

On the Chemistry and Physical Properties of Flux and Floating Zone Grown SmB₆ Single Crystals

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Figure S1. A movie showing the reconstructed X-ray computed tomography (CT) frames of the

¹⁵⁴Sm¹¹B₆ flux-grown crystal. It is obvious that this crystal is composed of two materials, one with a high atomic number (dark contrast) and low Z low atomic number (light contrast). **S4**

Figure S2. A X-ray diffraction histogram obtained laboratory X-ray diffraction. The asterisks and plus signs denote the reflections from aluminum present within the flux grown crystal of SmB₆ and the Si-standard powder, respectively. **S5**

Figure S3. Rietveld refinements to synchrotron X-ray diffraction data at $T = 295$ K collected on a) cut 2 b) cut 3, and c) cut 4 of the SmB₆ floating zone grown single crystal. The black crosses, red lines, and blue lines correspond to the collected data, refined model, and difference curve respectively. The higher angle data are multiplied by $\times 10$ ($25 \geq 2\Theta \geq 40$) and $\times 100$ ($40 \geq 2\Theta \geq 50$) to highlight the quality of the fit. **S6**

Figure S4. The (110), (310), and (640) reflections for cuts 1-4 where the peak positions were normalized along the x-axis relative to a silicon standard. All peak positions reside at higher angles when going from cuts 1 to 3 for each reflection, showing that the lattice parameters decrease with compositional variations

along the crystal. The overlapping of the (110), (310), and (640) peak positions for cuts 3 and 4 show that these cuts have roughly the same lattice parameters and compositions. **S7**

Figure S5. Concentration (ppm wt) of elements present in the starting material and cuts 1-3 versus Atomic Number. These semi-quantitative trace elemental analyses results were obtained from glow discharge mass spectrometry (GDMS) experiments and are tabulated in Table S1. **S8**

Figure S6. Rietveld refinements to X-ray diffraction data at $T = 295$ K collected on the vaporized material, which amounted only to approximately 1% of the total material from the SmB₆ floating zone single crystal growth. The black crosses, red lines, and blue lines correspond to the collected data, refined model, and difference curve respectively. It is obvious from fits to this data that this vaporized material is a multi-phase mixture of SmB₆ (gray ticks) and SmB₄ (orange ticks). **S9**

Figure S7. A plot of the resistance (R) normalized by the room temperature resistance values ($R_{300\text{ K}}$) versus temperature (T) for cuts 1'-3' (open circles) and cuts 1-3 (closed circles, Figure 5) from 0 to 10 K. To check the reproducibility of our resistance measurements, the authors removed the original platinum leads used to collect the data presented in Figure 5 for cuts 1-3, polished these three cuts, mounted new leads, and recollected the data (cuts 1'-3'). Additionally, resistances were measured using a new cut between the location of the original cut 1 and cut 2 (cut 2'') and a new cut beyond the location of the original cut 4 (cut 4''). Very much like the data for cuts 1-4 in Figure 5, the magnitude of the $R/R_{300\text{ K}}$ and the degree of plateauing decrease and the cut number gets larger for all cuts. Finally, the trend in lattice parameters of 4.1333(4) Å and 4.13284(3) Å for cut 2'' and cut 4'', respectively, agrees well with the resistance and lattice parameter trend for the original cuts 1 - 4. **S10**

Table S1. The concentration of differing elements present in the SmB₆ starting material, cut 1, cut 2, and cut3. **S11-13**

Table S2. Crystallographic parameters for the flux grown ¹⁵⁴Sm¹¹B₆ crystal obtained from model fits to the single crystal X-ray diffraction data. **S14**

Table S3. Atomic fractional coordinates, site occupancies, and ADPs for flux grown ¹⁵⁴Sm¹¹B₆ crystal obtained from model fits to the single crystal X-ray diffraction data. **S15**

Table S4. The percent abundance for the differing isotopes of boron and samarium in the flux grown doubly enriched SmB₆ crystal. **S16**

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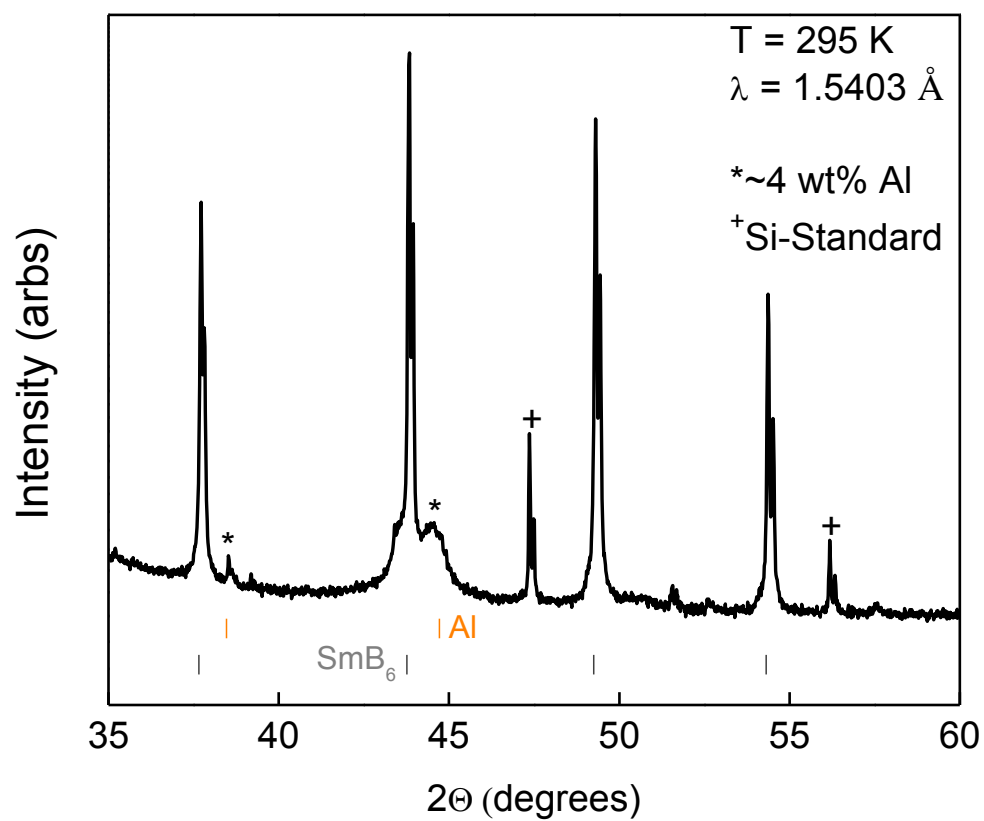


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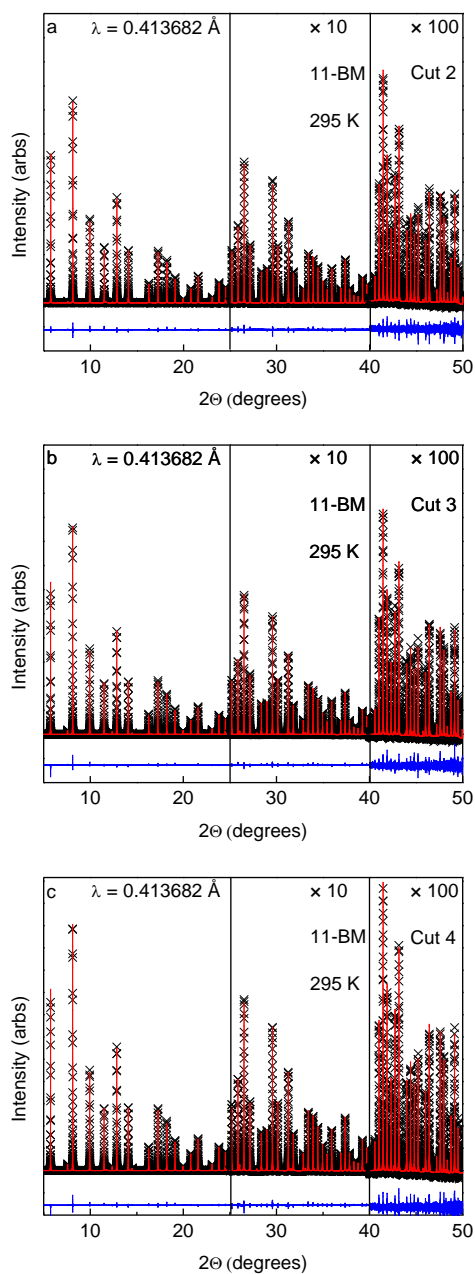


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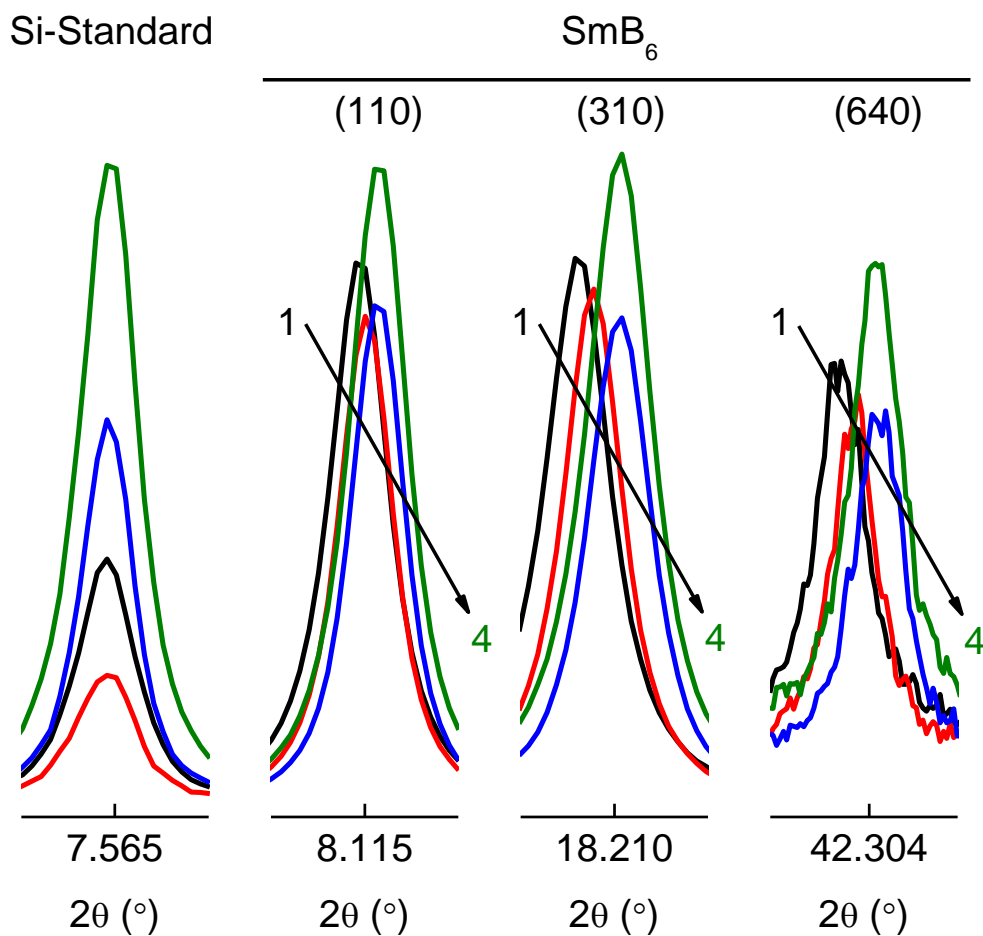


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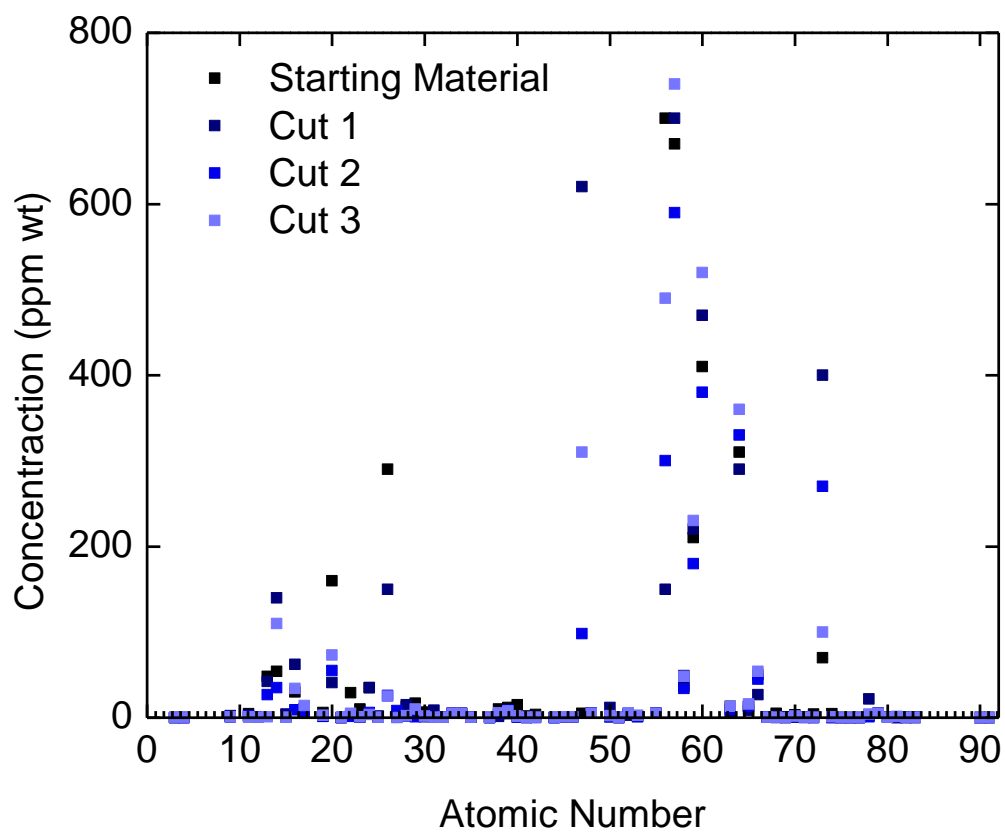


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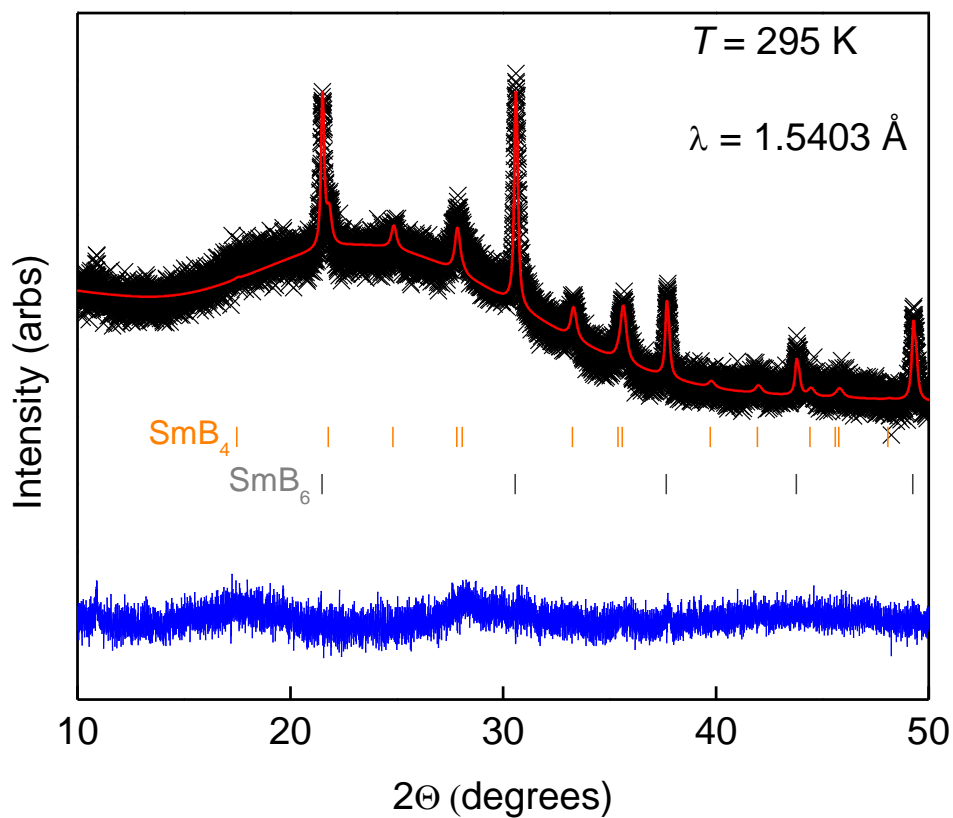


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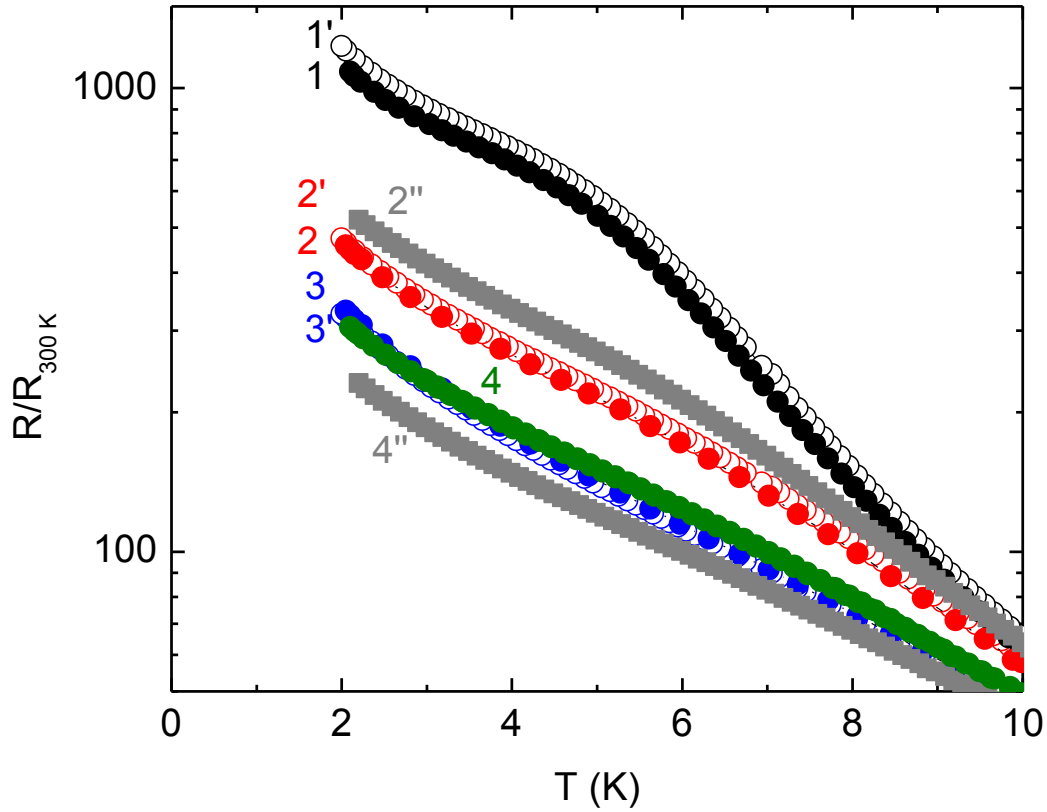


Table S1. The concentration of differing elements present in the SmB₆ starting material, cut 1, cut 2, and cut3.

Concentration (ppm wt)				
	Starting Material	Cut 1	Cut 2	Cut 3
Li	<0.05	<0.05	<0.05	<0.05
Be	<0.05	<0.05	<0.05	<0.05
B	Bulk	Bulk	Bulk	Bulk
F	<1	2.2	<1	<1
Na	4.7	0.97	2.4	0.83
Mg	0.95	1.1	0.91	0.47
Al	48	42	27	0.99
Si	54	140	35	110
P	0.78	4.3	1.1	0.52
S	30	62	9.1	34
Cl	~10	~11	~7	~14
K	6.1	1.3	1.8	2.6
Ca	160	41	55	73
Sc	0.06	0.08	<0.05	0.44
Ti	29	2.0	1.4	5.2
V	10	0.46	0.79	0.76
Cr	35	35	6.0	3.8
Mn	0.65	2.0	0.38	0.27
Fe	290	150	26	25
Co	0.64	2.4	8.2	0.19
Ni	13	15	1.9	2.1
Cu	17	4.3	0.81	9.8
Zn	6.5	0.72	1.9	1.8
Ga	<0.5	8.5	<0.5	<0.5
Ge	<0.5	<0.5	<0.5	<0.5
As	<5	<5	<5	<5
Se	<5	<5	<5	<5
Br	<0.5	<0.5	<0.5	<0.5
Rb	<0.05	<0.05	<0.05	<0.05

Table S1 continued. The concentration of differing elements present in the SmB₆ starting material, cut 1, cut 2, and cut3.

	Concentration (ppm wt)			
	Starting Material	Cut 1	Cut 2	Cut 3
Sr	10	1.5	3.8	5.5
Y	12	4.5	8.4	7.7
Zr	15	0.33	0.71	1.2
Nb	1.8	0.10	<0.05	0.29
Mo	3.8	0.45	<0.05	0.49
Ru	<0.05	0.16	<0.05	<0.05
Rh	<1	<1	<1	<1
Pd	<1	<1	<1	<1
Ag	5.2	620	98	310
Cd	<5	<5	<5	<5
In	Binder	Binder	Binder	Binder
Sn	<1	12	<1	2.3
Sb	<0.1	<0.1	<0.1	<0.1
Te	4.2	2.7	4.8	5.4
I	1.8	0.73	1.2	2.9
Cs	<5	<5	<5	<5
Ba	700	150	300	490
La	670	700	590	740
Ce	37	49	34	48
Pr	210	220	180	230
Nd	410	470	380	520
Sm	Bulk	Bulk	Bulk	Bulk
Eu	13	6.6	9.5	13
Gd	310	290	330	360
Tb	13	8.7	13	16
Dy	50	27	45	54
Ho	0.48	0.30	1.1	0.44
Er	5.1	0.15	0.35	0.61
Tm	0.95	<0.05	<0.05	0.11

Table S1 continued. The concentration of differing elements present in the SmB₆ starting material, cut 1, cut 2, and cut3.

	Concentration (ppm wt)			
	Starting Material	Cut 1	Cut 2	Cut 3
Yb	3.2	0.35	1.5	1.7
Lu	0.94	0.08	0.25	0.51
Hf	4.1	<0.1	<0.1	0.63
Ta	≤70	≤400	≤270	≤100
W	4.7	<0.1	<0.1	1.0
Re	<0.1	<0.1	<0.1	<0.1
Os	<0.1	<0.1	<0.1	<0.1
Ir	<0.1	<0.1	<0.1	0.19
Pt	1.8	22	1.1	3.9
Au	<5	<5	<5	<5
Hg	<0.5	<0.5	<0.5	<0.5
Tl	<0.01	0.10	0.93	1.5
Pb	<0.05	1.1	0.11	0.28
Bi	<0.05	1.1	0.11	0.24
Th	0.04	0.03	0.04	0.07
U	0.03	0.01	0.02	0.03

Table S2. Crystallographic parameters for the flux grown $^{154}\text{Sm}^{11}\text{B}_6$ crystal obtained from model fits to the single crystal X-ray diffraction data. The statistical uncertainties are given in parentheses.

Temperature (K)	293(2)
Composition	$\text{SmB}_{5.88}$
Space group	$Pm-3m$
a (Å)	4.13283(2)
V (Å ³)	70.590(1)
Z	1
Collected Reflections	7089
Independent Reflections	154
Goof	1.18
$R_I[F^2 > 2\sigma(F^2)]^a$	0.009
$wR_2(F^2)^b$	0.020
$\Delta\rho_{\text{max}}$ (e Å ⁻³)	1.00
$\Delta\rho_{\text{min}}$ (e Å ⁻³)	-1.36

$$^a R_I(F) = \sum ||F_o| - |F_c|| / \sum |F_o|; \quad ^b wR_2(F^2) = [\sum [w (F_o^2 - F_c^2)^2] / \sum [w (F_o^2)^2]]^{1/2}$$

Table S3. Atomic fractional coordinates, site occupancies, and ADPs for flux grown $^{154}\text{Sm}^{11}\text{B}_6$ crystal obtained from model fits to the single crystal X-ray diffraction data. The statistical uncertainties are given in parentheses.

$T = 293(2)$ K

atom	Wyckoff Site	x	y	z	Occupancy	$U_{11}(\text{\AA}^2)$	$U_{22}(\text{\AA}^2)$	$U_{33}(\text{\AA}^2)$
Sm1	$1a$	0	0	0	1	0.00776(4)	0.00776(4)	0.00776(4)
B1	$6f$	0.2000(1)	$\frac{1}{2}$	$\frac{1}{2}$	0.98	0.0036(2)	0.0036(2)	0.00523(14)

Table S4. The percent abundance for the differing isotopes of boron and samarium in the flux grown doubly enriched $^{154}\text{Sm}^{11}\text{B}_6$ crystal.

	<u>% Abundance of Boron</u>		<u>% Abundance of Samarium</u>						
	^{10}B	^{11}B	^{144}Sm	^{147}Sm	^{148}Sm	^{149}Sm	^{150}Sm	^{152}Sm	^{154}Sm
Doubly Enriched SmB_6 Crystal	3.6(1)	96.4(2.1)	0.0067(2)	0.044(1)	0.041(1)	0.0550(3)	0.0390(2)	0.305(2)	99.51(0.74)
