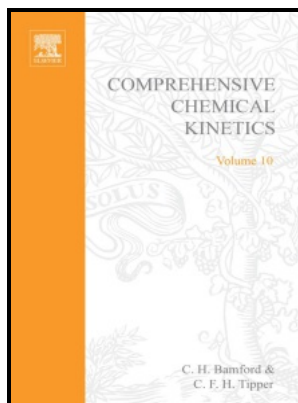


# Handbook of magnesium-organic compounds [by] S.T. Yoffe and A.N. Nesmeyanov.

- - Analytical Chemical Tools



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Magnesium organic compounds Handbook of magnesium-organic compounds [by] S.T. Yoffe and A.N. Nesmeyanov.

-Handbook of magnesium-organic compounds [by] S.T. Yoffe and A.N. Nesmeyanov.

Notes: Text in Russian. Comprises pt. 2 of the second series of Synthetic methods in the field of metallo-organic compounds, edited by A.N. Nesmeyanov and K.A. Kocheshkov. Foreword, by W. Wardlaw and Translations of pref. and introd. and notes by A.L. Mackay (31 p.) in pocket of v. 1.

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**Vol. 127, No. 3304, Apr. 25, 1958 of Science on JSTOR**

ELECTROPHILIC SUBSTITUTION It is apparent from simple valence bond considerations as well as from calculations of  $\pi$ -electron density, 9-12 that isoindoles should be most susceptible to electrophilic attack at carbon 1. Bornstein, F e t r a M o n Letters p.

**93644748**

Thesis, University of East Anglia, 1965. A compound described as 2-methyl-3-nitroindole 374 was said to be formed in low yield by the action of ethyl nitrate on 2-methylindole magnesium iodide.

**ADVANCES IN HETEROCYCLIC CHEMISTRY V10, Volume 10**

For the formation of haloalkoxybenzofuroxans from nitrobenzofuroxans, see Section VII, B.

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This, we feel, justifies the field being covered once more, and its separation from the monocyclic 1,2,5-oxadiazole oxides-the furoxans. Sections 11,C, 2,f and D, 1,a.

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Van der Walt, and A. Reactions in Acidic and Basic Media. Presentation of manuscripts Papers must be written in English.

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