

Department of Geomorphology and Quaternary Geology

Pigment concentration measurements: spectrophotometer

Consumables and instruments

Chemical reagents

- Acetone 100% (grade HPLC): solvent.
- Acetone **technical**: cleaning.
- Redistilled water (MilliQ).
- Neodisher Laboclean.

Neodisher LaboClean FLA: liquid, highly-alkaline cleaning, dispergator agent.

Equipment

• Laboratory fume hood.



Acetone

Acetone (C_3H_6O) is an organic solvent requiring work under the **running fume hood**. Acetone is highly **flammable** and is an **irritant**.

- Ultrasound bath.
- Vortex shaker.
- Laboratory furnace for glassware baking.
- Laboratory oven.
- Glass beakers for acetone pippeting.
- Acetone pipette 100–1000 µl and tips (for acetone).
- Acetone pipette $50-250 \mu l$ and tips (for acetone).
- Sample rack.
- Volumetric flasks:

1 ml, 5 ml, 10 ml, 25 ml, 50 ml, 100 ml.

Others

• Microcuvettes $400 \mu l$ with caps.

Hold only by the corners. Be sure there is no contamination.

• Glass Pasteur pipettes.

One-time use.

• Silicone nipple for pipettes.

Cleaning

- Clean the glassware manually with brush.
- Wash in the washing machine.

Glassware program and Neodisher Laboclean.

- Rinse with redistilled water (MilliQ) 3 times.
- Clean with technical acetone.
- Close with aluminum foil (cleaned with technical acetone).
- Bake in the furnace 500 °C.

Baking time

Minimum 3 h.

Disposal

- Acetone accordingly.
- Glass Pasteur pipettes need to be washed and recycled.

Spectrophotometry

Spectrophotometer run

- Turn on the instrument and PC.
- Wait minimum 1 h before measurements.
- Start **UVProbe** software.
- Press F4 and click Connect.
- Set parameters:
 - Range: **350** to **900 nm**.
 - Yellow button ??.

- Method: **spectrum**.

- Scan speed: **fast**.

- Sampling interval: **0.1 nm**.

- Scan mode: **Single**

- Click Ok.

Blank sample and baseline

Blank sample

• Pipette 400 µl acetone (100%, HPLC) to cuvette.

Correct pippeting

- Check if the solvent is contaminated.
- Be sure there are no air bubbles in the tip.
- Close cuvette with cap.

Avoid evaporation and instrument corrosion.

• Place cuvette in the measurement cell (front one).

V mark on the cuvette to the side.

Autozero (reference sample)

• Pipette 400 µl acetone (100%, HPLC) to cuvette.

Correct pippeting

- Check if the solvent is contaminated.
- Be sure there are no air bubbles in the tip.
- Close cuvette with cap.

Avoid evaporation and instrument corrosion.

• Place cuvette in the reference cell (back one).

V mark on the cuvette to the side..

• Click Baseline.

After this step absorption should show zero.

Reference sample stays in the cell for the entire measurement series.

• Run autozero before each sample.

Sample measurement

• Extracts are diluted so overall absorption is between **0.2** and **1.0** abs.

Limit of quantification of the instrument is linear in this range without the use of calibration curves.

Dilution

For sample dilution use small volumetric flask. Allows remeasurement of a sample at higher dilution. Pipette directly from the prepared sample into a new cuvette.

Example 1:10

- Shake amber vial with vortex.
- Pipette $2 \times 250 \mu l$ extract (sample) to 5 ml flask.
- Add acetone using glass Pasteur pipette. Fill till 5 ml.

Caution: meniscus is concave.

• Stir sample with glass Pasteur pipette.

Measurement

- Using pipette transfer $400 \mu l$ of extract into a cuvette.
- Close cuvette with cap.
- Place cuvette in the measurement cell (front one).

V mark on the cuvette to the side.

• Click Start.

Save and export data

Once analysis begins a window pops-up, where sample name can be put. After the sample analysis file needs to be **saved manually**. Otherwise data is lost.

Save data

• Use File > Save as > file_name.

Data export

• Use Data print.

A table will appear. Copy and paste data into spreadsheet.

Manually make columns wider so names are visible in full before copying.

Data evaluation

"Although in a sedimentary natural pigment sample you have a mixture of chlorins and carotenoids, you can already tell from the sample spectra and absorption peaks what kind of pigments you have in your sample. You can even compare single standard spectra (stored on PC) with your spectra by overlaying the spectra." - Paul Zander

Examples

- Chl a (absorption max. 430 and 663 nm)
- Chl b (absorption max. 463 and 648 nm)
- Pheophytin a (absorption max. 408 and 664nm)
- Carotenoids (420–450 nm)
- Bacteriochlorophyll a (absorption max. 364 and 770 nm)
 - Bacteriopheophytin a (absorption max. 357 and 746 nm)
- Bacteriochlorophyll b (absorption max. 373 and 795 nm)
 - Bacteriopheophytin b (absorption max. 367 and 776 nm)
- Bacteriochlorophyll c, d, e (absorption max. 434, 427, 469 and 666, 655, 654 nm)
 - Bacteriopheophytin c, d, e (absorption max. 412, 406, 435 and 666, 657, 665 nm)

Photometric calculations

Equation for Chlorophylls and Bacteriopheophytin a (needed for proxy-proxy-calibration for HSI data)

Correction of absorption values

Correct the absorption values in case of absorption values below or above **0** between **720 nm** and **900 nm** in the way to scale the absorption values at around **0 abs** in this range.

- To quantify total carotenoids and chloropigments, and bacteriochloropigments are not present in sample extract, use the absorption values at **750 nm** of wavelength to correct this.
- In presence of bacteriochloropigments, find a specific wavelength of absorption above **750 nm** which you can use to scale the absorption values to zero.

Procedure described in Sanchini and Grosjean (2020).

Calculation for bulk chloropigments a and Bphe a concentration from the absorption spectra

Calculate concentrations from the absorption spectra using the following formula

$$c = A_{\lambda}/(\alpha_{\lambda} \times l)$$

where:

1: width of the cuvettes (cm);

A: absorbance measured at a specific wavelength;

: specific molar extinction coefficient equal to $88.77 \times 10^{-3} \times L \times cm^{-1} \times mg^{-1}$ for chloropigments a at 666 nm (Jeffrey and Humphrey, 1975) oraz $52.855 \times 10^{-3} \times L \times cm^{-1} \times mg^{-1}$ for bacteriopheophytin a at 750 nm (Fiedor et al., 2002).

The result is a concentration measured in the cuvette (mg/l), which needs to be corrected to account for sample dilution and mass.

Sample calculations are available in the spreadsheed Spectrophotometer_calculator.xlsx

$$(peak\ absorption)/k \times (dilution\ factor) \times (sample\ mass)/(sample\ volume)$$

where:

k: constant, **0.080770** for chlorins with **666 nm** peak (Jeffrey and Humphrey, 1975) and **0.052855** for bacteriopheophytin peak at **750 nm** (Fiedor et al., 2002).

Calculate Chl a, b and c from absorption spectra

Calculate total carotenoids from absorption spectra

Following Lichtenthaler and Buschmann (2001)

In an extract of plant material containing carotenoids ($\mathbf{x} + \mathbf{c} = \mathbf{xanthophylls}$ and carotenes) in addition to Chls, $\mathbf{A_{470}}$ (the carotenoid region) is determined as the sum of specific absorbances for Chl a, Chl b, and total carotenoids:

• Acetone (pure, 100%):

$$c_{(x+c)} = (1000 \times A_{470} - 1.90_{C_a} - 63.14_{C_b})/214$$

• Acetone (20% water):

$$c_{(x+c)} = (1000 \times A_{470} - 1.82_{C_a} - 85.02_{C_b})/198$$

gdzie:

 A_{470} : absorbcja dla długości fali 470 nm;

 C_a : xxx

 C_b : xxx

Following Guilizzoni et al. (2011); based on Züllig et al. (1981)

Całkowite karotenoidy wyrażone są jako \mathbf{mg} \mathbf{TC} (total carotenoids) na gram suchej materii organicznej ($\mathbf{g}_{\mathbf{LOI}}$).

$$TC = ((E_{450} - 0.8 \times E_{665}) \times V \times 10) / E_{1cm}^{1\%} \times g_{LOI}$$

gdzie:

 $\mathbf{E_{xxx}}$: gęstość optyczna w kuwecie o ścieżce optycznej **1 cm** (szerokość) dla długości fali **450 nm** i **665 nm**, skorygowanych o gęstość optyczną dla długości fali **750 nm**;

V: objętość ekstraktu;

 $\mathbf{E^{1\%}_{1cm}}$: średni współczynnik ekstynkcji dla $\mathbf{1\%}$ roztworu ekstraktu z osadów (wagowo) dla karotenoidów przy ścieżce optycznej (szerokość) $\mathbf{1}$ cm. Równe $\mathbf{2.250}$ (Züllig 1981; Leavitt and Hodgson 2001);

 $\mathbf{E_{450}}$: główne pasmo absorbcji dla karotenoidów w zakresie widzialnym;

 $\mathbf{E_{665}}$: korekta na przybliżone pasmo absorbcji chlorofili i pochodnych chlorofili odpowiadające długości fali $\mathbf{665}$ nm;

 $\mathbf{g}_{\mathbf{LOI}}$: równoważnik strat na prażeniu dla mokrego osadu, obliczony na podstawie zawartości osadów i suchej masy po LOI.

Transfer function for total phosphorus

Credits

Andrea Sanchini, Giulia Wienhues and Paul Zander – University of Bern, Switzerland.

Changelog

- $\bullet~$ 09.12.2022, MZ initial version.
- 09.02.2023, MZ full version.
- $\bullet~17.03.2023,~\mathrm{MZ}-\mathrm{update}$ with filtering and call outs.

Maurycy Żarczyński, Department of Geomorphology and Quaternary Geology, Gdansk