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Finite-time thermodynamics: problems, methods, results

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Abstract. Insert your abstract here.

1 Introduction

The development of thermodynamics, beginning from the Carnots work, is closely linked with extremal problems considering limiting capabilities of thermodynamical systems. If Carnot had posed the heat engines maximal COP (coefficient of performance) problem mathematically strict: "To find such a law T(t) of the working bodys temperature changing, that a ratio ρ of the obtained work to the heat taken from the hot source would be maximal, where the heat engine takes the heat from the source with the temperature T_+ and gives it to the source with the temperature T_- ", he would not have got the solution of this problem. That is because the duration of the cycle τ is one of the sought-for variables in the given problem. The set of admissible values for this variable is bounded only by the non-negativity condition and therefore this set is not closed. Hence, according to the Weierstrass theorem (which Carnot could not know about) the problem may have no solutions. Really, the maximum of ρ does not exist. The upper bound of the COP (supremum) is reached in a limit as τ approaches the infinity.

This characteristic turned to be typical for many others extremal thermodynamical problems (problem of minimal mixture separation work, problem of maximal COP for systems with sources of the finite capacity). The classical solution of these problems gives reversible processes in which exchange flows are close to zero, so the exchange of the finite amount of a matter or an energy takes an infinite amount of time.

We must note that the Carnot's heat engine may have the finite power output, but only in the case when coefficients of the heat exchange between heat sources and the working body of the heat engine are arbitrarily large. In this case the Carnot COP is the ratio between the "reversible power output" and the flow of the heat taken from the hot source.

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The history of the finite-time thermodynamics

"The problem of the maximal power output" seems to be the first problem of the finite-time thermodynamics. This problem considers such form of the heat engine cycle that engine's power output would be maximal[1][2] and others.

This problem was very actual for the newborn atomic energetics in 1950s, because the cost price of nuclear power plants were high and the fuel cost was relatively small. Getting the maximal power output was far more important than getting the maximal COP in these conditions.

The problem of the maximal power output considers heat and power flows instead of the heat amount and the work. The heat exchange kinetics, coefficients of the heat exchange between the working body and sources are also taken into account. Authors of the above-mentioned books and many other studies often solved this problem independently but using the same scheme:

- 1. Flows of the heat exchange for the contact with every source were assumed to be proportional to the difference of the source and the working body temperatures (Newtonian kinetics).
- 2. The sought-for cycle assumed to be analogous the Carnot cycle: it consists of two isotherms and two adiabats, and temperatures of the working body contact with sources were selected using the condition of maximal power output in respect with the energy conservation law and the fact that the flow of the entropy coming from the hot source must be equal to the flow of the entropy given to the cold source. The latter condition means that processes inside of the working body were assumed to be reversible.

It turned up to be that the maximal power output of the heat engine, contacting with reservoirs having temperatures T_+ and T_- is

$$N_{max} = \frac{\alpha_1 \alpha_2}{\alpha_1 + \alpha_2} \left(\sqrt{T_+} - \sqrt{T_-} \right)^2, \tag{1}$$

where α_1 and α_2 are the coefficients of the heat exchange for the contact with sources. The COP of the cycle for the maximal power output does not depend on the heat exchange coefficients and is equal to

$$\eta = 1 - \sqrt{\frac{T_{-}}{T_{+}}} \tag{2}$$

These works did not answer the following questions:

- 1. Which kinetics of the heat exchange gives the cycle of the maximal power output, consisting of two isotherms and two adiabats?
- 2. Is the COP of a such cycle always independent of the heat exchange coefficients?
- 3. What is the form of the heat engine cycle, maximizing the COP for the given power output.

These questions were answered in [3][4] using the techniques of the average optimization developed in [5].

It turned up that

- 1. The cycle consists of two isotherms and two adiabats for every kinetics satisfying the condition of the equality of the heat flow direction and the sign of the difference between temperatures of contacting bodies.
- 2. The COP corresponding to such cycle depends on kinetics coefficients, in the most general case.

3. The cycle of the heat engine with the maximal COP for the given power output consists of no more than three isotherms and three adiabats. The condition for which the number of isotherms is exactly two was also obtained.

The development of the finite-time thermodynamics from the "maximal power output problem" to it's modern condition had begun after the work of Curzon and Ahlborn[2]. Talented researchers P. Salamon, B. Andresen, K. H. Hoffman worked with them in the department which head was R. S. Berry. They realized that in real thermodynamical, heat and mass exchange, chemical processes the limited duration of the process has very important role. It means we must consider so-called "irreversible processes" having additional condition of the finiteness of the time. This leads us to a new field of thermodynamics — the finite-time thermodynamics[6][7][8][9].

We think this term is not so well-chosen, we will see it in the typical problems overview in the next section. We will note that there is an important class of processes which are stationary in the time and distributed over the coordinate (steam turbine, stationary heat and mass exchange, processes with circulating working body and others). Maximal capabilities of these processes are found using general techniques based on the thermodynamics-specific optimization and optimal control. These techniques in total form the field of the "optimization thermodynamics" (the term is proposed by L.I. Rozonoer).

In the following sections we will take a look on the typical problems, general techniques of its solving and, finally, their solutions.

2 Formulations of the optimization thermodynamics problems

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