

A microscopic view of several red blood cells (erythrocytes) against a black background. The cells are biconcave discs, appearing as bright red, slightly irregular spheres with a darker red center. They are scattered across the frame, with some overlapping.

# Biochimie 1

## Les 12

# Vorige les

Allosterie

# Vandaag

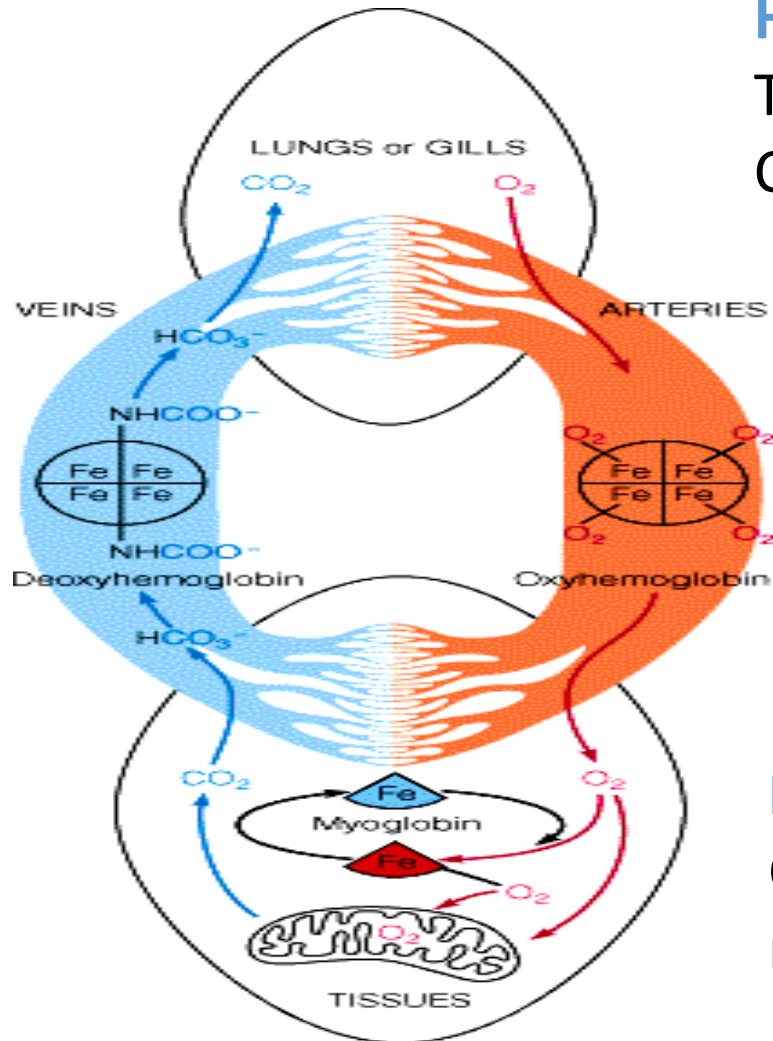
Voorbeeld van een allosterisch eiwit: hemoglobine

Vergelijking tussen hemoglobine en myoglobine

# Hemoglobin & Myoglobin

## Hemoglobin

Transport  $O_2$  from lungs to tissue and  $CO_2$  from tissue to lungs



## Myoglobin

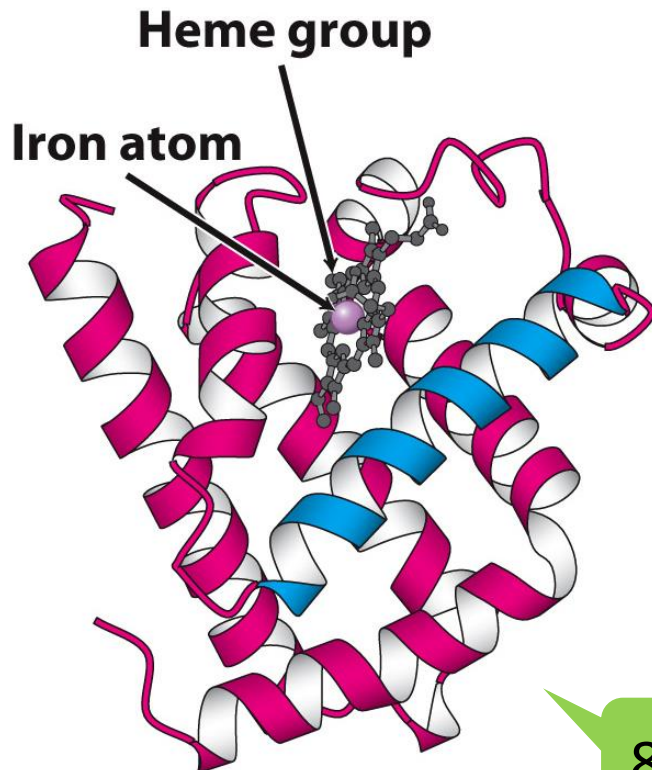
Carries  $O_2$  from cell membrane to mitochondria (in muscle cells)

# Myoglobin

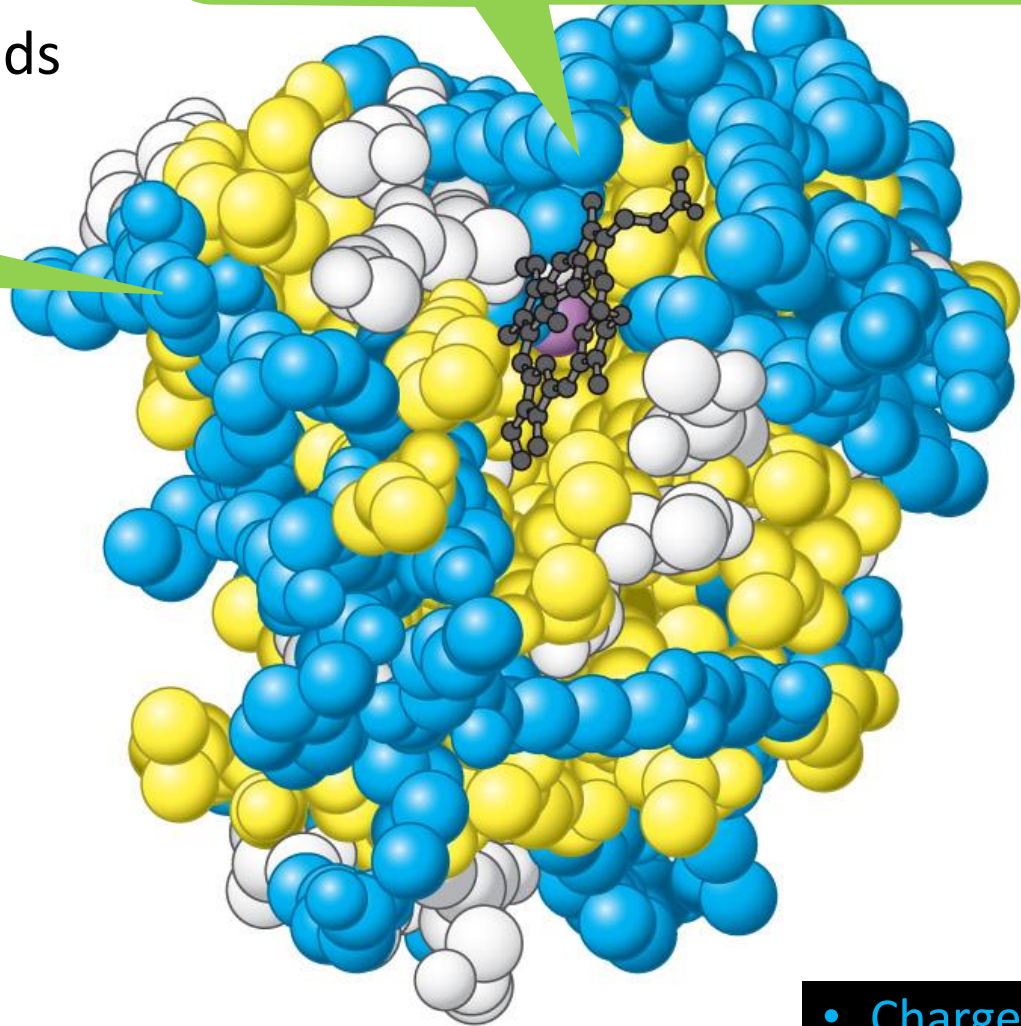
monomer, 153 amino acids

most polar side chains on the surface, nonpolar side chains folded to the interior

**heme group** in hydrophobic pocket, interacts with two His-residues.



8  $\alpha$ -helices

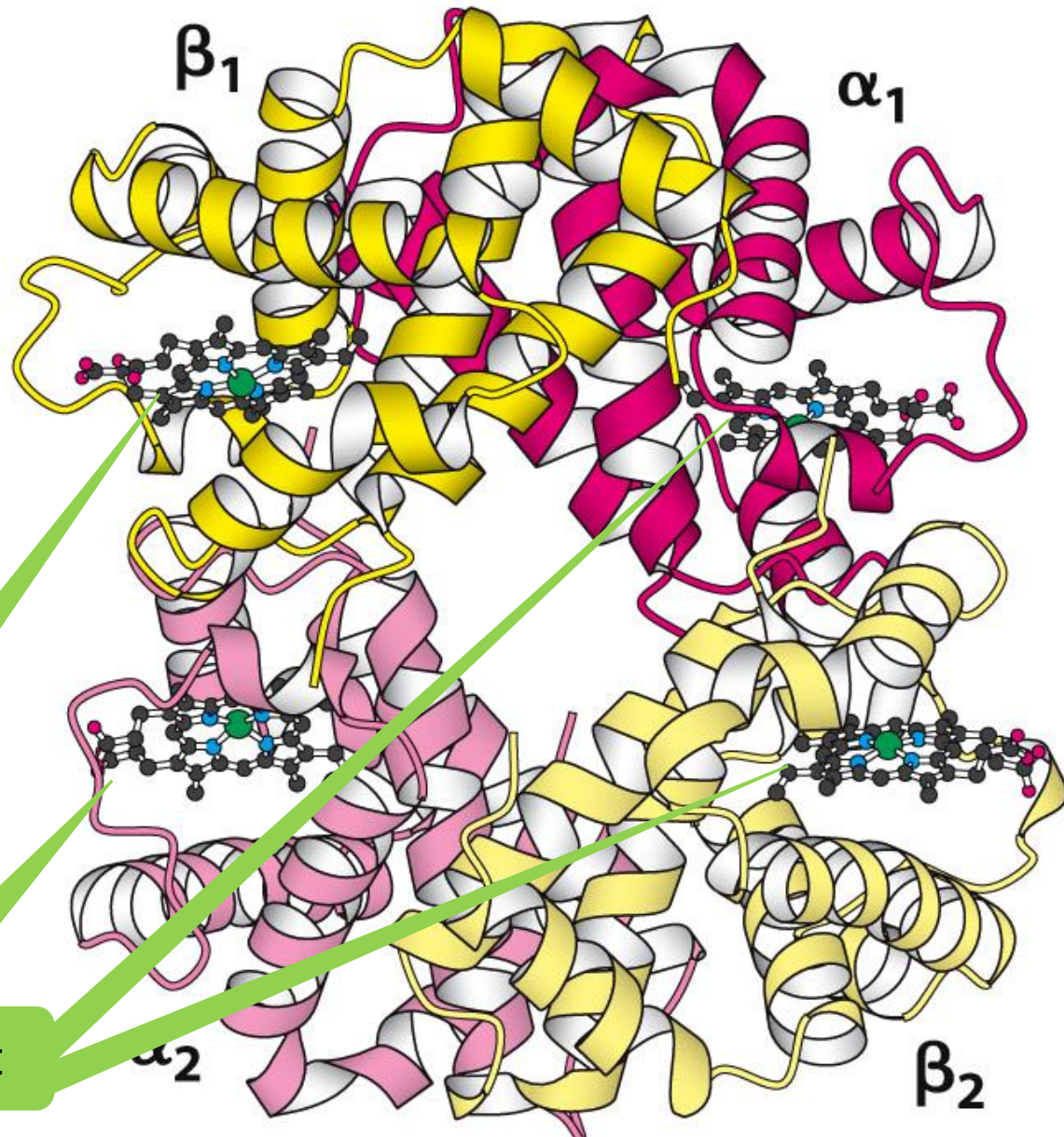


- Charged
- Hydrophobic
- Polar



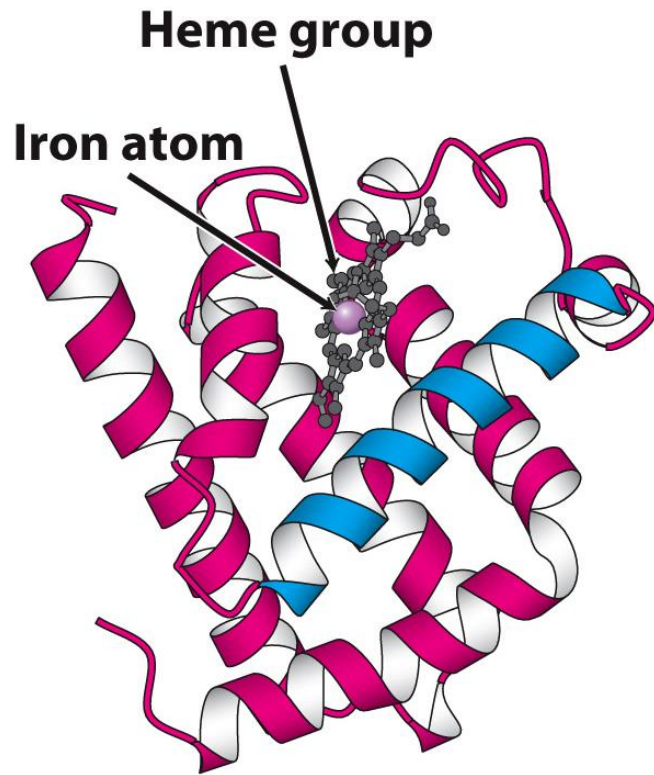
# Hemoglobin

tetramer of two  $\alpha$ -chains (141 amino acids each) and two  $\beta$ -chains (153 amino acids each);  $\alpha_2\beta_2$

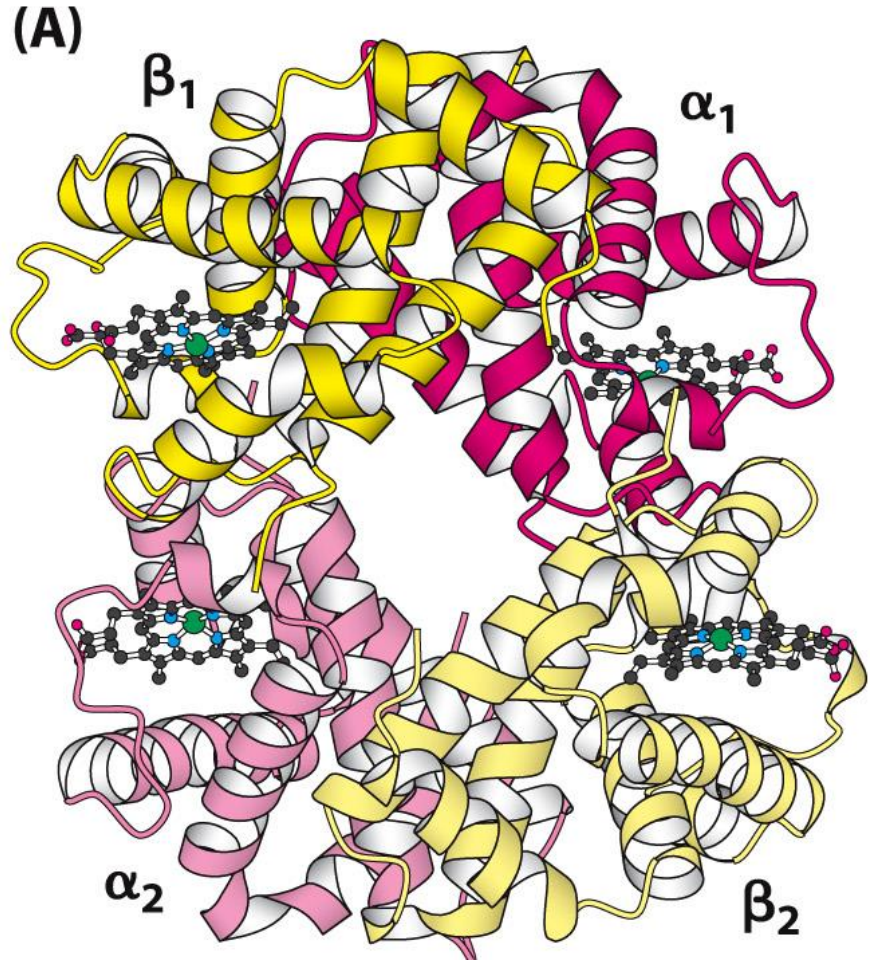


heme group in each subunit

# Quaternary Structures Mb and Hb



Myoglobin is monomeric



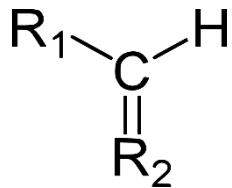
Hemoglobin is tetrameric

Globin fold

# Heme group

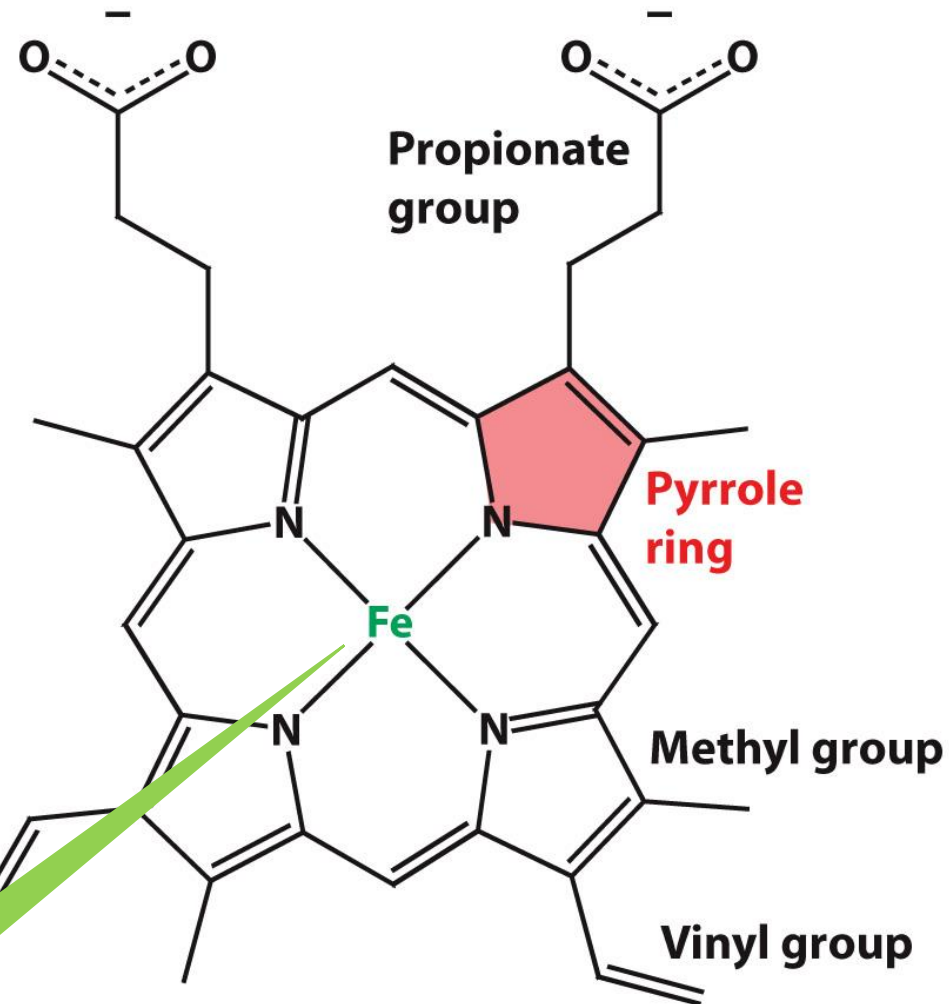
Iron atom + protoporphyrin

Protoporphyrin: 4 pyrrole rings linked by methine bridges (=C-)



Fe<sup>2+</sup> forms bonds with N atoms of the pyrrole rings

Fe(II): 6 coordination sites. 4 interactions with N's of protoporphyrin → 2 left

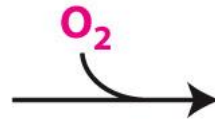
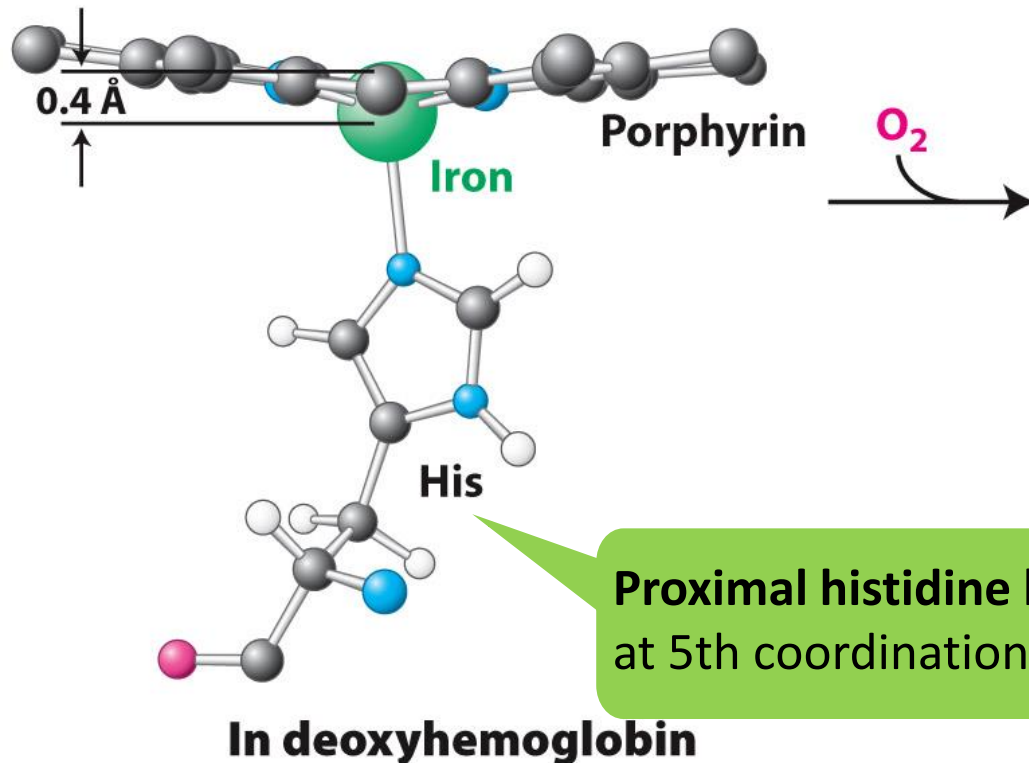


**Heme**  
**(Fe-protoporphyrin IX)**

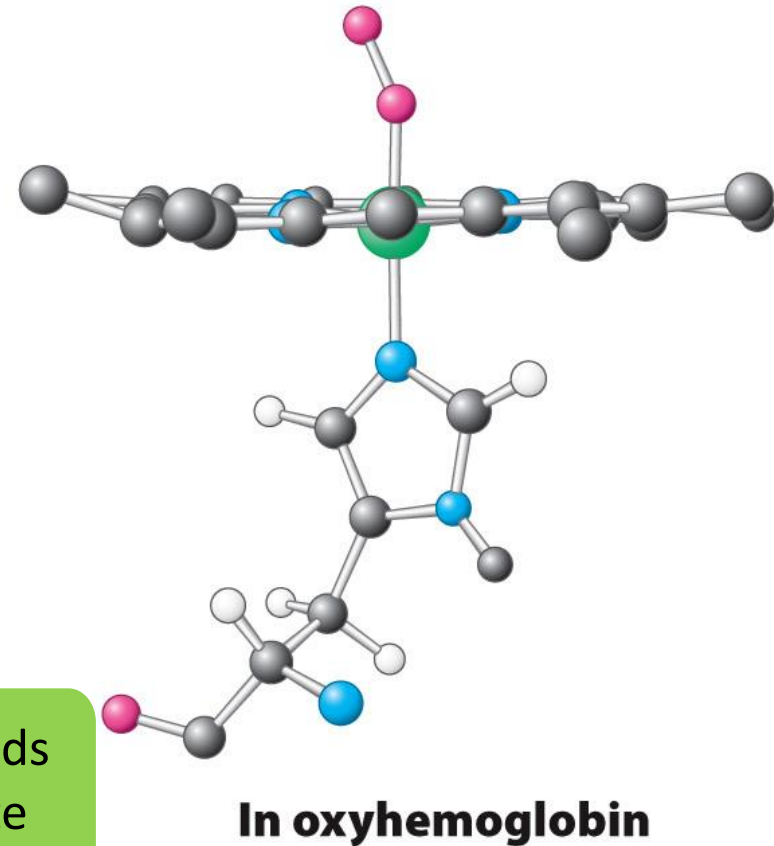


# Heme – oxygen binding

Not shown: distal histidine  
(prevents oxidation of  $\text{Fe}^{2+}$ )



Oxygen binds at 6th  
coordination site

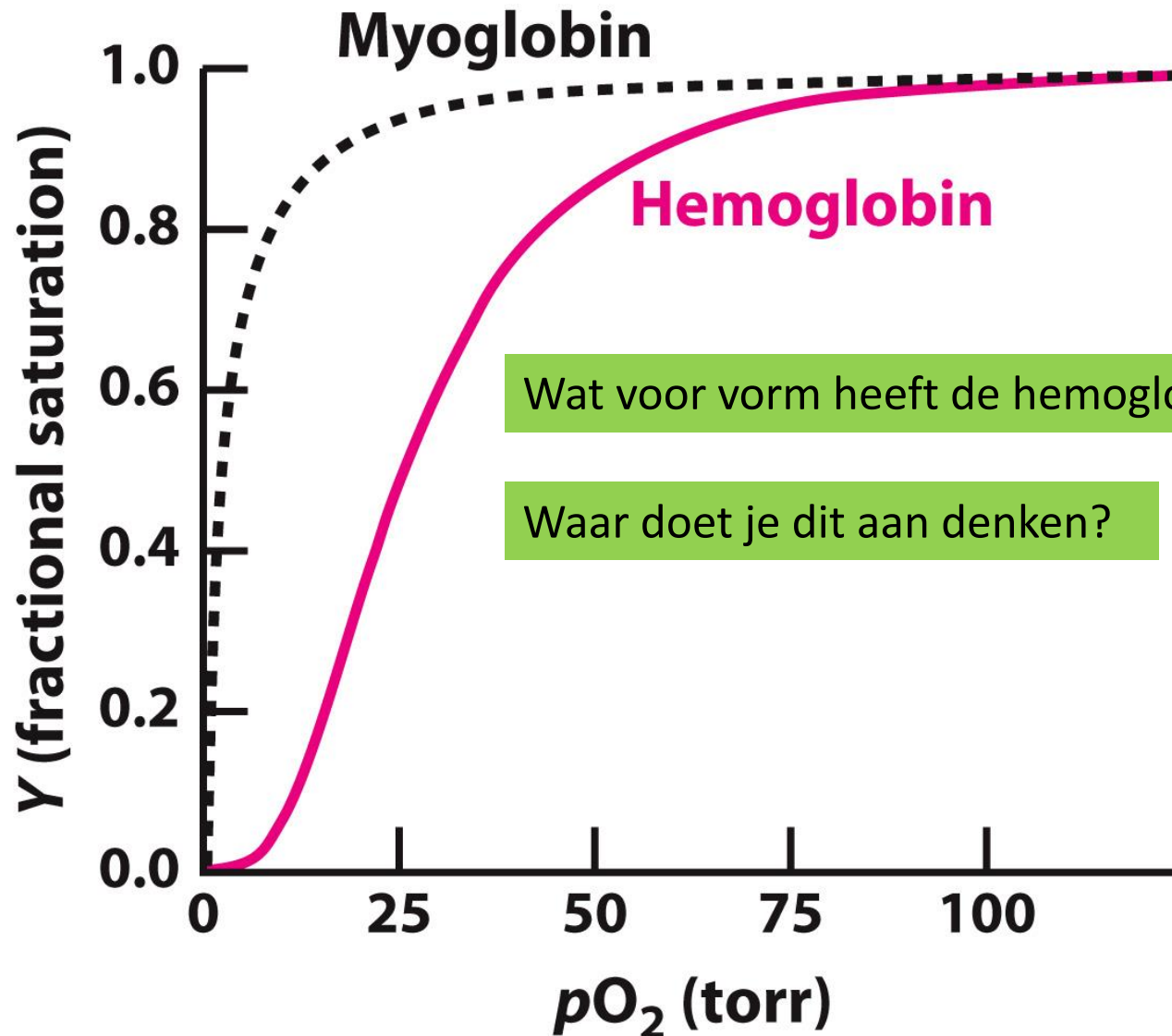


# Heme – oxygen binding

Zie ook:

<https://pdb101.rcsb.org/motm/41#tabs-2>

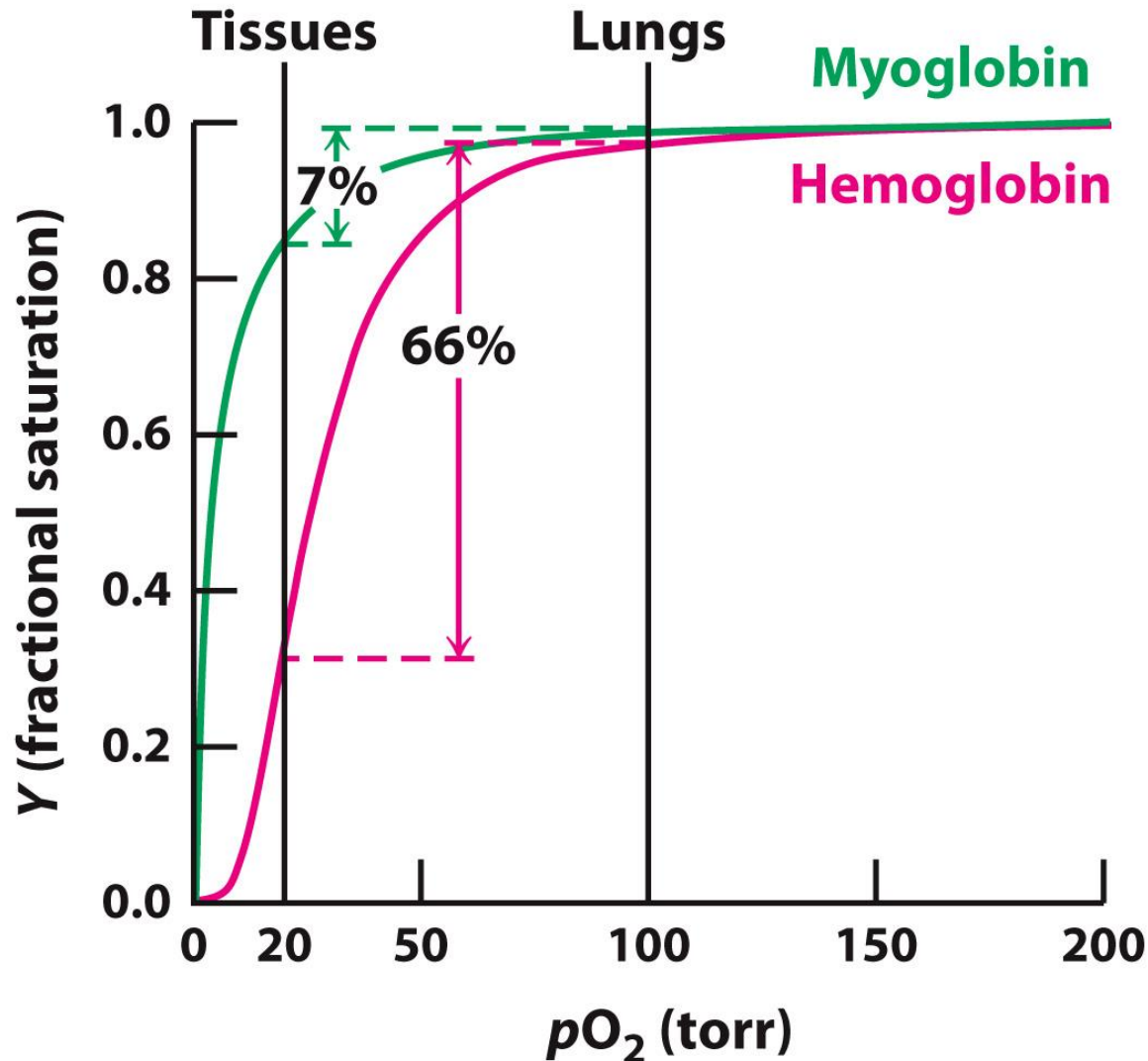
# Oxygen binding



Wat voor vorm heeft de hemoglobine curve?

Waar doet je dit aan denken?

# Oxygen binding





# Oxygen binding to hemoglobin

- each chain has one **heme group**; hemoglobin can bind up to 4 molecules of  $O_2$
- binding is **cooperative**; when one  $O_2$  is bound, it becomes easier for the next  $O_2$  to bind
- Hemoglobin is an **allosteric** protein: binding of  $O_2$  in one subunit brings changes in structure of other subunit(s)

# Hemoglobin

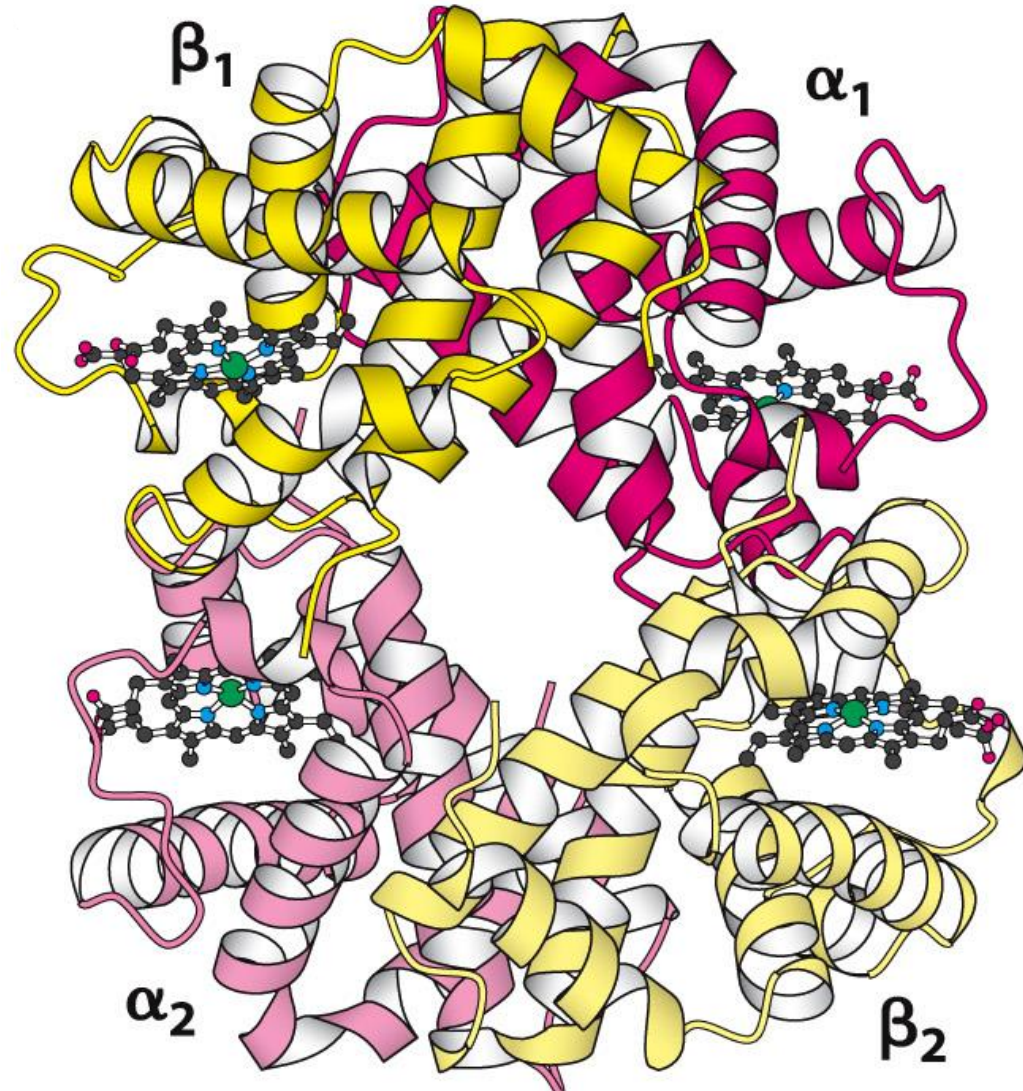
Allosteric protein

Deoxyhemoglobin  $\rightarrow$  T-state

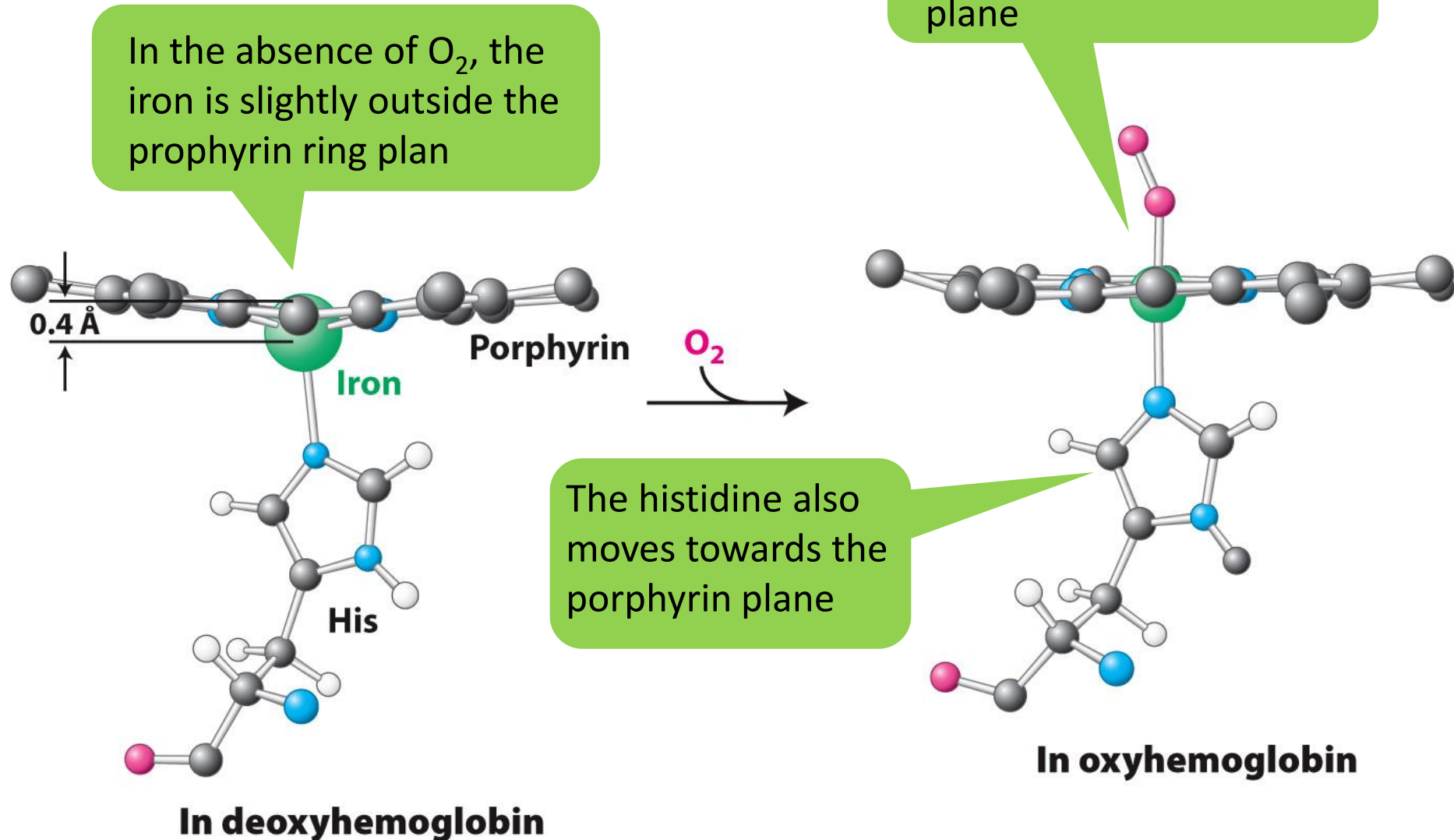
Oxyhemoglobin  $\rightarrow$  R-state

Oxygen binding: T  $\rightarrow$  R

How?



# Heme – oxygen binding

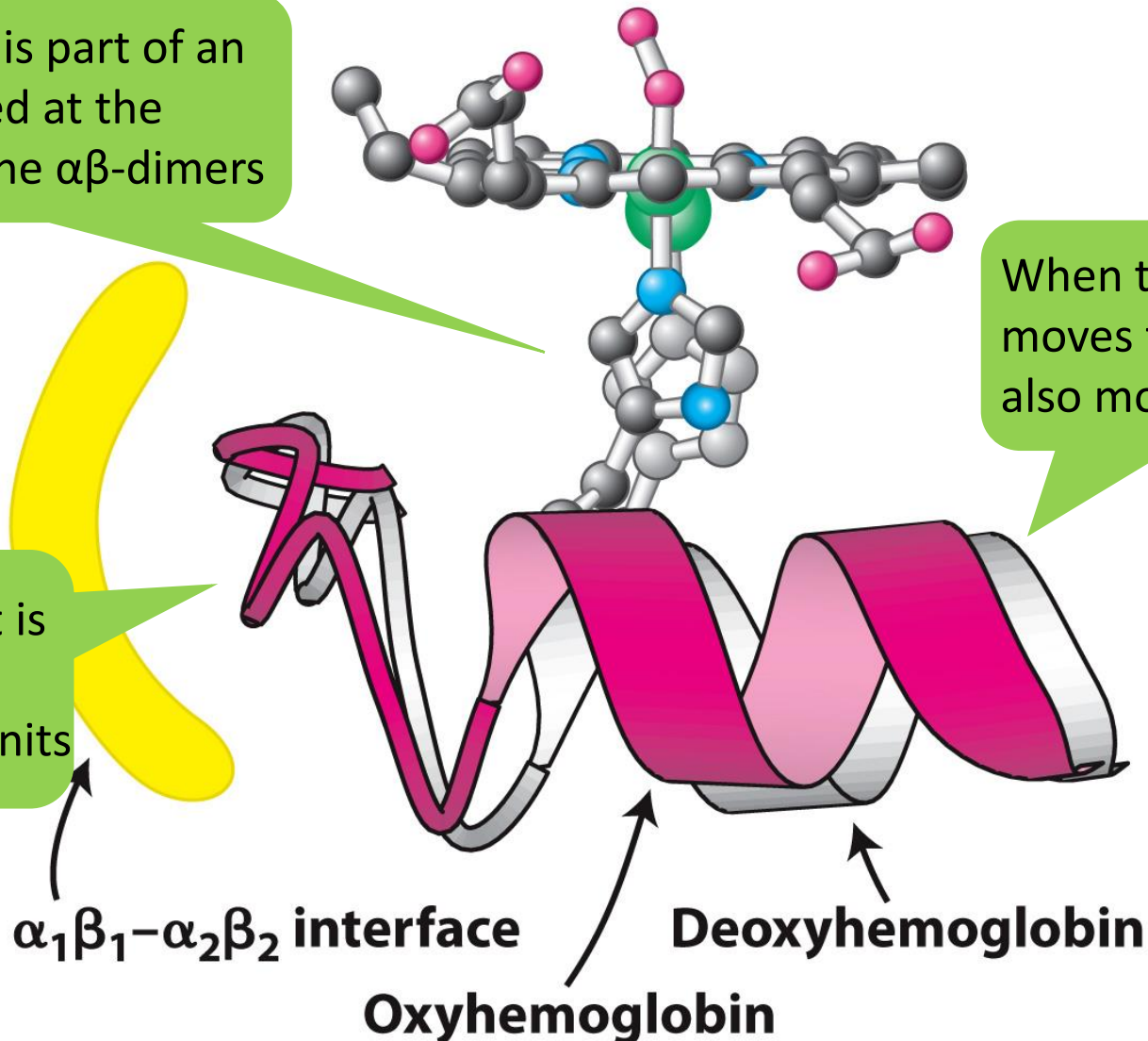


# Hemoglobin – oxygen binding

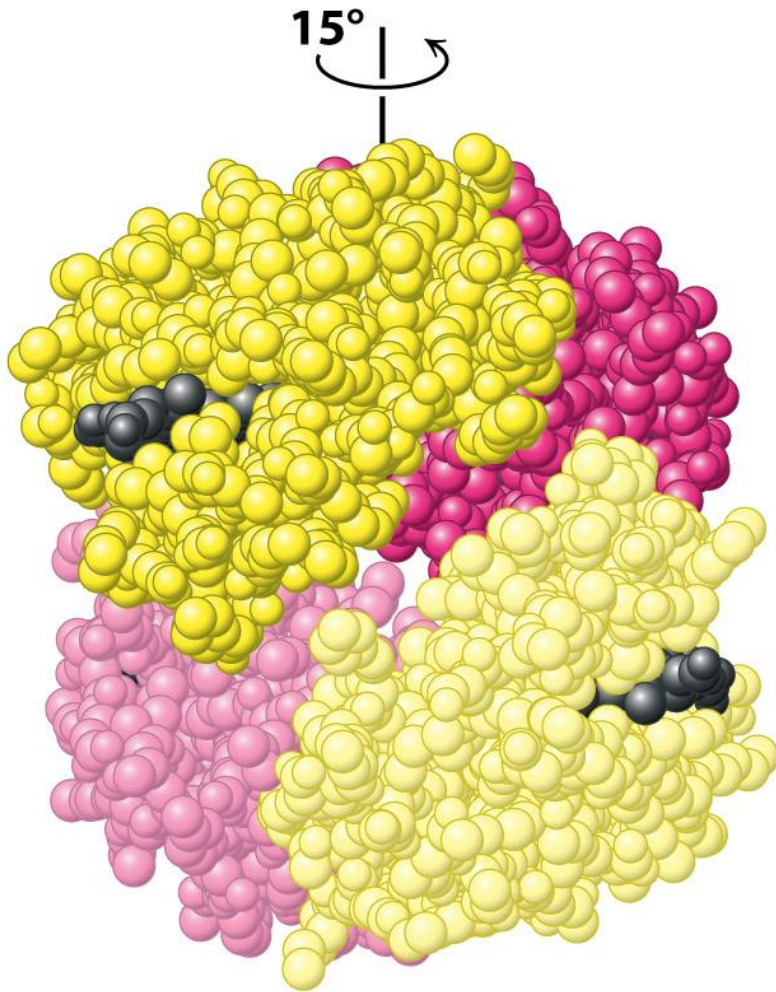
The histidine is part of an  $\alpha$ -helix located at the interface of the  $\alpha\beta$ -dimers

When the histidine moves the  $\alpha$ -helix also moves

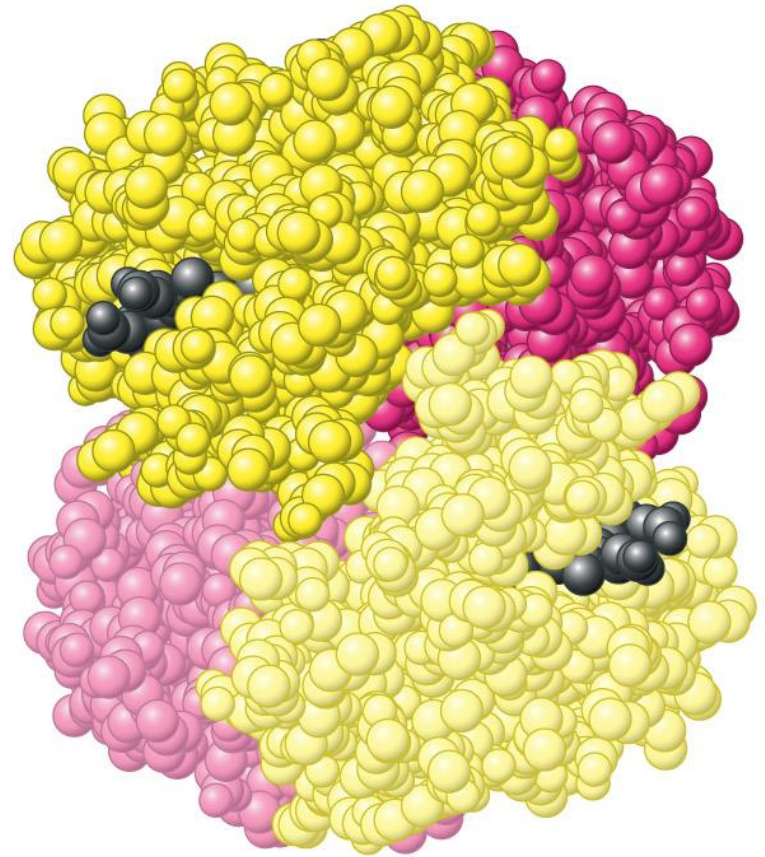
The movement is transmitted to the other subunits







**Deoxyhemoglobin**



**Oxyhemoglobin**

Zie ook:

<https://pdb101.rcsb.org/motm/41#tabs-2>

# Binding of Oxygen by Myoglobin and Hemoglobin

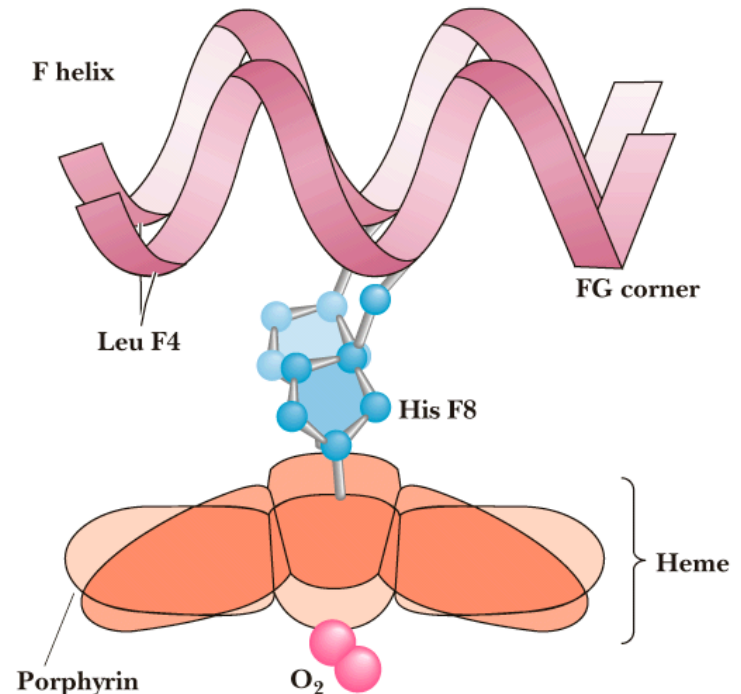
Hemoglobin must bind oxygen in lungs and release it in capillaries.  
Myoglobin must bind oxygen.

Hemoglobin:

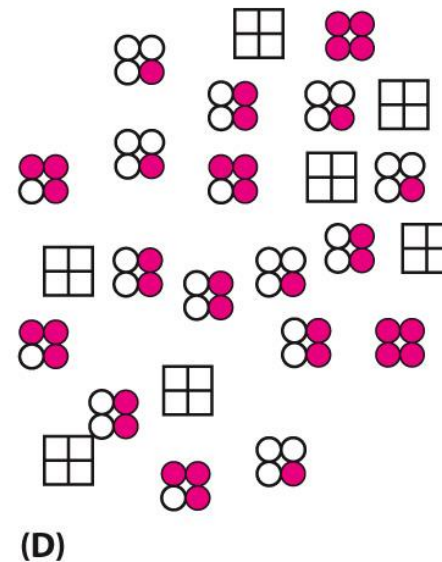
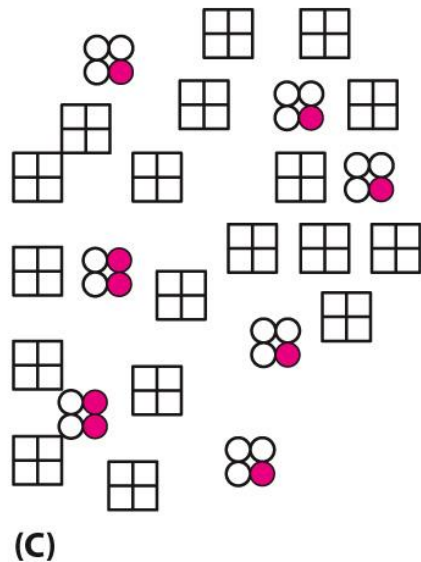
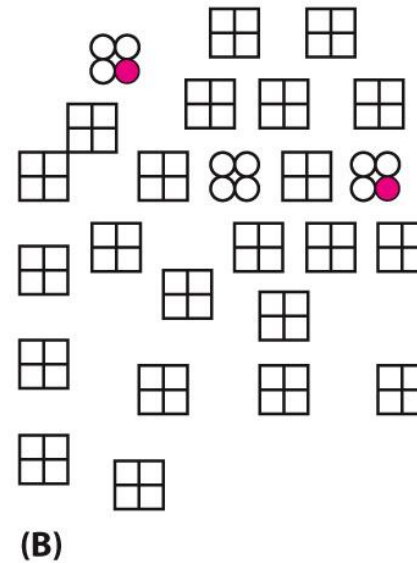
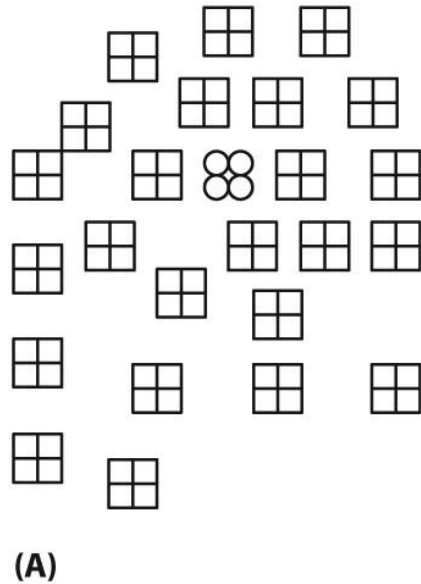
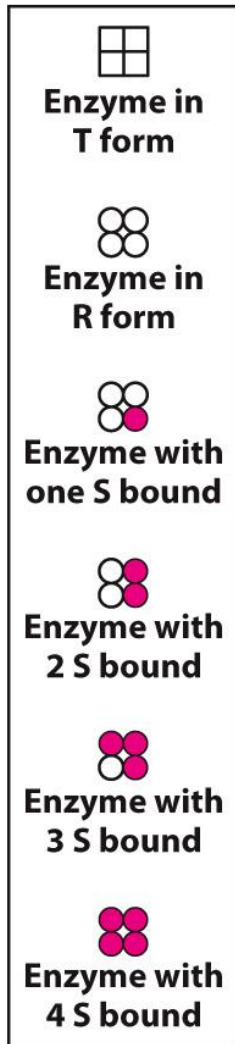
- Adjacent hemoglobin subunits' affinity for oxygen increases
- This is called **positive cooperativity** and does not happen in Myoglobin
- If Hemoglobin behaved like Myoglobin, very little oxygen would be released in capillaries
- The sigmoid, cooperative oxygen binding curve of Hemoglobin makes this possible!

# Recap: Hb conformational changes upon O<sub>2</sub> binding

- Without O<sub>2</sub> bound → Fe out of the heme plane
- O<sub>2</sub> binding → Fe pulled into the heme plane → His F8 (proximal histidine) pulled along → F helix moves
- Total movement of Fe is 0.29 Å.
- This change means little to Mb, but lots to Hb!
- Movement of Fe initiates a series of conformational changes to adjacent subunits
- T-to-R-state transition



# Vorige les: concerted model

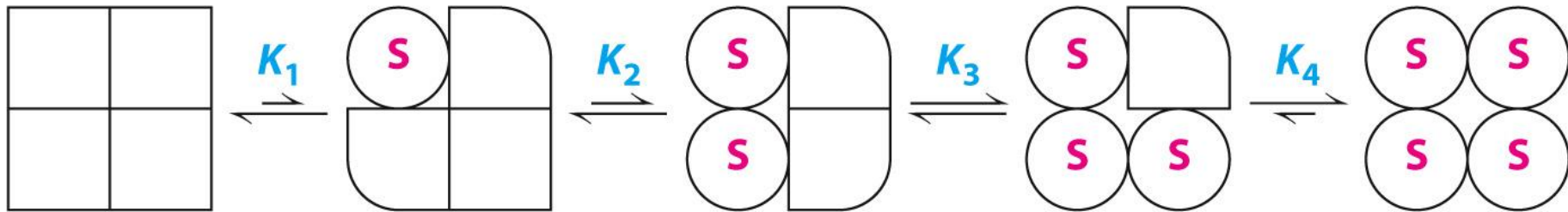




# Vorige les: sequential model

In het concerted model heeft een enzym 2 toestanden, T en R

Een alternatief is dat binden van S aan één subunit de conformaties van andere subunits verandert, zodat S beter bindt



Onderzoek naar allosterische enzymen suggereert dat veel van deze enzymen werken volgens een combinatie van beide modellen

# Hemoglobin

3 sites occupied by oxygen?

- quaternary structure almost always R state
- affinity for oxygen of free site 20x higher

Resembles  
concerted model

1 sites occupied by oxygen?

- quaternary structure: T state
- affinity for oxygen of free site 3x higher

Resembles  
sequential model

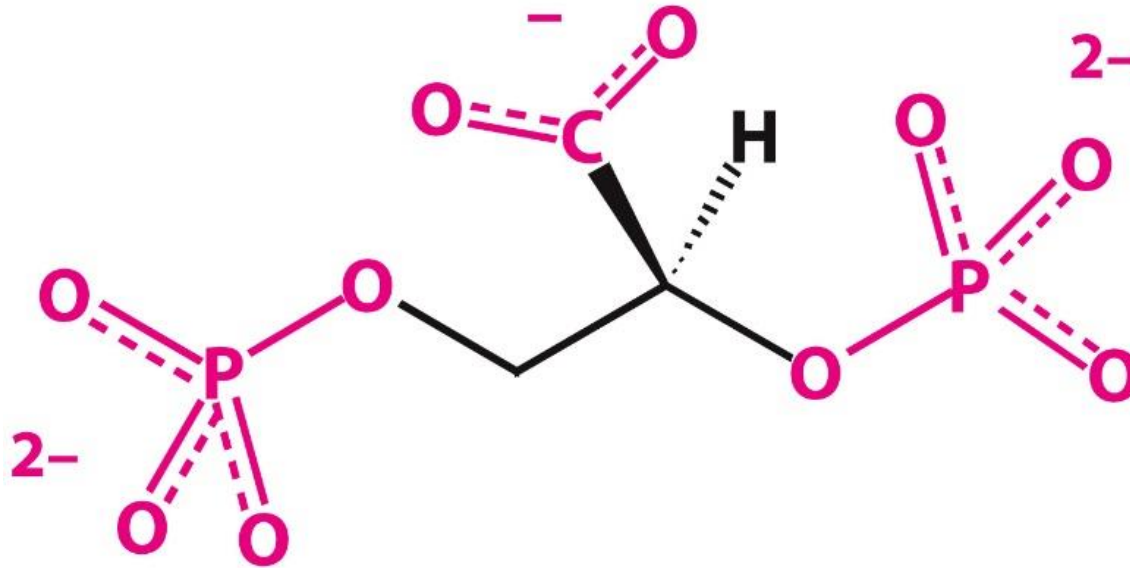
Concerted model/ sequential model: idealized cases.

# Vorige les: regulatiemoleculen beïnvloeden het evenwicht tussen T en R

- Positieve regulatie:
  - molecuul stabiliseert R
  - concentratie R  $\uparrow$
  - grotere kans op binding van S
- Negatieve regulatie:
  - molecuul stabiliseert T
  - concentratie T  $\uparrow$
  - kleinere kans op binding van S

# 2,3-Bisphosphoglycerate

- Hemoglobin in blood is bound to 2,3-BPG

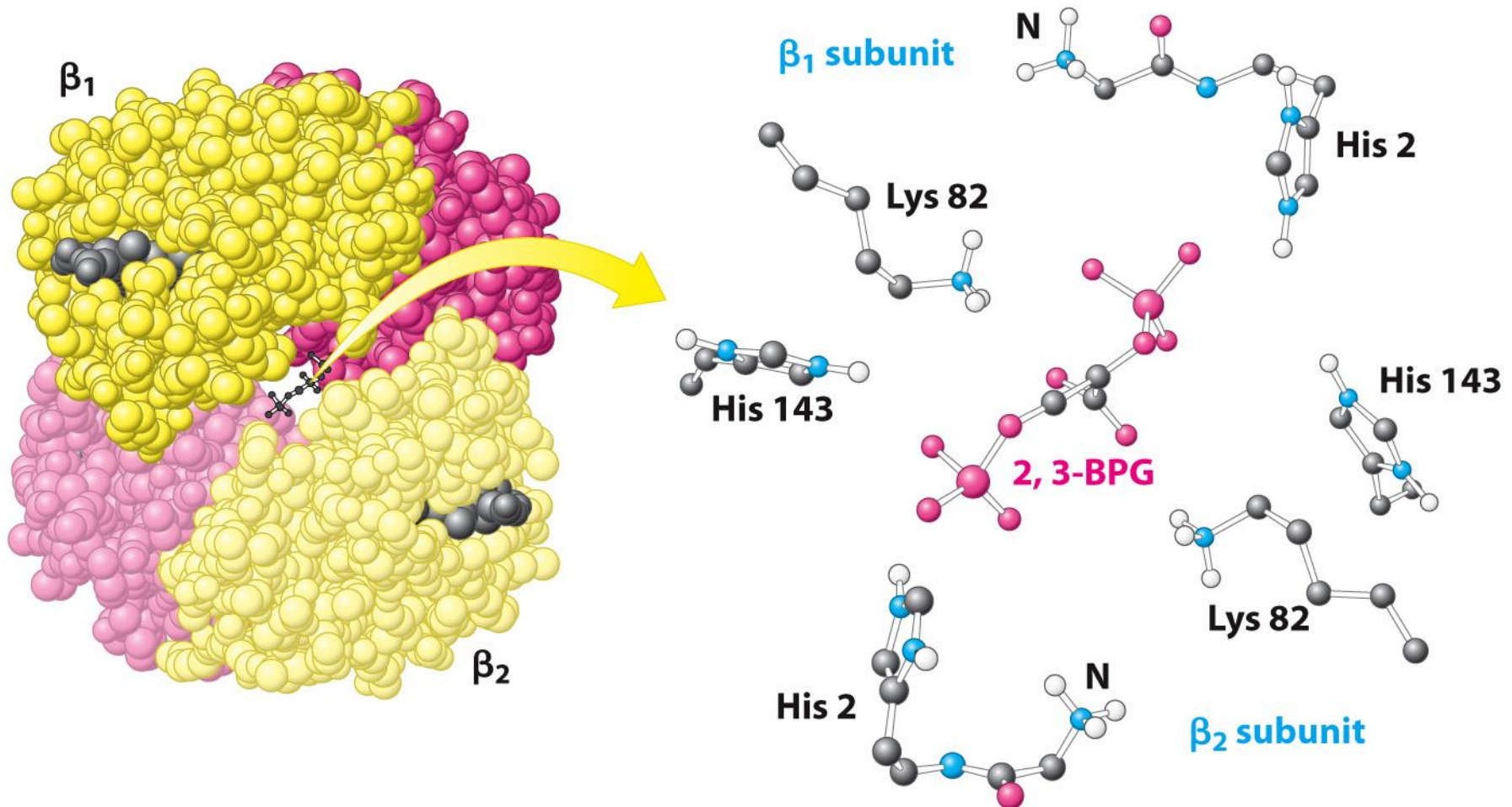


- Highly anionic compound
- Present in red blood cells at ~same concentration as hemoglobin
- 2,3-BPG binds at a site distant from the Fe where oxygen binds → **allosteric effector**



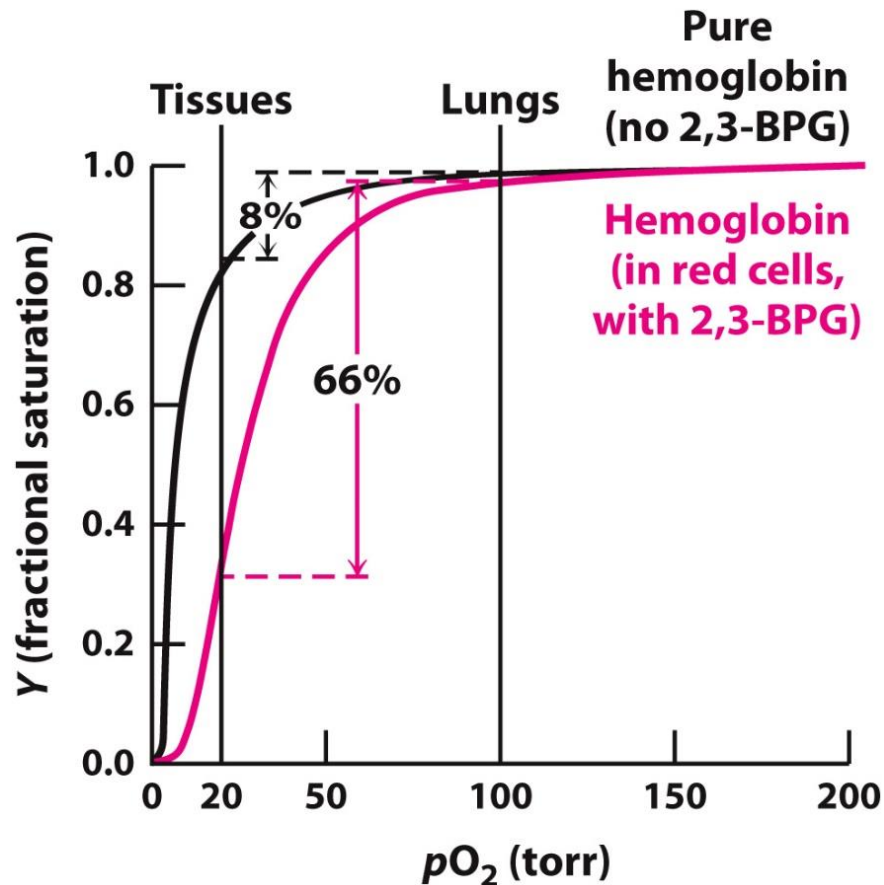
# 2,3-BPG and hemoglobin

2,3-BPG binds in a pocket only present in the T-form of Hg  
Negative charges interact with 2 Lys ( $\beta$  chain), 4 His (each subunit 1 His), 2 N-termini ( $\alpha$  chain)



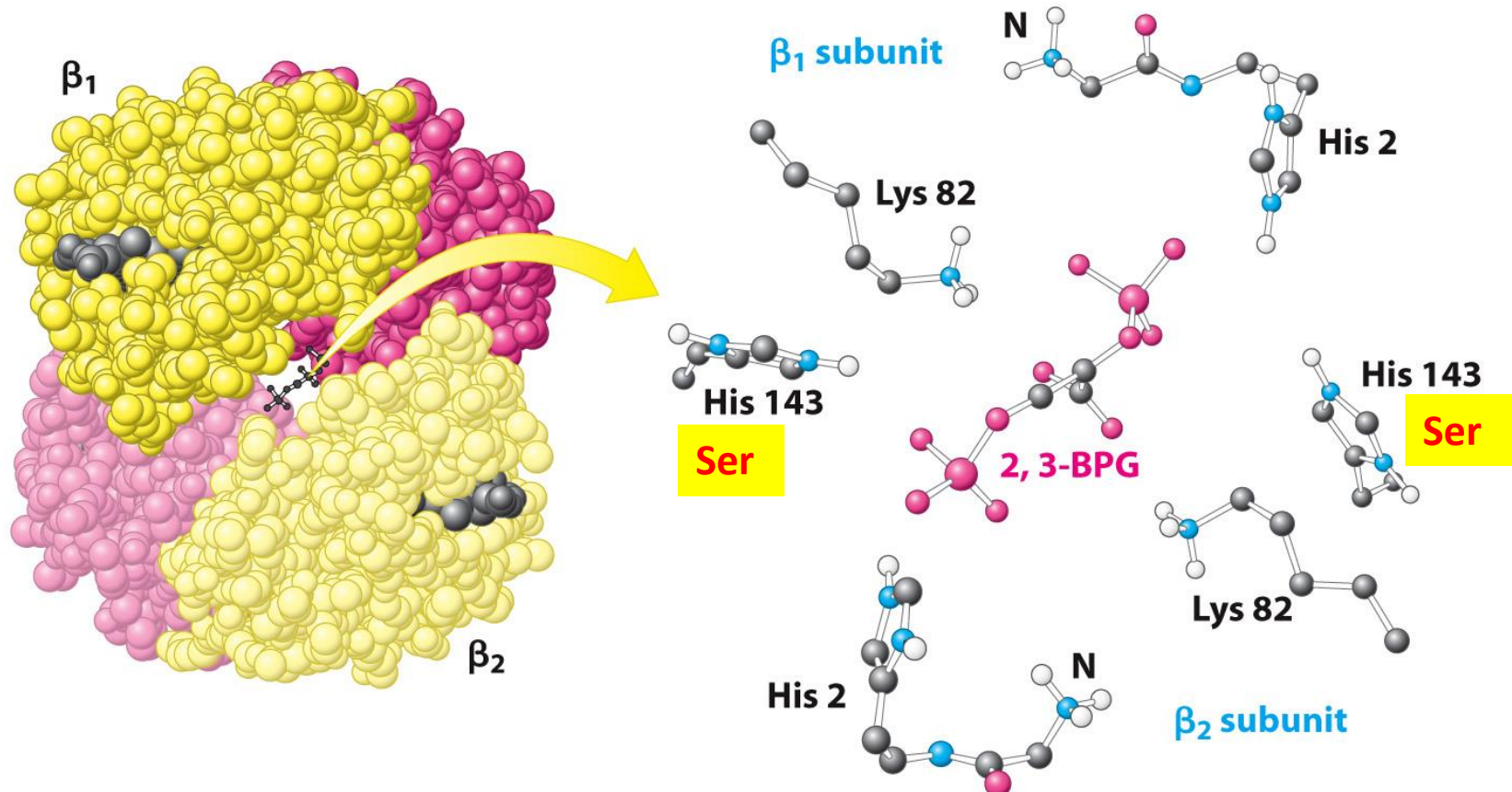
# 2,3-BPG and hemoglobin

- 2,3-BPG binds preferentially deoxyhemoglobin and stabilizes it  
→ 2,3-BPG **reduces** affinity for  $O_2$
- Hemoglobin stripped of 2,3-BPG remains saturated with  $O_2$



# Fetal Hemoglobin

- Two  $\alpha$ -chains and two  $\gamma$ -chains

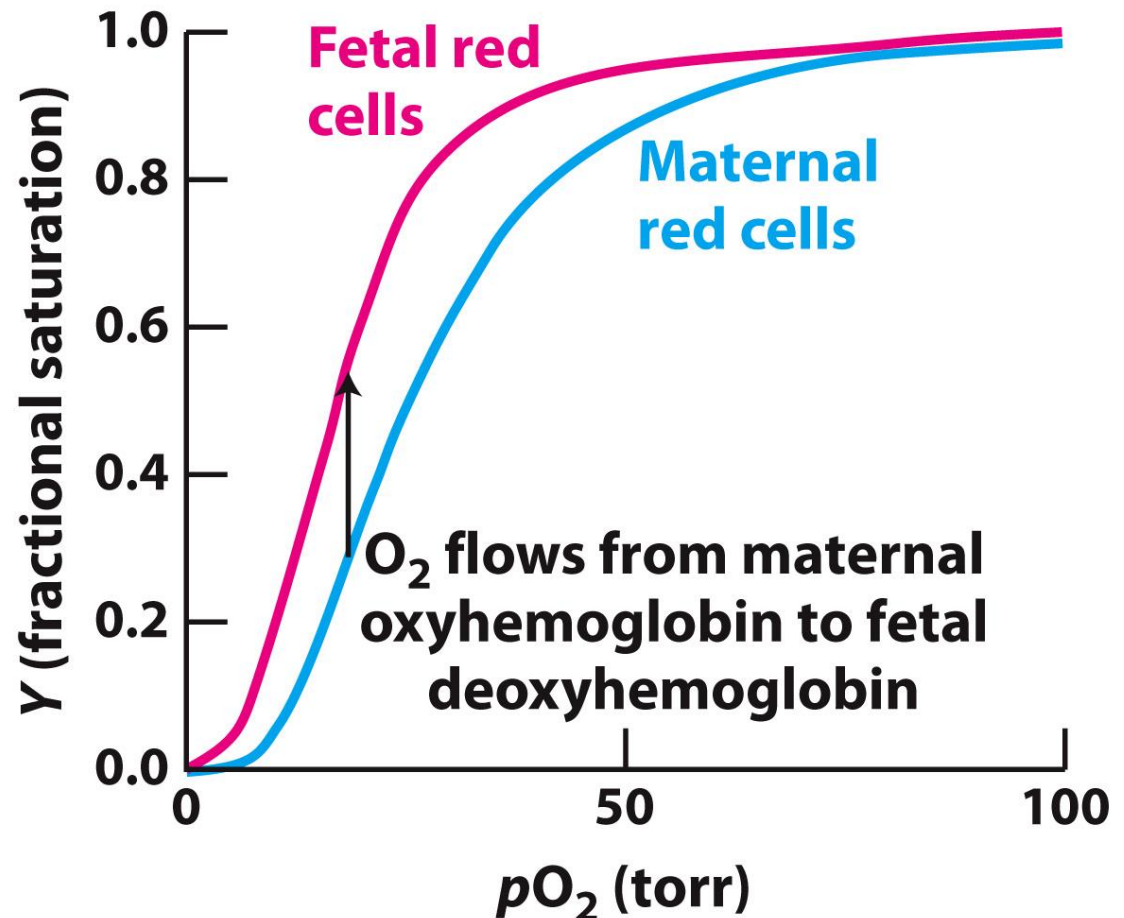


# Fetal Hemoglobin

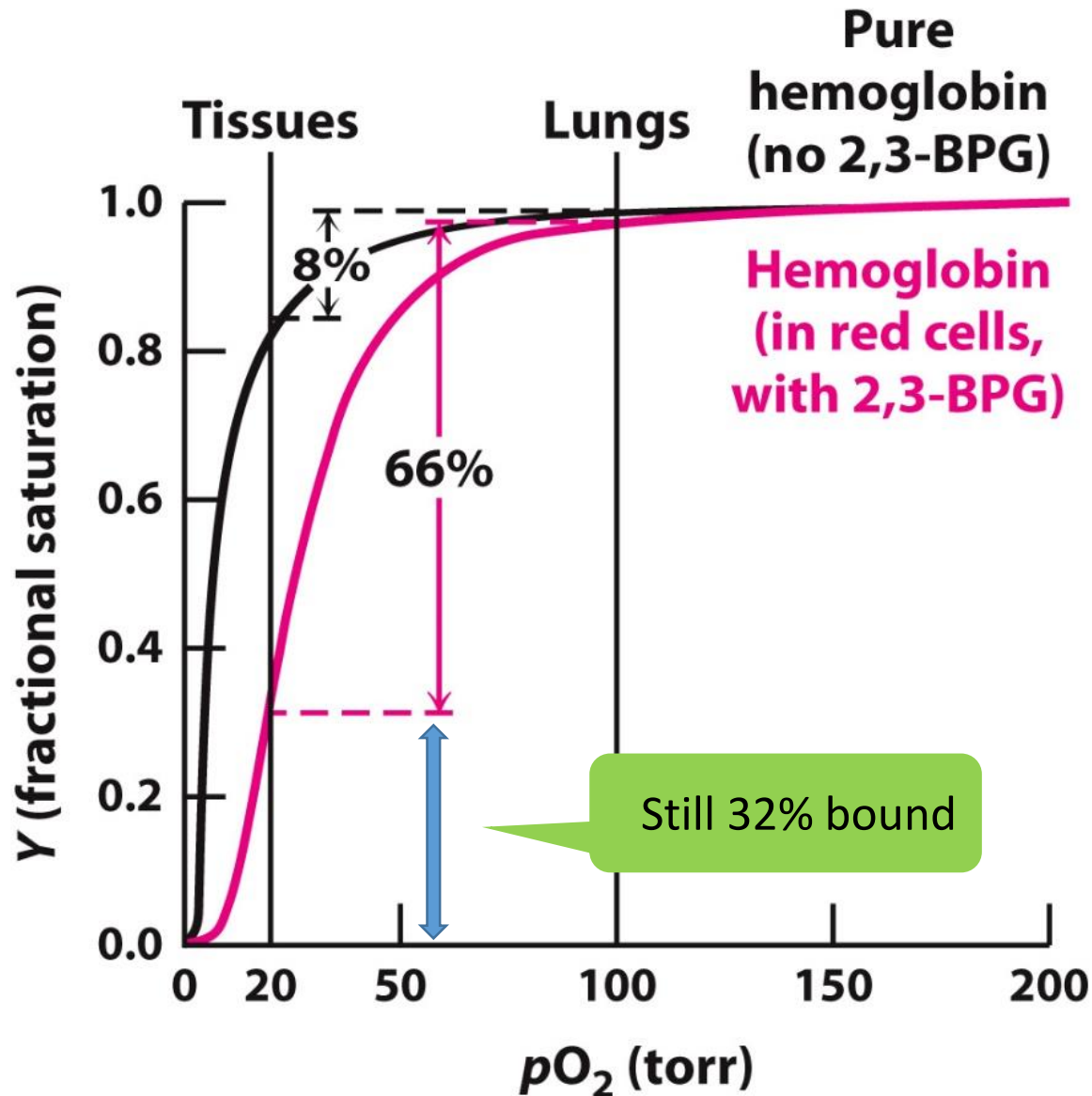
Lower affinity for 2,3-BPG

→ higher affinity for oxygen

→ oxygen from mother Hg to fetal Hg



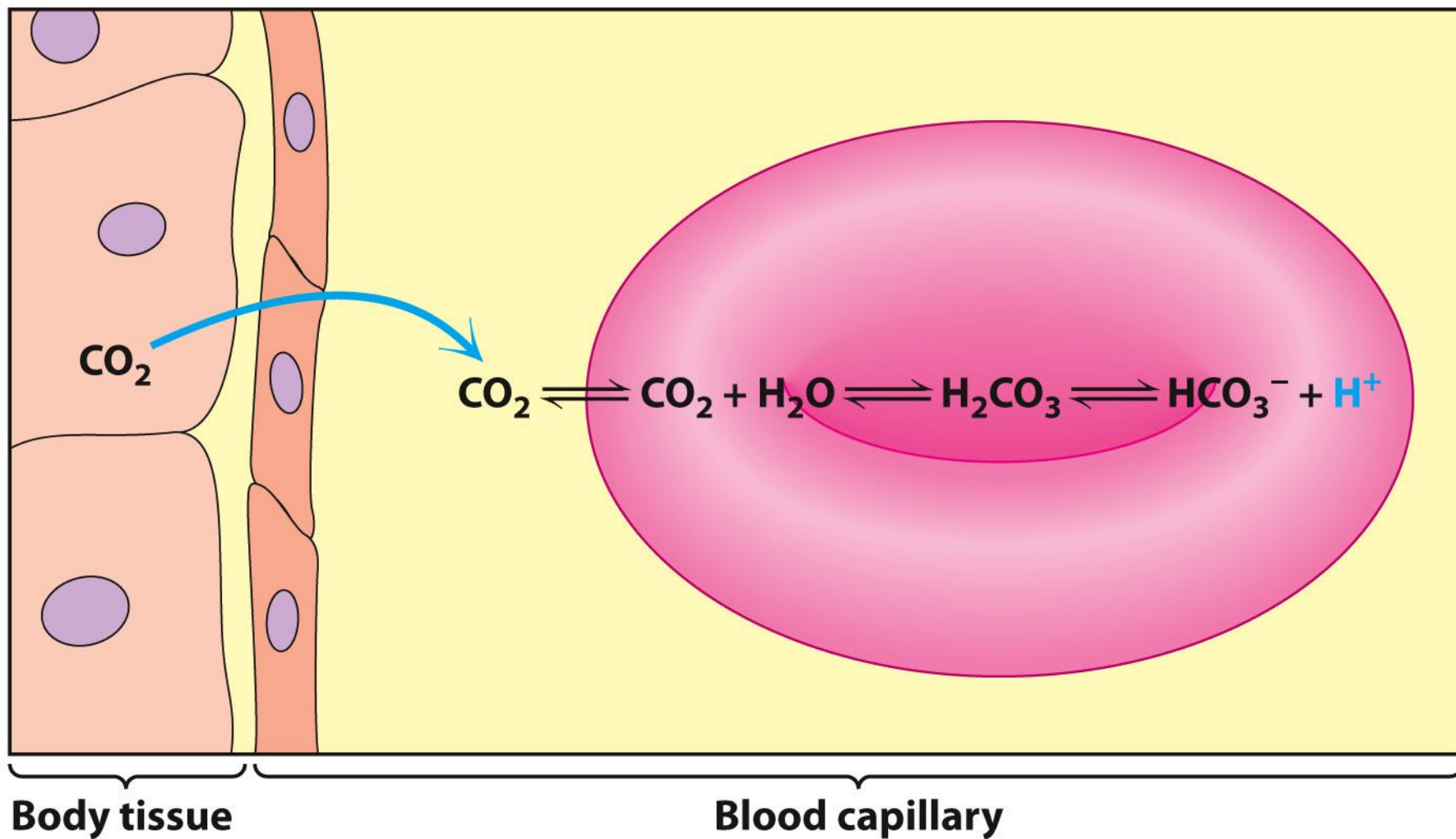
# Oxygen binding (so far)



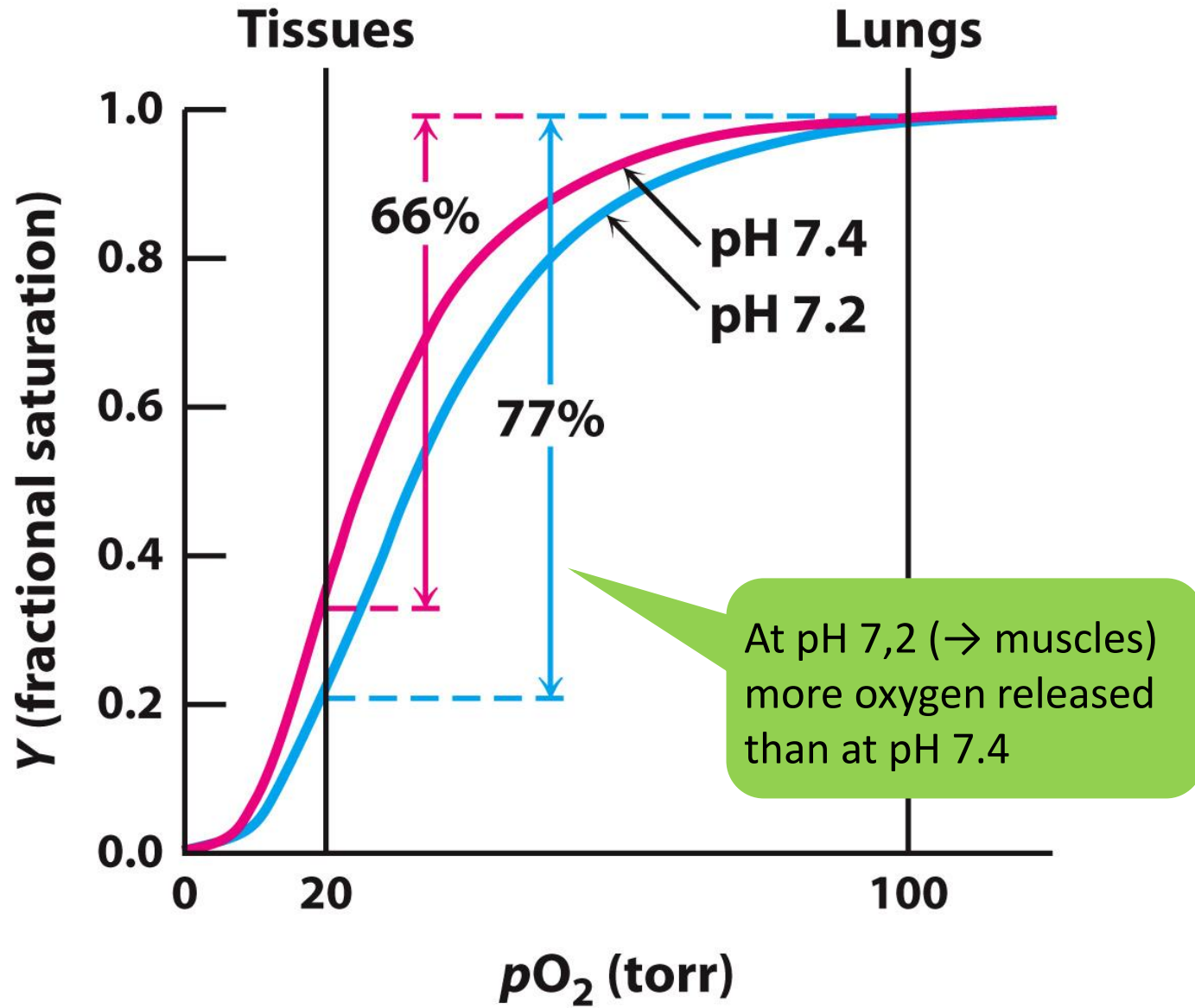
# The Bohr Effect

- Regulation of oxygen binding by  $\text{H}^+$  and  $\text{CO}_2$
- Discovered by Christian Bohr
- Binding of protons and  $\text{CO}_2$  diminishes oxygen binding
- Important physiological significance

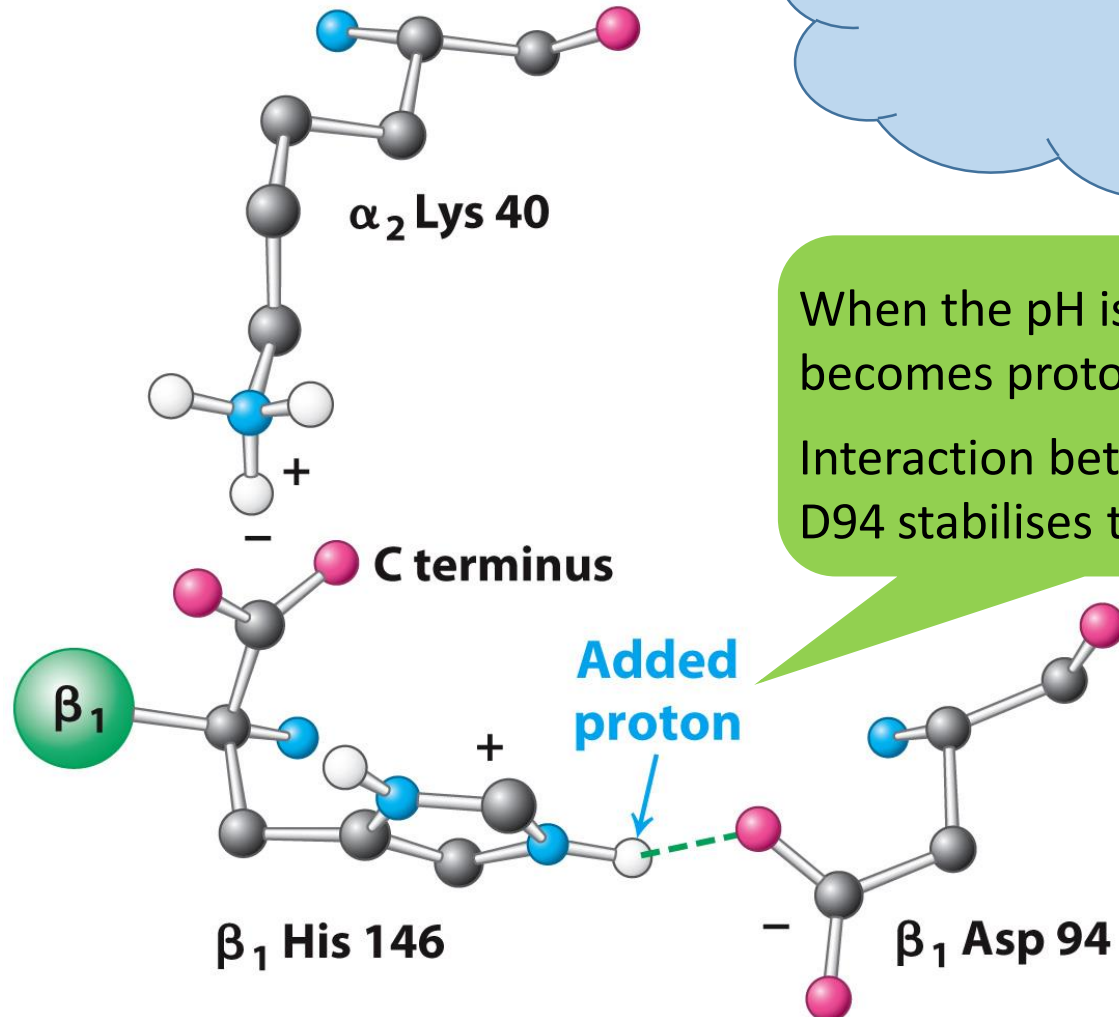




# Lowering the pH



# Lowering the pH

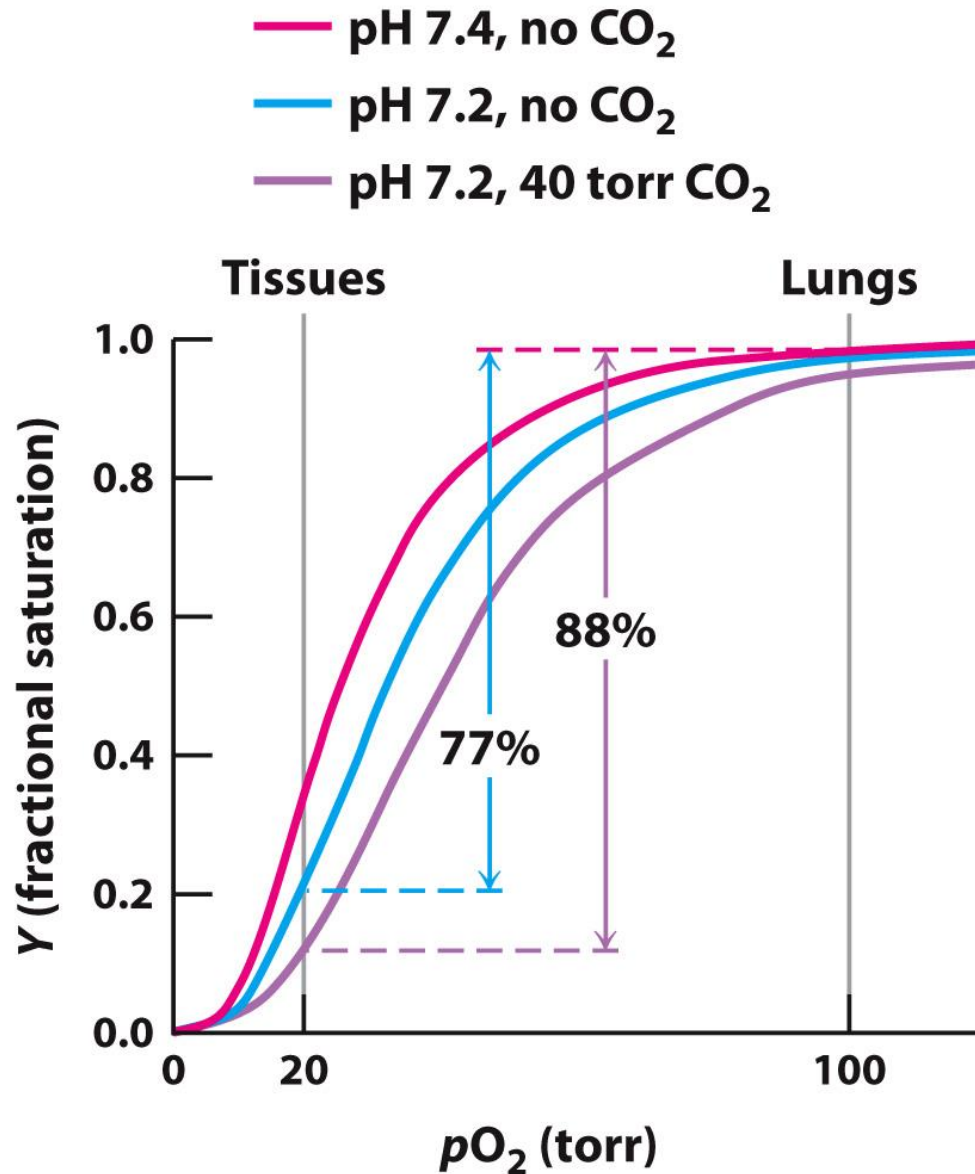


$$\text{pH} = \text{pK}_a + \log \frac{[\text{A}^-]}{[\text{HA}]}$$

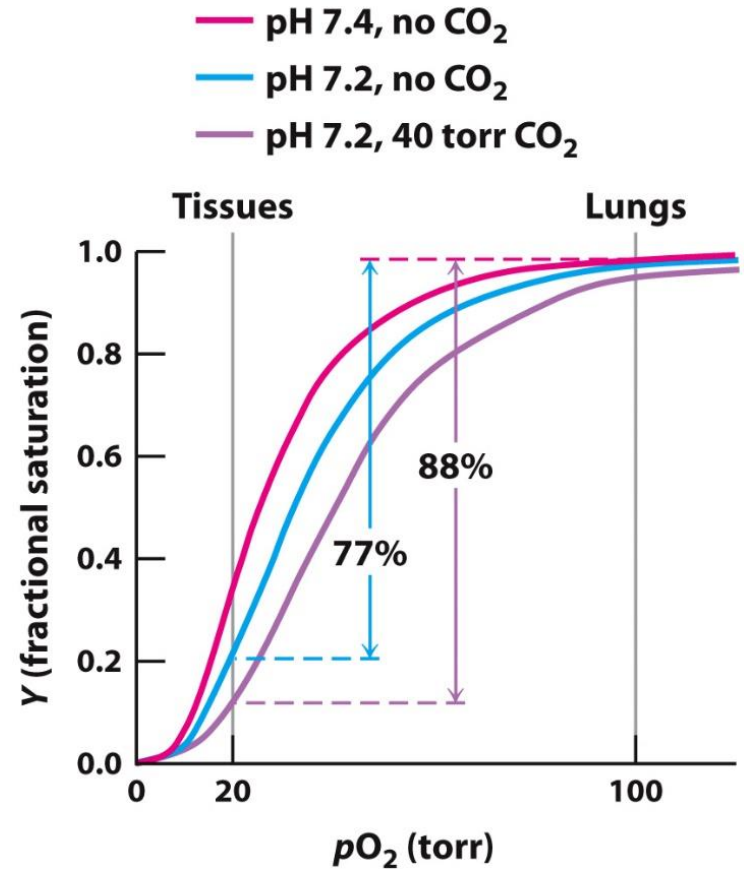
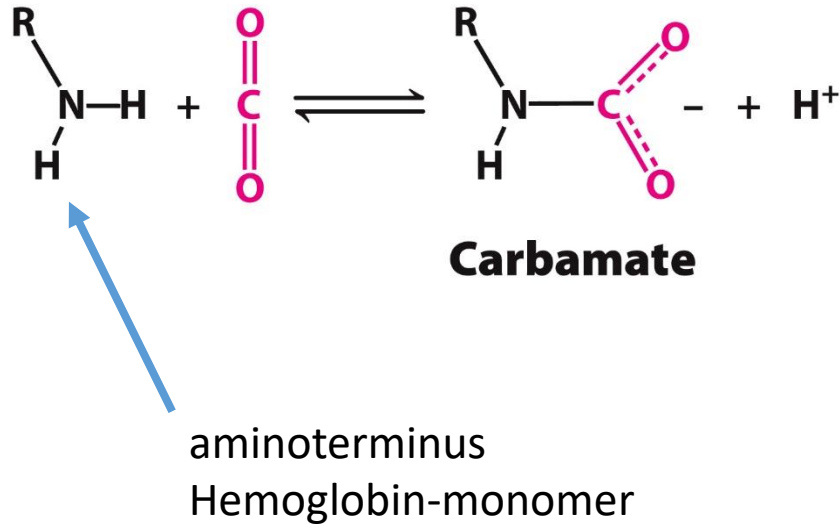
When the pH is lowered, H146 becomes protonated.

Interaction between H146 and D94 stabilises the T-state.

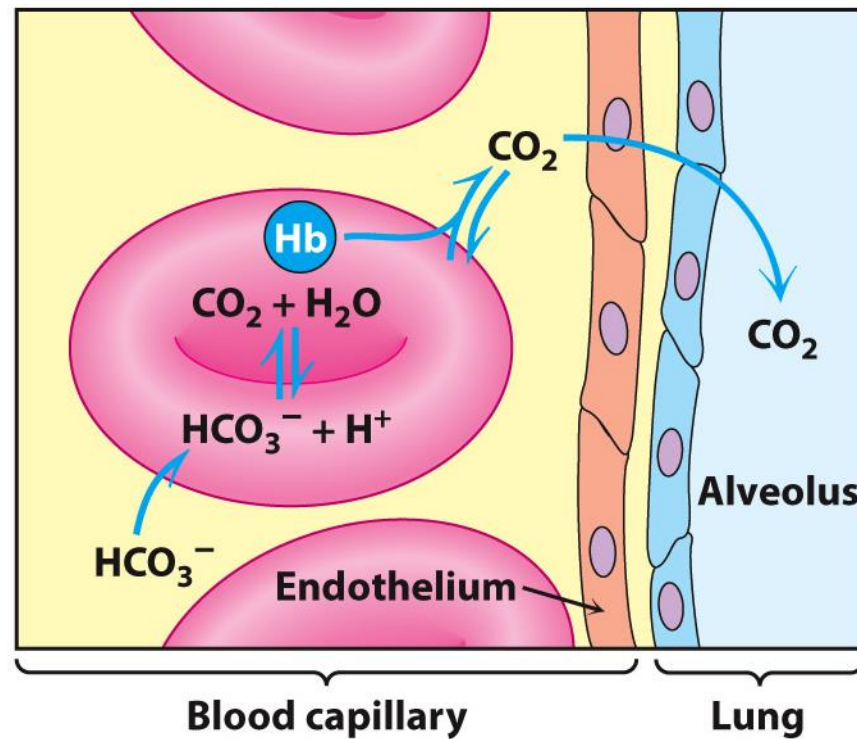
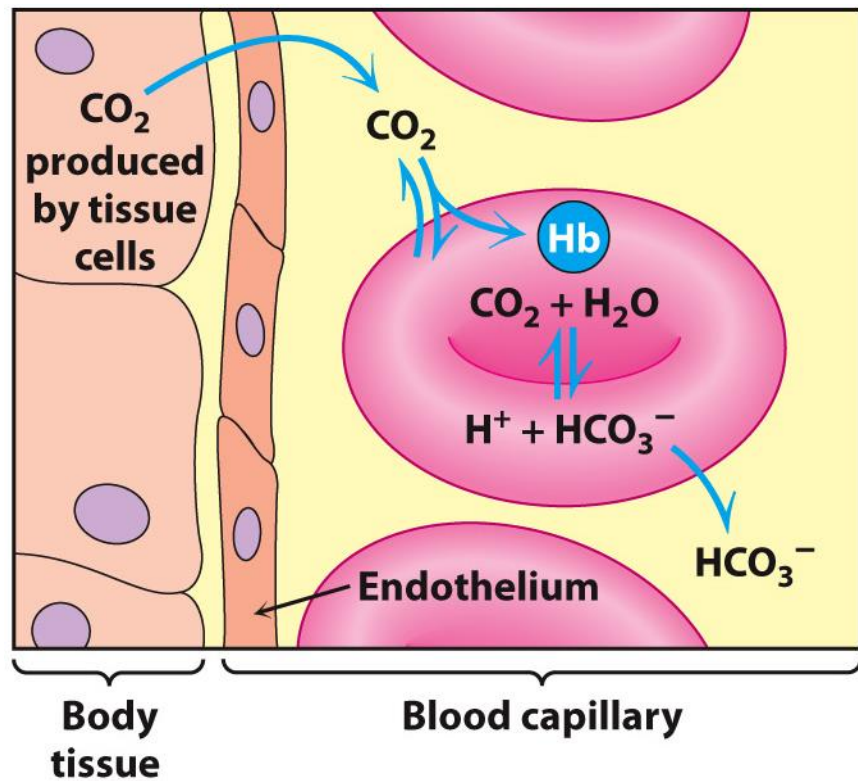
# CO<sub>2</sub> effects



# CO<sub>2</sub> effects



- Amino termini at the interface between  $\alpha\beta$  dimers
- Negatively charged carbamate groups participate in salt-bridges, characteristic of the T-state structure  $\rightarrow$  deoxyhemoglobin stabilized  $\rightarrow$  release of O<sub>2</sub> favored





# **Blackboard:**

Extra oefenopgaven over dit hoofdstuk.

Oefententamens

## **Volgende lessen:**

Hoofdstuk 11 en 12

Vragen