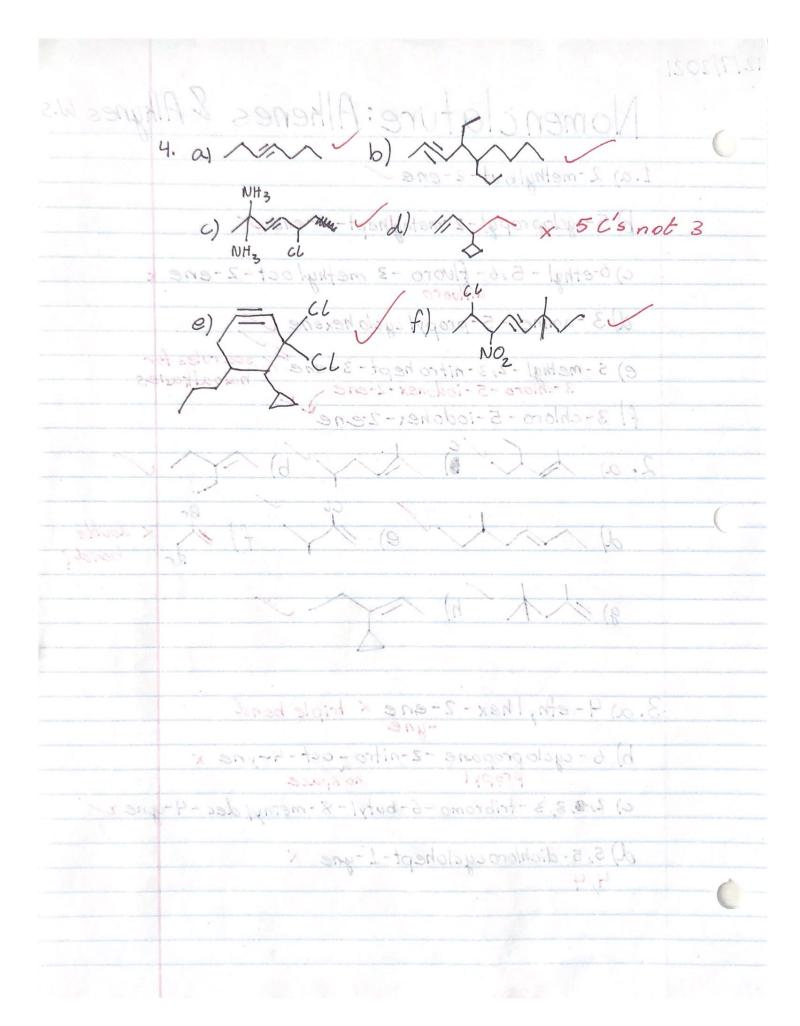
Naming Hydrocarbons W.S

- 1. octane ~ 2.2,5 dimethyloctane
- 3. 5,5 dimethylheptane x double bond
 5,5 dimethylhept 3 ene
- 4. 1,3-diethylogolopentane
- 5. non-4-ene
- 6. cyclopropane /
- 7.3-ethyl oct-6-yne x start numbering from end closest to the triple bond
- 8. 3-methylhexane 6-ethyloct-2-yne
- 9. 4-ethyl-2,3-dimethylheptane
- 10.5-ethyl-2,4,6-trimethyloctane
- 11. 3,4-diethylhex-2-ane x double bond means
 -ene not -ane
- 12. cyclo butene
- 13. benzene
- 14. 2, 7,8-trimethyldecome
- 15. hex 2-ene
- 16. 3,3 diethyl pentane
- 17. 3-ethyl-2-methyl pentane

Nomen clature: Alkenes & Alkynes w.s

- 1.a) 2-methylbut-z-ene
 - b) 5-cyclopropy 1-2-methylhept-3-ene
 - c) 6-ethyl 5,6-fluoro -3 methyloct-2-ene x
 - d) 3 amino 5 propyl cyclo hexene
 - e) 5-methyl-2,3-nitro hept-3-ene x see rules for 3-chloro-5-iodohex-2-ene haloalkanes
 - fl 3-chloro-5-iodohex-zene
- 2. a) (b) (c) (c)
- 3. a) 4-ethylhex-2-ene x triple bond
 - b) 6-cyclopropane -z-nitro-oct-4-yne x propyl no space
 - c) 2,4,3,3-tribromo-6-butyl-8-methyldec-4-yne
 - d) 5,5-dichlorocyclohept-1-yne x 4,4



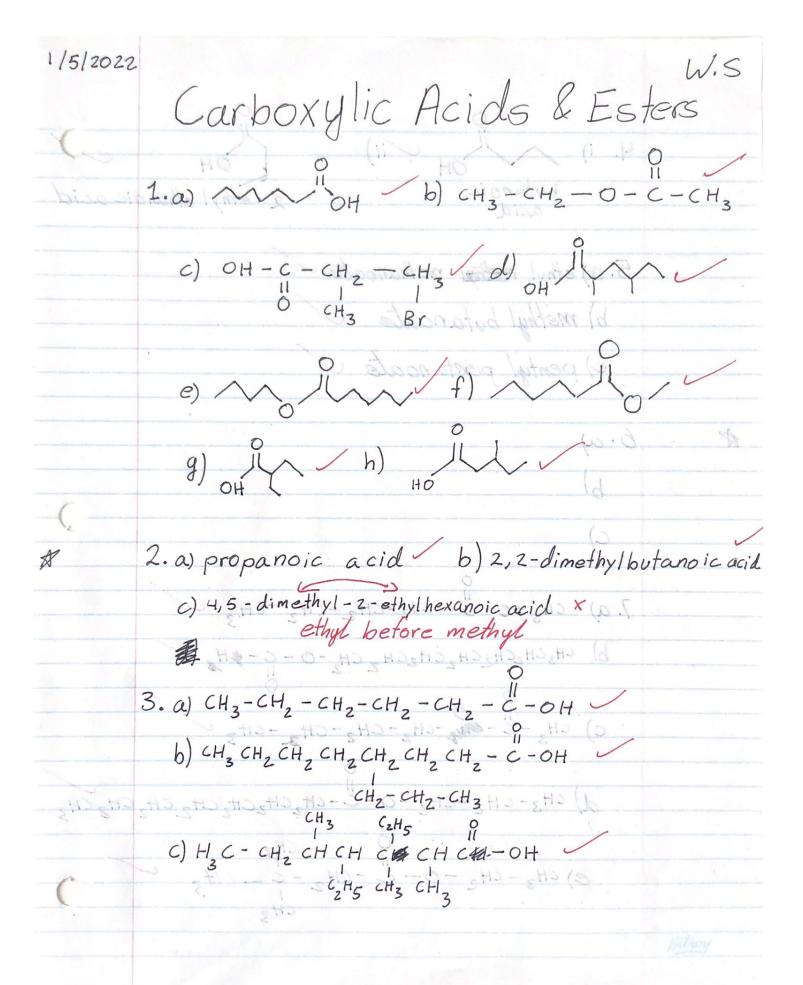


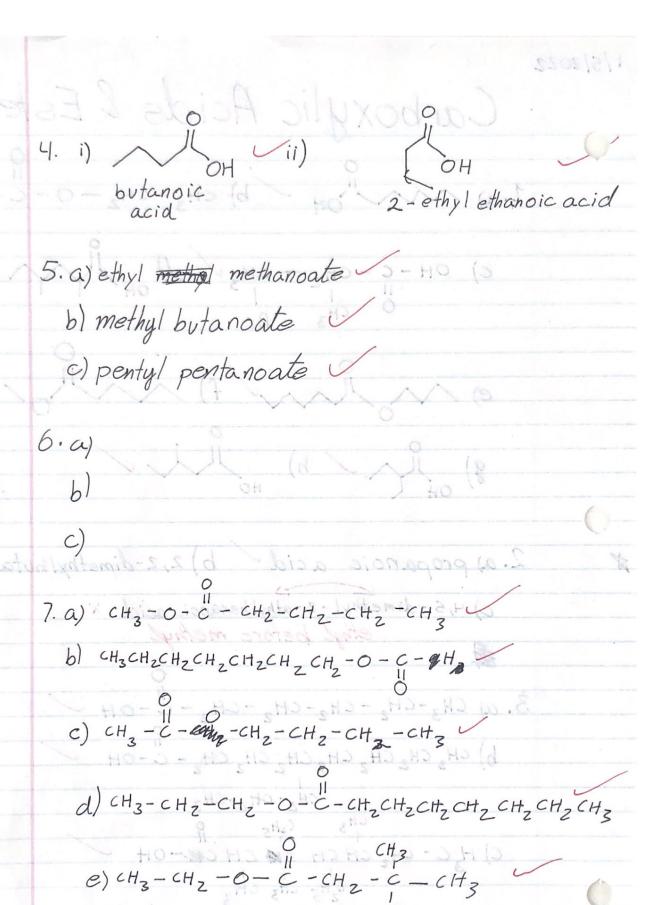
Alcohols, Aldehydes, Ethers & Ketones Ws

c)
$$CH_3 - CH_2 - CH_2 - CH_2 - C-H$$
OH OH

d)
$$CH_3 - CH_2 - CH_2 - C - H$$
 (1) $CH_3 - CH_2 - CH_2 - C - H$ (1) $CH_3 - CH_2 - CH_2 - C - H$ (1) $CH_3 - CH_2 - CH_2 - C - H$ (1)

	200
Alcohols, Aldehydes, Ethers & Ketones Ws	1702/01/2
2.a) 2,2-dimethylbutane	
b) 2-ethoxy butane Ma- Ha-Ma-Ma-Man	
c) penta-3-one	
d) buton-2-one	
e) but - 2-ene	
f) methoxy ethane g) proparal HO HO	
g) proparal	
h) butan-2-01	
i) ethane -1,2, diol	
j) 2-methylpropan-1-01	
K) butan-z-one	
U) 2-methylpent anal	
P (8 × 7) P	
100 · · ·	





Amines and Amides W.S

C)
$$CH_3 - CH - CH_2 - CH_2 - CH_3$$
 NH_3
 NH_3

$$d) CH_3 - CH_2 - CH - C = 0$$

$$CH_3$$