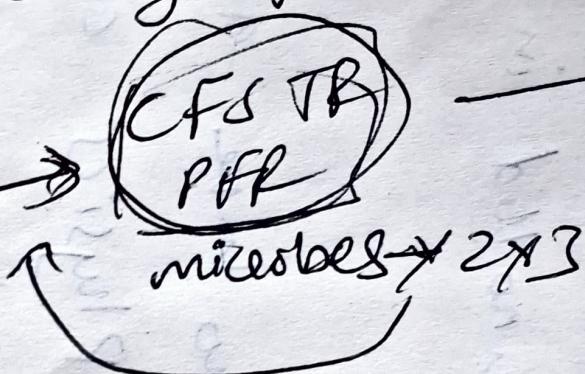


Waste
essence
of its good

A) Inorganic

even primary Clarification — Rags
separated sludge process

Air



Solid
waste

Manufactured

UV
Wetlands - walls (salt)
dissolved water treatment

Filtration

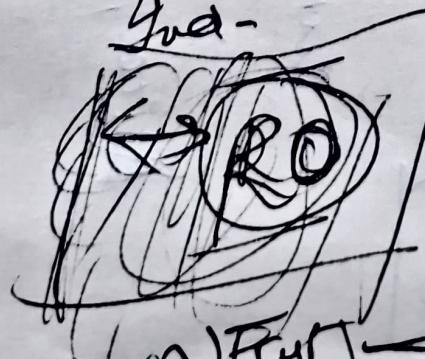
cloth

disc filters

Ind-

Municipal waste water

Waste



Hard

Lake Houston

pulls water into plant

desalination

water process

Cogeneration / flue gas

Production of NH₃ Methods

Direct Air —

SOC → total energy = IN and OUT

H₂ — free electricity available → what's

Baseline process

operating cost

(LCOH)

Scalable

Electrolysis

- Human health (carcogen)
- Environment

Growth
Acclimation
commu

Heat
Methods Desalination
Cheaper
Thermal collectrs.

(Direct electrolysis)

Baseline process

Baseline approach

(MEB)

wastewater

Electrolysis

Photo
voltaics

(batteries)

Renewable

Electrolysis

Fuel cell
battery storage

(CCAP for halve process)

(halve) → cheaper

Solar

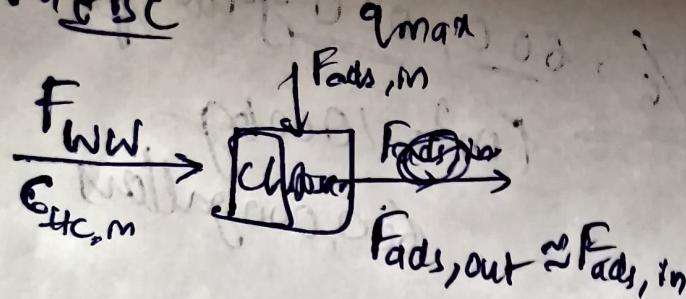
Hopewell at 30°

Ammonia is stored in hydroodes

Expensive
lithium hydrides

Lanthanum

MEBC



Flow: m³/h or gal/h
 $C_{HC,in}$ mg/l
 $C_{HC,out}$ mg/l
 q_{max} : Adsorption
 Organic capacity
 per g (mg HC per
 g adsorbent)

$$F_{HC, \text{ adsorbed}} = (C_{HC,in} - C_{HC,out}) F_{WW, \text{ in}}$$

$$175^3 = 269.13 \text{ gallons}$$

HAZOP study

$$\rightarrow 100 \text{ m}^3/\text{h}$$

634,000 gallons/day water

000

→ Define Volume water
 1000 gpm
 2000 gpm

intake

cost,
CAPEX

OPEx
 chemicals

warranty.

Toxicity

GCFI

Gulf Const

Water Authority

AB

300-500°C

1000 gpm

500°C

700-800°C

1) TCEQ guidelines
 $x^\circ\text{C}$ (surface water
 discharge point)

2) Deep well
 injection.

Antiscalant
 corrosion inhibitors

Prof. JB Powell

→ microbial fuel cell
water



(Mixed Resource)

① $\text{CO}_2\text{-eq}/\text{FU}$: Measures the climate impact (how much it contributes to global warming)

② $\text{Cu}\text{-eq}/\text{FU}$: Measures resource use (how much it affects material availability)

Judging?

Idea 3:

Photoelectrochemical → long-term form of t

3rd idea → Haber-Bosch process: (350°C)

AER Alkaline electrolysis

Heat from HB process - ?

Energy

TCA = electricity

size of battery

use of electrodes

PAP - $1g(4t)$

2.6 power scaling rule

NREL - After Prototype
Colorado test site

① H₂ - reactive, does not exist in free form. ② H₂ prod.
843 metric tons of CO₂ emitted
every year from fossil fuels

Fuel cell. Brown hydrogen [from coal] — Gasification —

Grey hydrogen (3/4 of all th) [Natural gas] — Steam reforming —

Blue H₂: [fossil fuels → carbon capture] — Gas + steam.

③ H₂ used widely in Oil refinery, steel Industry, methanol prod., ammonia prod.

High energy density

should be stored under high pressure

Storage & for short time: Battery

[California, Texas — so much electric
Renewable power on electric
grid.]

Electrolysis

Polymer Electrolyte Membrane Electrolyzers (PEM)

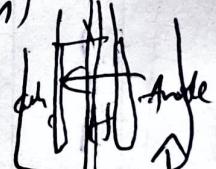
Alkaline Electrolyzers

Solid Oxide Electrolyzers

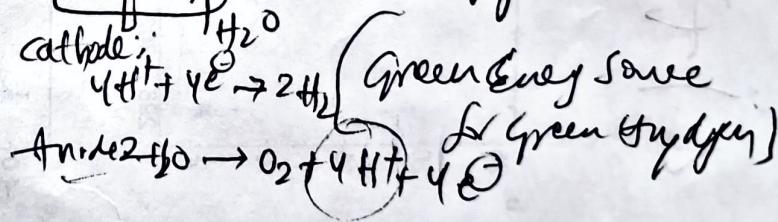
→ Needs to have cheap electrolyzer

Electrolyzer

+ Mitsubishi Power
Magnus development
Utah



cathode: 4H₂O + 4e⁻ → 2H₂



Entropy.

To do

① Compare H₂ power vs Nuclear Power. — Why H₂ so imp?

Seawater or ocean water?

② Clean for Sigma

③ Yield?

④ Electrolyzer — Electrolyzer?

JV · JV = 0.00

$$\frac{JV}{\sqrt{m^3}} \cdot 1m^3 = 1000 \text{ kJ} \cdot 1000 \times 10^3 \text{ m}^3 \cdot \frac{JV}{\sqrt{m^3}} \stackrel{\circ}{\text{C}} \text{ s}^{-1}$$

$\frac{JV}{\sqrt{m^3}} \times 10^6 \text{ J} \cdot 10^6 \text{ m}^3 \cdot 10^6 \text{ m}^3 \stackrel{\circ}{\text{C}} \text{ s}^{-1}$

$\text{J} = f, \exp\left(\frac{-Q}{RT}\right)$

pictures ·
Bad numbers

Habell Bosch plant Haldor-Topsøe

Solar Powerplants
PVPS - poly electrolyte

Cumulative cost?

→ Community impacts — people living closer by — sustainability

↳ Baytown (west Texas)
↳ Solar fine

② Out operations
MFC with
Integrate RO to OEC
add some blocks

③ Photo electrochemical water splitters.
+ add additional units

$$\text{CAPEX OPEX/kW} = \frac{\text{CAPEX} + \text{OPEX}}{\text{fuel rate}} = \text{MSP} (\text{min. selling price})$$

Estimate number for energy consumption → estimate carbon emissions
EPA (Environmental Protection Agency) — tends to some
numbers for CO2 from energy mix

Additional cost? ✓
Additional cost? ✓
Environmental cost? ✓
Short term vs medium vs long term
Renewable cost ✓
Time
cost of electrolysis
④ Calculation of solar & wind at
the site

~~Technological~~ ~~hydrogen~~ - global pressure Step 1 minimizes cost of electrolyzer

↪ Cost effectiveness.

Capital cost

electrolysis

cost of electricity from

literature review

Renewable Energy Sources

(economically
viable)

① Cost of Hydrogen

Techno-economic analysis.

{ CAPEX
OPEX

② Technology commercial scale ??
new technology

- PEM - most researchers

Step 2: RO

→ cost of RO?

Paper-RO

electrolysis to RO from Renewable

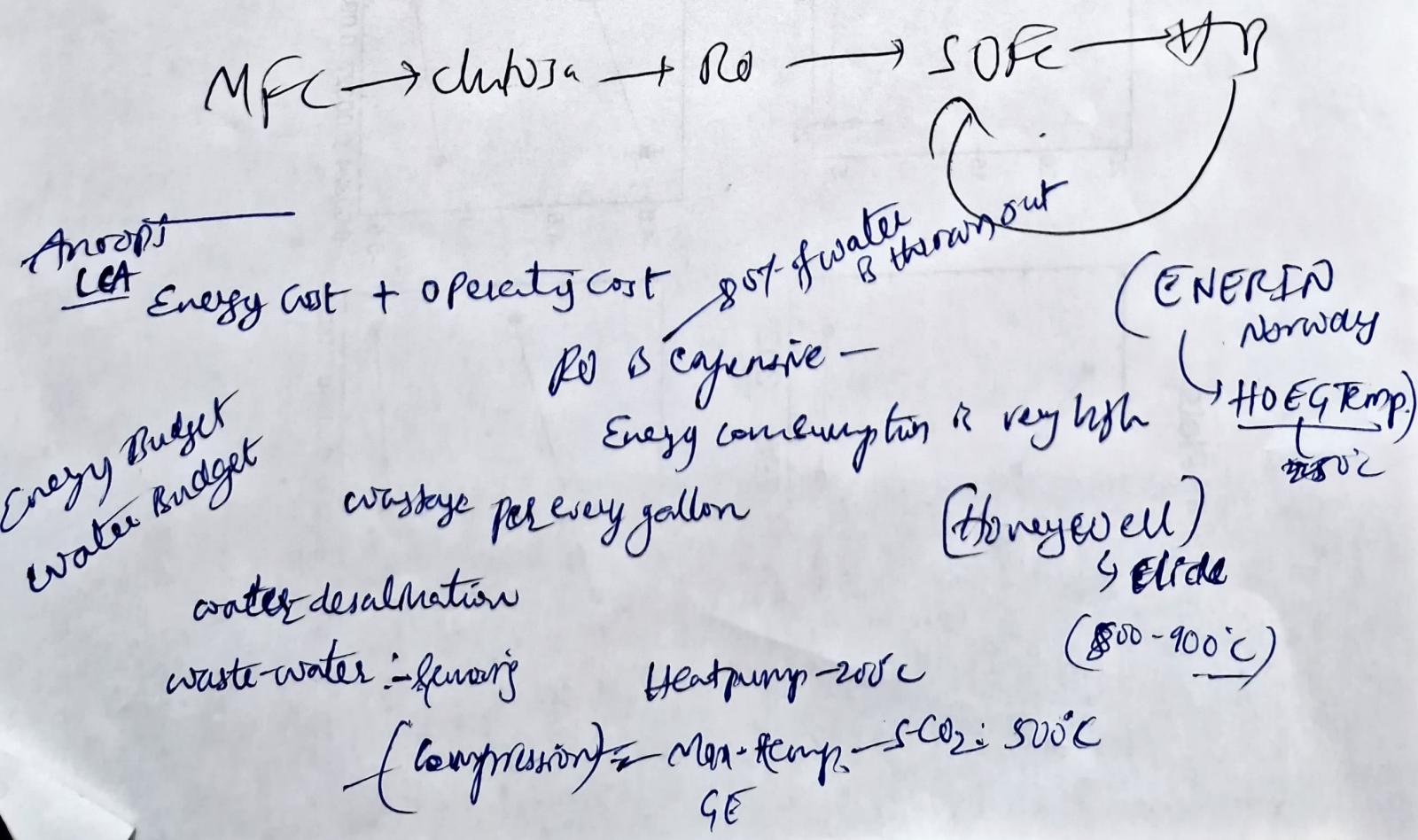
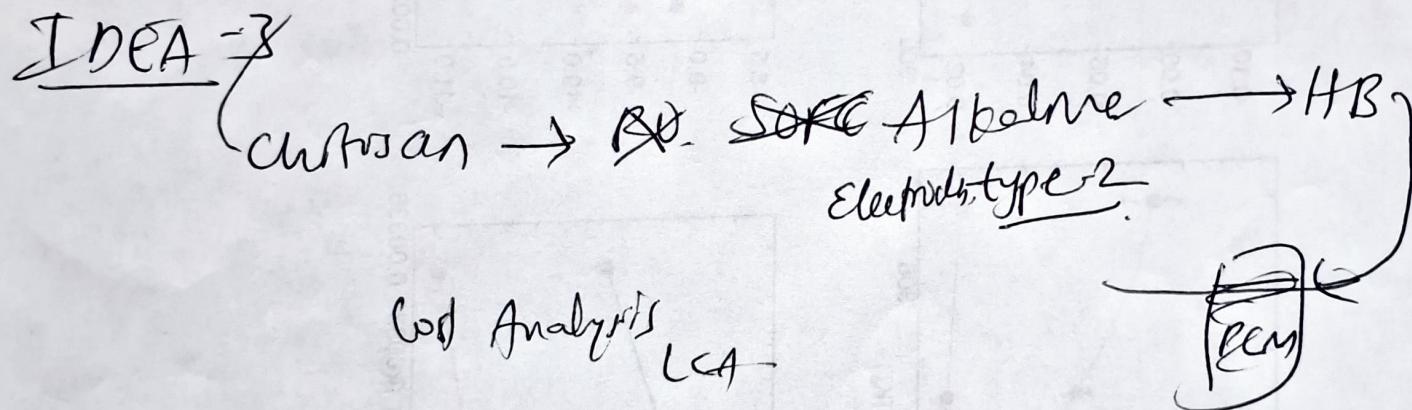
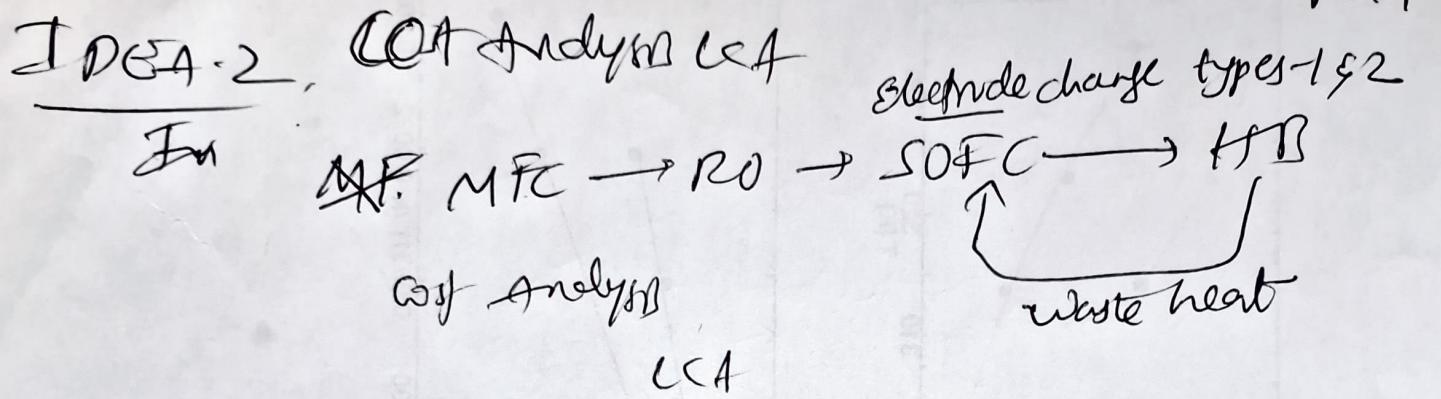
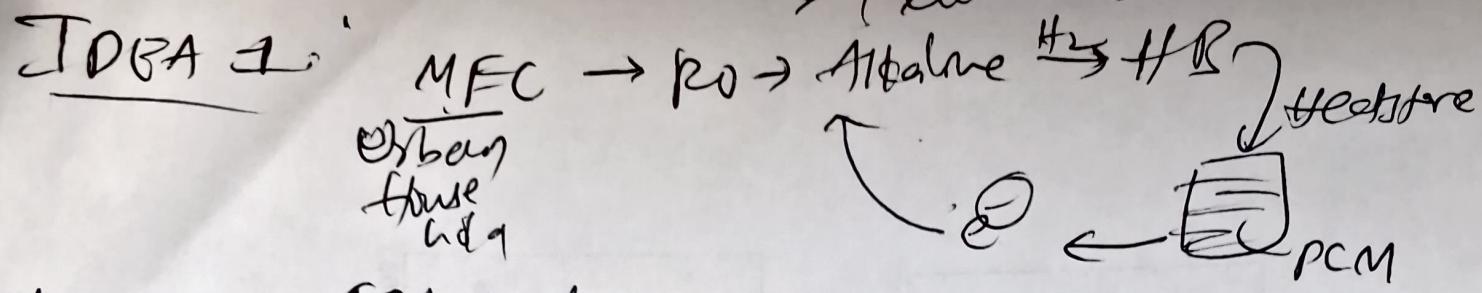
Blue H₂

Technical Assessment

2 scenarios 100% Renewable to produce Green H₂

Energy Balance

Elect from Grid - Plants
of Texas



\rightarrow \rightarrow \rightarrow

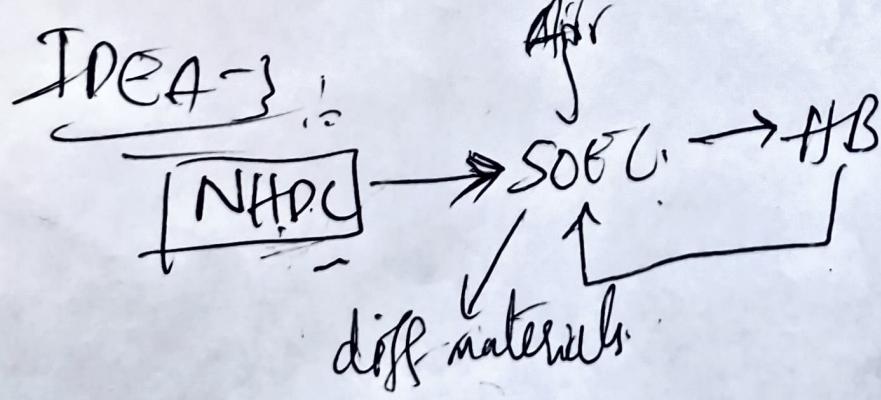
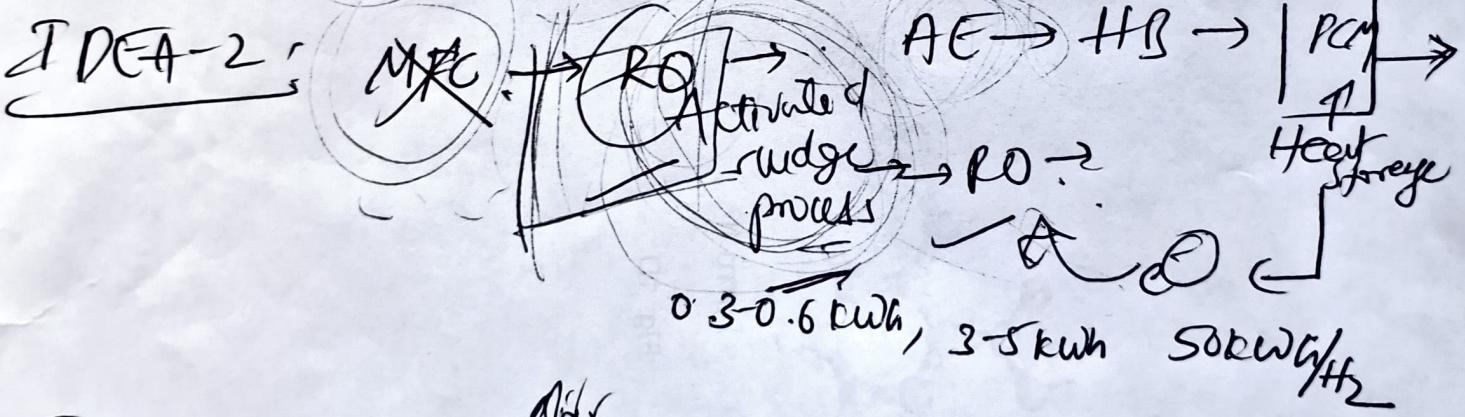
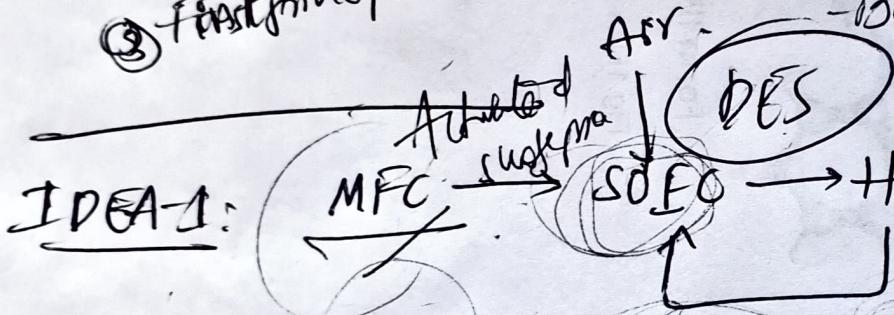
Hysys model - ONLINE

NREL, NEL, PNNL

① Models & -
Ansys model - prototype

③ First principles

- Baseline model -



Q2 - Life Cycle Assessment

SOFC, PEMEC

- 16 impact categories

Alkaline electrolyss : 60-90°C, nickel, stainless steel

PEM : 50-100°C: High power density, platinum

SOFC: 600-900°C due to favorable kinetics

SOFC: - built on oxygen conducting and ceramic membranes

Oxygen electrode

Electrolyte

Hydrogen electrolyte

Interconnect

Frame

Pewrite solar cells:

Perfluorosulfonic acid (PFSA) - Nafion membrane

Copolymers: Tetrafluoroethylene +
fluorinated vinyl ether

T

electrode material?

Electrode Charge

Electrolyzer
Electrolysis

10 pounds steel
LCA - Harm factors
tools free?
Flame Retardant
Catalyst

Life cycle Assessment?

Software

Euro Eco indicator 99

9 diff factors

LCA for solar

colleagues

or carbon composites

Entire Plant

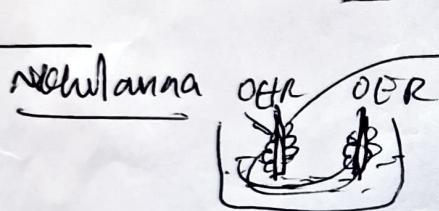


to monitor
electrolyte pH
additive
too weak



(D)

General Atomic



Functional
Pt Rutherford
oxide

(D) Goal

PEM, AEM, solid Oxide

→ Energy

nuclear power

(a)

→ green ammonia going to Japan. (U.S. govt by development
January)

→ Ammonia + Electrolyzed -
One ammonia RAN, [oxide electrolytes] solid oxide Ele

→ Wind is flowing - futuristic. Idaho labs.

Intermittency PNNL - Oldham area

Solar, wind, intermittency - simple science experiment (bubbles)
so bad - waves the nail (diamond)
efficiency of electrolyzer

$$F = \frac{72}{180} = 0.4$$

Energy:

(1) Cata
Thermal Decomposition of water at high temp.
 $\text{conc H}_2\text{SO}_4 - \xrightarrow{\text{SVDIC}} \text{H}_2 + \text{O}_2$

Westinghouse process at 800°C
decompose H_2 to O_2

catalysis
reduce thermal
solar heat difficult
good & possible

(2) Using wastewater \rightarrow Fuel cell

Reverse Osmosis - expensive

Membrane - ΔP - require

70-80°C

water desalination - top effect evaporators

thermal collector & purify water

- (photovoltaic) high temps

photovoltaic devices
photo voltaic
multiple layers of light
absorbing materials.

perovskite-silicon
tandem.

Gallium Cadmium cells? ~~efficiency~~
Cadmium cells ~~efficiency~~
Tandem cells - higher efficiency

higher efficiency
lower temp,
better perform

Plant

2 pounds/day

Scrubber - use H_2 store & electricity

1 pound for supply / m

carbon

1 pair of sack to electrodes

carbon-composite tanks - with high stored pressure - store H_2

long fiber composite
light weight

stainless-oxysine

Add

5 ± 1

$\frac{2}{5}$ $\frac{1}{6}$

0.4 1

6 ± 2

$7 \pm$

$$0. \frac{00014}{1.4 \times 10^{-4} \times 10^1} \times 10$$

~~140×10~~

4.9×10 ~~10~~

~~140×10~~

8.9×10

~~10~~

~~$1 > 2 > 3$~~

C_{O_2}

I
II
III

27