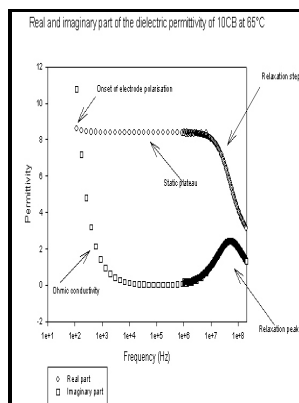


Dielectric relaxation and molecular structure

Research Institute of Applied Electricity, Hokkaido University - Microwave Absorption and Molecular Structure in Liquids. LIII. Hydrogen Bonding and Dielectric Properties in Chloroform Mixtures



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Crystal Structure, Stoichiometry, and Dielectric Relaxation in Bi_{3.32}Nb_{7.09}O_{22.7} and Structurally Related Ternary Phases

The Journal of Chemical Physics 1975, 62 4 , 1253-1268. Dielectric relaxation and hydrogen bonding in liquids.

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The electrical conductivity σ for pyrazine-water mixture its evident that formation complexes may be observed from the characteristic behaviour of the conductivity within the concentration range by equation. Hence, the study of molecular dynamics through dielectric relaxation of liquid and glassy phase of pharmaceuticals are very common Grzybowska et al.

Dielectric Relaxation and Molecular Structure. II. Notes on Analysing Dielectric Data of Dilute Solutions

The potential energy of orientations aligned with the field is lowered and orientations aligned against the field are raised.

Influence of Water on the Structure and Dielectric Properties of the Microcrystalline and Nano

Its growth testifies to the increase of the average number of water molecules, which surround these groups on the surface of MCC crystallites. The density of the liquid mixture increases with increasing the concentrations of brompheniramine.

Dielectric dispersion, relaxation and molecular interaction of pyrazine binary mixtures

KEYWORDS: Brompheniramine; Dielectric Relaxation; 1-Butanol; 1-Hexanol; 1-Pentanol Copy the following to cite this article: Elangovan S, Garbi T. Materials and Methods Brompheniramine and alcohols Purity 99%, AR grade were procured from E-Merck India.

Dielectric Relaxation Studies Between Brompheniramine with 1

Crystal nucleation requires motion of the entire molecule and Johari Goldstein JG β -relaxation is credited to the initial process of crystallization Ngai.

Dielectric Relaxation and Molecular Structure. IV. Intermolecular Interaction between Dichloroethane and Benzene

The gauche molecules appear to be more stabilized than the trans molecules in the medium of aromatic solvent. The crystallization follows Avrami equation and the parameters are obtained. Further at mid concentration the thermal parameters are lower indicating exchange of heat with surroundings minimized.

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