# 5500 Problem Set 3 Nikko Cleri September 21, 2020

### Question I.

Consider the chemical reaction  $nA + mB \iff A_nB_m$ . As the double arrow indicates, chemical reactions go both ways. Suppose this reaction happens at a fixed temperature and pressure. Show that the equilibrium conditions is  $n\mu(A) + m\mu(B) = \mu(A_nB_m)$ .

Since G is at a minimum at equilibrium

$$dG = n\mu(A) dN_A + m\mu(B) dN_B + \mu(A_n B_m) dN_{AB}$$
$$0 = n\mu(A) + m\mu(B) + \mu(A_n B_m)$$

Combining these give the desired  $n\mu(A) + m\mu(B) = \mu(A_n B_m)$ .

#### Question II.

A variation of the infamous lattice gas. Suppose we have a big volume V divided into cells of volume v. Each cell can accommodate at most one diatomic molecule. There are N (N >> 1, N << V/v) molecules, and by assumption there are no other degrees of freedom for a molecule except that it is in one of the cells. However, if supplied the dissociation energy D, a molecule will break up into two atoms, each of which can similarly occupy one of the cells. There are no constraints for an atom and a molecule occupying the same cell.

(a) Find the free energy (can't say if it is Helmholtz or Gibbs, as there are no volume/pressure type thermodynamic parameters in this problem) at temperature T, assuming that the gas is either all atoms, or all molecules.

We can find the number of microstates as

$$g = {V/v \choose N}$$

$$S = k \ln g$$

$$= k[\ln n! - \ln N! - \ln (n - N)!]$$

$$= k[\frac{V}{v} \ln \frac{V}{v} - N \ln N - (\frac{V}{v} - N) \ln (\frac{V}{v} - N)]$$

We want this as

$$\begin{split} \frac{\mathrm{d}S}{\mathrm{d}U} &= \frac{1}{T} \\ E &= N\epsilon \\ &= U \\ S(U) &= k [\frac{V}{v} \ln \frac{V}{v} - \frac{U}{\epsilon} \ln \frac{U}{\epsilon} - (\frac{V}{v} - \frac{U}{\epsilon}) \ln (\frac{V}{v} - \frac{U}{\epsilon})] \\ \frac{\mathrm{d}S}{\mathrm{d}U} &= \frac{k}{\epsilon} [\ln \frac{n\epsilon}{U} - 1] \\ &= \frac{1}{T} \\ U &= \frac{n\epsilon}{e^{\frac{\epsilon}{kT}} + 1} \end{split}$$

(b) Take it as given that there is a phase transition in which the molecules dissociate. What is the transition temperature?

From the first question of this homework we can say

$$\mu(A) + D = \mu(B)$$

$$\mu(A) = kT \ln \frac{N_A}{n - (N_A + N_B)}$$

$$\mu(B) = kT \ln \frac{N_B}{n - (N_A + N_B)}$$

$$kT \ln \frac{N_A}{n - (N_A + N_B)} + D = 2kT \ln \frac{N_B}{n - (N_A + N_B)}$$

The temperature follows.

#### Question III.

Take it as given that two hermitian operators may be diagonalized simultaneously if and only if they commute.

(a) As already noted, every operator A may be decomposed trivially in the form  $A = A_1 + iA_2$ , where  $A_1$  and  $A_2$  are hermitian. Suppose we have a normal operator N with the corresponding components  $N_1$  and  $N_2$ . Verify the following items: (i)  $[N_1, N_2] = 0$ . (ii) N may be diagonalized.

(i) For a normal operator N

$$N = N_1 + iN_2$$

$$[N, N^{\dagger}] = 0$$

$$NN^{\dagger} = (N_1 + iN_2)(N_1 - iN_2)$$

$$= N_1N_1 + iN_2N_1 - iN_1N_2 + N_2N_2$$

$$N^{\dagger}N = N_1N_1 + iN_2N_1 - iN_1N_2 + N_2N_2$$

$$[N_1, N_2] = NN^{\dagger} - N^{\dagger}N$$

$$= 0$$

(ii) If we suppose N is diagonalizable, we can act with N on  $|\psi\rangle$  as follows:

$$N |\psi\rangle = N\mathbb{I} |\psi\rangle$$

$$= N \left( \sum_{n} |n\rangle\langle n| \right) |\psi\rangle$$

$$= N \sum_{n} |n\rangle\langle n| |\psi\rangle$$

$$= \left( \sum_{n} c_{n} |n\rangle\langle n| \right) |\psi\rangle$$

Thus we know that N is diagonalizable since an operator with a spectral representation is diagonalizable.

(b) Conversely, suppose that an operator N can be diagonalized, with the eigenvalues  $c_n$  (not necessarily real) and the orthonormal eigenvectors  $u_n$ . Verify the following items: (i)  $(u_n, N^{\dagger}u_m) = c_n^* \delta_{nm}$ . (ii)  $N^{\dagger}u_m = c_m^* u_m$ . Therefore,  $N^{\dagger}$  can also be diagonalized, eigenvalues and eigenvectors  $c_n^*$  and  $u_n$ . (iii) N is normal.

(i)

$$\langle u_n | N^{\dagger} | u_m \rangle = (\langle u_m | N | u_n \rangle)^*$$

$$= \langle u_m | c_n^* | u_n \rangle$$

$$= c_n^* \langle u_m | u_n \rangle$$

$$= c_n^* \delta_{nm}$$

(ii)

$$Nu_m = \sum_n c_n |u_n\rangle\langle u_n| |u_m\rangle$$
$$N^{\dagger}u_m = \sum_n c_n^* |u_n\rangle\langle u_n| |u_m\rangle$$
$$= c_m^* u_m$$

(iii)

$$NN^{\dagger} = \sum_{n} c_{n} |n\rangle\langle n| \sum_{m} c_{m}^{*} |m\rangle\langle m|$$

$$= \sum_{nm} c_{n} c_{m}^{*} |n\rangle\langle n|m\rangle\langle m|$$

$$= \sum_{n} c_{n}^{*} c_{n}^{*} |n\rangle\langle n|$$

$$= N^{\dagger} N$$

Thus N is normal.

We have, again, the result that normal, and only normal, operators can be diagonalized.

## Question IV.

(a) Show that all eigenvalues of a unitary operator have unit modulus.

$$U |u_i\rangle = u_i |u_i\rangle$$
$$U |u_i\rangle = u_i |u_i\rangle$$

We can then take

$$\langle u_j | U^{\dagger} U | u_i \rangle = u_i u_j^* \langle u_j | u_i \rangle$$

$$(1 - u_i u_j^*) \langle u_j | u_i \rangle = 0$$

$$\therefore \langle u_j | u_i \rangle = \delta_{nm}$$

(b) Show that an operator U is unitary if and only if there is a Hermitian operator A such that  $U=e^{iA}$ .

$$\begin{split} U &= e^{iA} \\ &= \sum_{m=0}^{\infty} \frac{(iA)^m}{m!} \\ U^{\dagger} &= (e^{iA})^{\dagger} \\ &= \sum_{m=0}^{\infty} \frac{[(iA)^m]^{\dagger}}{m!} \\ &= \sum_{m=0}^{\infty} \frac{(-iA^{\dagger})^m}{m!} \\ &= \sum_{m=0}^{\infty} \frac{(-iA)^m}{m!} \\ &= e^{-iA} \\ U^{\dagger}U &= e^{-iA}e^{iA} \\ &= 1 \end{split}$$

Thus U is unitary.

In the other direction, assume U is unitary and find some A for which  $U=e^{iA}$  is Hermitian. We know that for some diagonal operator D we can say  $U=VDV^{\dagger}$ . We can then say that for some set of real numbers  $\{\theta_i\}$  that

$$D = \begin{bmatrix} e^{i\theta_1} & & \\ & \ddots & \\ & & e^{i\theta_1} \end{bmatrix}$$

We can then let

$$H = V \begin{bmatrix} \theta_1 & & \\ & \ddots & \\ & & \theta_1 \end{bmatrix} V^{\dagger}$$

$$e^{iH} = V \begin{bmatrix} e^{i\theta_1} & & \\ & \ddots & \\ & e^{i\theta_1} \end{bmatrix} V^{\dagger}$$

$$= U$$

$$H = H^{\dagger} :: \{\theta_i\} \in \mathbb{R}$$

So H is a Hermitian operator.