| Name Period |
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| Thermochemistry Vocabulary |
| Look up "chemistry" definitions for the following terms. This whole unit focuses on the interactions between: |
| 1) Energy |
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| 2) Matter |
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| 2) Heat |
| 3) Heat |
| |
| 4) Work |
| |
| 5) Temperature |
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| Having a clear picture of what we are studying is very important: |
| 6) System |
| |
| 7) Surroundings |
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| What flows in and what flows out of the system can be confusing to keep track of: |
| 8) Exothermic |
| |
| 9) Endothermic |
| |
| Energy and matter are not the same so: |
| Energy and matter are not the same so: 10) Open System |
| |
| 11) Closed System |
| |
| 12) Isothermal System |
| 12) Isothermal System |
| |
| 13) Adiabatic System |

| We need to be able to measure these changes so: 14) Joule |
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| 15) calorie |
| 16) kcal |
| 17) Calorie |
| 18) Calorimeter |
| We also need to be able to calculate these changes: 19) Heat Capacity |
| 20) Specific Heat |
| Energy comes in many forms: 21) Kinetic Energy |
| 22) Potential Energy |
| Many of these thermodynamics functions have similar properties: 23) State Function |
| 24) Enthalpy |
| 25) Entropy |
| 26) Free Energy |