# REPO LINK: https://github.com/neal-p/CHEM279

# HW3

# **Compilation Instructions**

- 1. Go to the top level repo directory CHEM279/
- 2. mkdir -p build
- 3. cd build
- 4. cmake ...
- 5. make hw3 bond\_energy

### Run Instructions

Once you have compiled, there will be three executables: - CHEM279/HW3/hw3 - CHEM279/HW3/bond\_energy - CHEM279/HW3/basic\_rxn

#### Discussion

Here, the hw3 executable will run an Extended Huckel calculation on the provided coordinate file (argument 1) and basis set file (argument 2). The coordinate file should be in xyz format with '' as the first line and atomic numbers for the elements. The basis set file must be in Gaussian format such as pulled from the BasisSetExchange website. Note! The system must have an EVEN number of electrons.

The program reads in the information, calculates the AO overlap matrix, the Huckel hamiltonian, and finally the total energy. A few of the crucial matricies are printed in the process.

The sample\_input/ folder contains the provided sample input, as well as a test that the program properly terminates when an odd number of electrons is requested. You can execute run\_samples.sh to run ./hw3 <file>basis\_set.txt for each sample and pipe the output to a corresponding file in my\_output.

An example output is shown below: ./hw3 sample\_input/C2H2.txt basis\_set.txt

```
Read in the following:
```

## Simulation:

Overlap Matrix S:

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1.00000
          0.48501 -0.46382
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                                               0.07295 -0.09755
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          1.00000
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                                               0.46629 -0.45219
 0.48501
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Huckel Hamiltonian Matrix H:
-13.60000 -14.85358
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Matrix X:
                   0.39220
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                                     0.00000 -0.08612 0.07231
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 1.26865 -0.29295
-0.29295
          1.48017
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-0.01205 -0.08612 -0.07231 0.00000 0.00000 -0.29295 -0.39220
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# Check orthogonalizatoin X^t \* S \* X:

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## is identity check: 1

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Matrix V:
 0.23344
          0.45938
                   0.43216
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                                      0.00000
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                                                        0.00000
                                                                  0.53754 - 0.50867
                                                                                     0.00437
 0.65220
          0.37966 -0.03310
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                                      0.00000
                                               0.00000
                                                         0.00000 -0.32851
                                                                           0.27118
                                                                                     0.49793
 0.14192 -0.38056 -0.55870
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                                                                  0.32114 -0.40953
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Matrix C:
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                                                                  0.90554 -0.87300
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 0.53539
          0.37953 -0.03931
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                                               0.00000
                                                         0.00000 -0.52740
                                                                           0.58491
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 0.03575 -0.23725 -0.42622
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                                                                  0.62436 -0.66014
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                   0.00000 -0.14837
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 0.09151 -0.33303 0.37688 0.00000
                                      0.00000
                                               0.00000 0.00000 -0.90554 -0.87300 -0.13580
```

## Orbital energies Vector E:

- -26.79253
- -19.63008
- -15.44271
- -13.35975
- -13.35975
- -7.78142
- -7.78142
- 6.54517
- 10.46392
- 56.05331

# TOTAL ENERGY:

# -177.16963

Using the extended Huckel code, I have a program bond\_energy that given coordinates of H2 and a basis set will calculate the bond energy relative to infinitely separated hydrogen atoms. Below, I show the command to run this program and its output:

./bond\_energy sample\_input/H2.txt basis\_set.txt

Computed energy of H2: -35.3099 eV

Given energy of H: -13.6 eV H2 bond energy: -8.1099 eV

Lastly, I calculate the reaction enthalpy of C2H2 + H2 -> C2H4 with the program basic\_rxn. While my total energies for the individual components match the given expected values, our very basic EHT model doesn't give great agreement to the known enthalpy of -174 kJ/mol. I thought it might be a geometry problem, so visualized the structures and every so slightly tweaked the geometries, but unfortunately, it seems EHT does not capture the required interactions to model the bonding correctly. The program use and output is shown below:

./basic\_rxn basis\_set.txt sample\_input/C2H4.txt sample\_input/H2.txt sample\_input/H2.txt

Computed energy of C2H4: -211.48 eV Computed energy of C2H2: -177.17 eV Computed energy of H2: -35.3099 eV

dH (C2H2 + H2  $\rightarrow$  C2H4): 0.999345 eV or 96.4218 kJ/mol