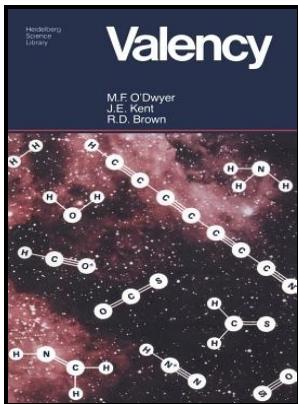


Wave mechanics and valency.

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This edition was published in 1966



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At the initial stage, as the two hydrogen atoms approach each other, the attractive forces are stronger than the repulsive forces and the potential energy decreases.

Wave Mechanics and Its Applications

Consider a situation wherein two hydrogen atoms H a and H b are separated by infinite distance.

Wave Mechanics and Valency by Linnett, J W

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In good all round condition. Previous owners inscription on front endpaper. The wave theory penetrated into organic chemistry, for it was found capable of explaining in general terms the properties of double bonds and extended molecular groups containing alternate single and double bonds.

Wave Mechanics and Its Applications

The internuclear distance at this stage gives the H-H bond length and is equal to 74 pm. Soon the same ideas were extended to molecules, and for the first time there emerged a theory of chemical valency with a quantitative basis. Please note the Image in this listing is a stock photo and may not match the covers of the actual item, 300grams, ISBN:.

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