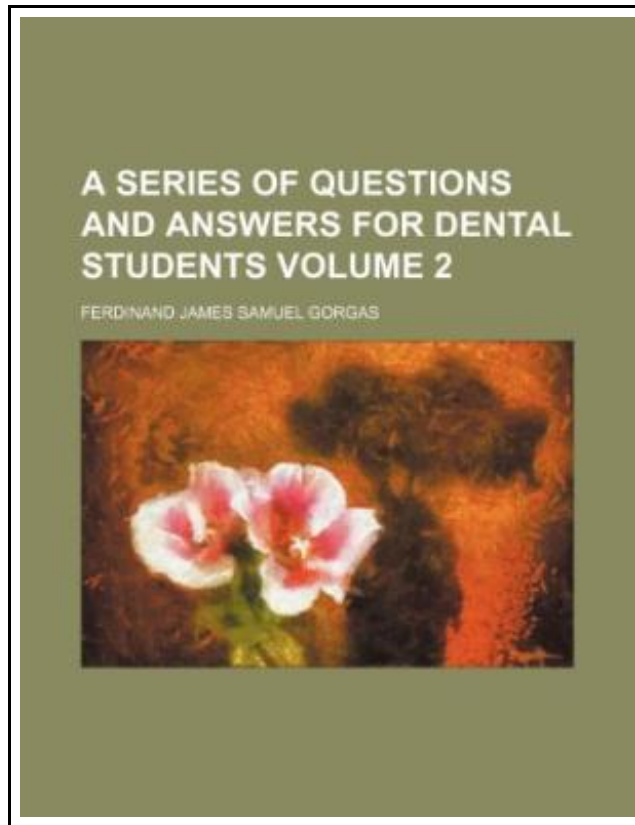


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1892 edition. Excerpt: .and molecular formulae, with examples. A. Atomic is the number of atoms in an element; this collection is written, at the right and a little below as Ha, being 2 atoms of H. Molecular is the number of molecules in an element or compound, and is written just in front, as 4 Ha O, being 4 molecules of Ha O. Q. Describe the leading properties of Oxygen. A. An inodorless, tasteless gas, only sparingly soluble in Ha O, has great affinity for nearly all other elements, and combines directly with nearly all substances; it does not combine with flourine; the union of substances with O forms oxicles. Q. What is the action produced by animals and by plants on the air. A. Animals breath in O and give off C Oa to the air. Plants give off O and take in C Oa, so the supply of O is kept up. Q. What is the composition of Atmosphere. A. Oxygen 20.60; Nitrogen 77.95; Carbon dioxide.04; Moisture 1.40; Nitric acid and Ammonia, traces. The oxygen supports animal life; nitrogen dilutes the oxygen. Q. Why is it necessary that the oxygen should be diluted. A. To prevent too rapid vital or life functions which would soon exhaust and destroy animals. Q. How does Ozone differ from Oxygen. A. Ozone is allotropic form of oxygen, made when non-luminous electric discharges pass through air or oxygen; when phosphorous partially covered with Ha O is exposed to air; also during...

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