| Name | Period |
|--|---------------------|
| Honors Chemistry Aufbau Homework | |
| 1) Write the electron orbital diagram (with arrows) for the fo | llowing elements: |
| a) Na | |
| b) F | |
| c) O | |
| d) Ne | |
| e) V | |
| 2) Write the electron configurations (not with arrows) for the | following elements: |
| a) N | |
| b) Al | |
| c) Mg | |
| d) Cl | |
| e) K | |

Ions!

| 3) Write the electron orbital diagram (with arrows) for the following ions: |
|---|
| a) Na ⁺ |
| b) O ^{2—} |
| c) Mg^{2+} |
| d) Al ³⁺ |
| e) F [—] |
| 4) Write the electron configurations (not with arrows) for the following ions |
| a) K ⁺ |
| b) S ^{2—} |
| c) Ca ²⁺ |
| d) P ³⁻ |
| e) Cl [—] |