

# CHEN 364 HW7

April 10, 2023

```
[ ]: import matplotlib.pyplot as plt
import numpy as np

from scipy.integrate import solve_ivp, odeint

import matplotlib_inline
%matplotlib inline
matplotlib_inline.backend_inline.set_matplotlib_formats('png', 'pdf')
```

## 1 Problem 1

Mole balances:

$$\frac{dF_A}{dV} = -r_1 v_0 - r_2 v_0$$

$$\frac{dF_B}{dV} = -2r_1 v_0$$

$$\frac{dF_C}{dV} = 2r_1 v_0 - r_2 v_0$$

$$\frac{dF_D}{dV} = 2r_2 v_0$$

Reactions:

$$r_1 = k_1(T)C_A C_B^2$$

$$r_2 = k_1(T)C_A C_C$$

Temperature:

Adiabatic  $Q_r = 0$

$$\frac{dT}{dV} = \frac{r_1 \Delta H_1 + r_2 \Delta H_2}{F_A C_{P,A} + F_B C_{P,B} + F_C C_{P,C} + F_D C_{P,D}}$$

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[ ]: ode_kwargs = {
    'method': 'Radau',
    'atol': 1e-8,
    'rtol': 1e-8,
}

def p1_ode(t, y):
    f = y*0
```

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F_A = y[0]
F_B = y[1]
F_C = y[2]
F_D = y[3]
T = y[4]

v_0 = 10
C_PA = 20
C_PB = C_PA
C_PC = 60
C_PD = 80
H_1 = 20000
H_2 = -10000

k_1 = 0.001 * np.exp(5000 * 4.184 / 8.314 * (1 / 300 - 1 / T))
k_2 = 0.001 * np.exp(7500 * 4.184 / 8.314 * (1 / 300 - 1 / T))

r_1 = k_1 * F_A / v_0 * (F_B / v_0)**2
r_2 = k_2 * F_A / v_0 * F_C / v_0

f[0] = (-r_1 - r_2) * v_0
f[1] = -2 * r_1 * v_0
f[2] = (2 * r_1 - r_2) * v_0
f[3] = 2 * r_2 * v_0
f[4] = (r_1 * H_1 + r_2 * H_2) / (F_A * C_PA + F_B * C_PB + F_C * C_PC +
↪F_D * C_PD)

return f

T_range = np.linspace(300, 600, 13)

p1_sols = []

for i, val in enumerate(T_range):
    ode_args = (
        p1_ode,
        [0, 10],
        [20, 40, 0, 0, val],
    )
    p1_sols.append(solve_ivp(*ode_args, **ode_kwargs))

for i in range(0, 7):
    plt.plot(p1_sols[i].t, p1_sols[i].y[2]/10, label=rf"$T_0=${T_range[i]} K")

plt.xlabel("Reactor volume (L)")
plt.ylabel("C Concentration (mol/L)")
plt.title(r"C Production for $T_0$ in Range 300K to 450K")

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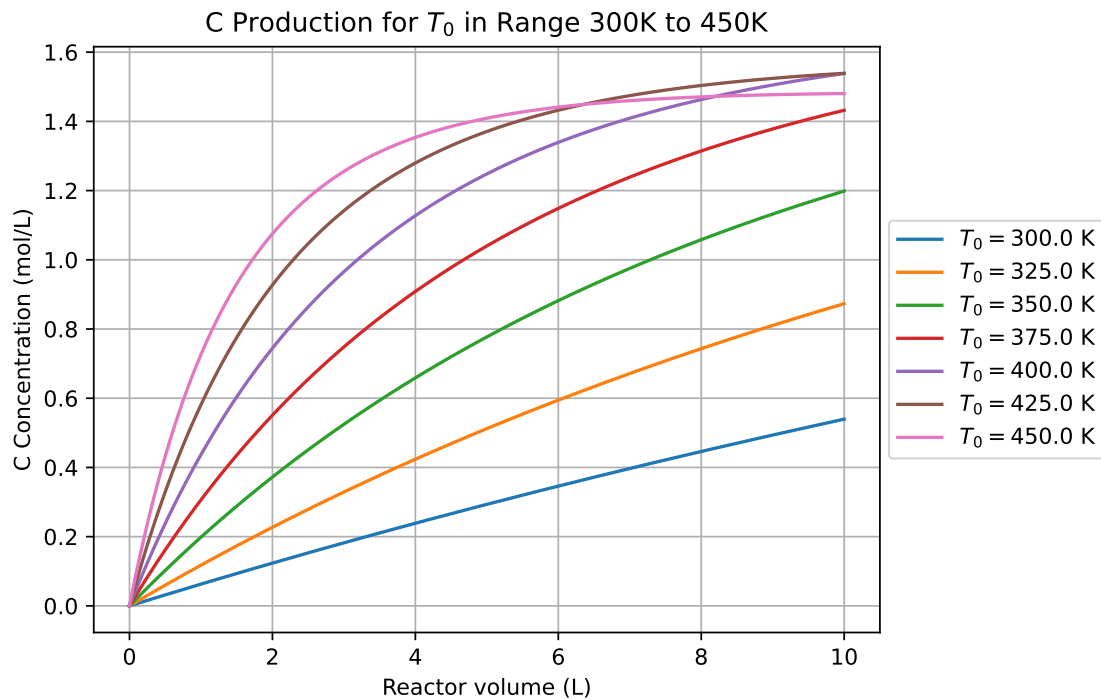
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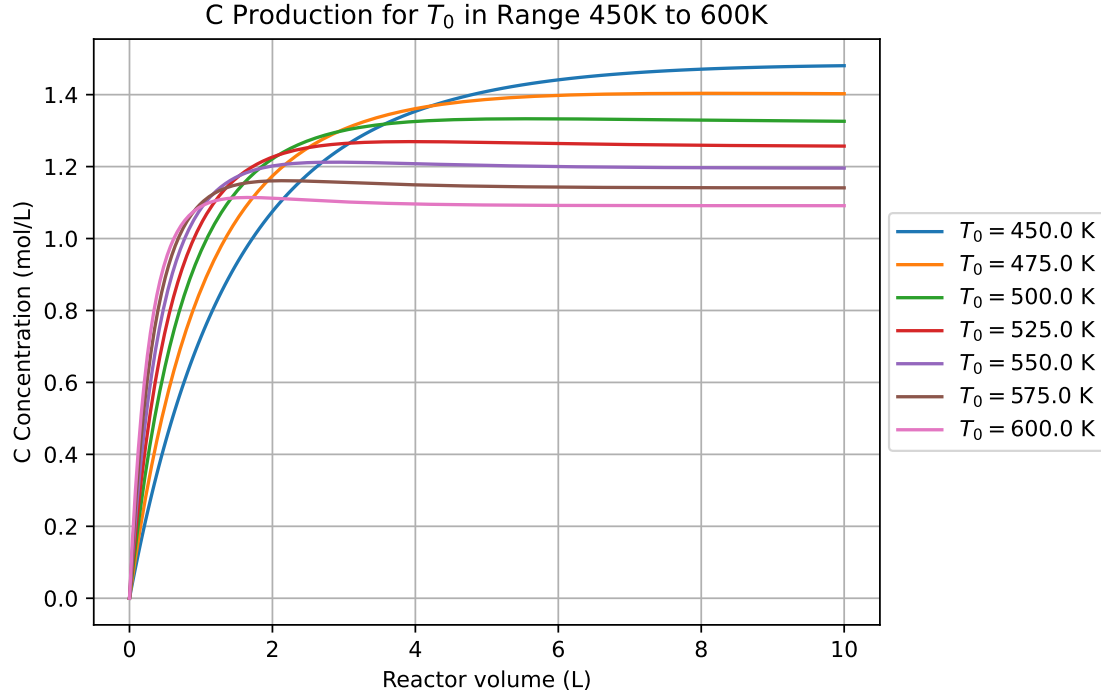
plt.legend(loc="right", bbox_to_anchor=(1.3, 0.5))
plt.grid(which='both', axis='both')
plt.show()

for i in range(6, len(p1_sols)):
    plt.plot(p1_sols[i].t, p1_sols[i].y[2]/10, label=rf"$T_0=${T_range[i]} K")

plt.xlabel("Reactor volume (L)")
plt.ylabel("C Concentration (mol/L)")
plt.title(r"C Production for $T_0$ in Range 450K to 600K")
plt.legend(loc="right", bbox_to_anchor=(1.3, 0.5))
plt.grid(which='both', axis='both')
plt.show()

```





An optimal inlet temperature is one between 400K and 425K. Both 400K and 425K inlet temperatures have about the same rate of production of C.

## 2 Problem 2

$$\epsilon = \frac{1}{2} - 1 = -\frac{1}{2}$$

Concentrations in terms of X

$$C_A = C_{A0} \left( \frac{1-X}{1-\frac{X}{2}} \right) p^{\frac{T_0}{T}}$$

$$C_C = C_{A0} \left( \frac{X}{1-\frac{X}{2}} \right) p^{\frac{T_0}{T}}$$

Rate law

$$r_A = - \left( k(T) C_A^2 - \frac{k(T)}{K(T)} C_C \right)$$

$$k(T) = 0.1 \exp \left[ \frac{8000}{8.314} \left( \frac{1}{450} - \frac{1}{T} \right) \right]$$

$$K(T) = 10000 \exp \left[ \frac{-20000}{8.314} \left( \frac{1}{450} - \frac{1}{T} \right) \right]$$

Design equation:

$$\frac{dX}{dW} = \frac{r_A}{F_{A0}}$$

Pressure drop

$$\frac{dp}{dW} = - \frac{\alpha(1-\frac{X}{2})}{2p} \frac{T}{T_0}$$

Energy balance

$$\frac{dT}{dW} = \frac{r_A \Delta H_{R_x} - Ua(T - T_a)}{F_{A0}(\sum \Theta_j C_{p,j} + X \Delta C_P)}$$

$$\sum \Theta_j C_{p,j} = C_{P,A} = 40$$

$$\Delta C_P = \frac{1}{2} \cdot 20 - 40 = -30$$

$$\frac{dT}{dW} = \frac{r_A \Delta H_{R_x} - Ua(T - T_a)}{F_{A0}(40 - 30X)}$$

Adiabatic case

$$\frac{dT}{dW} = \frac{r_A \Delta H_{R_x}}{F_{A0}(40 - 30X)}$$

Changing  $T_a$  case

Parallel flow:

$$\frac{dT_a}{dW} = \frac{Ua(T - T_a)}{\dot{m}_c C_{P,c}}$$

Countercurrent flow:

$$\frac{dT_a}{dW} = -\frac{Ua(T - T_a)}{\dot{m}_c C_{P,c}}$$

```
[ ]: C_A = lambda X, p, T: 1.9 * (1 - X) / (1 - X / 2) * p * 450 / T
C_C = lambda X, p, T: 1.9 * (X) / (1 - X / 2) * p * 450 / T
dpdW = lambda X, p, T: -0.005 * (1 - X / 2) / 2 / p * T / 450
k = lambda T: 0.1 * np.exp(8000 / 8.314 * (1 / 450 - 1 / T))
K = lambda T: 10000 * np.exp(-20000 / 8.314 * (1 / 450 - 1 / T))
r_A = lambda C_A, C_C, T: -k(T) * C_A**2 + k(T) / K(T) * C_C
F_A = lambda X: 5 * (1 - X)
F_C = lambda X: 5 * (X / 2)
```

```
[ ]: def p_2_adiabatic_ode(t, y):
    f = y*0

    X = y[0]
    p = y[1]
    T = y[2]

    f[0] = -r_A(C_A(X, p, T), C_C(X, p, T), T) / 5
    f[1] = dpdW(X, p, T)
    # f[2] = -20000 * r_A(C_A(X, p, T), C_C(X, p, T), T) / (F_A(X) * 40 +
    ↪ F_C(X) * 20)
    f[2] = (-20000 * r_A(C_A(X, p, T), C_C(X, p, T), T)) / (40 - 30 * X) / 5

    return f

def p_2_const_T_a_ode(t, y):
    f = y*0

    X = y[0]
```

```

p = y[1]
T = y[2]

r = r_A(C_A(X, p, T), C_C(X, p, T), T)

T_a = 500

UA = 0.001 * 3600

f[0] = -r / 5
f[1] = dpdW(X, p, T)
f[2] = (-20000 * r - UA * (T - T_a)) / (40 - 30 * X) / 5

return f

def p_2_parallel_ode(t, y):
    f = y*0

    X = y[0]
    p = y[1]
    T = y[2]
    T_a = y[3]

    r = r_A(C_A(X, p, T), C_C(X, p, T), T)

    UA = 0.001 * 3600

    f[0] = -r / 5
    f[1] = dpdW(X, p, T)
    f[2] = (-20000 * r - UA * (T - T_a)) / (40 - 30 * X) / 5
    f[3] = UA * (T - T_a) / 0.05 / 4200

    return f

def p_2_counter_ode(t, y):
    f = y*0

    X = y[0]
    p = y[1]
    T = y[2]
    T_a = y[3]

    r = r_A(C_A(X, p, T), C_C(X, p, T), T)

    UA = 0.001 * 3600

    f[0] = -r / 5

```

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f[1] = dpdW(X, p, T)
f[2] = (-20000 * r - UA * (T - T_a)) / (40 - 30 * X) / 5
f[3] = -UA * (T - T_a) / 0.05 / 4200

return f

```

```

[ ]: p2_options = {
    'Adiabatic': {
        'args': (p_2_adiabatic_ode, [0, 90], [0, 1, 450]),
        'kwargs': {
            'method': 'Radau',
            'atol': 1e-8,
            'rtol': 1e-8,
        }
    },
    'Constant $T_a$: {
        'args': (p_2_const_T_a_ode, [0, 90], [0, 1, 450]),
        'kwargs': {
            'method': 'Radau',
            'atol': 1e-8,
            'rtol': 1e-8,
        }
    },
    'Parallel Flow': {
        'args': (p_2_parallel_ode, [0, 90], [0, 1, 450, 500]),
        'kwargs': {
            'method': 'Radau',
            'atol': 1e-8,
            'rtol': 1e-8,
        }
    },
    'Countercurrent Flow': {
        'args': (p_2_counter_ode, [0, 90], [0, 1, 450, 717.0987]),
        'kwargs': {
            'method': 'Radau',
            'atol': 1e-8,
            'rtol': 1e-8,
        }
    },
}

```

```

[ ]: for key in p2_options:
    p2_sol = solve_ivp(*p2_options[key]['args'], **p2_options[key]['kwargs'])
    fig, ax = plt.subplots(3, figsize=[6.40, 4.80*1.8])
    ax[0].plot(p2_sol.t, p2_sol.y[0], 'tab:blue', label="Conversion")
    ax[0].set_ylabel("Conversion")
    ax[1].plot(p2_sol.t, p2_sol.y[1], 'tab:orange', label="Pressure drop ratio")

```

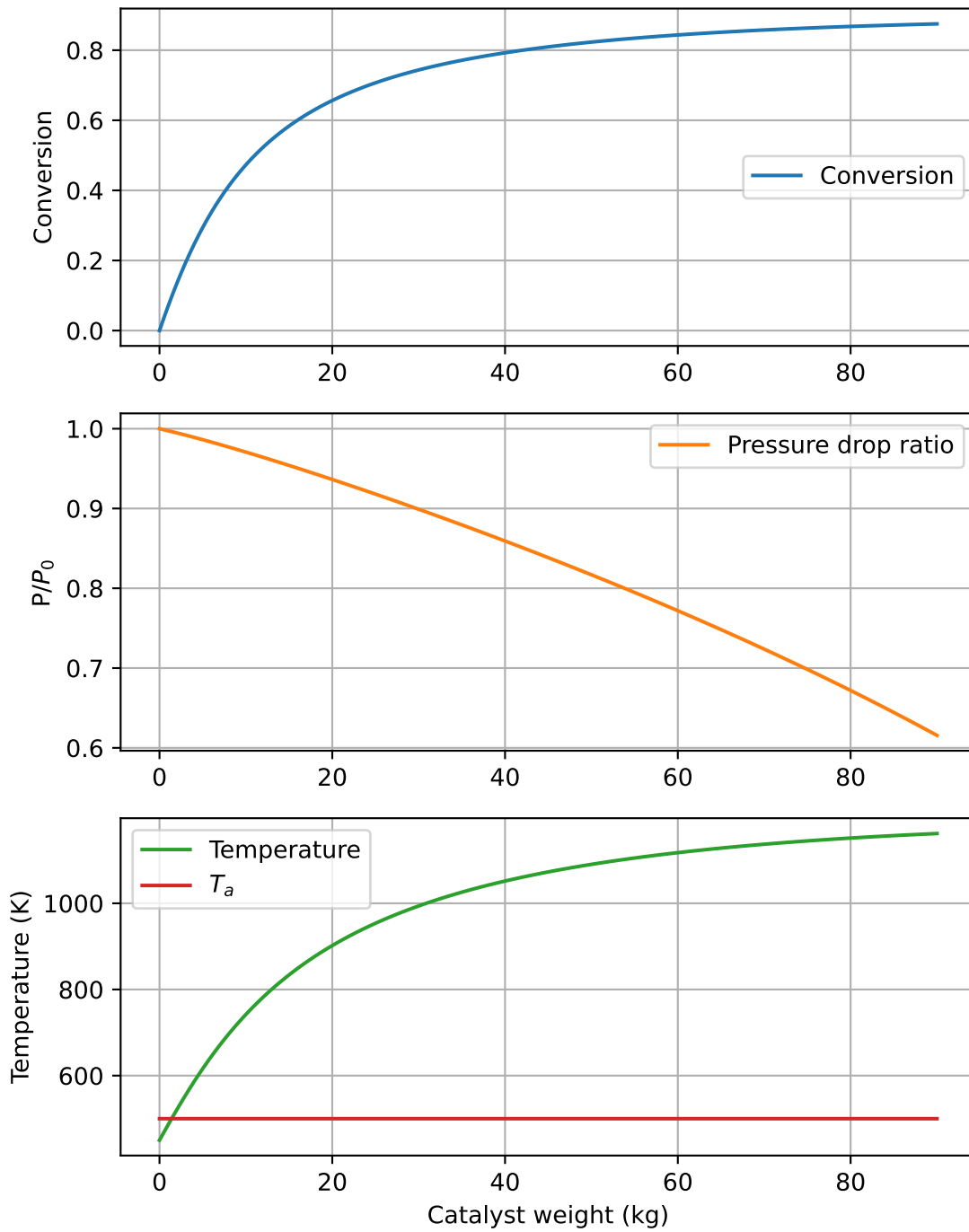
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ax[1].set_ylabel(r"P/$P_0$")
ax[2].plot(p2_sol.t, p2_sol.y[2], 'tab:green', label="Temperature")
ax[2].set_ylabel("Temperature (K)")
try:
    ax[2].plot(p2_sol.t, p2_sol.y[3], 'tab:red', label=r"$T_a$")
except:
    ax[2].plot(p2_sol.t, np.ones(p2_sol.t.shape[0])*500, 'tab:red',
    ↪label=r"$T_a$")
for a in ax:
    a.legend()
    a.grid(which='both', axis='both')
ax[0].legend(loc='right')
ax[0].set_title(rf"Problem 2 {key} Reactor")
plt.xlabel("Catalyst weight (kg)")
plt.show()

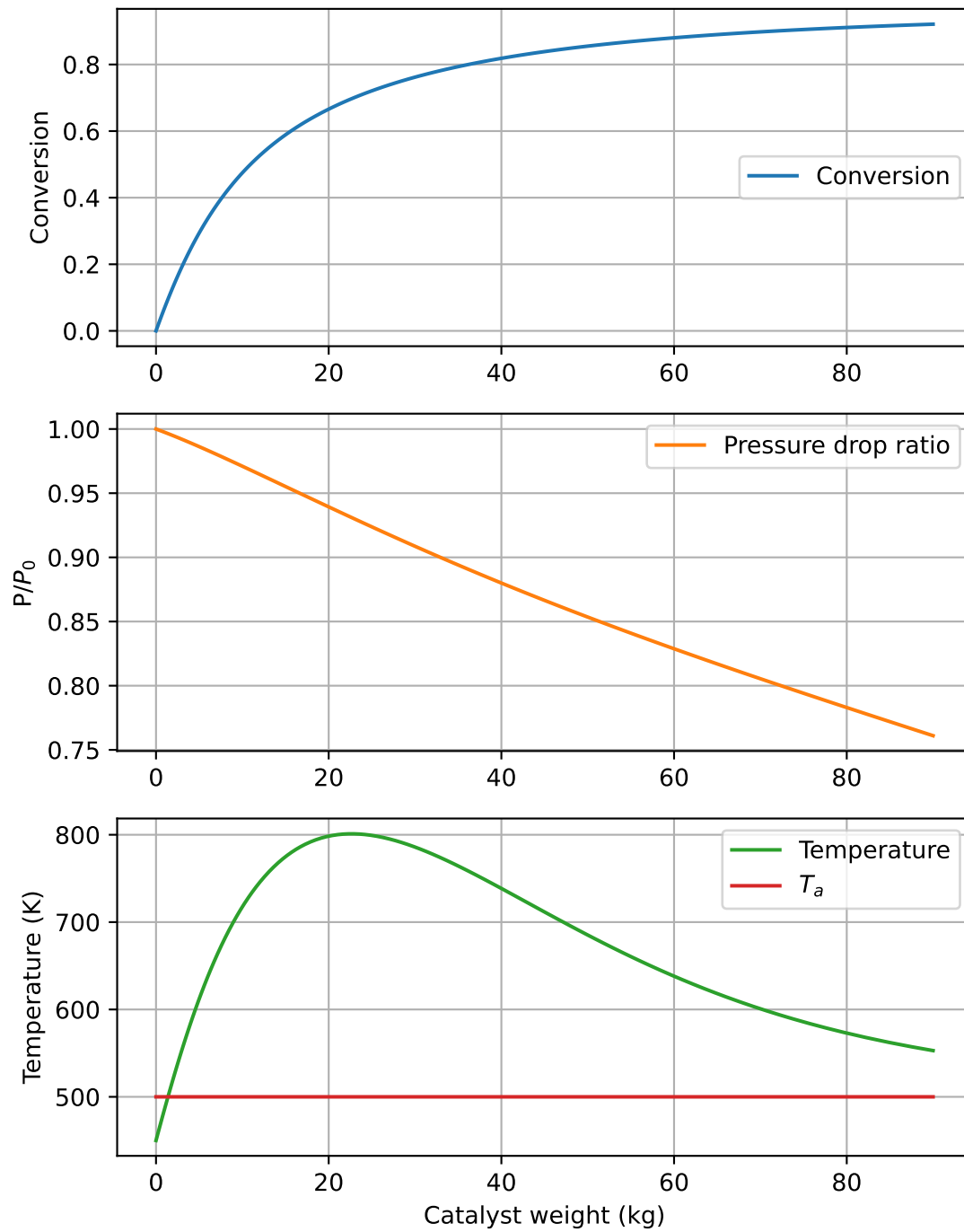
```



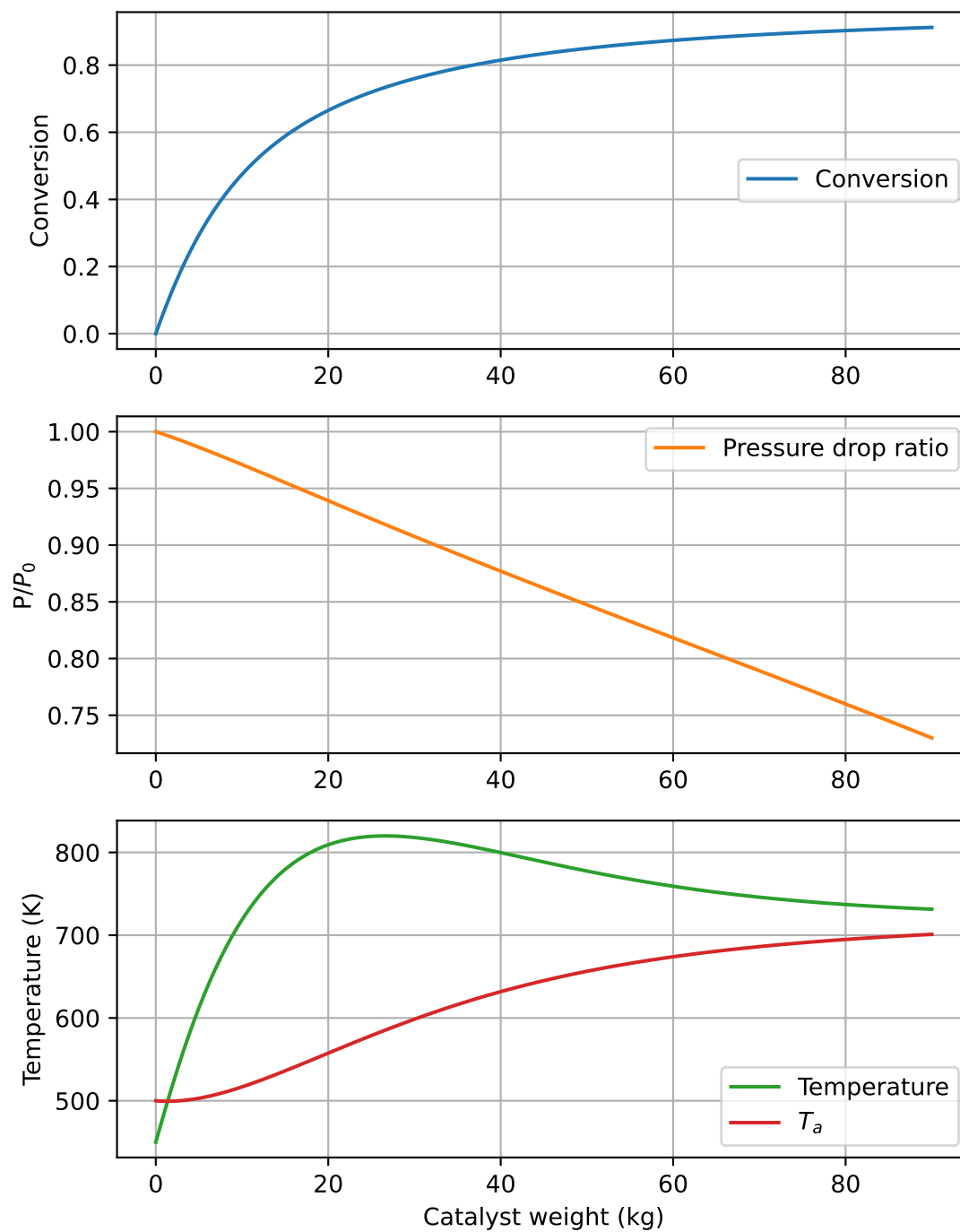
### Problem 2 Adiabatic Reactor

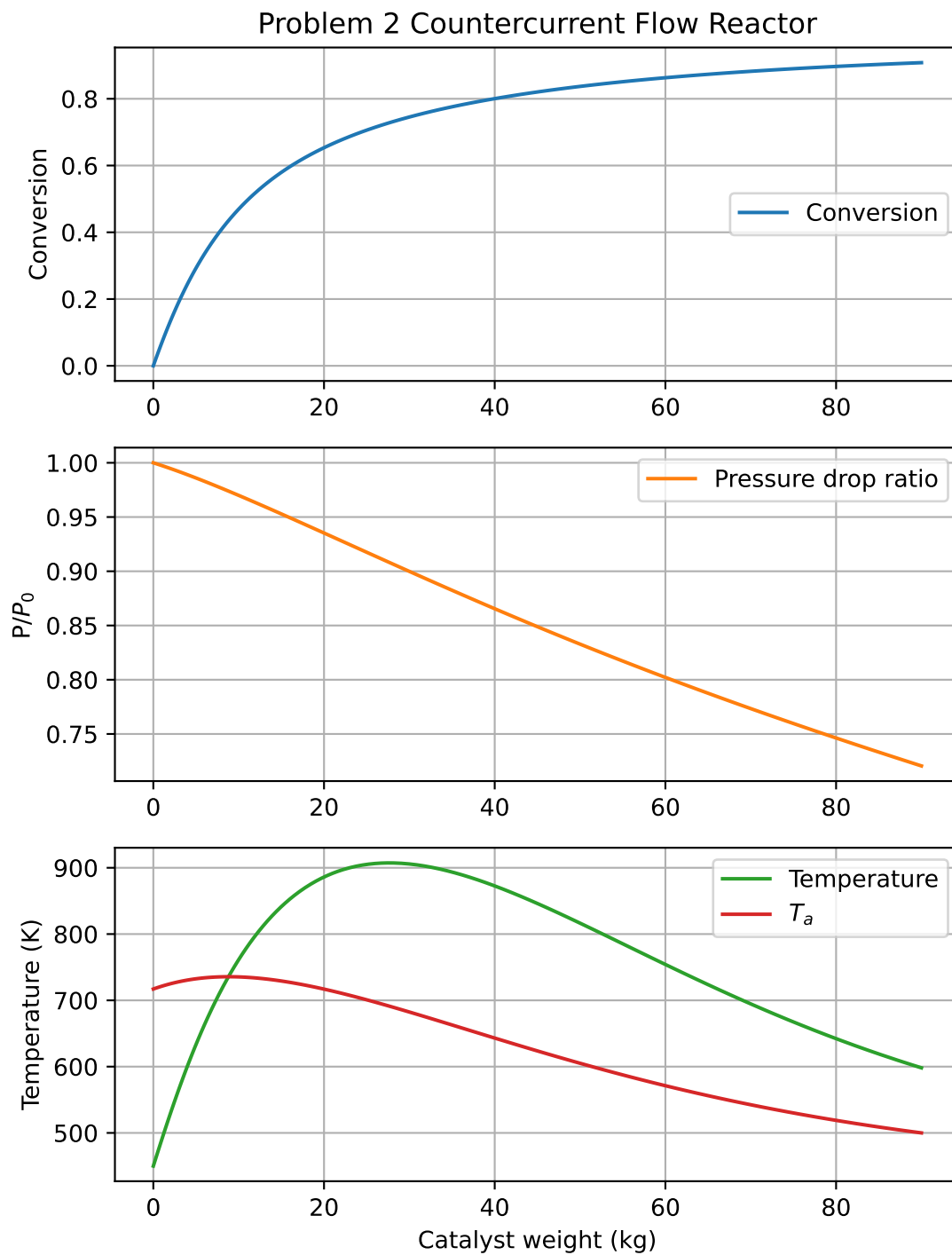


Problem 2 Constant  $T_a$  Reactor



### Problem 2 Parallel Flow Reactor





### 3 Problem 3

Rate law:

$$r_A = k \frac{N_A}{V}$$

Design equation:

$$\frac{dN_A}{dt} = -r_A V = -k N_A$$

Analytical solution:

$$\int \frac{dN_A}{N_A} = \int -k dt$$

$$\ln \frac{N_A}{N_{A0}} = -kt$$

$$N_A = N_{A0} e^{-kt}$$

For  $\frac{dT}{dt} = 0$ , energy balance must be  $Q_g = Q_r$

$$Q_g = r_A V \Delta H_{rx}^\circ = k N_{A0} e^{-kt} \Delta H_{rx}^\circ$$

$$Q_r = F_c C_{P,c} (T - T_0)$$

$$F_c = \frac{k N_{A0} e^{-kt} \Delta H_{rx}^\circ}{C_{P,c} (T - T_0)}$$

At 2 h

$$F_c = 3.16 \text{ lb/s}$$