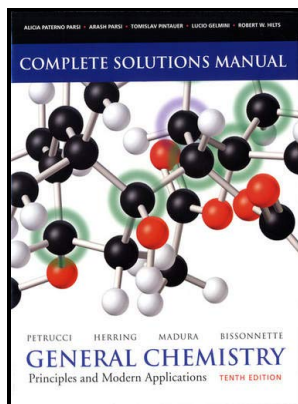


General chemistry

Scientific American Books - CHEM101: General Chemistry I



Description: -

- ChemistryGeneral chemistry
-General chemistry

Notes: Includes index.

This edition was published in 1989



Filesize: 22.75 MB

Tags: #GENERAL #CHEMISTRY #I

GENERAL CHEMISTRY I

It is quite remarkable that in 430 BC, of ancient Greece proclaimed that the atom was the simplest unit of matter, and that all matter was composed of atoms. For example, sodium differs from neon and potassium by one atomic number, but it resembles potassium, not neon.

General Chemistry Notes

Click the link below for source of data. How can those electron-dot diagrams showing shared electrons happily sitting between the nuclei be consistent with the principle that opposite charges attract? In 2013, I created a that I considered worth recommending to others. Clarity rating: 5
The text uses a readable prose style.

General Chemistry for Students

Consistency rating: 3 The format of the textbook is fairly consistent: within most topics is a learning objective set apart with a colored box , text, followed by examples highlighted with a different colored background with a self-consistent format , key equations another colored background , summary, key takeaway, concept and numerical problems both with answers. PDF format; Dan Dill, Boston U - this excellent tutorial covers all the major topics commonly encountered at the general chemistry level, with an unusually thorough treatment of buffer systems.

General Chemistry for Students

Also, to increase the readability of the material, adding in colorful pictures, diagrams, tables and examples would be wonderful: the text is rather dry at the moment, which I find disenchanting since the material is so fraught with wonder and awe. A nice rejoinder to those who accuse chemistry of polluting the world.

General Chemistry 1 — Open & Free

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