

Carbohydrates and Fatty Acids

September 13, 2016



Primary Objectives

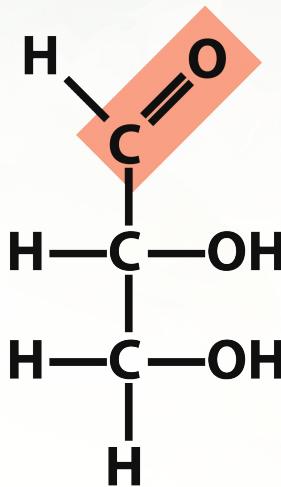
- Understand the relationship between different sugars.
- Interconvert between Fischer and Haworth structures.
- Understand the glycosidic bond.
- Comprehend the structural features of fatty acids and triacylglycerols.
- Understand the hierarchy of lipids.



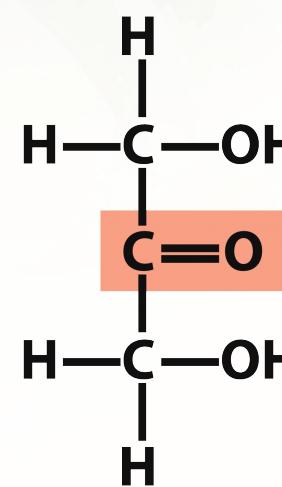
Section 1

Carbohydrates

Carbohydrates Are Polyhydroxy Biomolecules



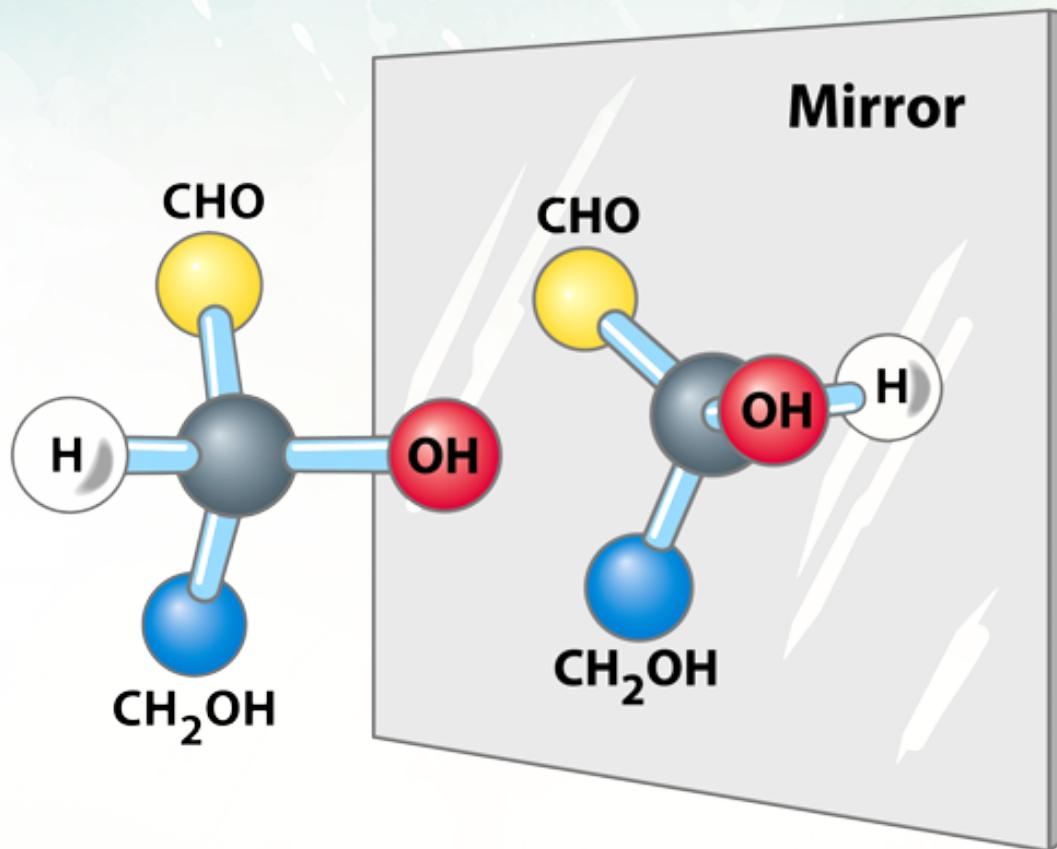
Glyceraldehyde,
an aldotriose



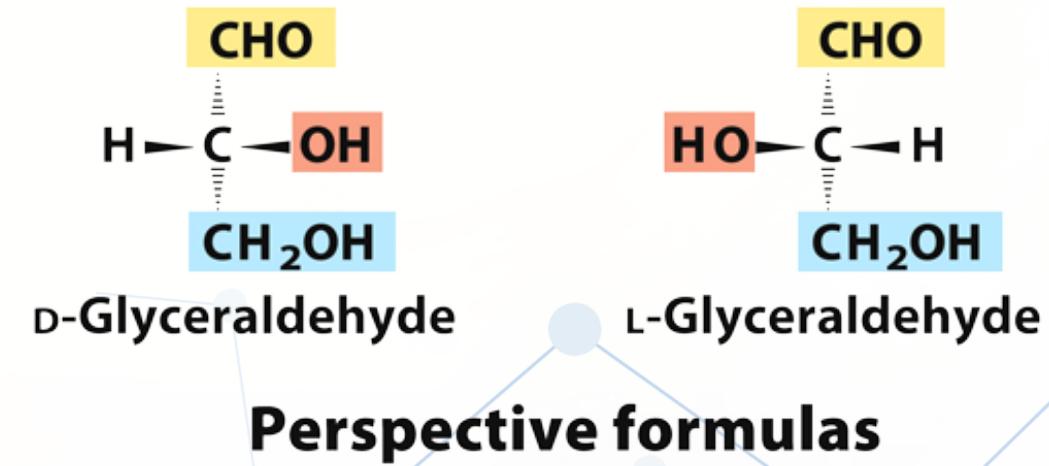
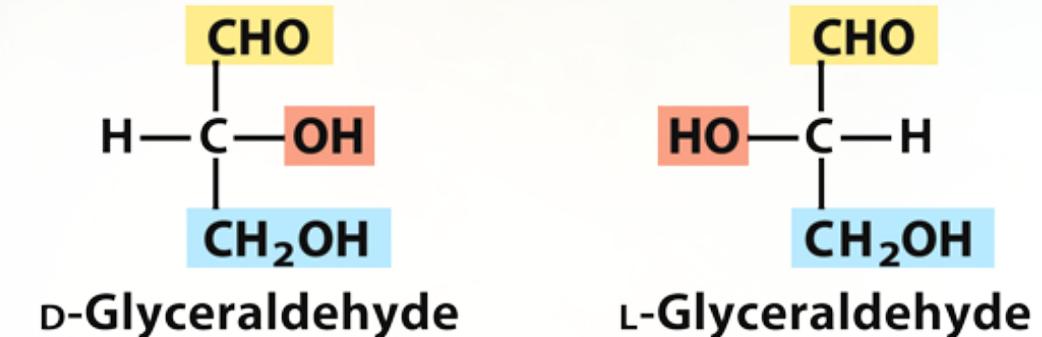
Dihydroxyacetone,
a ketotriose

- Empirical formula: $(\text{CH}_2\text{O})_n$
- Polyhydroxy aldehydes or ketones
 - Aldose → carbonyl group is in an aldehyde group
 - Ketose → carbonyl group is in a ketone group
- Naming conventions are important

Carbohydrates Exist As Enantiomers, Designated D- or L-



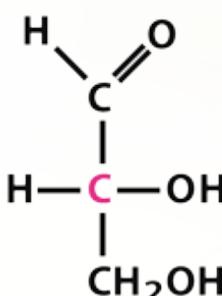
Ball-and-stick models



The Aldoses Are Carbohydrate Aldehydes

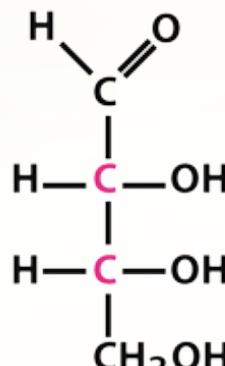
D-Aldoses

Three carbons

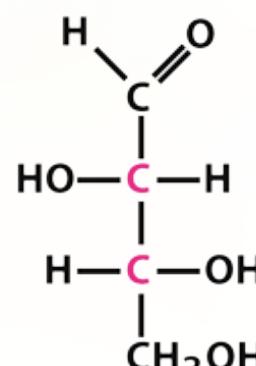


D-Glyceraldehyde

Four carbons



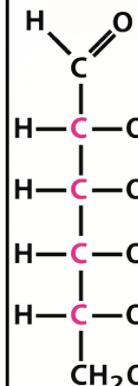
D-Erythrose



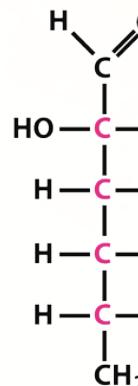
D-Threose

D-Aldoses

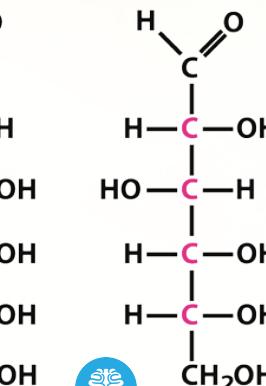
Six carbons



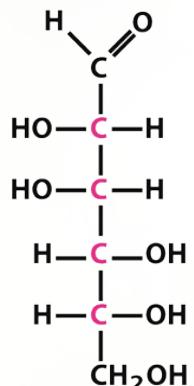
D-Allose



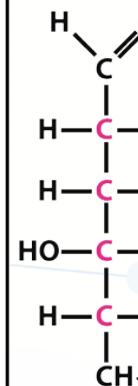
D-Altrose



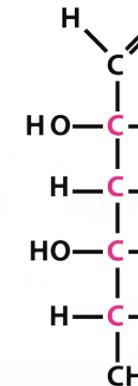
D-Glucose



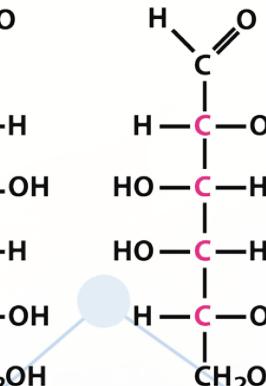
D-Mannose



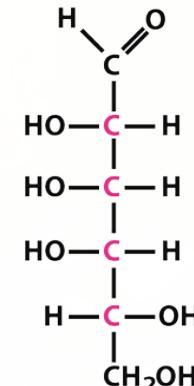
D-Gulose



D-Idose



D-Galactose

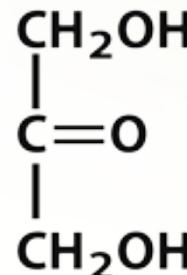


D-Talose

The Ketoses Are Carbohydrate Ketones

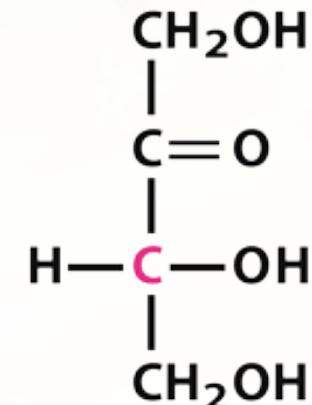
D-Ketoses

Three carbons



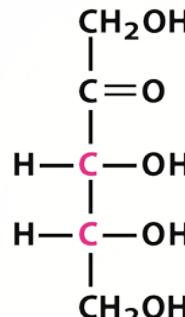
Dihydroxyacetone

Four carbons

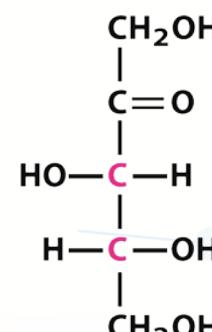


D-Erythrulose

Five carbons



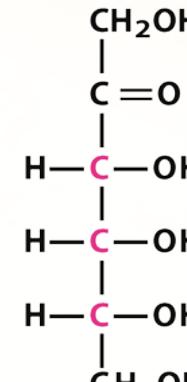
D-Ribulose



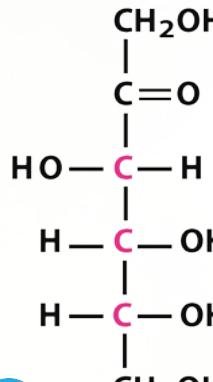
D-Xylulose

D-Ketoses

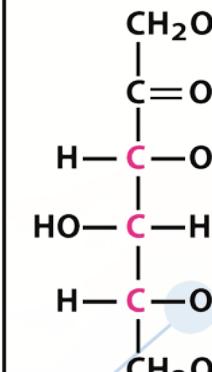
Six carbons



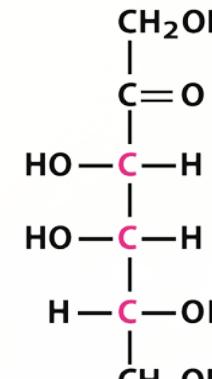
D-Psicose



D-Fructose

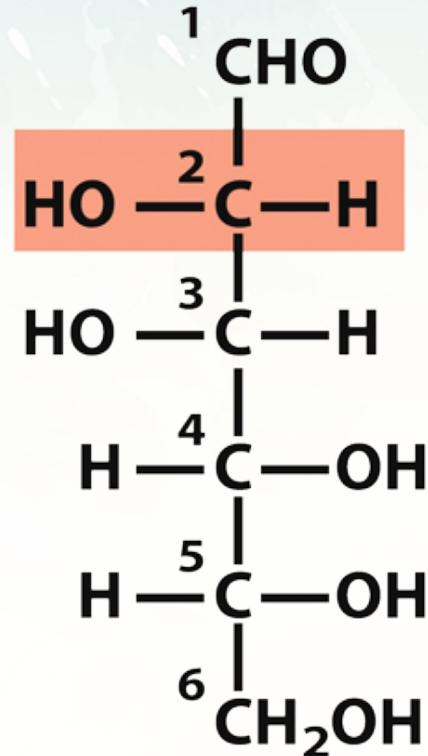


D-Sorbose

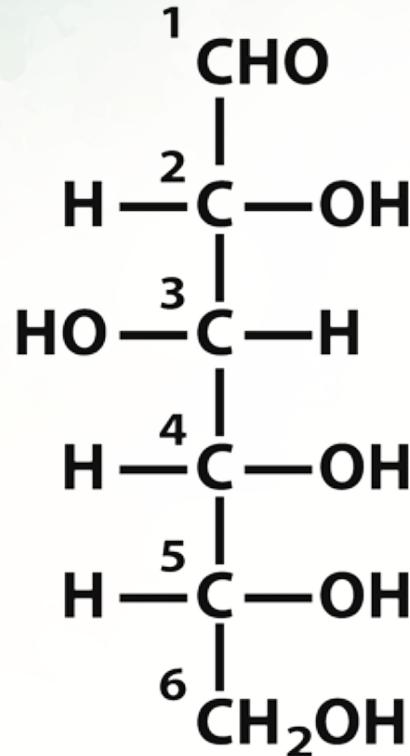


D-Tagatose

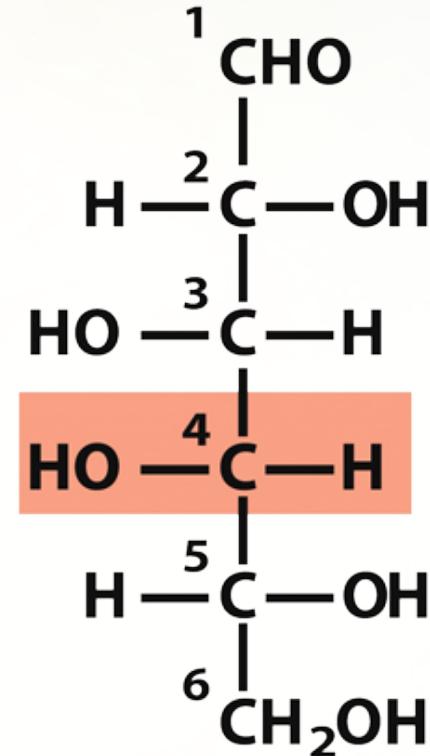
Epimeric Pairs Differ From One Another at a Single Chiral Carbon



D-Mannose
(epimer at C-2)

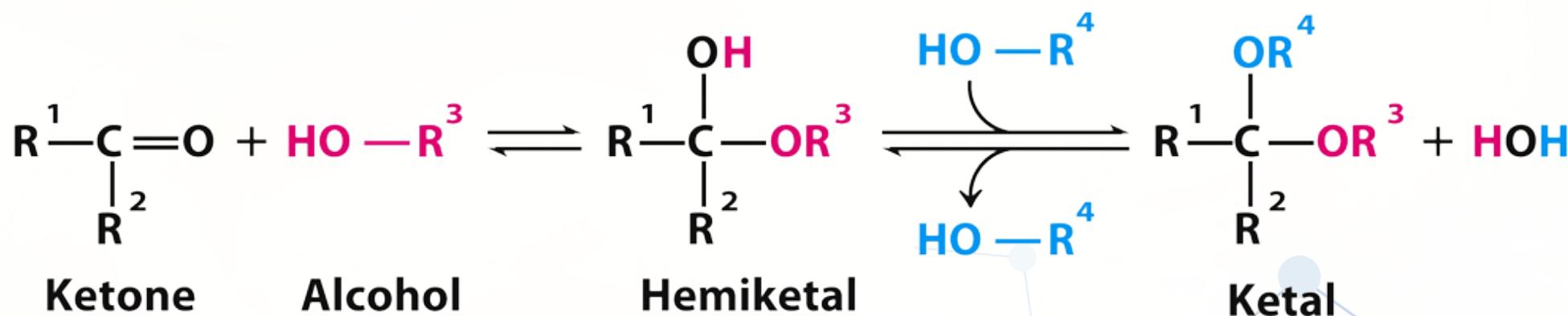
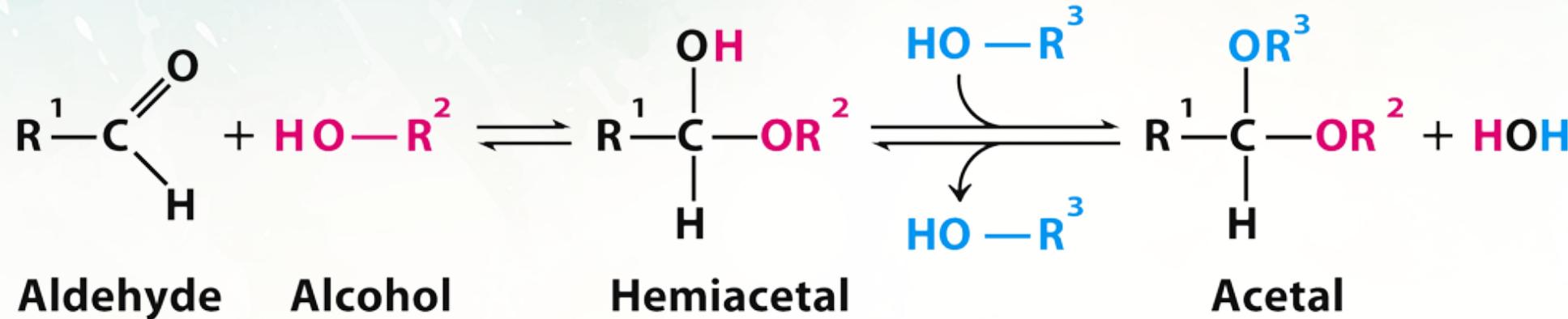


D-Glucose

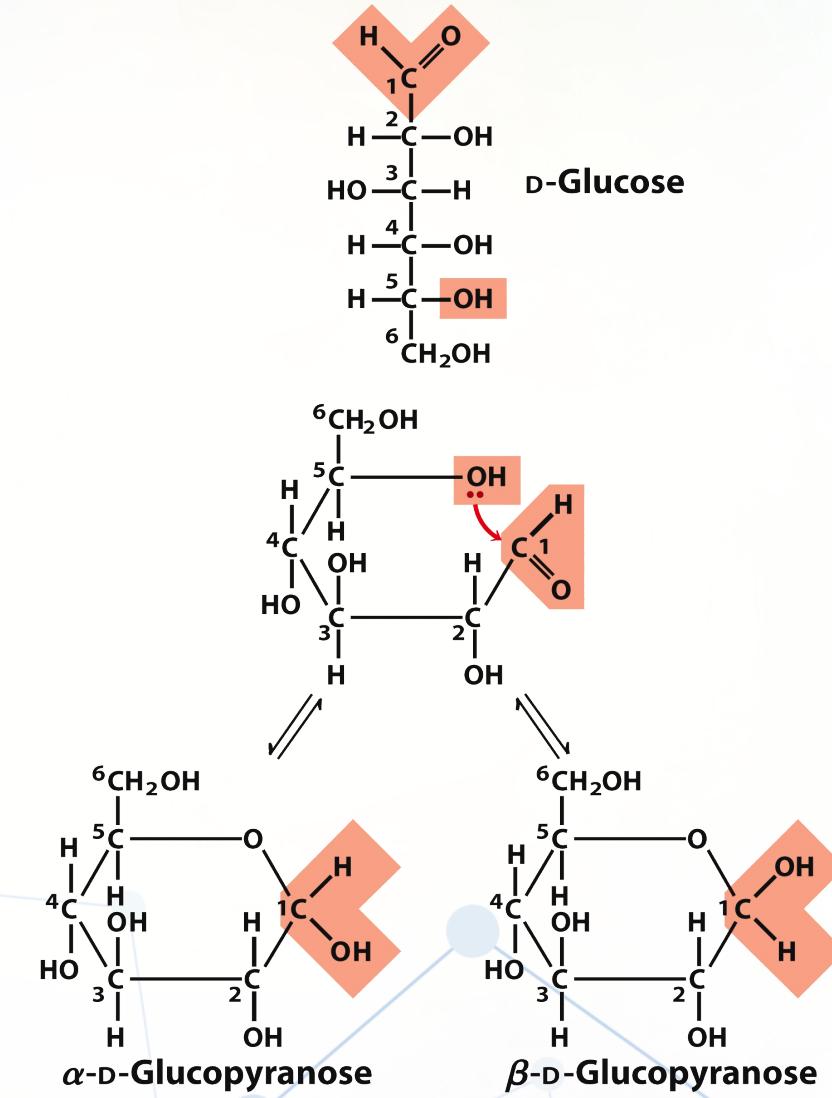
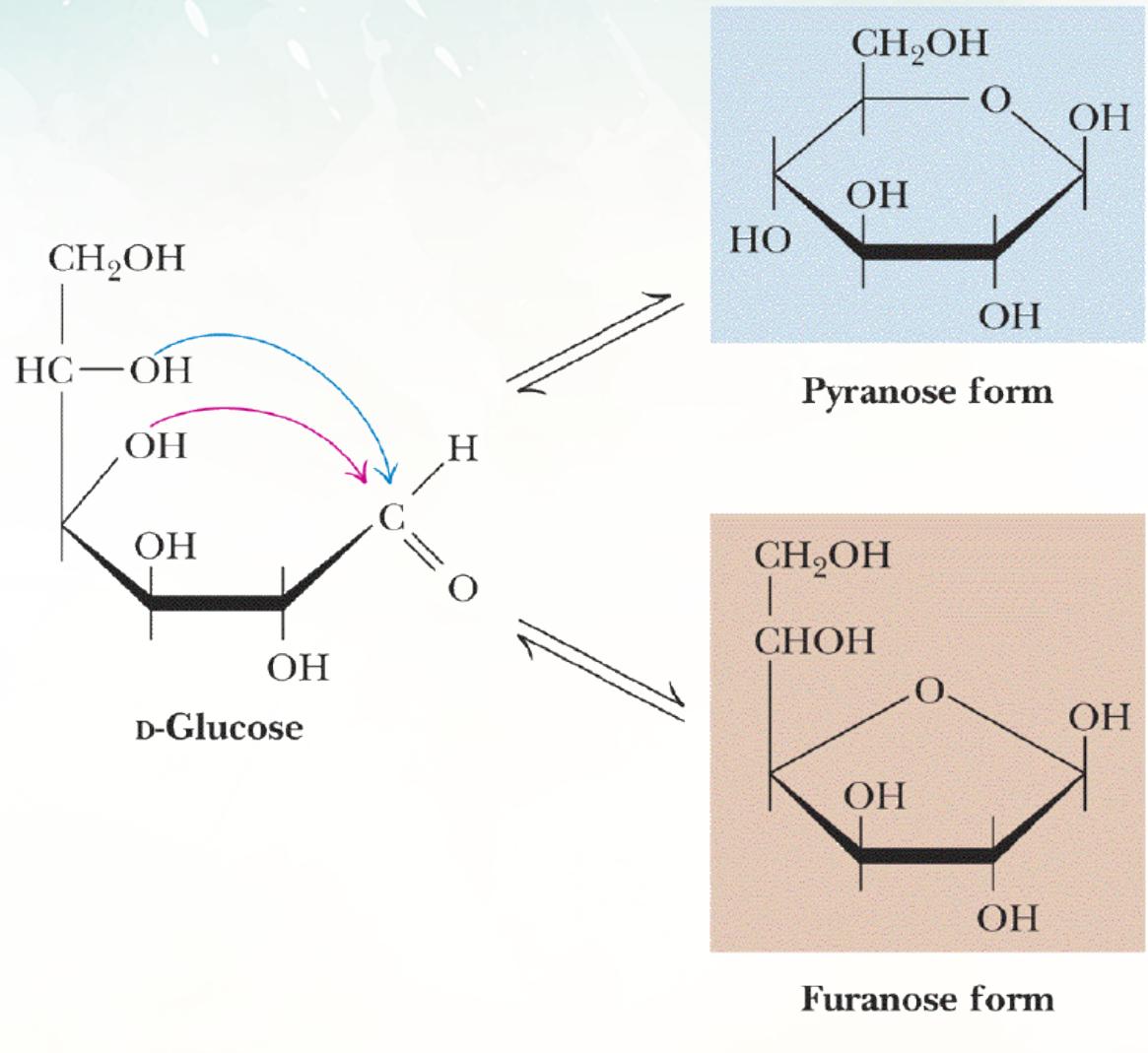


D-Galactose
(epimer at C-4)

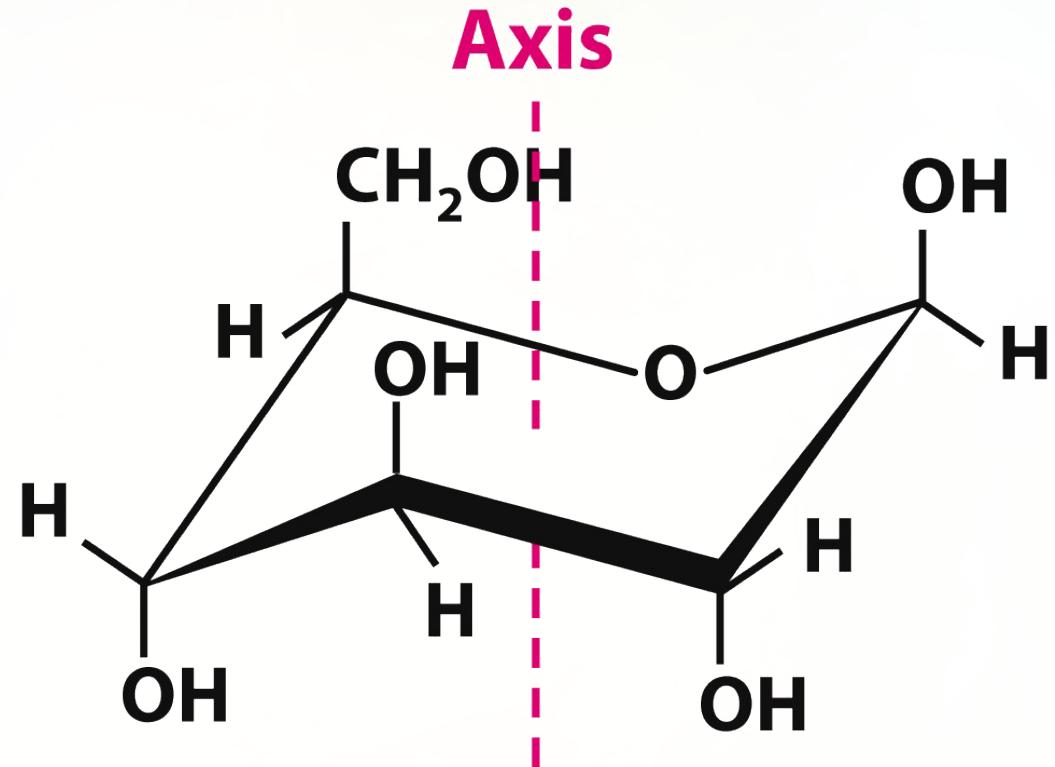
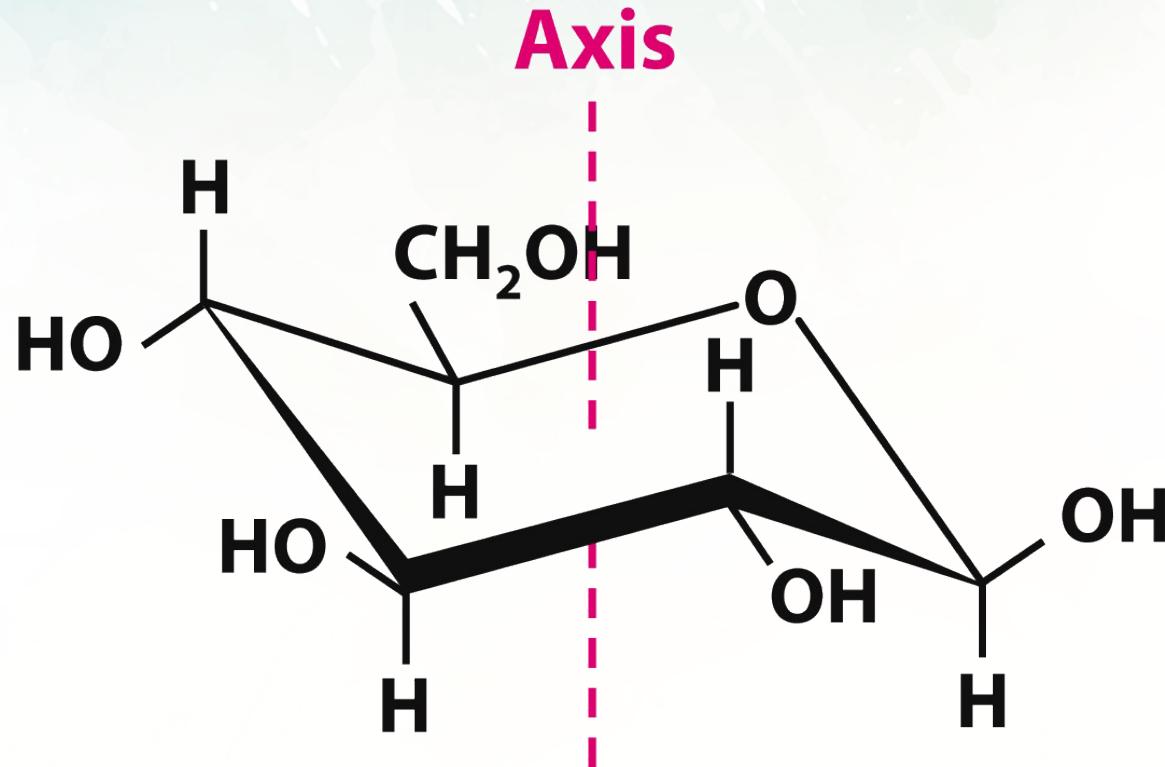
Carbohydrate Cyclization Follows the Hemiacetal and Hemiketal Mechanism



There Are Multiple Possible Cyclic Forms of a Single Carbohydrate

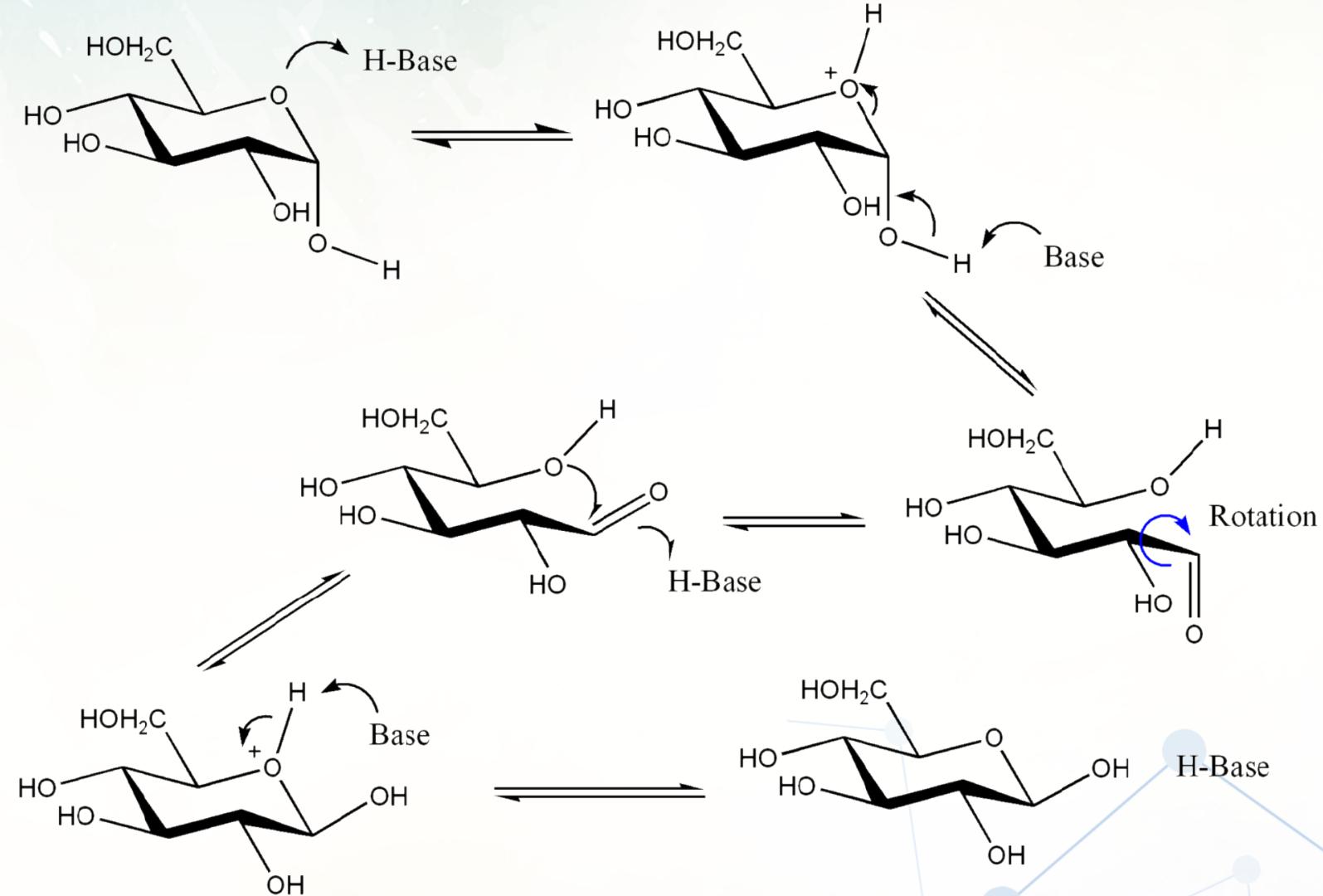


Positions of the Hydroxyl Groups Stay Constant Regardless of Chair Form

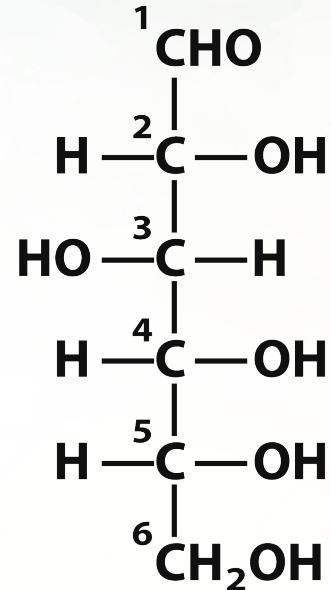


Two possible chair forms of β -D-glucopyranose

Mutarotation Switches the Configuration of the Anomeric Carbon's Hydroxyl Group

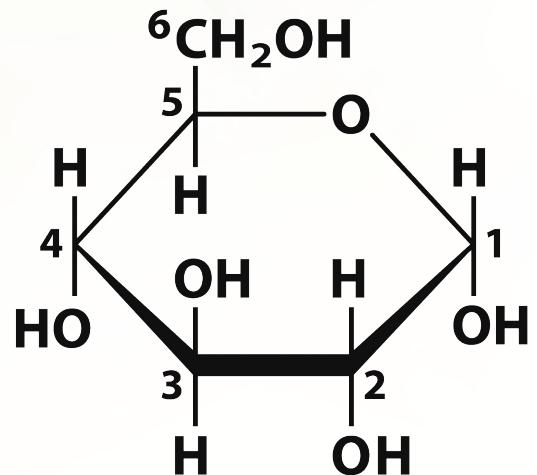


Fischer Projection and Haworth Perspective Are Two Ways To Draw Carbohydrate Structure



D-Glucose

Fischer projection

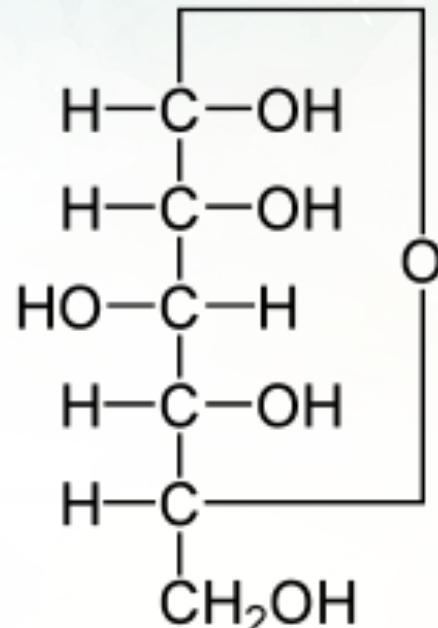


α -D-Glucopyranose Haworth perspective

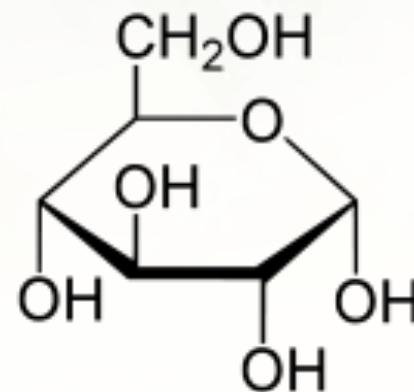
Converting to Haworth:

- Place the nucleophilic oxygen at the top (or top right) of the ring
 - Number carbons clockwise beginning with anomeric carbon
 - Place hydroxyl groups
 - Up → OH points left in Fischer projection
 - Down → OH points right in Fischer
 - Place terminal CH_2OH group for terminal group
 - Up → for D-enantiomer
 - Down → for L-enantiomer
 - Note that placement of the terminal CH_2OH group is very specific to this class, as we deal exclusively with D-sugars
 - Hydroxyl group on anomeric carbon can point up (β) or down (α)

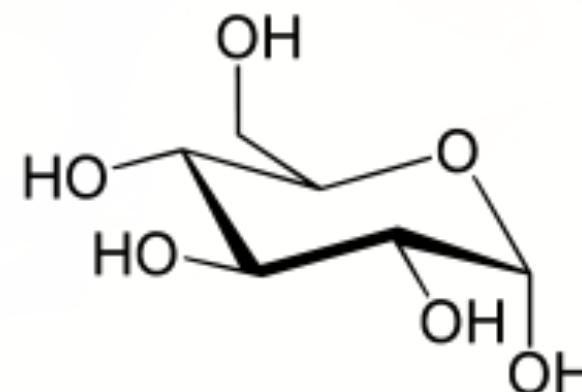
Not All Projections of α -Glucopyranose Are Easily Comprehensible



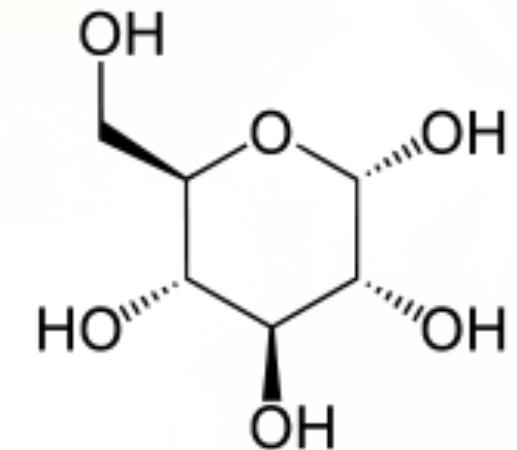
1



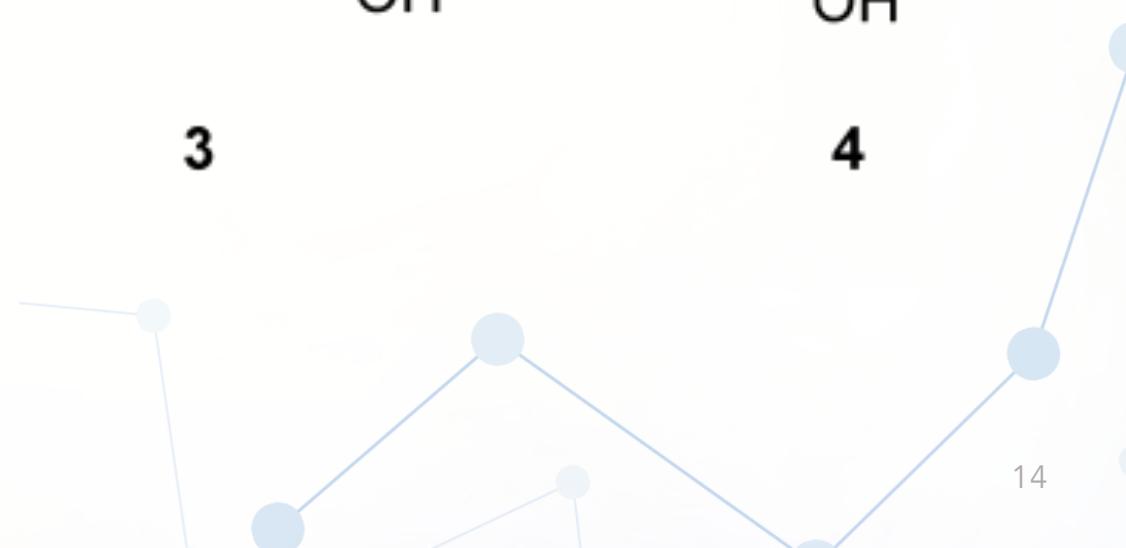
2



3



4



14

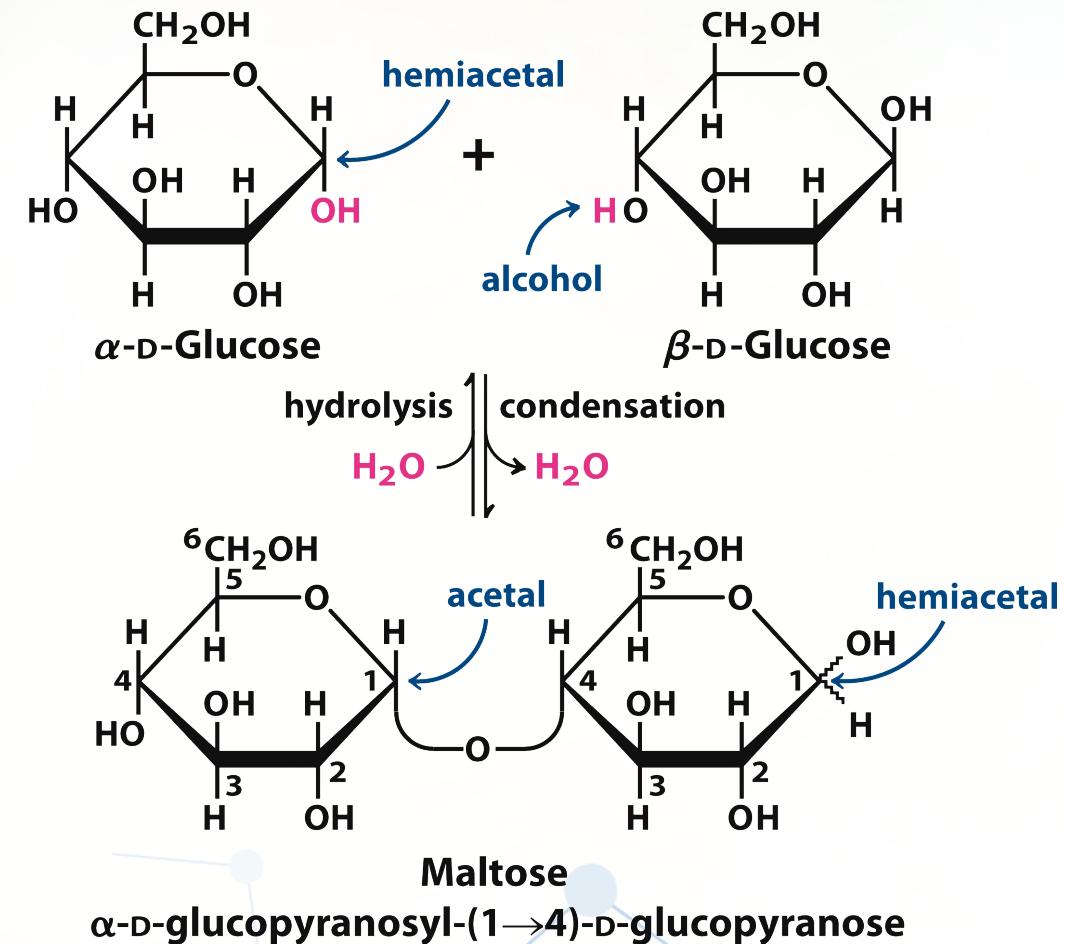
IN-CLASS EXERCISE:

- Draw D-glucose in its four cyclic forms as Haworth projects.



Disaccharide Formation is a Condensation Reaction

- Formation → condensation reaction
- Breakdown → hydrolysis reaction
- Once formed, interconversion between linear and ring form is prevented
- The end of a chain with a free anomeric carbon is called reducing end

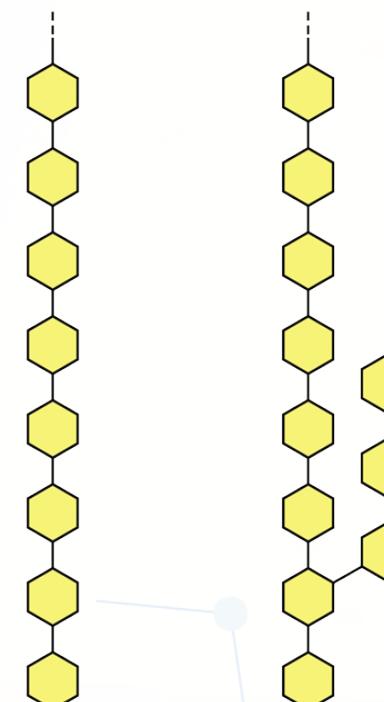


Polysaccharides Come in Two General Forms

- Repeated condensation reactions result in multiple monosaccharides linked together
- Two general kinds of polysaccharides:
 - Homopolysaccharides
 - Heteropolysaccharides

Homopolysaccharides

Unbranched Branched

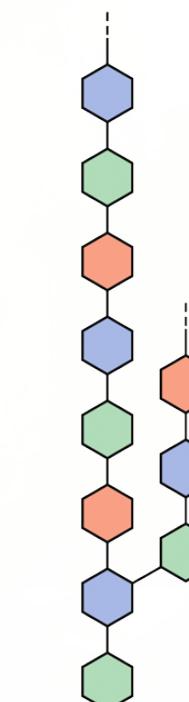


Heteropolysaccharides

Two monomer types, unbranched



Multiple monomer types, branched

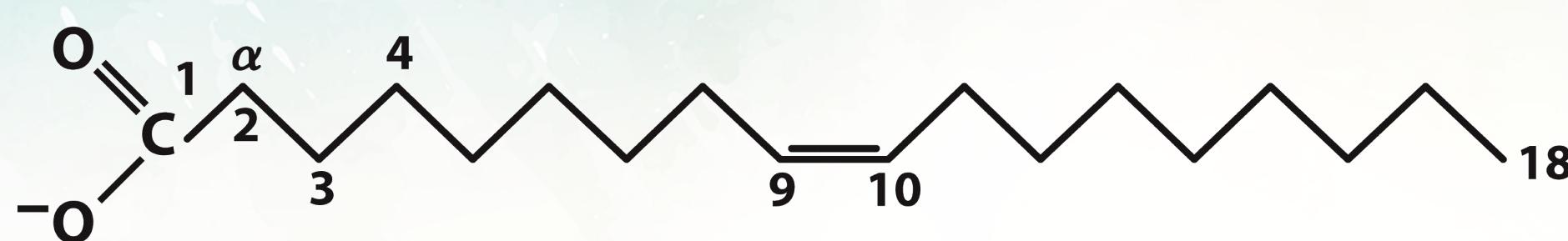


Section 2

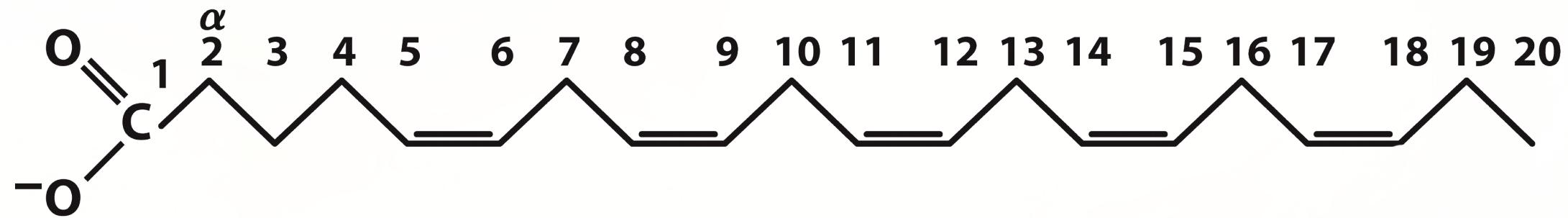
Fatty Acids



Fatty Acids Are Long Hydrocarbon Chains



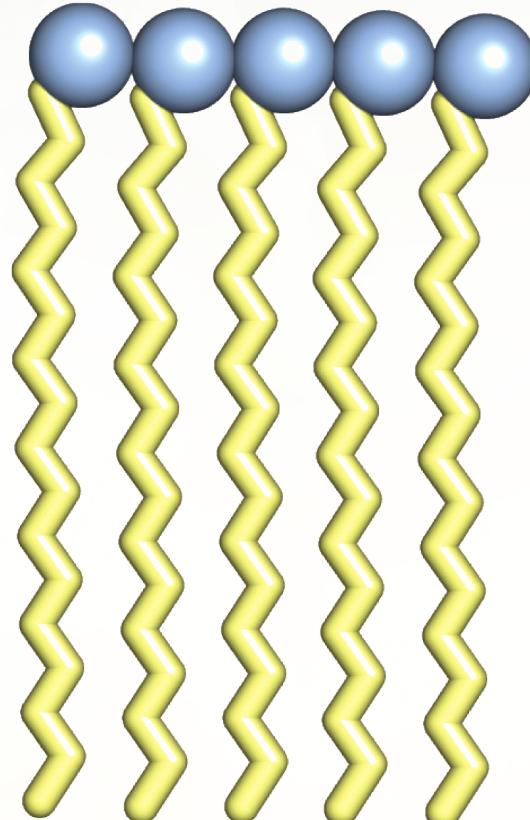
18:1(Δ^9) *cis*-9



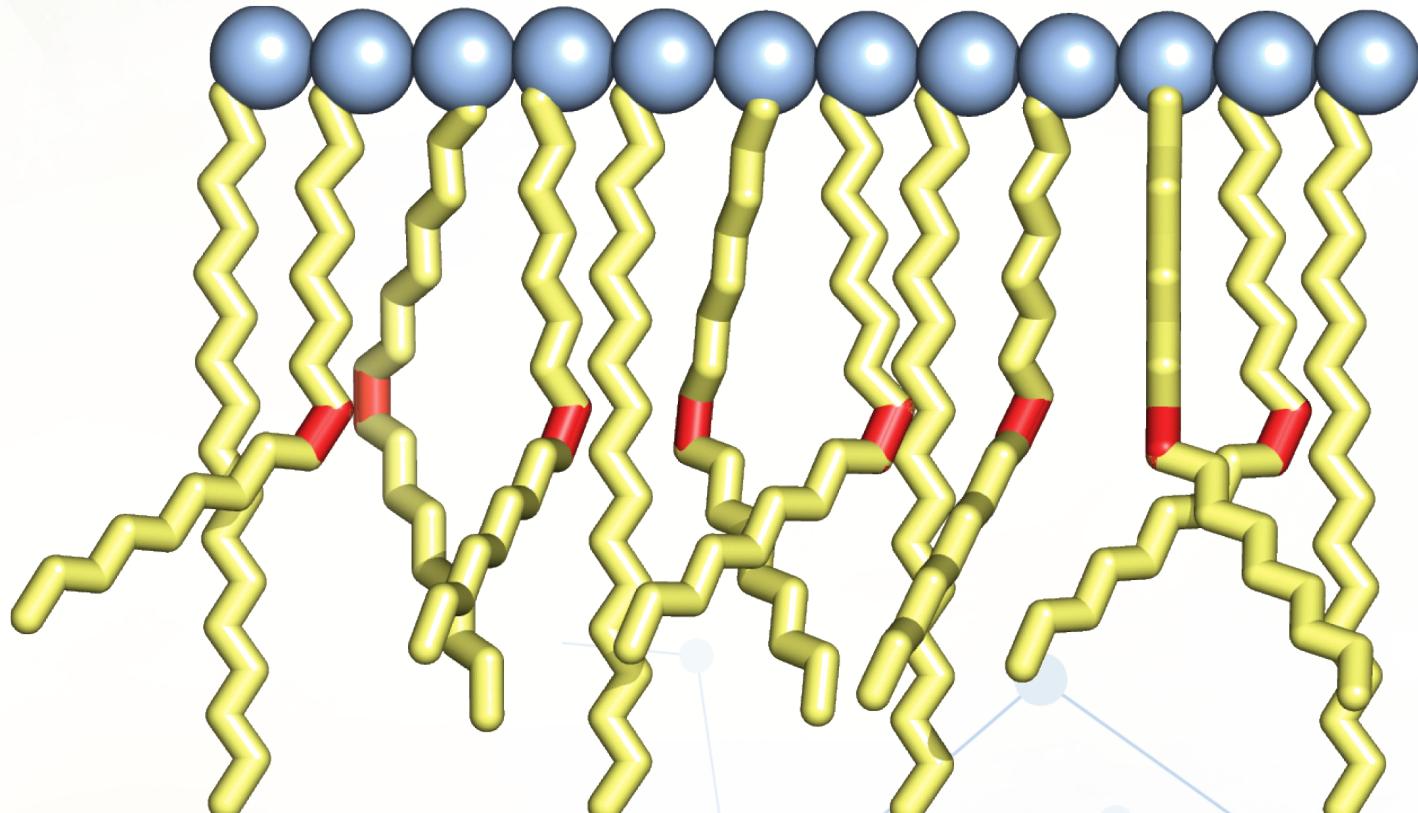
20:5($\Delta^5,8,11,14,17$) *cis*-,*cis*-,*cis*-,*cis*-5,8,11,14,15

Saturated and Unsaturated Fatty Acids Have Different Properties

Saturated fatty acids



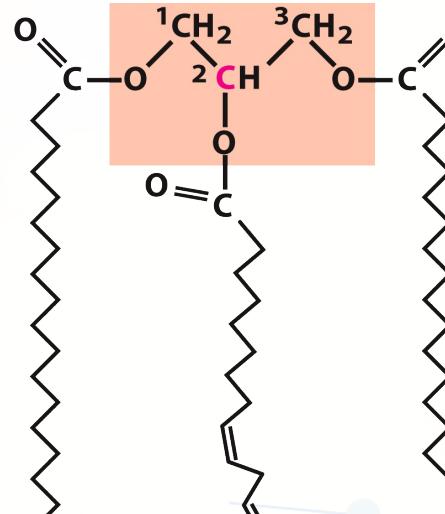
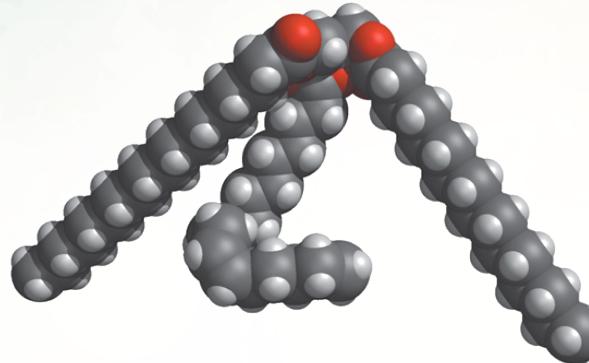
Mixture of saturated and unsaturated fatty acids



Triacylglycerols Consist of Three Fatty Acids Connected to a Glycerol

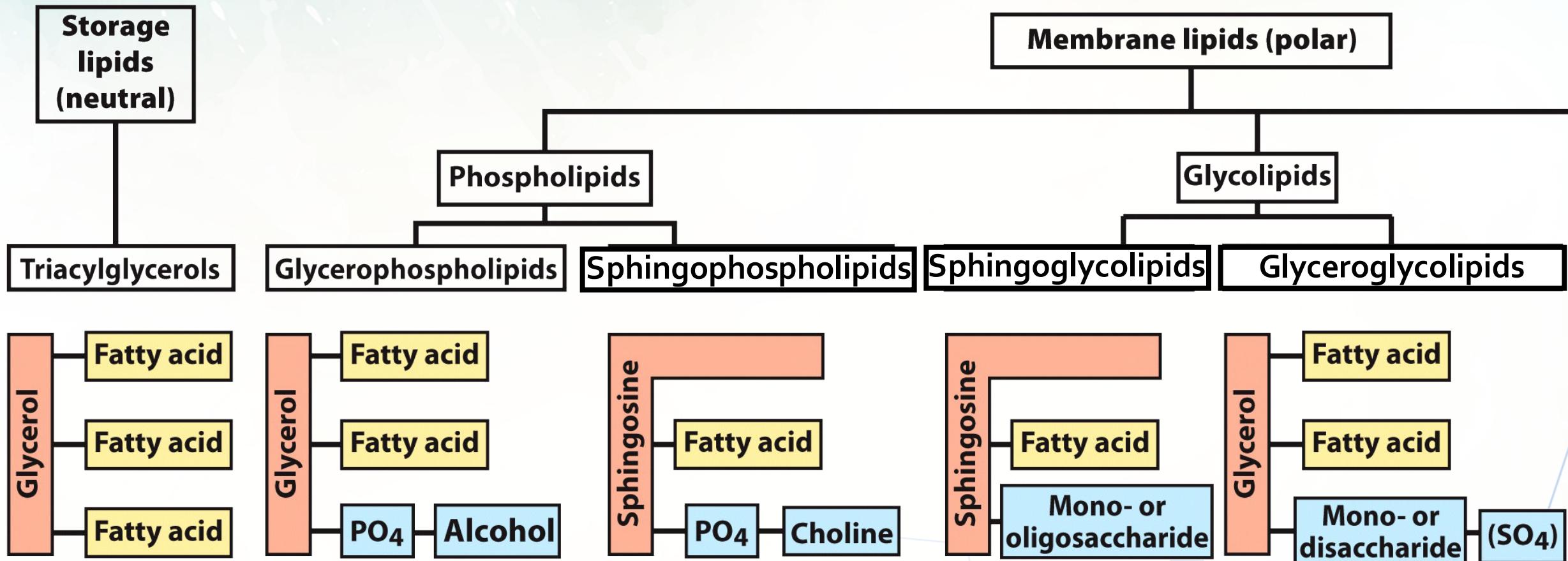


Glycerol

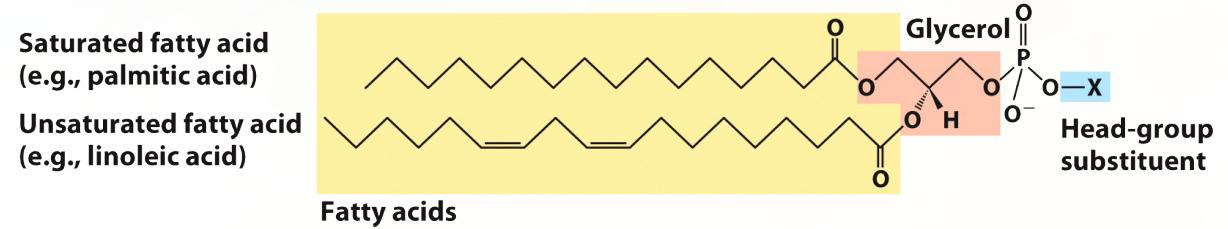


1-Stearoyl, 2-linoleoyl, 3-palmitoyl glycerol,
a mixed triacylglycerol

Classification of Lipids Is Hierarchical

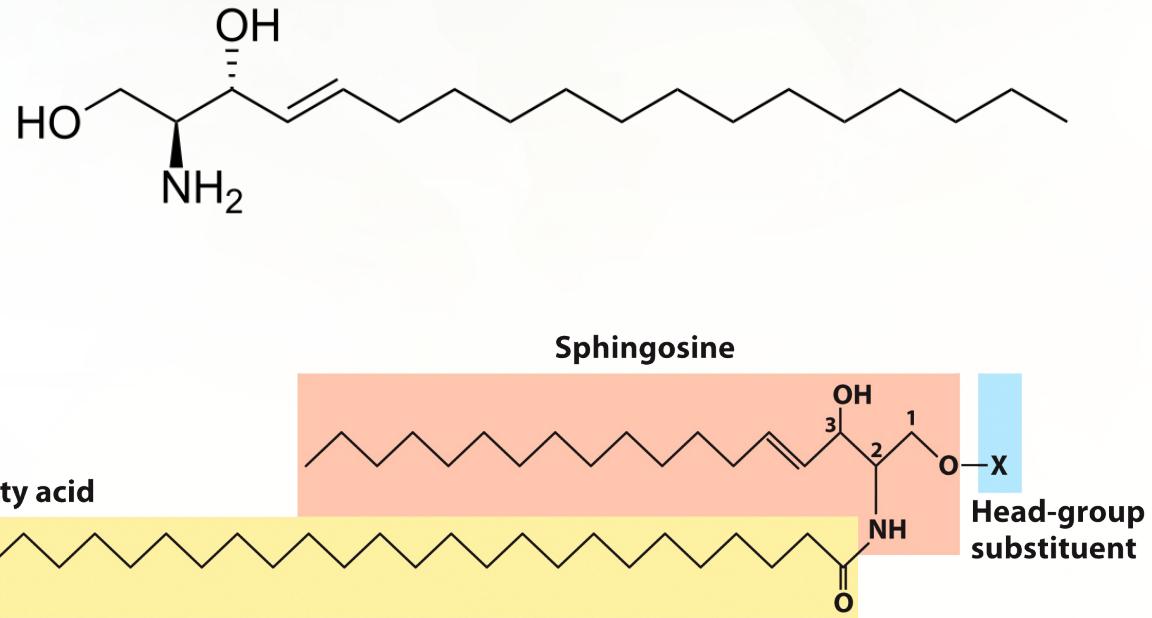


Glycerophospholipids Can Have Different Head-Group Substituents



Name of glycerophospholipid	Name of X — O	Formula of X	Net charge (at pH 7)
Phosphatidic acid	—	— H	-2
Phosphatidylethanolamine	Ethanolamine		0
Phosphatidylcholine	Choline		0
Phosphatidylserine	Serine		-1
Phosphatidylglycerol	Glycerol		-1
Phosphatidylinositol 4,5-bisphosphate	myo-Inositol 4,5-bisphosphate		-4*
Cardiolipin	Phosphatidyl-glycerol		-2

Sphingolipids Can Have Signaling Polysaccharides As Head-Group Substituents



Name of sphingolipid	Name of X—O	Formula of X
Ceramide	—	—H
Sphingomyelin	Phosphocholine	$\text{--P}(\text{O}^-)(\text{CH}_2\text{CH}_2\text{N}^+(\text{CH}_3)_3)$
Neutral glycolipids Glucosylcerbroside	Glucose	 A six-membered pyranose ring with carbons numbered 1 to 6. Hydroxyl groups are at C2, C3, and C6, and methyl groups are at C1 and C4.
Lactosylceramide (a globoside)	Di-, tri-, or tetrasaccharide	 A horizontal line with two circles, one blue and one yellow, representing a disaccharide linkage.
Ganglioside GM2	Complex oligosaccharide	 A horizontal line with three distinct shapes: a blue circle, a yellow circle, and a purple diamond, representing a more complex branched saccharide structure.