

# PS-4 Solution

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This is my GitHub link: [Nana Ama's GitHub Link](#)

## 1 Question 1

The heat capacity of a solid at temperature  $T$ , as expressed by Debye's theory, is given by

$$C_v = 9V\rho k_B(T/\theta_D^3) \int_0^{\theta_D/T} (x^4 \exp(x))/(\exp(x) - 1)^2 dx \quad (1)$$

where  $V$  is the volume,  $\rho$  is the number density of atoms,  $k_B$  is Boltzmann's constant and  $\theta_D$  is the Debye temperature.

At  $V = 1000 \text{ cm}^3$ ,  $\rho = 6.022 \times 10^{28} \text{ m}^{-3}$  and  $\theta_D = 428 \text{ K}$  at  $N = 50$  sample points, the function  $cv(T)$  was created using Gaussian quadrature which yielded the heat capacity at a specific temperature.

A plot of heat capacity against temperature ranging from  $T = 5 \text{ K}$  and  $T = 500 \text{ K}$  was made using the function above. The result is as shown in Figure 1.

## 2 Question 2

Using

$$T = \sqrt{8m} \int_0^a dx / \sqrt{V(a) - V(x)} \quad (2)$$

Gaussian quadrature was used to evaluate the integral within the function using  $N = 20$  points and a graph of the period against amplitude was plotted from  $a = 0$  to  $a = 2$ . The graph is shown below.

It can be seen from the graph in Figure 2 that as amplitudes increase, the oscillator gets faster.

## 3 Question 3

For  $n = 0, 1, 2, 3$  at  $x = -4$  to  $x = 4$ , the graph of  $\psi_n$  against  $x$  yields the result in Figure 3

For  $n = 30$  at  $x = -10$  to  $x = 10$ , the graph of  $\psi_n$  against  $x$  yields The uncertainty was found to be 2.3452078737858177

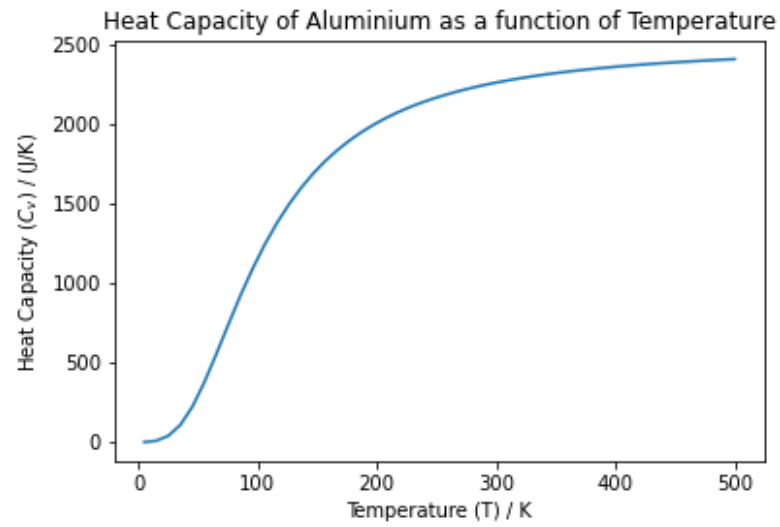


Figure 1: A graph of the heat capacity of aluminium against temperature

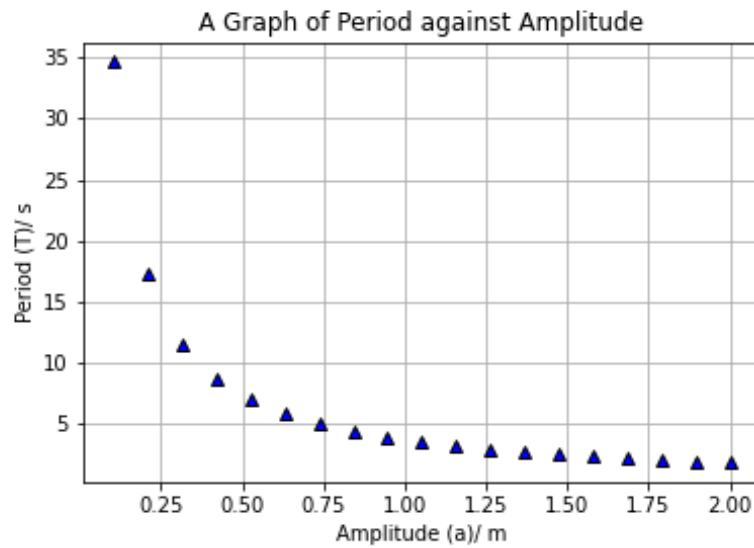


Figure 2: A graph of period against amplitude of an harmonic oscillator

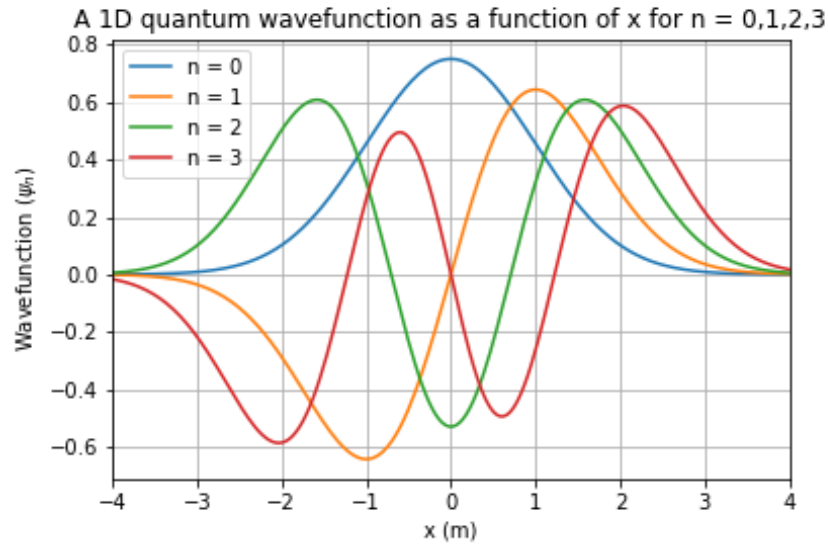


Figure 3: A graph of first 4 wavefunction plots of  $n$  against  $x$

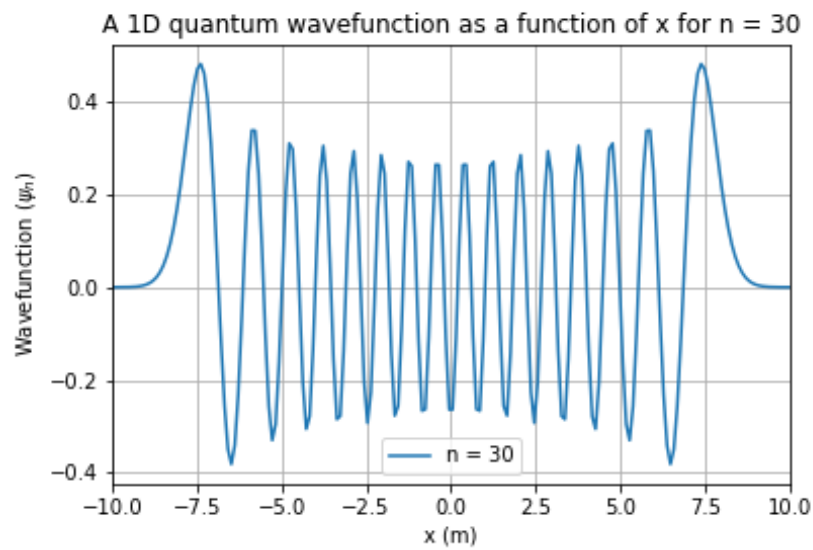


Figure 4: A plot of the wavefunction at  $n = 30$  against  $x$