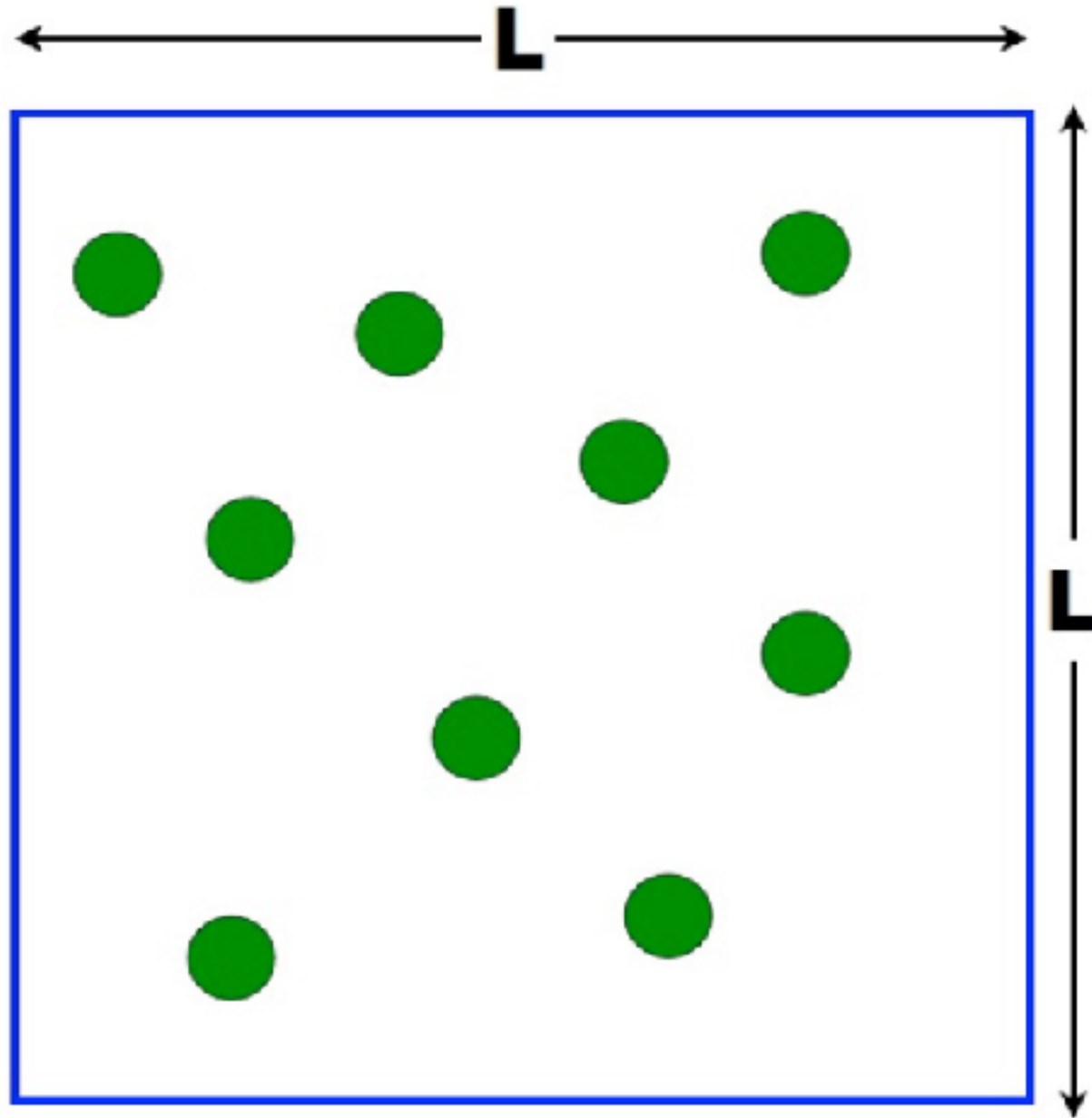


Considere un gas ideal bidimensional de N partículas indistinguibles que se mueven dentro de un cuadrado de lado L , como muestra la figura.

N moléculas



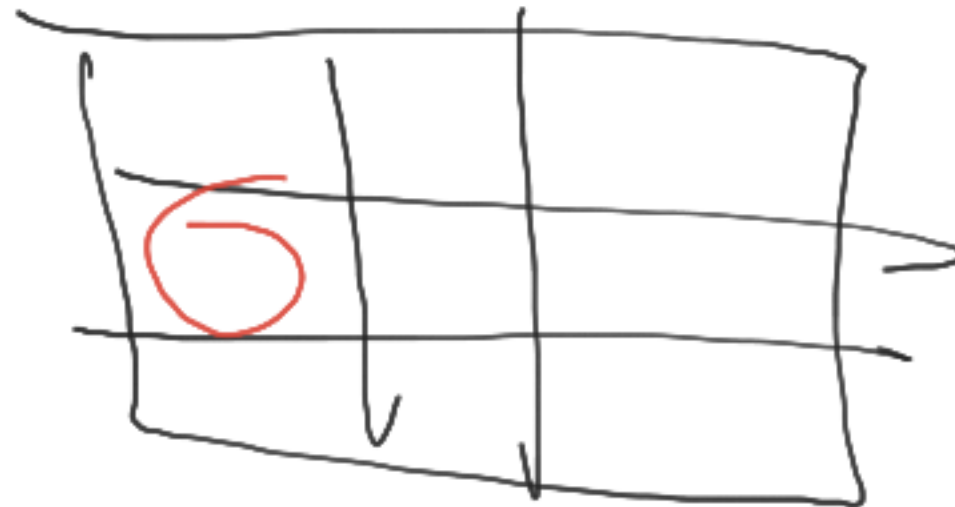
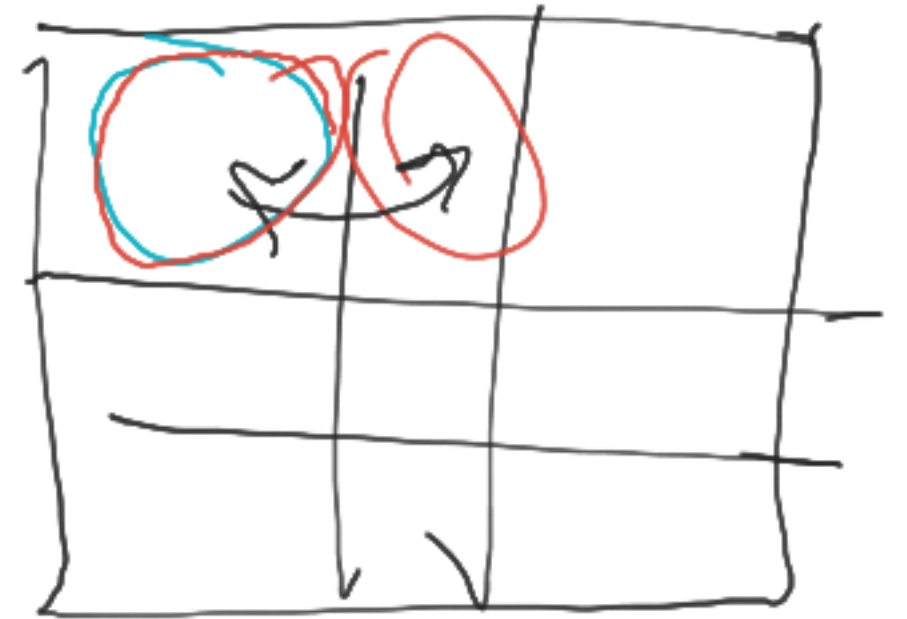
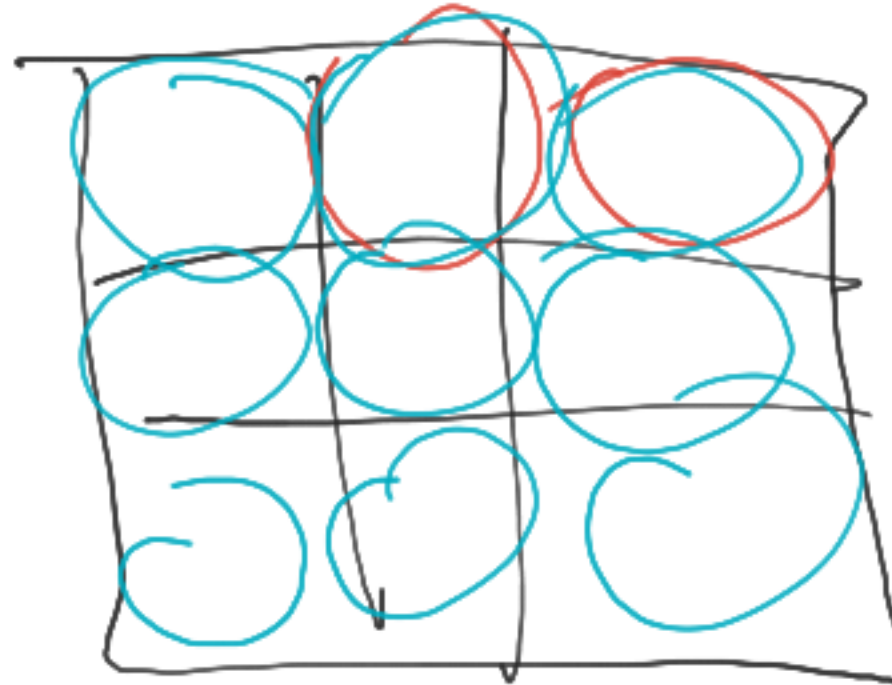
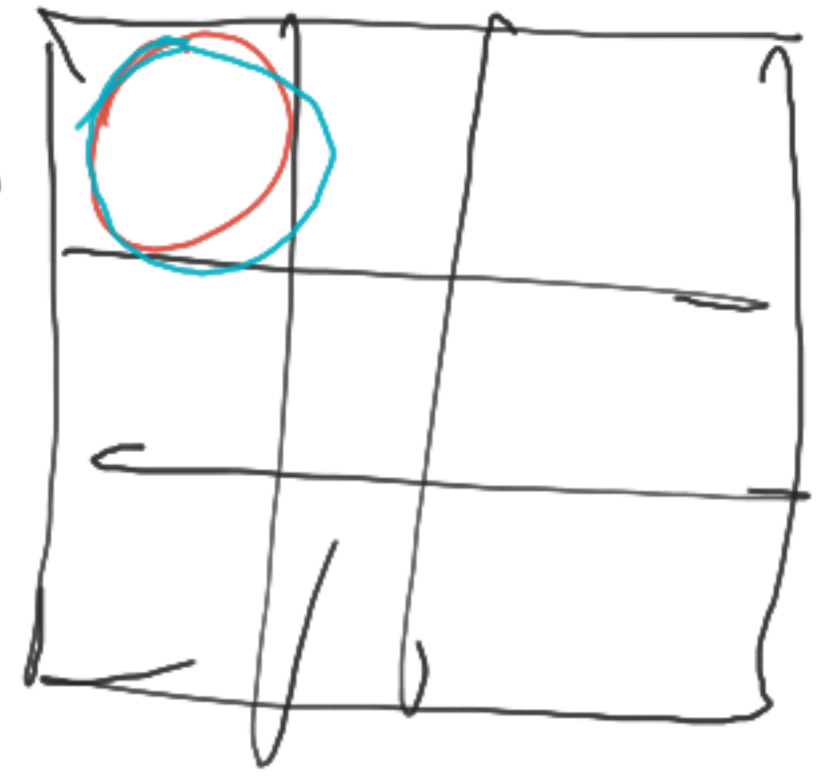
Recordando que el volumen de una esfera de D dimensiones es

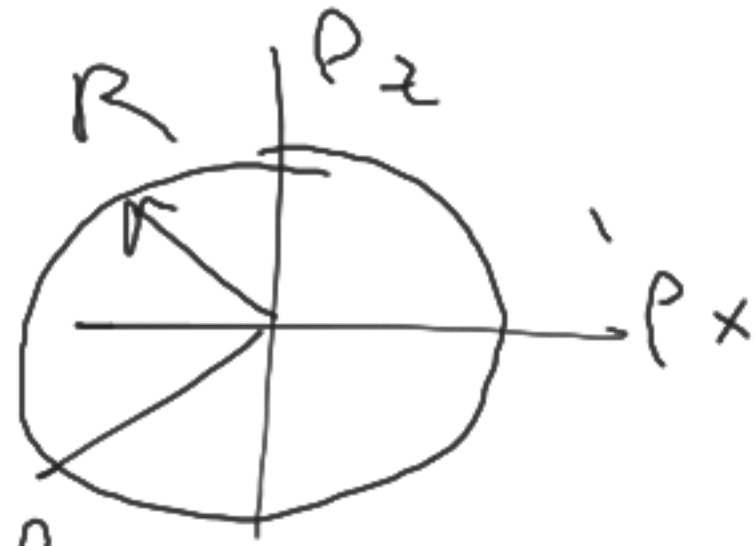
$$V_D(R) = \frac{\pi^{D/2}}{\Gamma(\frac{D}{2} + 1)} R^D$$

El número de estados con energía entre E y $E + \Delta E$ es:

Seleccione una:

$$\left(\frac{L^2}{2\pi} \right)^N \frac{2\pi}{L^2}$$





$$R = \sqrt{2mE}$$

$$L^{2N}$$

$$= \frac{p_x^2}{2m} + \frac{p_y^2}{2m} = \frac{p^2}{2m}$$

$$p^2 = p_x^2 + p_y^2$$

$$\Sigma = \left(L^{2N} \right) \left(V_{D=2N} \left((2mE)^{1/2} \right) \right)$$

$$N!$$

$$\Sigma = \frac{L^{2N} \cdot V_{D=2N} (2mE)^{1/2}}{N!}$$

$\rightarrow \Sigma_{\text{Todas las energías} \leq E} = L^{2N} \left(\frac{\pi^N (2mE)^N}{\Gamma(N+1)} \right)$

$\rightarrow \frac{1}{N!} \frac{\partial \Sigma}{\partial E} \bigg|_{E, E+\Delta E} = \frac{\partial}{\partial E} \left(\frac{L^{2N} (2\pi m)^N E^N \Delta E}{h^{2N} N! \Gamma(N+1)} \right)$

$\Gamma(N+1) = N!$

$$\Omega = \frac{L^{2N} (2m\pi)^N E^{N-1}}{h^{2N} N! (N-1)!} \Delta E$$

↓

$$\Omega = \left(\frac{1}{N!} \right) \frac{\partial \Sigma}{\partial E} \Delta E$$

