## Introduction to mass spectrometry

### Heyden - 11.06: Introduction to Mass Spectrometry

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Notes: Bibliography: p. 128-133. This edition was published in 1966



Filesize: 33.14 MB

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#### Mass spectrometry

Firstly, metabolites have to carry a positive or negative charge so that they can be easily manipulated in a magnetic or electrical field based on their mass to charge ratio.

#### **Introduction to Mass Spectrometry**

The matrix also serves as a proton donor and acceptor, acting to ionize analyte in both positive and negative ionization modes, respectively. Further examples of functional group influence on fragmentation are provided by a selection of compounds that may be examined by clicking the left button below.

#### **Short Introduction to Mass Spectrometry**

Each has a different mass resolution capability. Once the sample is ionized, it is passed through some form of electric or magnetic field. Usually the results are compared with databases to unravel the sequence of the sample.

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The fragmentations leading to the chief fragment ions will be displayed by clicking on the appropriate spectrum. The ions then leave the influence of the electric field, and travel through what is called the drift region. Fenn and Tanaka were awarded the Nobel Prize for chemistry in 2002 for the development of electrospray ionisation.

#### Mass Spectrometry

Science, 246 4926, 64-71. The following mass spectrum of cocaine demonstrates how a forensic laboratory might determine the nature of an unknown street drug. Stable isotope labeling with amino acids in cell culture SILAC requires growing mammalian cells in specialized media that is deficient in lysine and arginine.

## Introduction to Mass Spectrometry

Assume that the isotopes a	are all the most	abundant in nature.
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