

Fig. 1. Dissociation some aren diazonium cations

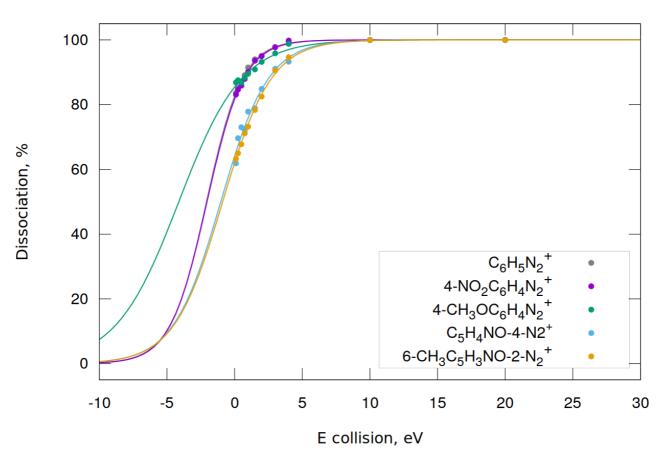


Fig. 2. Dissociation some NO-pyridines and aren diazonium cations

$$K_{D} = \frac{[Ar^{+}]}{[ArN_{2}^{+}]} = \exp(B + \frac{E_{collision}}{\Delta G})$$
 (1)

Table 1. Parameters of aproximation equation (1) for some diazonium salts

N₂	Diazonium salts	ΔG, eV	В
1.	$C_6H_5N_2^+TfO^-$	1.33	1.56
2.	$2\text{-NO}_2\text{C}_6\text{H}_4\text{N}_2^+\text{TfO}^-$	1.71	-0.84
3.	$3-NO_2C_6H_4N_2^+TfO^-$	1.60	1.83
4.	$4\text{-NO}_2\text{C}_6\text{H}_4\text{N}_2^+\text{TfO}^-$	1.36	1.50
5.	$4\text{-NO}_2\text{C}_6\text{H}_4\text{N}_2^+\text{TsO}^-$	1.42	1.56
6.	$4-NO_2C_6H_4N_2^+BF_4^-$	1.27	1.56
7.	$4\text{-CH}_3\text{OC}_6\text{H}_4\text{N}_2^+\text{TfO}^-$	2.33	1.77
8.	$4-BrC_6H_4N_2^+TfO^-$	1.77	0.80
9.	$4\text{-HCO}_2\text{C}_6\text{H}_4\text{N}_2^+\text{TfO}^-$	1.16	2.63
10.	$4\text{-Br-2-HCO}_2\text{C}_6\text{H}_3\text{N}_2^+\text{TfO}^-$	2.12	-0.07
11.	$C_5H_4NO-4-N_2^+$ TfO	1.76	0.58
12.	6-CH ₃ C ₅ H ₃ NO-2-N ₂ ⁺ TfO ⁻	1.80	0.49
13.	$4-Br-2-NO_2C_6H_3N_2^+TfO^-$	2.11	-0.21
14.	$C_{10}H_7$ -2- N_2^+ TfO	1.29	2.92