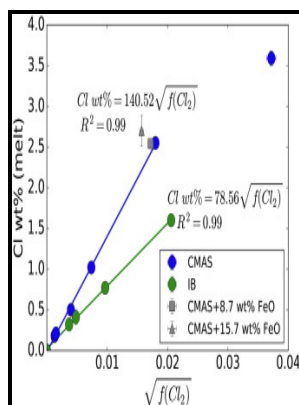


# Solubilities of copper-nickel alloys in silica - saturated iron silicate melts.

University of Birmingham - US3682623A



Description: -

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## 82. Metal Processing and Metal Working Industry

The process takes place at temperatures ranging from 40 to 85°C and pressures ranging from atmospheric to 2. The invention is further directed to such a process wherein the refining slag is subjected to further treatment which comprises the additional step of mixing said refining slag with a solid material containing metallic iron to form a slag recovery bath; selectively reducing combined copper contained in the slag portion of said slag recovery bath to the elemental metallic state with solid-state metallic iron while subjecting the slag recovery bath to a high degree of agitation of the character described above, thus promoting reoxidation by copper oxide of such lead or tin as may be temporarily reduced from any combined lead or tin contained in said refining slag and preventing the volatilization of major proportion of lead or tin compounds, thereby producing a black copper and an extracted slag of low copper content; and separating said extracted slag from said black copper.

### The solubilities of copper

It is associated with carpal tunnel syndrome and degenerative changes in the joints. This rod product is shipped to wire mills where it is extruded into various sizes of copper wire.

### Aluminium control of argon solubility in silicate melts under pressure

Al coordination changes in high-pressure aluminosilicate liquids. The plasma torch produces a considerable amount of metal fumes, ozone, nitrogen oxide and ultraviolet radiation, and generates high levels of noise. In addition, the trace elements Ag, Au, Ga as Ga<sub>2</sub>O<sub>3</sub>, Ge, In, and Sb were simultaneously added as powders into the copper—slag mixtures, each assaying 1 wt% of the mass of copper.

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Reduction is carried out, for example, in cobalt and nickel production with gas.

Gold and copper in volatile saturated mafic to intermediate magmas: Solubilities, partitioning, and implications for ore deposit

## formation

In this situation, the ladle should be equipped with a mobile exhaust hood. The bottom half of the pattern is placed in the bottom flask the drag, and first fine sand and then heavy sand are poured around the pattern. The metal is kept from contact with the solid fuel.

## 82. Metal Processing and Metal Working Industry

Silico-tuberculosis has long been reported in foundry workers. Process materials inputs and pollution outputs for lead smelting and refining Process Material input Air emissions Process wastes Other wastes Lead sintering Lead ore, iron, silica, limestone flux, coke, soda, ash, pyrite, zinc, caustic, baghouse dust Sulphur dioxide, particulate matter contain-ing cadmium and lead Lead smelting Lead sinter, coke Sulphur dioxide, particulate matter contain-ing cadmium and lead Plant washdown wastewater, slag granulation water Slag containing impurities such as zinc, iron, silica and lime, surface impoundment solids Lead drossing Lead bullion, soda ash, sulphur, baghouse dust, coke Slag containing such impurities as copper, surface impoundment solids Lead refining Lead drossing bullion Â Zinc Zinc concentrate is produced by separating the ore, which may contain as little as 2% zinc, from waste rock by crushing and flotation, a process normally performed at the mining site. Thus, essentially any slag containing copper, lead, tin or nickel, even in relatively small proportions, may be reduced by mixing such slag with a solid material containing metallic iron and subjecting the resulting bath to the degree of agitation employed in this process as described above.

## 82. Metal Processing and Metal Working Industry

The high degree of agitation employed additionally provides substantially uniform temperature distribution throughout the reaction masses, thereby avoiding the formation of hot spots which would tend to promote volatilization of lead or tin oxide. All workers involved in the Bayer process should be well informed of the hazards associated with handling caustic soda. Production of alumina from bauxite by the Bayer process in which bauxite is digested at high temperature and pressure in a strong solution of caustic soda.

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