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Practice Set 3 Solution

## Chemistry, English & COMPUTER PRACTICE

**Topics:** 

Chemical Equations: Writing a Chemical Equation, Balanced Chemical Equations, Types of Chemical Reactions: Combination Reaction, Decomposition Reaction, Displacement Reaction, Double Displacement Reaction, Correction of Incorrect Words and Sentences, Basics of Computer System, Introduction to Internet HTML, Using MS - Word/MS-Excel/MS-Power Point

DDCET final exam weightage of this topic:

13 Questions (26 Marks)

**Total Practice sets** of this topic:

15 (sets) x 30 (questions) = 450 Questions

Total Practice tests of this topic:

5 ( exams ) x 20 ( questions ) = 100 Questions

Offline / Online during lecture :

4 (lectures) X 50 (Questions) = 200 Question



## Section 1:

Physical properties of Metals and Non-metals, Chemical Properties of Metals, How do Metals and Non-metals React?: Properties of Ionic compounds, Corrosion of Metals.

- 1. In which reaction does heat cause breakdown?
- A) Thermal decomposition  $\checkmark$
- B) Electrolytic decomposition
- C) Photo decomposition
- D) Catalytic reaction
- 2. What is required to start most decomposition reactions?
- A) Light
- B) Heat or electricity  $\checkmark$
- C) Water
- D) Air
- gas evolved is:
- A) NO
- B) NO2 **✓**
- C) N2O
- D) NH3
- 4. Which of the following is a displacement reaction?
- A)  $Zn + CuSO4 \rightarrow ZnSO4 + Cu \checkmark$
- B) NaOH + HCl → NaCl + H2O
- C)  $H2 + O2 \rightarrow H2O$
- D) CaCO3  $\rightarrow$  CaO + CO2
- 5.  $Zn + H2SO4 \rightarrow ZnSO4 + H2$  is an example
- A) Double displacement
- B) Combination
- C) Decomposition
- D) Displacement

- 6. Reaction: Cu +  $2AgNO3 \rightarrow 2Ag + Cu(NO3)2$  is:
- A) Combination
- B) Decomposition
- C) Displacement 🗸
- D) Double displacement
- 7. What is observed when iron nail is dipped in CuSO4 solution?
- A) Solution turns green
- B) Copper is deposited on nail
- C) Blue color fades
- D) All of the above  $\checkmark$
- 3. In decomposition of lead nitrate, the brown 8. Why does no reaction occur when Ag is added to CuSO4?
  - A) Ag is less reactive
  - B) Cu is more reactive
  - C) Both a and b 🗸
  - D) Ag and Cu react easily
  - 9. Which of the following is not a displacement reaction?
  - A) Mg + HCl  $\rightarrow$  MgCl2 + H2
  - B)  $Cu + AgNO3 \rightarrow Cu(NO3)2 + Ag$
  - C) FeSO4 + Cu  $\rightarrow$  No change
  - D) NaCl + AqNO3 → AqCl + NaNO3 ✓
  - 10. A student adds Fe to ZnSO4 solution. What happens?
  - A) Fe displaces Zn
  - B) No reaction  $\checkmark$
  - C) Zn deposits on Fe
  - D) Green solution forms



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## Section 1:

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- 11. Choose the correct spelling.
- A) Hightlight
- B) Highlight 🗸
- C) Hightlite
- D) Hightlite
- 12. Choose the correct spelling.
- A) Illusion 🗸
- B) Illution
- C) Illousion
- D) Illussion
- 13. Choose the correct spelling.
- A) Hight
- B) Height 🗸
- C) Hite
- D) Heigt
- 14. Choose the correct sentence.
- A) My career in marketing is going well.
- B) My career in marketing are going well.
- C) My careers on marketing are going well.
- D) My career on marketing going well.
- 15. Choose the correct sentence.
- A) My job is very demanding.  $\checkmark$
- B) My jobs is very demanding.
- C) My job are very demanding.
- D) My job were very demanding.
- 16. Choose the correct spelling.
- A) Fustration
- B) Frustration  $\checkmark$
- C) Frustrition
- D) Fustrition

- 17. Choose the correct sentence.
- A) The hospital is located on the hill.  $\checkmark$
- B) The hospital are located on the hill.
- C) The hospitals are located on the hill.
- D) The hospital was located on the hill.
- 18. Choose the correct spelling.
- A) Forign
- B) Foreign 🗸
- C) Foregin
- D) Foriegn
- 19. Choose the correct sentence.
- A) He has been promoted many times.  $\checkmark$
- B) He was promoted many times.
- C) He is promoted many times.
- D) He have been promoted many times.
- 20. Choose the correct sentence.
- A) I have sent my resume for the job application.  $\checkmark$
- B) I send my resume for the job application.
- C) I have sent my resume to the job application.
- D) I have sending my resume for the job application.
- 21. What is the function of a debugger in programming?
- A) To execute code
- B) To find and fix errors in code  $\checkmark$
- C) To compile code
- D) To optimize code



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## Section 1:

Physical properties of Metals and Non-metals, Chemical Properties of Metals, How do Metals and Non-metals React?: Properties of Ionic compounds, Corrosion of Metals.

- 22. What does the <code> tag represent in HTML?
- A) To create a code block
- B) To display inline code 🗸
- C) To define an external script
- D) To embed a font style
- 23. What is the function of the <iframe> tag in HTML?
- A) To define a form
- B) To embed another document within the page  $\checkmark$
- C) To create an image gallery
- D) To define a section
- 24. What does the <label> tag do in an HTML form?
- A) It groups form elements
- B) It provides a label for an input element  $\checkmark$
- C) It defines the form action
- D) It submits the form data
- 25. What is the correct HTML element to display a tooltip?
- A) <tooltip>
- B) <title> ✓
- C) <span>
- D) <mark>
- 26. India's first super computer is
- A) Saga 220
- B) Eniac
- C) Param 8000 **✓**
- D) Param 6000

- 27. Which one of these is called as the fastest memory?
- A) CD/DVD
- B) Cache Memory
- C) Floppy Disk
- D) External Hard Disk
- 28. What is the purpose of a subnet mask?
- A) To divide an IP network into smaller subnetworks  $\checkmark$
- B) To store data
- C) To execute programs
- D) To manage hardware
- 29. What is the function of a web crawler?
- A) To index web pages for search engines  $\checkmark$
- B) To browse the internet
- C) To execute applications
- D) To store data
- 30. allows the users of WWW to exchange information found on webpages on the Internet.
- A) SMTP
- B) TCP/IP 🗸
- C) TTP
- D) FTP