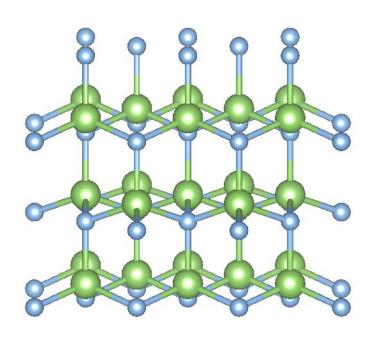


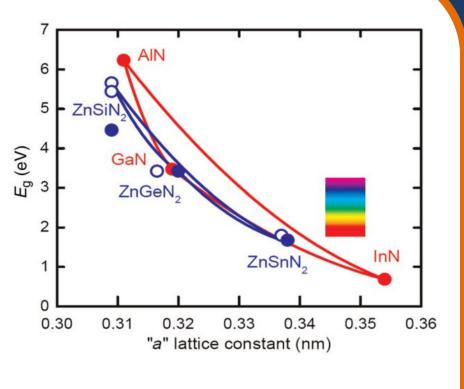
Defects and Doping in ZnGeN₂

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Motivation





Orthorhombic ZnGeN₂

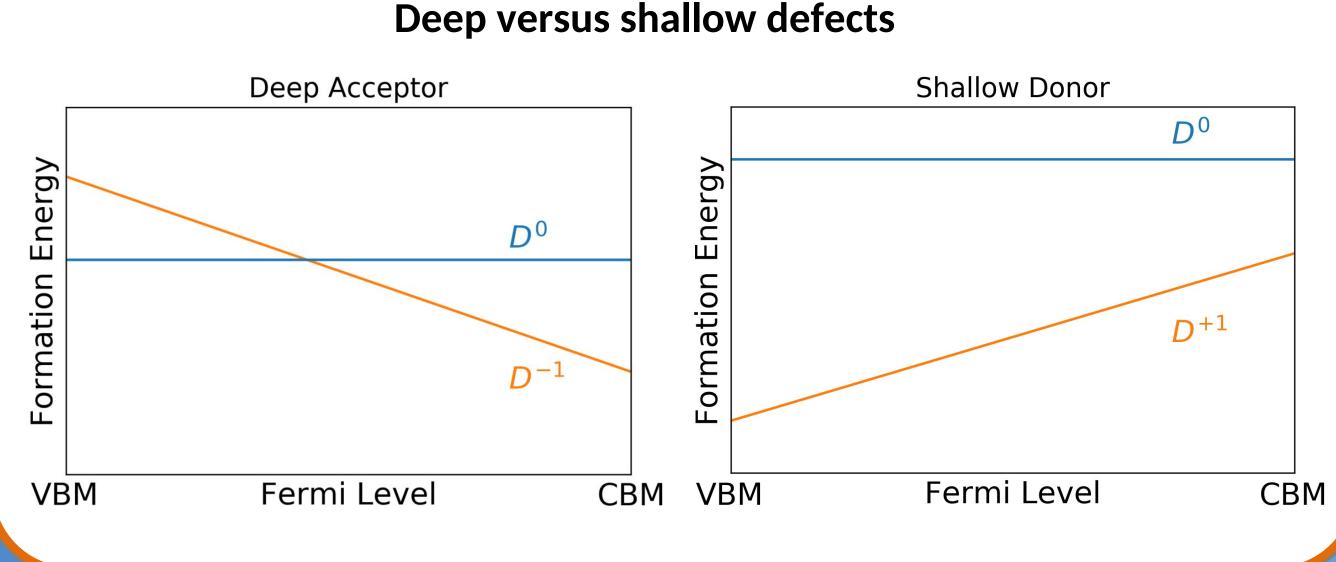
Wurtzite GaN

Quayle, P.C., et al., MRS Comm., 3(3), 135-138. (2013).

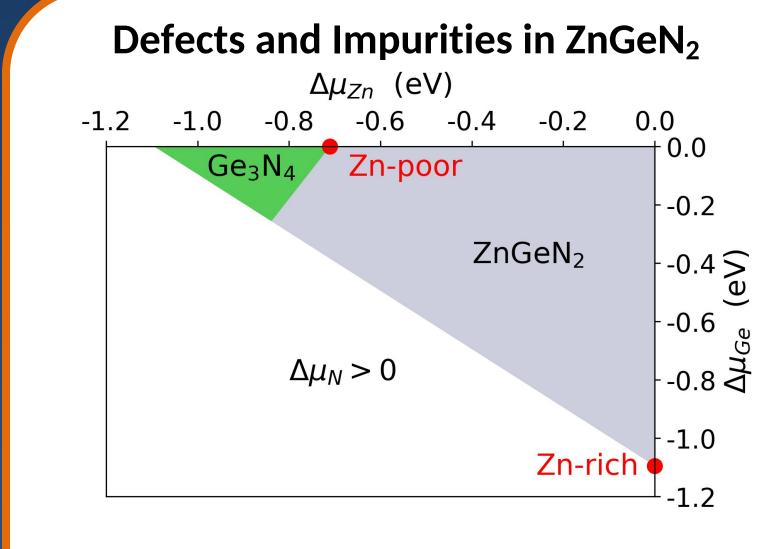
- II-IV-nitrides are a promising class of materials formed from earth-abundant materials that exhibit a wide range of band gaps and spontaneous and piezoelectric polarization moments.
- Potential for optoelectronic and electronic devices.
- Orthorhombic II-IV-nitride unit cell derived from cation mutation of wurtzite III-nitrides: group III ? group II + IV
- What are the properties of defects and prospects for controlled doping in these materials?

Methodology

- Density functional theory with the Heyd-Scuseria-Ernzerhof^[1] functional implemented in VASP^[2]
- [1] J. Heyd, G. Scuseria, M. Ernzerhof, J. Chem. Phys. **124**, 219906 (2006) [2] G. Kresse and J. Hafner, Phys. Rev. B **54**, 11169 (1996)
- Defect formation energies calculated using a 128-atom supercell with electronic corrections. C. Freysoldt et al., Phys. Status Solidi B 248, 1067 (2011)



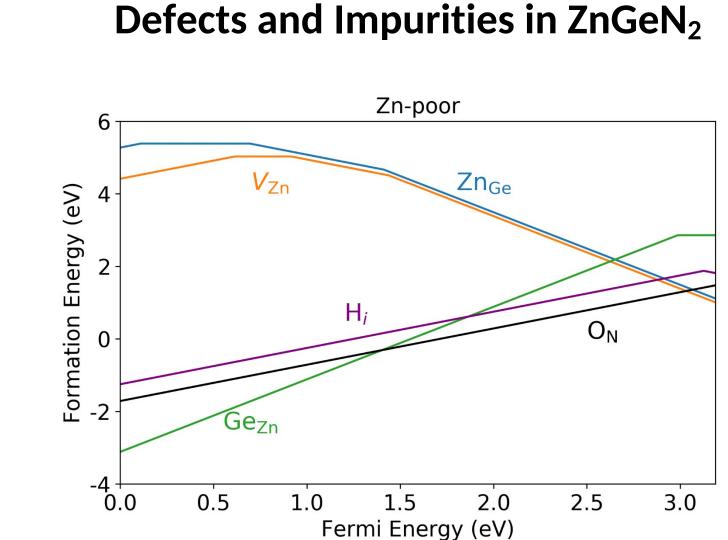
Native point defects and impurities in ZnGeN₂



Variation in Fermi-level pinning due

to O_N and Zn_{Ge}

 $\Delta\mu_N > 0$

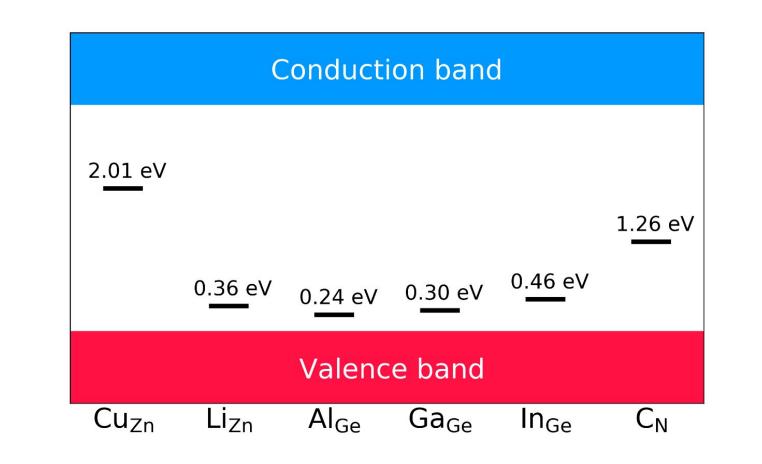


Oxygen and hydrogen are donors in ZnGeN₂, strong compensation by vacancies (V_{Zn}) and antisites (Zn_{Ge})

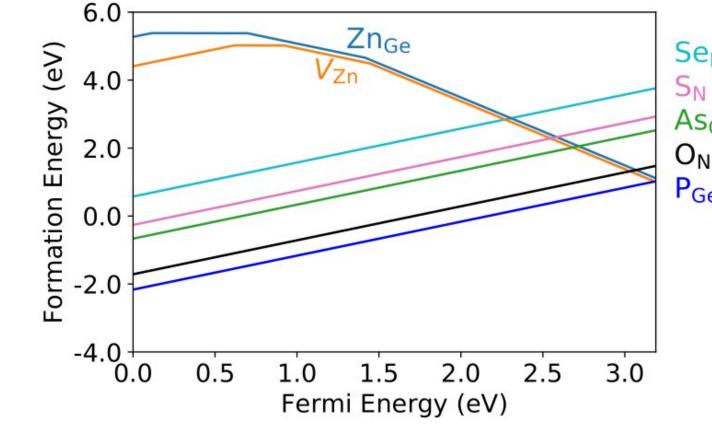
Fermi level is closest to the CBM (3.19 eV) under Zn-poor conditions.

Candidate p-type and n-type dopants in ZnGeN₂





- All candidate acceptors have transition levels within the band gap
- Al_{Ge} has a transition level 0.24 eV from the VBM; similar to Mg in GaN
- Formation energy of candidate donors All candidate donors are shallow
 - P_{Ge} has the lowest formation energy, and least risk of compensation



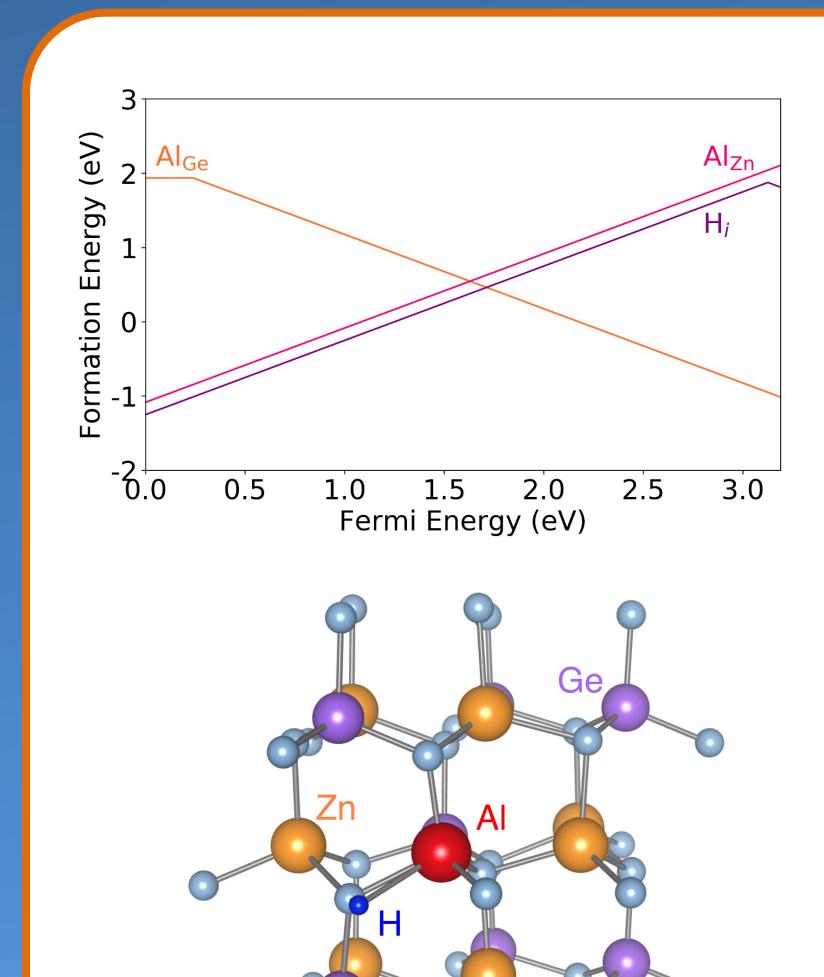
Hydrogen-assisted p-type doping

3.05

2.90

-0.4 0

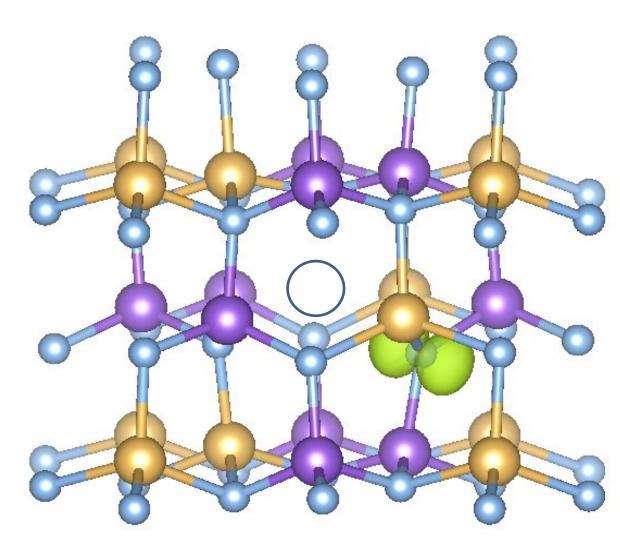
8.0-VHGe



- Al incorporates on Ge and Zn sites
- ? strong compensation ? limits p-type doping by Al_{Ge}
- Al_{Ge} forms neutral complexes with H similar to Mg_{Ga} – H in GaN
- Al_{Ge} H binding energy calculated to be 0.55 eV
- Diffusion barrier for H in ZnGeN₂ is 1.2 eV
- H can be removed in a post-growth anneal

Conclusions

- Native point defects in ZnGeN₂ lead to insulating material.
- Al_{Ge} can lead to *p*-type doping in ZnGeN₂.
- H-assisted doping increases Al substitution on the Ge site.
- P_{Ge} is a promising n-type dopant.



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