

Chemical thermodynamics at a glance

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Thermodynamics
Chemical thermodynamics at a glance
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Nuclear Energy Agency (NEA)

The only system that meets this criterion is a perfect crystal at a temperature of absolute zero 0 K, in which each component atom, molecule, or ion is fixed in place within a perfect crystal lattice. Assume, for the moment, that a thermometer immersed in a beaker of water on a hot plate reads 73.

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It does not, however, say anything about whether an energetically feasible reaction will actually occur as written, and it tells us nothing about the reaction rate or the pathway by which it will occur described by chemical kinetics. We are indebted to the staff of Seeley-Mudd Science and Engineering Library for their assistance in obtaining resource materials. The internal energy change of a process is equal to the heat change if it is measured under conditions of constant volume, as in a closed rigid container such as a bomb calorimeter.

Energy, Enthalpy, and the First Law of Thermodynamics

For a gas, we use partial pressure in place of concentration.

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