Using DFT to model TiO$_2$ nanoparticles

Nicolas Gastellu[1] and Rustam Z. Khaliullin[1, *]

[1]*Department of Chemistry, McGill University, 801 Sherbrooke St. West, Montreal, QC H3A 0B8, Canada*

(Dated: August 28, 2018)

TABLE I. Parameters used for evaluating the MA potentials.

Element	Z ($|e|$)	A (Å)	B(Å)	C (Å^{3}kJ$^{1/2}$mol$^{-1/2}$)
Ti	+2.196	1.1823	0.077	22.5
O	-1.098	1.6339	0.117	54.0

INTRODUCTION

SIMULATION DETAILS

Classical MD

All classical MD simulations described in this work were ran in the NpT ensemble, using the Matsui-Akaogi (MA) potential to model the interactions between pairs of atoms. The MA potential was originally developed for classical MD simulations of the four main polytypes of crystalline TiO$_2$ and has been shown by previous studies to be the most adequate force field for predicting the structure and properties of its liquid and amorphous phases. The MA potential describes the short-range interaction between atoms with a Buckingham potential and their long-range electrostatic interaction using the traditional Coulomb term:

$$V_{ij}(r) = f_0 \cdot (B_i + B_j) \cdot \exp\left(\frac{A_i + A_j - r}{B_i + B_j}\right) - \frac{C_i C_j}{r^6} + \frac{e\, Z_i\, Z_j}{4\pi\epsilon_0 r}, \quad (1)$$

where r denotes the distance between the two interacting ions i and j, e is the elementary charge, Z_i is the dimensionless ionic charge of ion i, f_0 is a standard force of $4.184\,\text{kJmol}^{-1}\text{Å}^{-1}$, and A_i, B_i, and C_i are parameters corresponding to i. The numerical values of the constants listed above are given in table I.

The atomic structures of the various $a - \text{TiO}_2$ nanoparticles described in this work were obtained in multiple steps. We started by melting rutile ($r - \text{TiO}_2$) nanocrystals of 198, 390, 768, and 1842 atoms respectively whose structures were previously optimized at the BLYP/DZVP-GTH level of theory (using a plane wave energy cutoff of 2100 Ry) using classical MD in the NPT ensemble with $T = 2000\,\text{K}$ and $P_{\text{ext},0} = 0.0\,\text{Pa}$. We ran this first round of simulations for 40000 timesteps of $\Delta t = 0.5\,\text{fs}$. The resulting melt was then cooled in three steps; classical MD simulations using the same potentials, ensemble, and number of steps were ran using $T = 1500\,\text{K}, 750\,\text{K}$, and $300\,\text{K}$ successively, all with $\Delta t = 2.0\,\text{fs}$. This process simulated the annealing of the melted nanocrystals into a glass which we then studied using Kohn-Sham DFT.

Seeing as the MA potential was originally elaborated to describe the describe the structural properties of $r - \text{TiO}_2$, we also generated a set of conformations of a rutile lattice comprised of 72 atoms in the NPT ensemble at $T = 300\,\text{K}$, using the same potential as described above. Doing so provided us with energy values which we expect to be well correlated with the ones we obtain with DFT methods, thus giving us a reference data set to which we could compare the rest of our results.

A. DFT calculations

Having obtained equilibriated atomic structures for nanoparticles of different sizes, we sampled 100 conformations from the last 40 ps of the last cooling run, at which point all four nanoparticles were in equilibrium with $T = 300\,\text{K}$ thermal bath. We then ran single-point energy calculation using KS DFT at the PBE/DZVP level of theory, using a plane wave cutoff of 2000 Ry. We also ran similar DFT calculations the different configurations of the rutile lattice that we mentioned in the previous section.

RESULTS AND DISCUSSION

For each nanoparticle, we compare the classically evaluated energy of every configuration against the energy obtained through a DFT calculation for that some configuration. We plot our results in figures 1-5. This allows us to gauge the accuracy of MA forcefield; if the classical and quantum mechanical results are well correlated, then the purely classical representation of the forces in the nanoparticles is sufficiently accurate to discriminate between slight variations in a given system's configuration and can therefore be used to simulate this system nanoparticles instead of DFT, which is much more computationally expensive.

Comparing the energies obtained using classical MD and those calculated using DFT reveals that a description of the forces inside an $a - \text{TiO}_2$ nanoparticle using only the two-body MA potential is not precise enough to yield an accurate representation of its potential energy surface (PES) (with respect to its atoms' positions). Indeed, plotting the energies yielded by both calculation methods reveals that DFT calculation methods are much more sensitive to a change in a given nanoparticle's atomic configuration than classical methods.

The system for which this is the most obvious is the 768 atom nanoparticle, for which all classically eval-
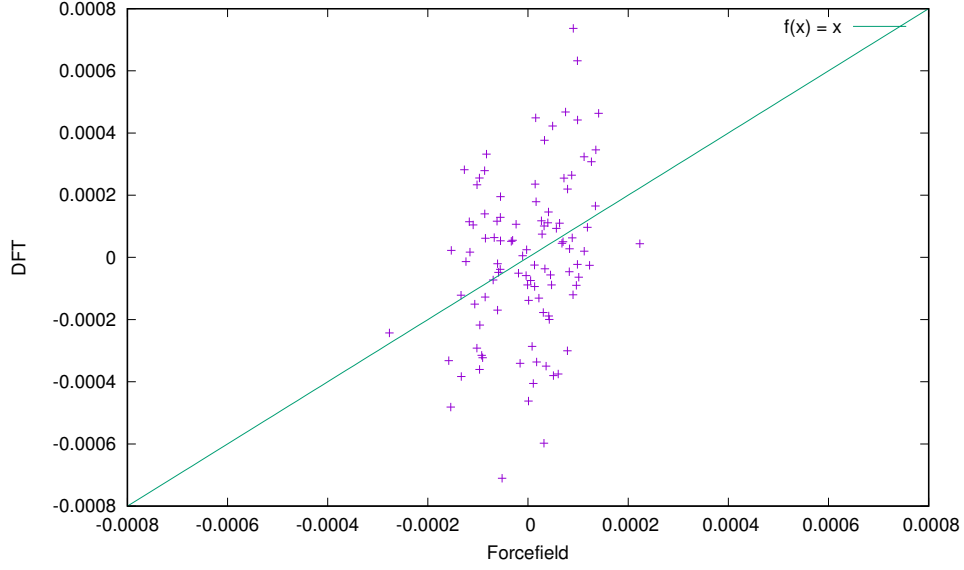
FIG. 1. DFT energy vs. classical energy for 101 conformations of an $a - \mathrm{TiO_2}$ nanoparticle with 198 atoms.
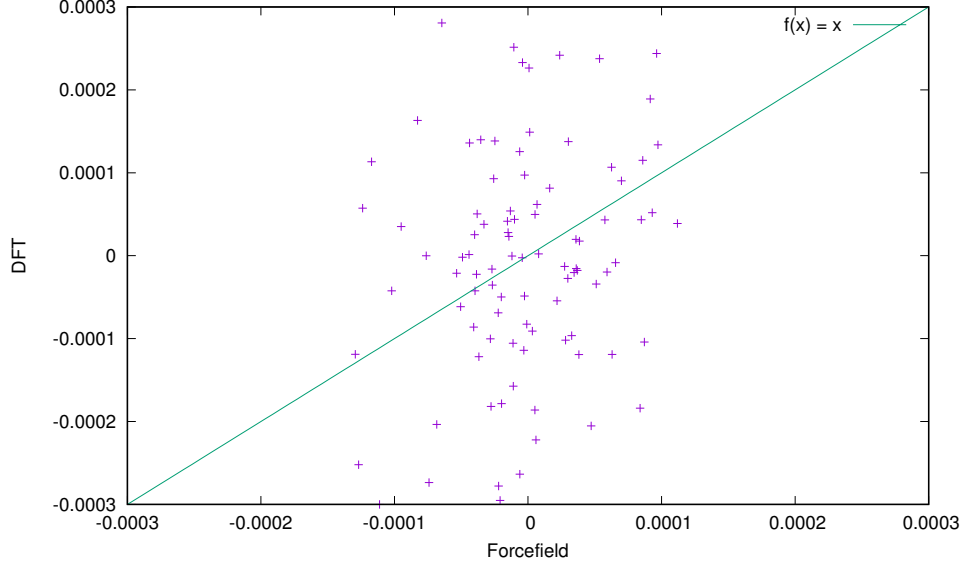

FIG. 2. DFT energy vs. classical energy for 101 conformations of an $a - \mathrm{TiO_2}$ nanoparticle with 390 atoms.

uated energies lie within $\sim 10^{-4}\,\mathrm{Ha}$ of each other, while the energies obtained usinq quantum mechanical methods vary by $\sim 10^{-1}\,\mathrm{Ha}$. While this effect is most dramatic for the 768 atom system, every other nanoparticle on which we ran similar calculations exhibit significant clustering of the energies obtained using the MA potential about their mean value, while their DFT energies spread out over a much larger interval[1].

All four nanoparticles we ran these calculations exhibit weak correlation between the DFT-calculated energy values and the energies obtained using the MA potential. Even more surprisingly we find that the two sets of energy values of the different conformations of the ru-

tile lattice were even less correlated than the ones obtained from the amorphous nanoparticles. Indeed, as can be seen in table A, our simulated rutile lattice yielded the lowest value of ρ of all the systems we studied.

Having observed that DFT calculations are necessary to properly sample from an $a - \mathrm{TiO_2}$ nanoparticle's PES, we are now faced with the problem posed by the computing power necessary to run such calculations on large systems (i.e. nanoparticles containing more than ~ 1000 atoms). Seeing as computational expense is one of DFT's main drawbacks, a number of optimisation schemes have been proposed over the years to reduce both the amount of memory and the time necessary to run
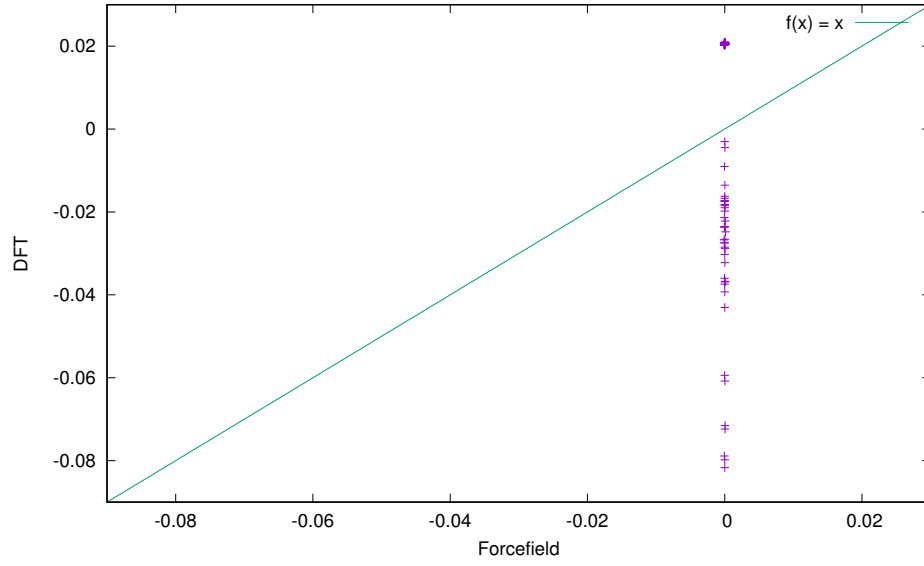
FIG. 3. DFT energy vs. classical energy for 101 conformations of an $a - \mathrm{TiO}_2$ nanoparticle with 768 atoms.
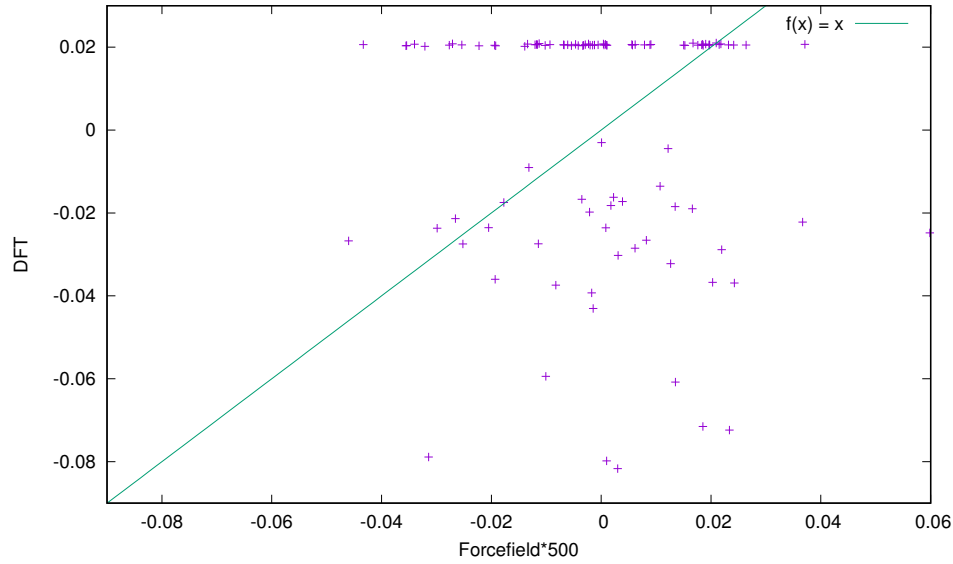

FIG. 4. Rescaled version of figure A.

such calculations. Among them, we focus our attention to the absolutely localised molecular orbitals (ALMO) method which partitions the system being simulated into smaller fragments (usually these are the individual atoms or molecules that compose it) and runs traditional DFT on each fragment in parallel. This has the advantage of having a runtime that scales linearly with the number of fragments for large systems. Seeing as ALMO DFT makes a number of large DFT calculations much more feasible, we decide to test its accuracy on our various $a - \mathrm{TiO}_2$ nanoparticles. As before, we evaluate the accuracy of the ALMO DFT method by comparing the energy it outputs for a given configuration of a nanoparticle with the energy obtained through classical DFT.

CONCLUSIONS

System	$a - TiO_2$ (198 atoms)	$a - TiO_2$ (390 atoms)	$a - TiO_2$ (768 atoms)	$a - TiO_2$ (1842 atoms)	$r - TiO_2$ (72 atoms)
ρ	0.3066474	0.3497706	-0.0496755	TBC	0.0141203
RMSE	0.0486056	0.0628487	22.21090	TBC	0.0518275

TABLE II. Pearson correlation coefficient ρ and RMSE between the energy values obtained through DFT and those obtained using the two-body MA potential on various configurations of different systems.

---

* rustam.khaliullin@mcgill.ca

[1] Daniele Selli, Gianluca Fazio, and Cristiana Di Valentin, "Using density functional theory to model realistic tio2 nanoparticles, their photoactivation and interaction with water," Catalysts **7**, 357 (2017).