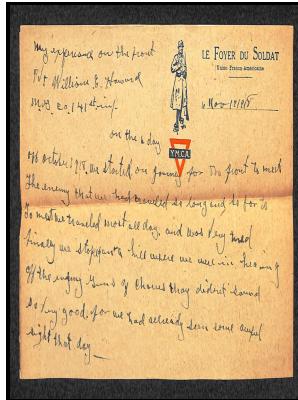


Magnesia refractories produced from chemically modified periclase grains and Mg(OH)₂ slurries

United States Dept. of the Interior, Bureau of Mines - Modeling the cyanide heap leaching of cupriferous gold ores

Description: -



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Trial practice -- United States -- Technological innovations
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Wales -- Guidebooks.
England -- Guidebooks.
Slurry.
Magnesium oxide.
Refractory materials.Magnesia refractories produced from chemically modified periclase grains and Mg(OH)₂ slurries

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8848.
Report of investigations (United States. Bureau of Mines) ;
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Report of investigations ;Magnesia refractories produced from chemically modified periclase grains and Mg(OH)₂ slurries
Notes: Includes bibliographical references.
This edition was published in 1984



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Magnesia refractories produced from chemically modified periclase grains and Mg(OH)₂ slurries /, by James P. Bennett et al.

The current density measurements are further used to determine the signature of pinhole formation in the dielectric film. Chang, Department of Metallurgical and Materials Engineering, University of Wisconsin, Madison, WI 53706. Soffa, University of Pittsburgh, Pittsburgh, PA 15261; J.

Fabrication and properties of in situ intergranular CaZrO₃ modified microporous magnesia aggregates

Chromite Deposits Along the Border Ranges Fault, Southern Alaska.

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The residue from the methanol leach is water-leached to recover the remainder of the NaOH and the Na₂CrO₄ product. It has been found that the rate of reaction is a function of not only temperature and sulfur partial pressure, but also the partial pressure of H₂S.

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The influence of concentration of lithium and calcination temperature on the structure of Li₇La₃Zr₂O₁₂ were investigated by X-ray diffraction XRD and scanning electron microscopy SEM in detail. Values for the latter typically range between 40 and 50 per Cr₂O₃.

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