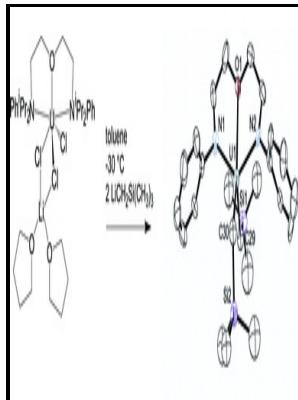


Triamidoamine complexes of the f-elements

-- Synthesis of a highly strained uranacycle: molecular structures of organometallic products arising from reduction, oxidation and protonolysis



Description: -

-Triamidoamine complexes of the f-elements

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Six coordinate capped trigonal bipyramidal complexes

Further, when KAsD 2 ref. Bis hydroxylamino triazines: High Selectivity and Hydrolytic Stability of Hydroxylamine-Based Ligands for Uranyl Compared to Vanadium V and Iron III. Applications The authors studied in schools headed by pioneers in rare earth chemistry, have a combined experience of one hundred and fifty years in inorganic chemistry, rare earth complex chemistry, nuclear and radiochemistry of rare earths and supramolecular chemistry.

Complexes of zirconium with aryl substituted triamidoamines: molecular structures of amide and alkyl derivatives

The thorium—arsenic single bond in 2 and double bond in 3 can be considered to be long when compared to the additive sum of the single bond radii of thorium and arsenic; this trend has been observed with the related U—P, U—As and Th—P complexes and can be related to the steric demands of the Tren TIPS ligand and for 3 to the anionic formulation of the Th-component „,

Behaviour of complexes of f

Solvent Extraction and Ion Exchange 2021, 39 2 , 128-151. Development of Targeted Alpha Particle Therapy for Solid Tumors.

24.12: Carbene and Carbyne Complexes

Daane, The chemistry of the Rare Earth Elements by N. Recent developments in synthesis and structural chemistry of nonaqueous actinide complexes. Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Modern Aspects of Rare Earths and Their Complexes

This Feature Article presents a current account of the chemistry of Tren—uranium complexes.

Six coordinate capped trigonal bipyramidal complexes

Journal of the American Chemical Society 2013, 135 41 , 15338-15341. Despite the developed nature of uranium—ligand multiple bond chemistry „,—that is burgeoning due to a pressing need to improve our fundamental understanding of the chemical bonding of actinide An elements „,, and how this might impact reactivity and nuclear waste clean-up,—the chemistry of other An—ligand multiple bonds, aside from actinyls, AnO²⁻ ref. Thorium—ligand multiple bond chemistry is an attractive area for studying chemical bonding because of the intrinsic features of metal— ligand multiple bonds, but it is far less developed than uranium „,, being restricted to a few carbene „„„, imido „, and chalcogenido „„, complexes.

Uranium triamidoamine chemistry

Although Sc and Y belong to the d-block, their properties are similar to those of lanthanoids. This results, in the case of phosphine complexes, in a large J_{PH} NMR coupling constant of 50—100 Hz between the cis hydride and phosphorus nuclei.

Reduction of Dinitrogen to Ammonia at a Well

Low-Valent Vanadium Complexes of a Pyrrolide-Based Ligand.

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