data are missing for some of the systems so that they cannot be described entirely.

## Solid

The second author entered the preparation of the volume when it was realized that besides graphics and tables a fast access to data was required.

## [PDF] Sea

But also up to about 100 0 C data are missing for some of the systems so that they cannot be described entirely.

## [PDF] Sea

## **References Cited**

Thus the goal of the present volume is at the same time scientific and educational. The procedure is applied to isothermal evaporation of Black Sea water at 25 °C, and the balance of the crystallization sequence is...

## **Phase equilibria in the sea**

But also up to about 100 °C data are missing for some of the systems so that they cannot be described entirely. The polythermal solubility diagram of the system  $K_2SO_4$ — $MgSO_4$ — $H_2O$  presents the formation of three double salts, picromerite, leonite, langbeinite appearing in this order with increasing temperature.

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