

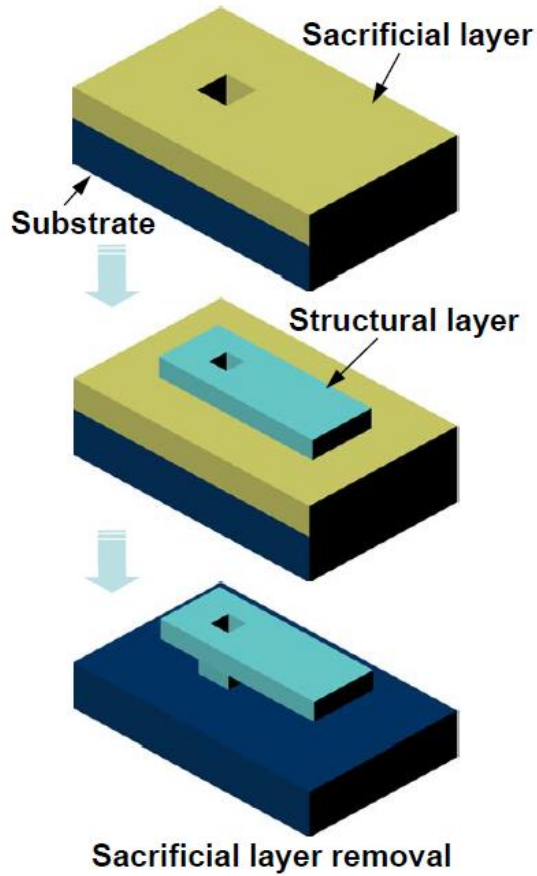
# Lecture#7:

## Etching (1)

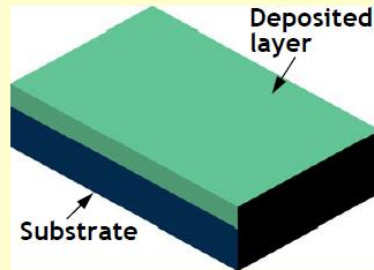
### :Bulk Micromachining

# Reminder: General Step

## Etching (1): Bulk Micromachining

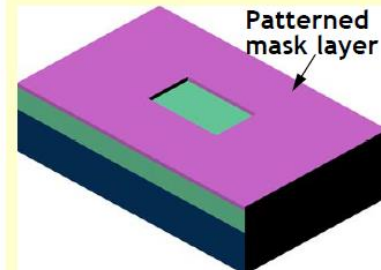


### Deposition



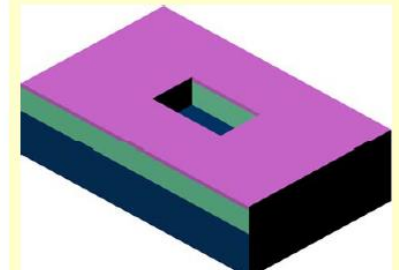
- CVD
- Coating
- Sputtering
- Evaporation
- Bonding
- Plating

### Patterning



- UV lithography
  - Stepper
  - Contact aligner (Double side)
- E-beam lithography
- X-ray lithography

### Etching



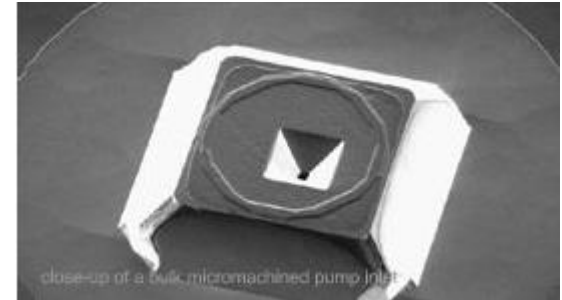
- Dry etching
  - Anisotropic
  - Isotropic
- Wet etching
  - Anisotropic
  - Isotropic

## Etching (1): Bulk Micromachining

### ● Bulk micromachining

; Bulk micromachining was developed in the late 70's essentially for the fabrication pressure sensor membranes from a silicon monocrystal.

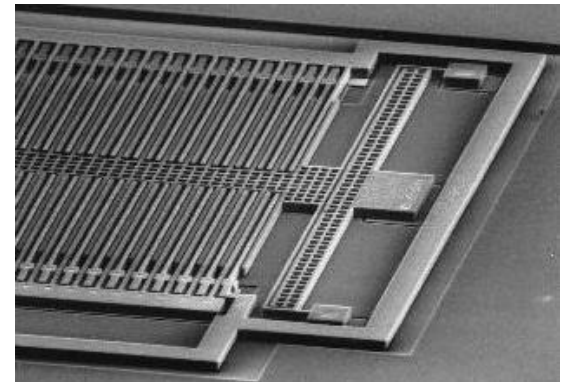
; It is nowadays commonly used to fabricate a variety of mechanical structures (e.g. membranes, valves, and cantilevers) and fluid handling devices (e.g. channels, cavities)



### ● Surface micromachining

; Surface micromachining uses several deposition, lithography, and etching processes to build up mechanical features, layer by layer, on a wafer surface.

; It usually requires the use of a sacrificial layer to form cavities or free-standing and moving structures.



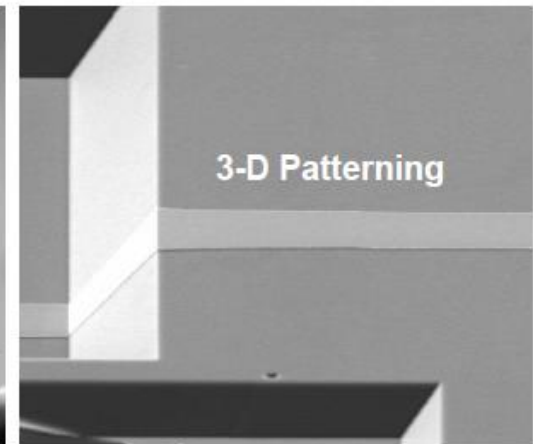
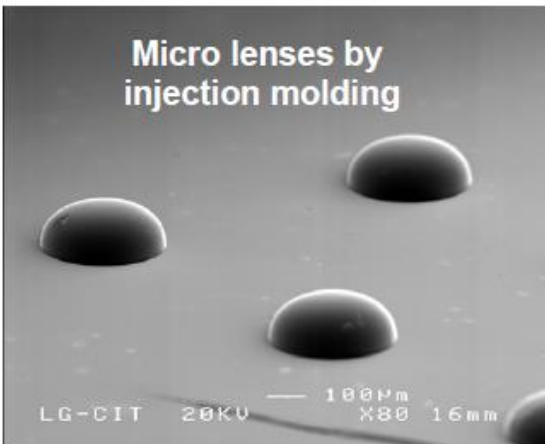
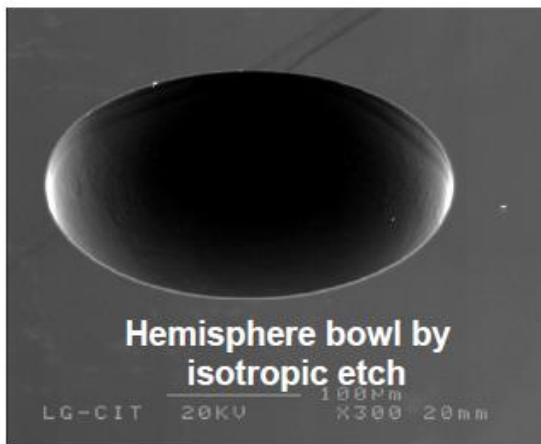
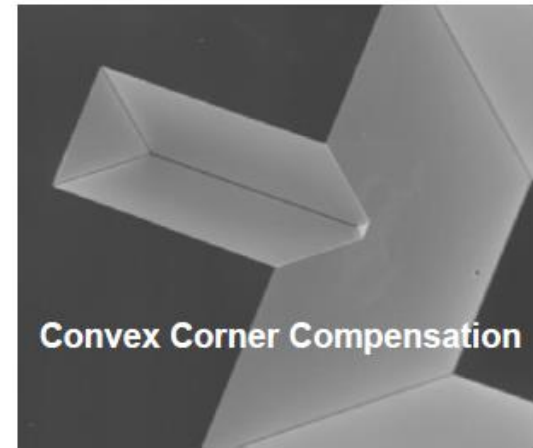
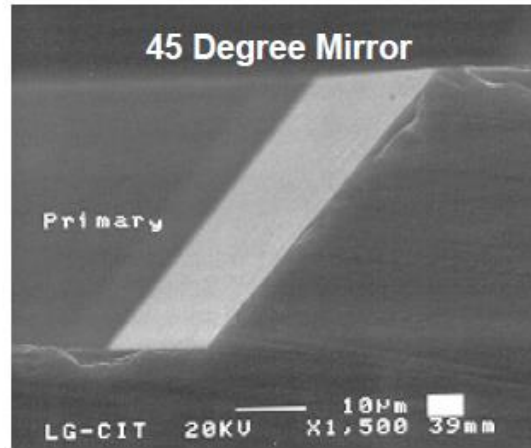
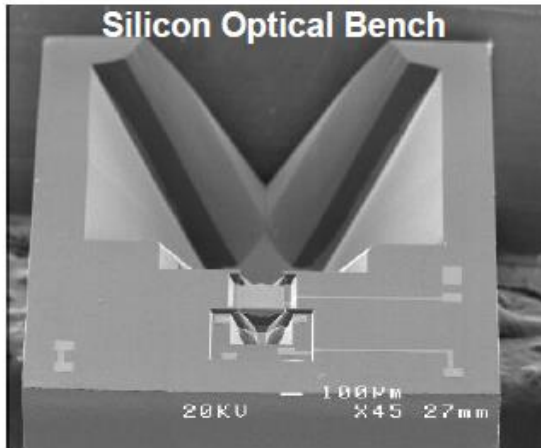
## Etching (1): Bulk Micromachining

### ● Summary: micromachining

	Bulk-micromachining	Surface-micromachining
Birth	1960's	1980's
Structure (Material)	Single crystal silicon wafer	Thin film on wafer
Etching	Wet/Dry etching	Wet/Dry etching
Etch control	Crystal orientation, Diffusion layer	Material selectivity

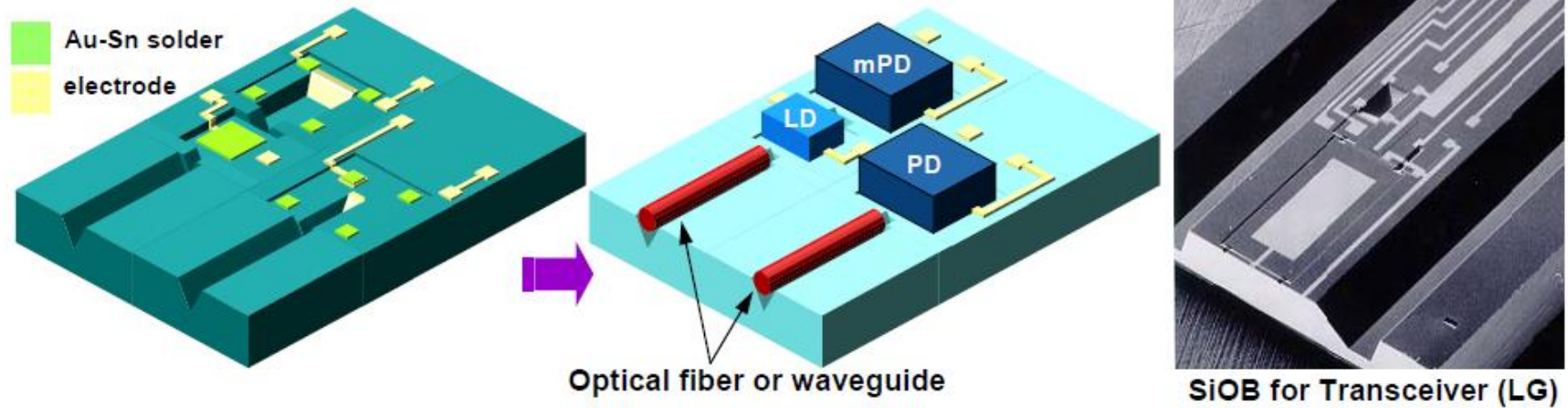
## Etching (1): Bulk Micromachining

### ● Applications

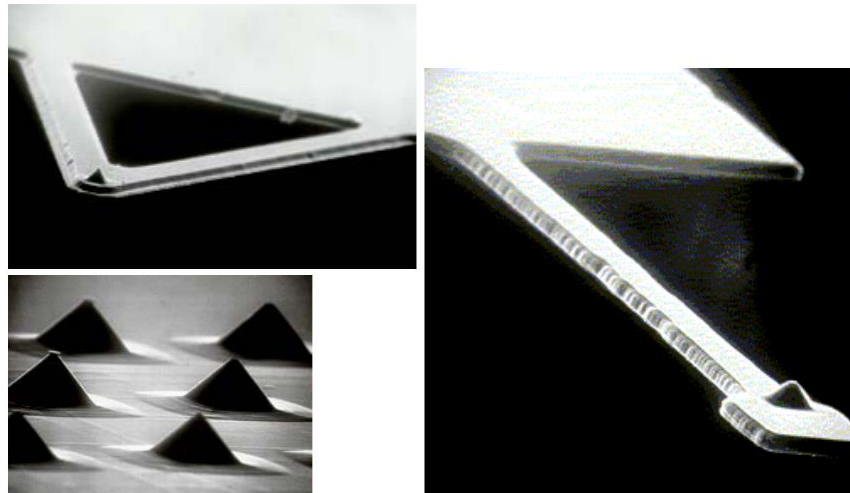


## Etching (1): Bulk Micromachining

### ■ SiOB for transceiver



### ■ AFM probe tip

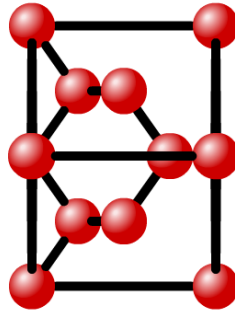
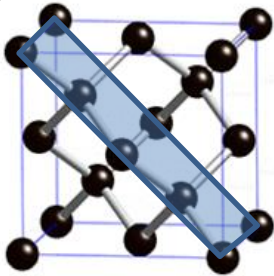




# Reminder: Silicon Crystallography

## Etching (1): Bulk Micromachining

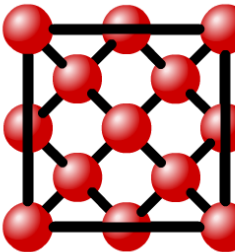
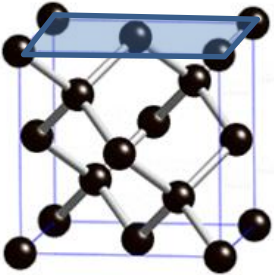
(110) plane



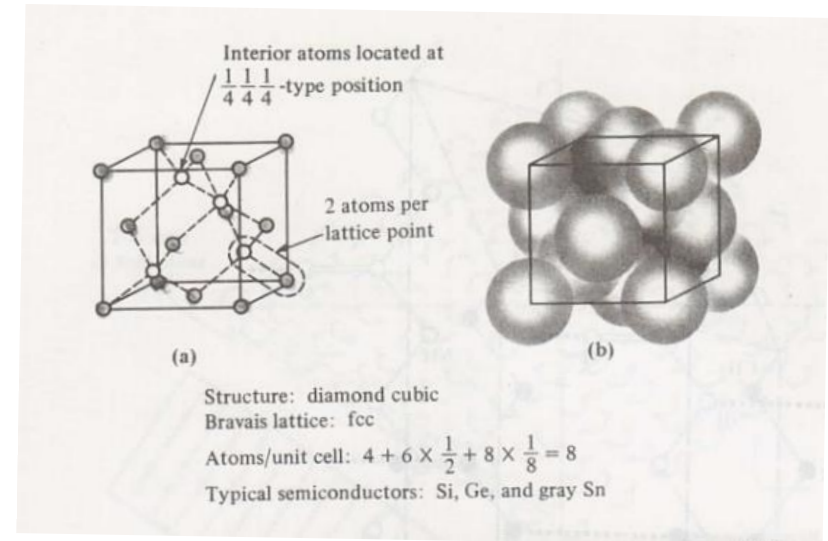
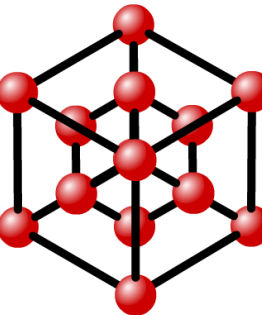
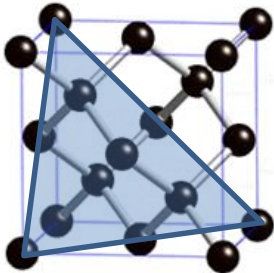
● Silicon wafer

Single crystal → Surface density difference

(100) plane



(111) plane



$$(100) \rightarrow 6.8 \times 10^{14}/\text{cm}^2$$

$$(110) \rightarrow 9.6 \times 10^{14}/\text{cm}^2$$

$$(111) \rightarrow 11.8 \times 10^{14}/\text{cm}^2$$

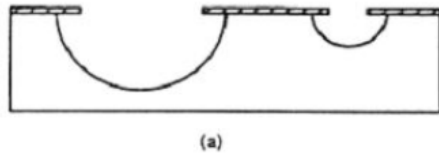
<http://www.dawgSDK.org/crystal/en/library/diamond#0001>

## Etching (1): Bulk Micromachining

### Wet etching

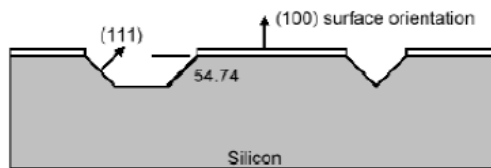
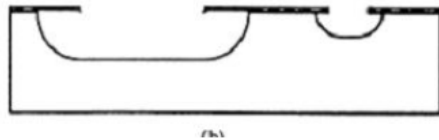
- **Immersion etching**; Wafer is immersed in the etch solution
- Wet chemical etching provides a higher degree of selectivity and faster than dry etching technique.

### Etching profile

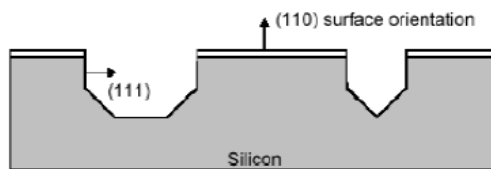


- **Isotropic etching**: etch in all crystallographic directions at the same rate

- *Note: agitation during etching creates more rounded features, but does not significantly change the undercutting.*



- **Anisotropic etching**: directionally dependent on crystallographic directions





# Wet Chemical Etching: Isotropic (1)

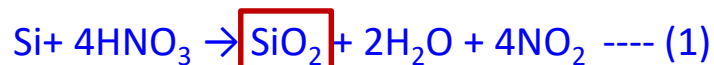
## Etching (1): Bulk Micromachining

### ● Isotropic etching

- They usually are acidic, such as HF/HNO<sub>3</sub>/CH<sub>3</sub>COOH (hydrofluoric acid / nitric acid / acetic acid), called **HNA system** and lead to rounded isotropic features in single crystalline silicon.

### ● HNA system

- The overall reaction of HNA system with silicon is



Etchant (Diluent)	Reagent Quantities	Temp. °C	Etch Rate (μm/min)	(100)/(111) Etch Ratio	Dopant Dependence	Masking Films (etch rate)
HF	10 ml				≤10 <sup>17</sup> cm <sup>-3</sup> n or p reduces etch rate ≈150×	
HNO <sub>3</sub>	30 ml	22	0.7 to 3.0	1:1		SiO <sub>2</sub> (30 nm/min)
(water, CH <sub>3</sub> COOH)	80 ml					
HF	25 ml					
HNO <sub>3</sub>	50 ml	22	4	1:1	no dependence	Si <sub>3</sub> N <sub>4</sub>
(water, CH <sub>3</sub> COOH)	25 ml					
HF	9 ml					
HNO <sub>3</sub>	75 ml	22	7	1:1	---	SiO <sub>2</sub> (70 nm/min)
(water, CH <sub>3</sub> COOH)	30 ml					

# Wet Chemical Etching: Isotropic (2)

## Etching (1): Bulk Micromachining

### ● Masking for isotropic silicon etchant

- $\text{SiO}_2$  has an appreciable etch rate of 300 to 800 Å / min in the HNA system.
- A mask of nonetching Au or  $\text{Si}_3\text{N}_4$  is needed for deeper etching.
- Photoresists do not stand up to strong oxidizing agents such as  $\text{HNO}_3$ .

### <Masking materials for acidic etchants>

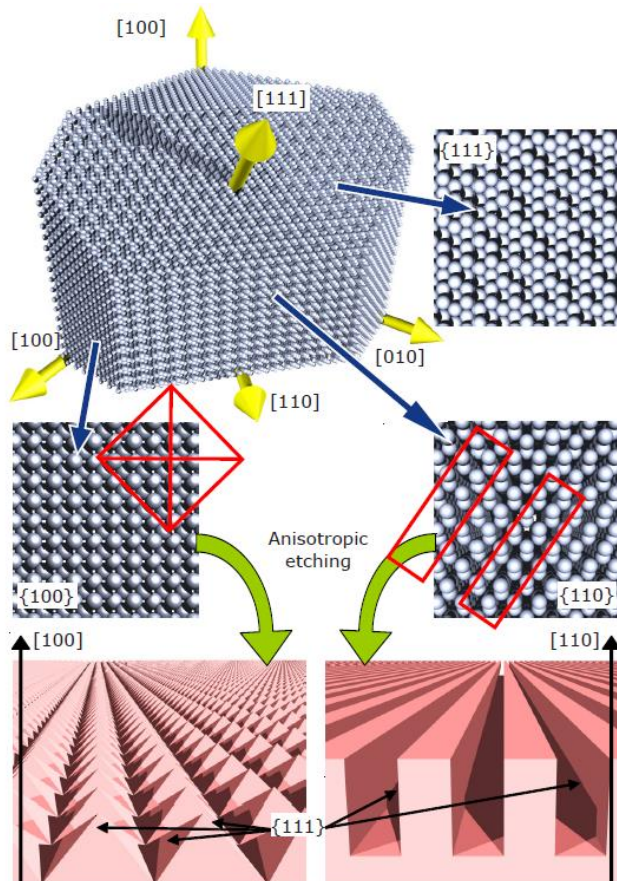
Masking	Piranha (4:1, $\text{H}_2\text{O}_2$ : $\text{H}_2\text{SO}_4$ )	Buffered HF (5:1 $\text{NH}_4\text{F}$ : conc. HF)	HNA
Thermal $\text{SiO}_2$		0.1 $\mu\text{m}/\text{min}$	300–800 Å /min. Limited etch time, thick layers often are used due to ease of patterning.
CVD (450°C) $\text{SiO}_2$		0.48 $\mu\text{m}/\text{min}$	0.44 $\mu\text{m}/\text{min}$
Corning 7740 glass		0.063 $\mu\text{m}/\text{min}$	1.9 $\mu\text{m}/\text{min}$
Photoresist	Attacks most organic films	OK for short while	Resists do not stand up to strong oxidizing agents like $\text{HNO}_3$ and are not used.
Undoped Si polysilicon	Forms 30 Å of $\text{SiO}_2$	0.23 to 0.45 Å/min	0.7 to 40 $\mu\text{m}/\text{min}$ at RT [at a dopant concentration $< 10^{17} \text{ cm}^{-3}$ (n or p)].
Black wax			Usable at room temperature.
Au/Cr	OK	OK	OK
LPCVD $\text{Si}_3\text{N}_4$		1 Å/min	Etch rate is 10–100 Å/min. Preferred masking material.

Note: The many variables involved necessarily mean that the given numbers are approximate only.

## Etching (1): Bulk Micromachining

### ● Anisotropic etching

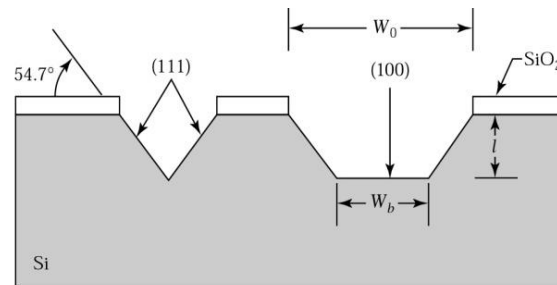
- In the usual application, the wafer is selectively thinned with precisely controlled lateral dimensions and a thickness control.



e.g.) Etchant : 19wt % KOH in deionized (DI) water at 80°C

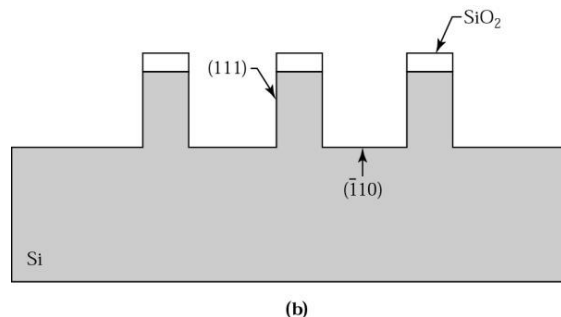
Etch rate (100) : (110) : (111) = 100 : 16 : 1

### ● Orientation dependent (<100>-oriented silicon)



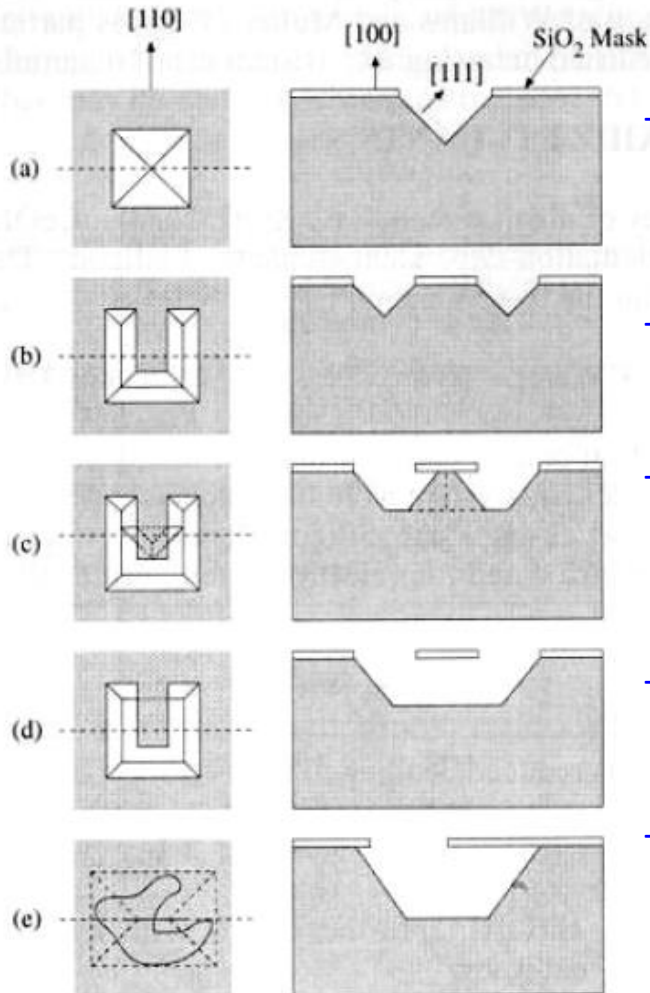
$$W_b = W_0 - 2l \cot(54.7^\circ)$$

$$W_b = W_0 - \sqrt{2}l$$



## Etching (1): Bulk Micromachining

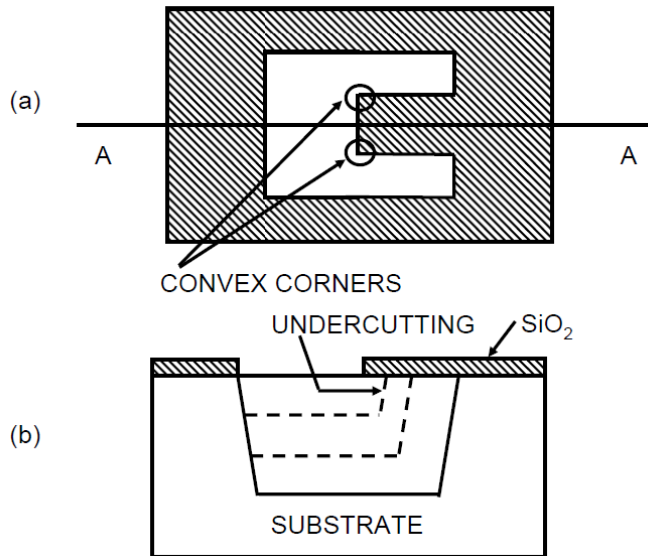
### ● Various etching profile



- (a) Typical pyramidal pit, bounded by the (111) planes, etched into (100) silicon with an anisotropic etch through a square hole in an oxide mask.
- (b) Type of pit which is expected from an anisotropic etch with a slow convex undercut rate.
- (c) The same mask pattern can result in a substantial degree of undercutting using an etchant with a fast convex undercut rate.
- (d) Further etching of (c) produces a cantilever beam suspended over the pit.
- (e) Illustration of the general rule for anisotropic etch undercutting assuming a “sufficient time”

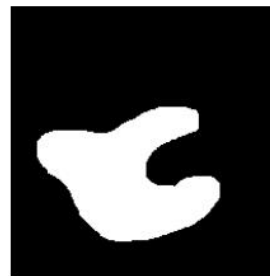
## Etching (1): Bulk Micromachining

### ● Convex corner compensation

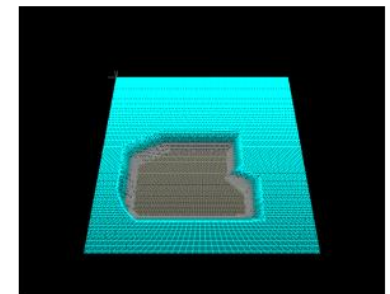
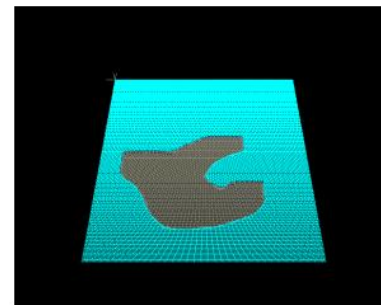


- For **convex corners**, the **fastest etching planes** dominate the three-dimensional shape.
- For **concave corners**, the **slowest etching planes** dominate the three dimensional shape.

- For an arbitrary shaped mask with transparent opening, the pyramidal shape is bound by the outside boundary.



mask



# Wet Chemical Etching: Anisotropic (4)

## Etching (1): Bulk Micromachining

### Mainly used anisotropic etchants

Etchant/diluent/ additives/temperature	Etch stop	Etch rate (100), $\mu\text{m}/\text{min}$	Etch rate ratio	Remarks	Mask (etch rate)
KOH (water) 85°C 44 g/100 mL	$B > 10^{20} \text{ cm}^{-3}$ reduces etch rate by 20	1.4	400 for (100)/(111)	IC incompatible, avoid in eyes, etches oxide fast, lots of $\text{H}_2$ bubbles	Photoresist (shallow etch at room temperature); $\text{Si}_3\text{N}_4$ (<1 nm/min) $\text{SiO}_2$ (28 Å/min)
Ethylenediamine pyrocatechol (water) 115°C 750 mL /120 g/240 mL (EDP)	$= 7 \times 10^{19} \text{ cm}^{-3}$ reduces the etch rate by 50	1.25	35 for (100)/(111)	Toxic, ages fast, $\text{O}_2$ must be excluded, few $\text{H}_2$ bubbles, silicates may precipitate	$\text{SiO}_2$ (2–5 Å/min) $\text{Si}_3\text{N}_4$ (1 Å/min) Ta, Au, Cr, Ag, Cu are not attacked Al at a 0.33 $\mu\text{m}/\text{min}$
Tetramethyl ammonium hydroxide (TMAH) (water) 90°C	$> 4 \times 10^{20} \text{ cm}^{-3}$ reduces etch rate by 40	1	from 12.5 to 50 (100)/(111)	IC compatible, easy to handle, smooth surface finish, few studies	$\text{SiO}_2$ etch rate is four orders of magnitude lower than (100) LPCVD $\text{Si}_3\text{N}_4$
$\text{N}_2\text{H}_4$ (water, isopropyl alcohol) 100°C 100 mL/100 mL	$> 1.5 \times 10^{20} \text{ cm}^{-3}$ practically stops the etch	2.0	10(100)/(111)	Toxic and explosive, OK at 50% water	$\text{SiO}_2$ (<2 Å/min) and most metallic films; does not attack Al

Note: Given the many possible variables, the data in the table are only typical examples.

- **KOH:** (Good) High etching rate, Nontoxic  
(Bad) Contaminate the MOS process (Potassium ion), Temperature sensitive, Stratified etching rate
- **EDP:** (Good) MOS process compatible (Sodium free), Good uniformity, Abrupt stop at etching stop  
(Bad) Toxic, Low etching rate
- **TMAH:** (Good) Nontoxic, MOS process compatible (Sodium free), Will not attack the silicon doped Aluminum, Stable (will not decompose under 130 degree C)  
(Bad) Bad surface roughness on the bottom <100>, High undercutting ratio



## Etching (1): Bulk Micromachining

### ● Etch stop methods

#### 1) Dopant dependent etch stop

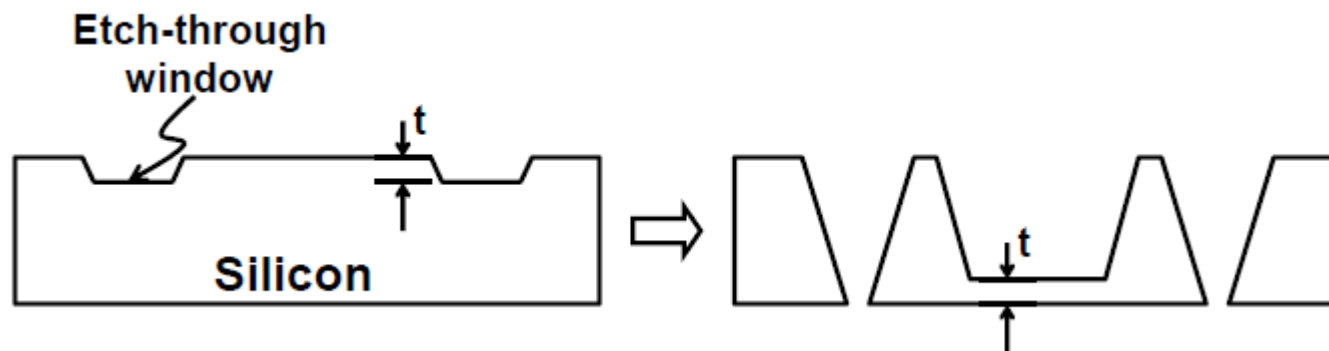
→ EDP : Boron  $\left\{ \begin{array}{l} 10^{20} / \text{cm}^3 - 1:50 (\text{Si} : \text{p}+) \\ 7 \times 10^{19} / \text{cm}^3 - 1:20 (\text{Si} : \text{p}+) \end{array} \right.$

#### 2) Electrochemical etch stop

#### 3) Timed etch stop

#### 4) Etch-through window

→ Thickness of a commercial silicon wafer =  $525 \mu\text{m} \pm$



## Etching (1): Bulk Micromachining

### ● Etching for $\text{SiO}_2$ , $\text{Si}_3\text{N}_4$ , poly-Si (a-Si), and Al

- **$\text{SiO}_2$  etching:** Hydrofluoric acid (HF)
  - : Buffered oxide etch (BOE or BHF)
  - HF +  $\text{NH}_4\text{F}$  (ammonium fluoride)

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- **$\text{Si}_3\text{N}_4$  etching:** HF + boiling  $\text{H}_3\text{PO}_4$  (phosphoric acid)
  - : BOE or BHF + boiling  $\text{H}_3\text{PO}_4$
  - 85%  $\text{H}_3\text{PO}_4$  at  $180^\circ\text{C}$
  - etching selectivity for  $\text{SiO}_2$
  - ; PR has problems as etching mask
  - Using  $\text{SiO}_2$  mask

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- **Poly & a-Si:** Same etchant of Single crystal Si
  - : Etch rate is faster than SC Si
  - : Almost isotropic etching

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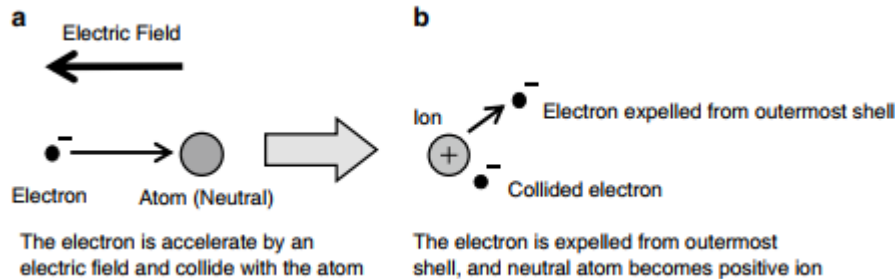
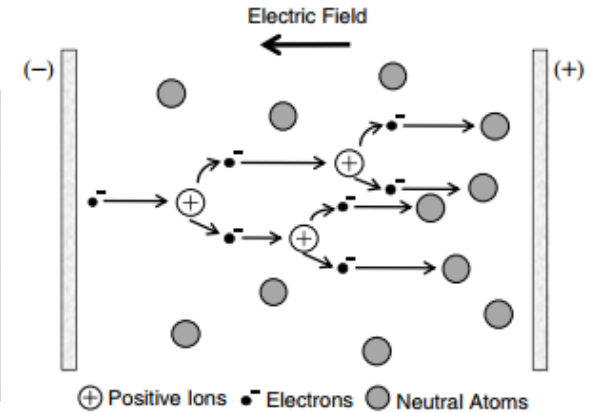
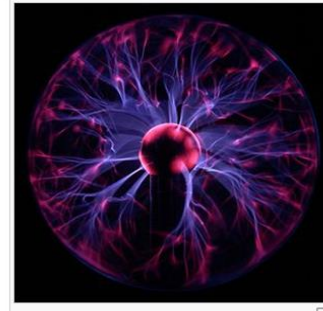
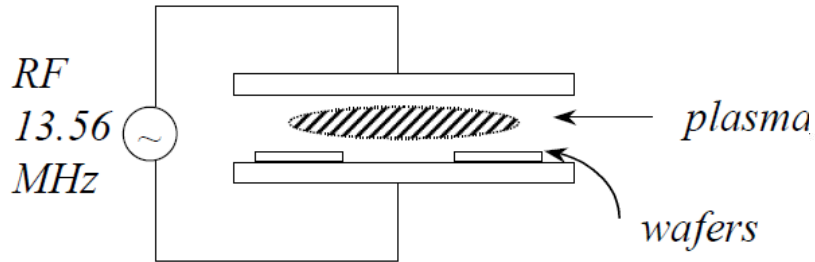
- **Al etching:** 73%  $\text{H}_3\text{PO}_4$  + 4%  $\text{HNO}_3$  (nitric acid) + 3.5%  $\text{CH}_3\text{COOH}$  (acetic acid) + 19.5% DI water
  - : Generally, developer can etch pure Al easily
  - Al alloy can solve this problem (AlNd)

Ref: Michael Köhler "Etching in Microsystem Technology" Wiley

# Dry Etching: Anisotropic (1)

## Etching (1): Bulk Micromachining

### ● Dry etching (Plasma assisted etching)

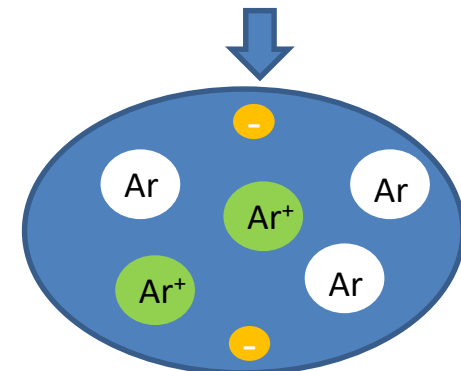


Plasma :

1. Initiated by free electron (such like field emission from negative biased electrode)
2. The electron gain kinetic energy from electrical field
3. The electron collide with gas molecules → Ion + electron

Plasma :

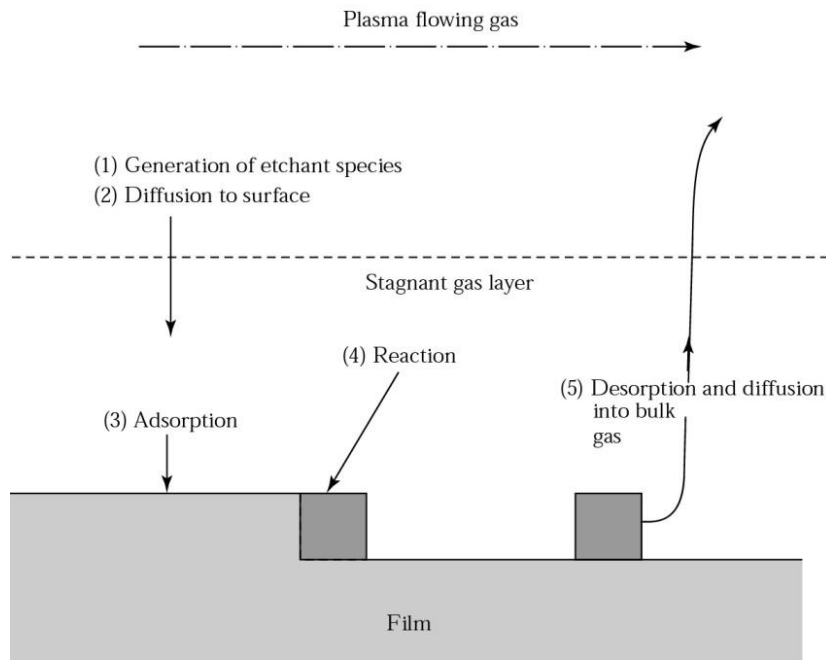
The ionized gas composed of equal numbers of positive ions and negative electrons and unionized molecules



# Dry Etching: Anisotropic (2)

## Etching (1): Bulk Micromachining

### Sequence: mechanism



### Process

1. The etchant species is generated by plasma
2. The reactant moves to the surface of substrate
3. The reactant is absorbed on the surface
4. Chemical (Physical) reaction
5. The compounds are detached

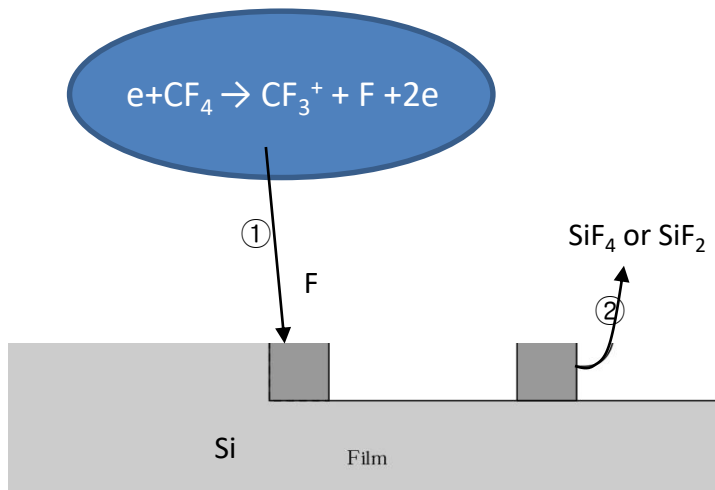
### <Etching gas for dry method>

SOLID	ETCH GAS	ETCH PROD.
Si, SiO <sub>2</sub> , Si <sub>3</sub> N <sub>4</sub>	CF <sub>4</sub> , SF <sub>6</sub> , NF <sub>3</sub>	SiF <sub>4</sub>
Si	CCl <sub>2</sub> , CCl <sub>2</sub> F <sub>2</sub>	SiCl <sub>2</sub> SiCl <sub>4</sub>
Al	BCl <sub>3</sub> , CCl <sub>4</sub> , SiCl <sub>4</sub> , Cl <sub>2</sub>	AlCl <sub>3</sub> , Al <sub>2</sub> Cl <sub>6</sub>
Organic solid	O <sub>2</sub> ,	CO, CO <sub>2</sub> , H <sub>2</sub> O
	O <sub>2</sub> + CF <sub>4</sub>	CO, CO <sub>2</sub> , HF
Refractory Metal (W, Ta, Mo)	CF <sub>4</sub>	WF <sub>6</sub> , ....

# Dry Etching: Anisotropic (3)

## Etching (1): Bulk Micromachining

### Physical mode & Chemical mode



- ① Sputtering effect : Ion bombardment  
→ Physical mode
- ② Pure chemical reaction  
→ Chemical mode

<100mTorr

~100mTorr

Higher Pressure

#### Physical Sputtering (and Ion Beam Milling)

- Physical momentum transfer
- Directional etch (anisotropic) possible
- Poor selectivity
- Radiation damage possible

#### Reactive Ion Etching (RIE)

- Physical (ion) and Chemical
- Directional (~ anisotropic)
- More selective than sputtering

#### Plasma etching

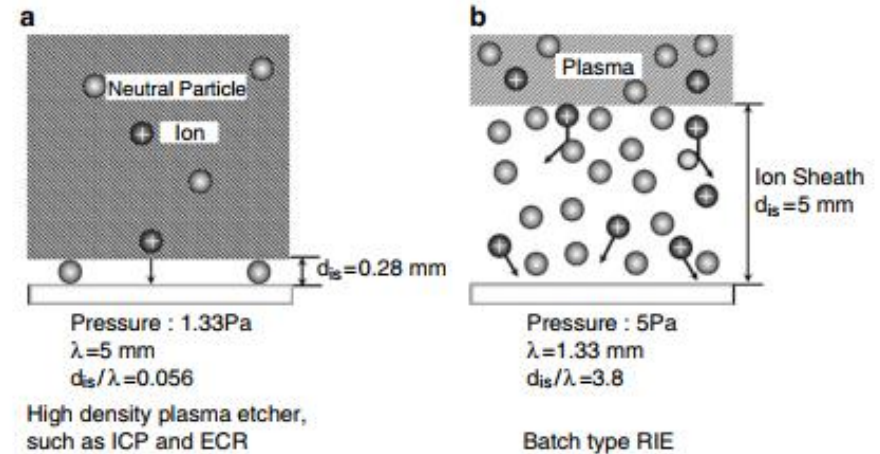
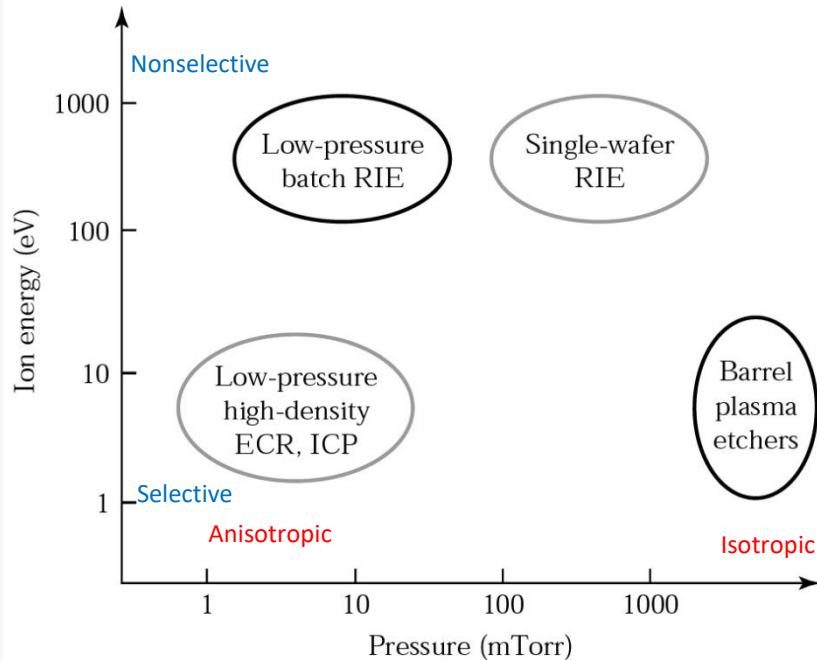
- Chemical, thus faster by 10-1000x
- Isotropic
- More selective
- Less prone to radiation damage

Higher  
Excitation  
Energy

# Dry Etching: Anisotropic (4)

## Etching (1): Bulk Micromachining

### Comparison of ion energy and operating pressure ranges for different types of plasma reactors.

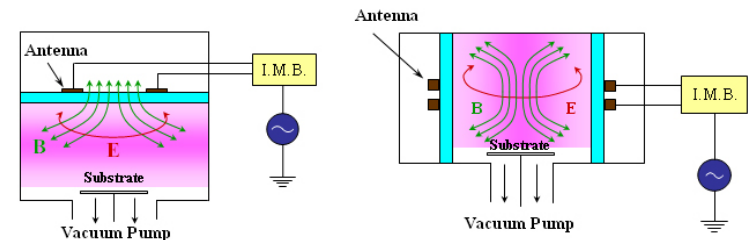


\*ICP: inductively coupled plasma  
ECR: electron cyclotron resonance

Anisotropic etching & High selectivity : Ultimate target

→ - Low pressure

- High plasma density with low power



<TCP: transformer coupled plasma>

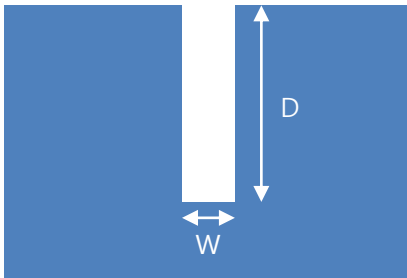
<ICP>



# Dry Etching: Anisotropic (5)

## Etching (1): Bulk Micromachining

### Trench structure

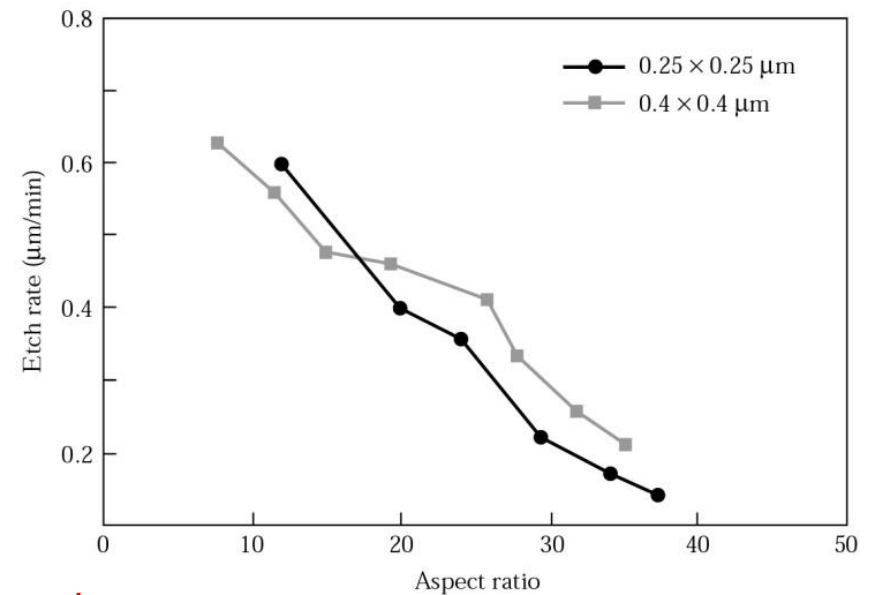


$$\text{Aspect ratio} = \frac{D}{W}$$

High aspect ratio & Small dimension

→ Low etching rate

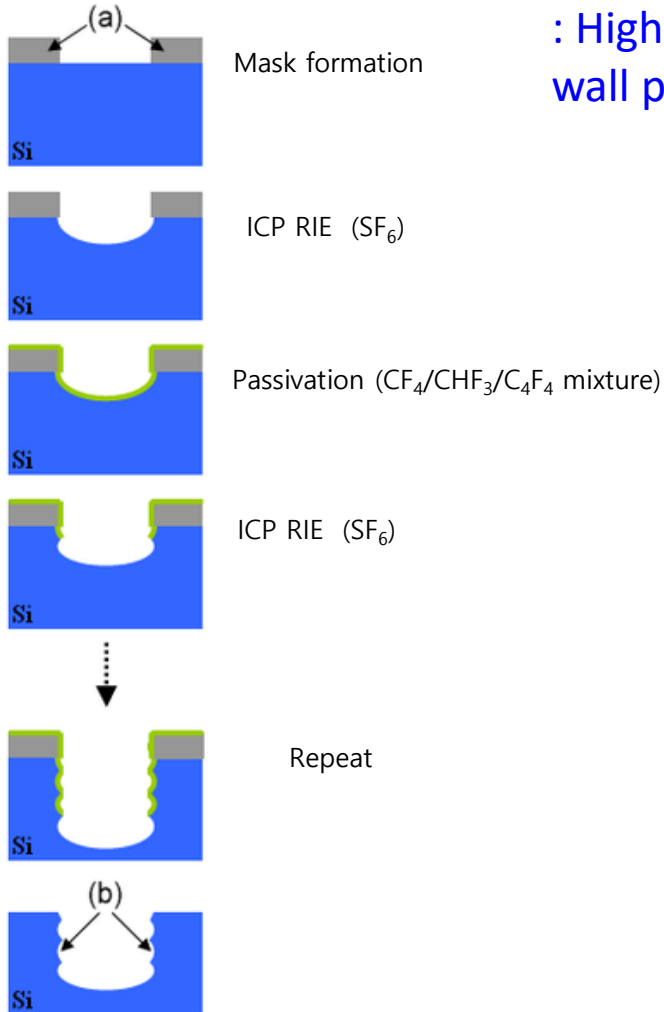
:limitation of ion and neutral transport in trench



# Dry Etching: Anisotropic (6)

## Etching (1): Bulk Micromachining

### New method (1): Bosch Process (Deep RIE)



Mask formation

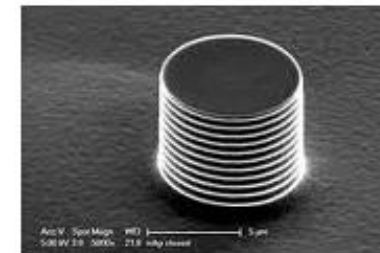
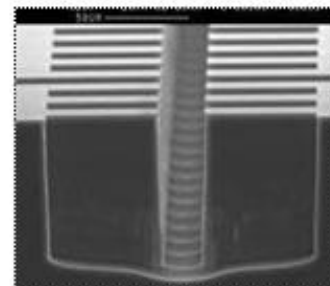
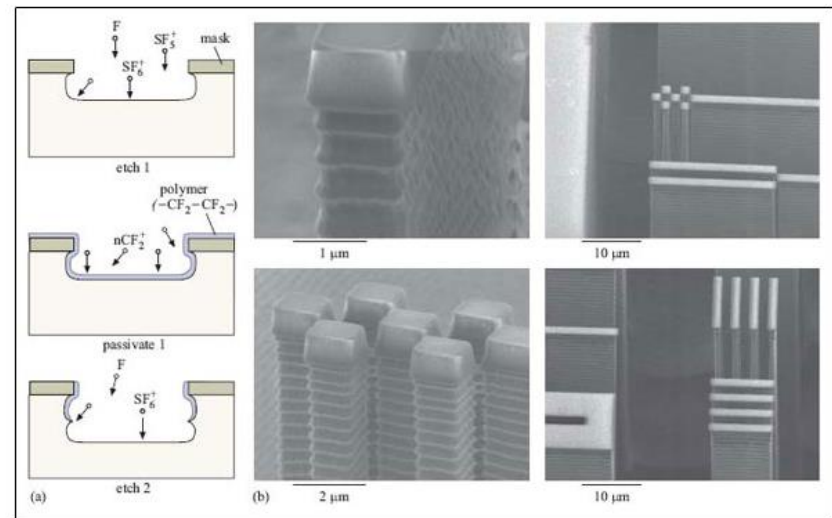
ICP RIE ( $\text{SF}_6$ )

Passivation ( $\text{CF}_4/\text{CHF}_3/\text{C}_4\text{F}_4$  mixture)

ICP RIE ( $\text{SF}_6$ )

Repeat

: High aspect ratio trench or pillar structure  $\rightarrow$  Side wall passivation



## Etching (1): Bulk Micromachining

### ● New method (2): Atomic Layered Etch (ALE)

: self-limiting chemical modification steps -> affect only the top atomic layers of the wafer -> etching steps which remove only the chemically-modified areas, allows the removal of individual atomic layers

