

## VG100: INTRODUCTION TO ENGINEERING

### Thermal Engineering

Dr. Qiang Zhang



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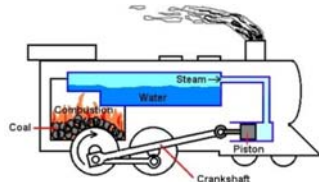
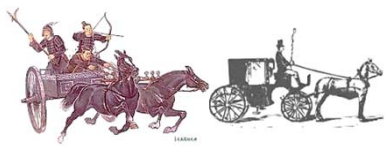
## Preview

- Combustion (Invited talk by Prof Lipo Wang)
- Thermal energy & power generation
- Thermoelectric Power Generation
- Introduction to Conduction



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## Revolution in Engineering



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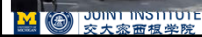
## Steam Powered...



(1769)



(1886)



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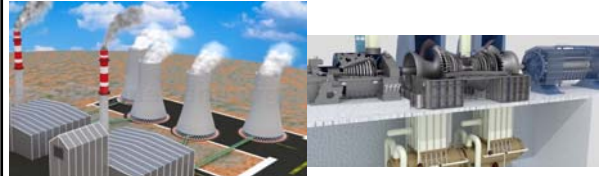
## Steam Powered...



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## Thermal power generation

Heat sources: coal, nuclear, solar, biofuels, etc.



- Efficiency is governed by the laws of thermodynamics.
- low 40% - 62%

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## Thermal engineers are in great need!

Year	2001	2050	2100
Global population (in billions of people)	6.15	9.4	10.4
Global energy consumption, TW	13.5	27.6	43.0

- Energy and water head the list of humanity's top 10 problems and 21<sup>st</sup> century innovation topics.
- Smalley R. E. Future global energy prosperity: the terra-watt challenge. *MRS Bulletin*. 2005. Vol. 30. No. 6. P. 412-417.
- S. Meieran E. S. *Predicting Challenges of the 21st Century*. Materials Engineering Impact, Purdue University, West Lafayette, IN, USA, 2006. P. 8-9.

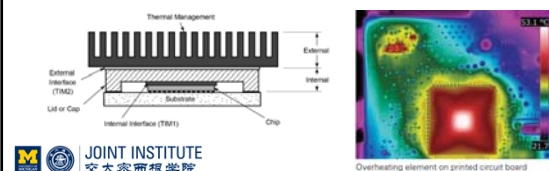
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## New challenges to next generation thermal engineers:

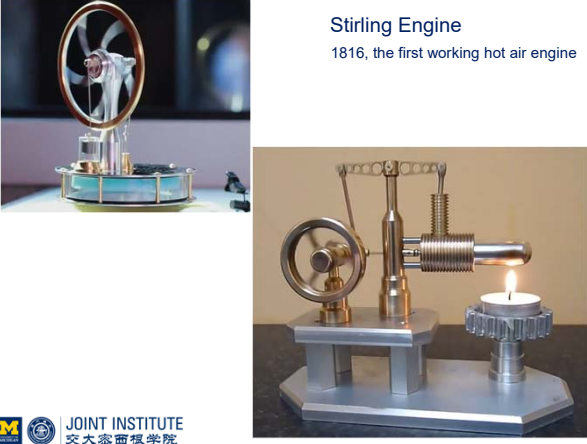
- Better understanding of the fundamentals!
- Multi-disciplines

### Areas to contribute:

- promote new efficient energy conversion concepts
- Energy friendly building systems ("net zero energy building")
- Renewable energy utilization.
- New technological challenges



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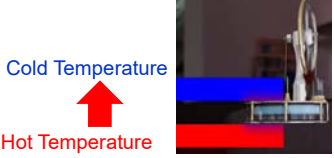


**Stirling Engine**  
1816, the first working hot air engine

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### Entropy, again...




- The more possible micro-states in a macro-state, the higher the entropy. The configuration with the highest entropy (i.e. the most micro-states) is the most probable.
- **Energy is always spreading out as entropy will always increase!**

*"The increase of disorder or entropy is what distinguishes the past from the future, giving a direction to time." - Stephen Hawking*

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### Entropy, again...



- What if you spin the wheel?
- Will this be against the 2<sup>nd</sup> law?

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### Back to our lab ...



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## The Seebeck Effect

- Thomas Seebeck (1821) found that an electric current would flow continuously in a closed circuit made up of two dissimilar metals, if the junctions of the metals were maintained at two different temperatures.



- Seebeck Coefficient:

$$S = dV / dT$$



- If these materials had been used at that time to construct a thermoelectric generator, it could have had an efficiency of around **3%** - similar to that of contemporary steam engines.

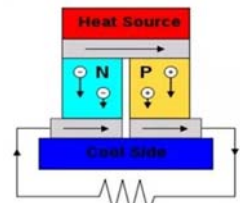
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## The Seebeck Effect

The good thermoelectric materials should possess

1. Large Seebeck coefficients  $S = dV / dT$
2. Electrical conductivity ?
3. Thermal conductivity ?

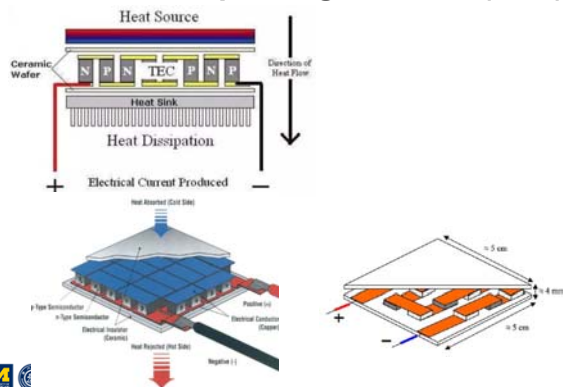
- Thermoelectric power generation (TEG) devices typically use special semiconductor materials, which are optimized for the Seebeck effect.



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## Thermoelectric power generation (TEG)



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## Thermoelectric power generation (TEG)

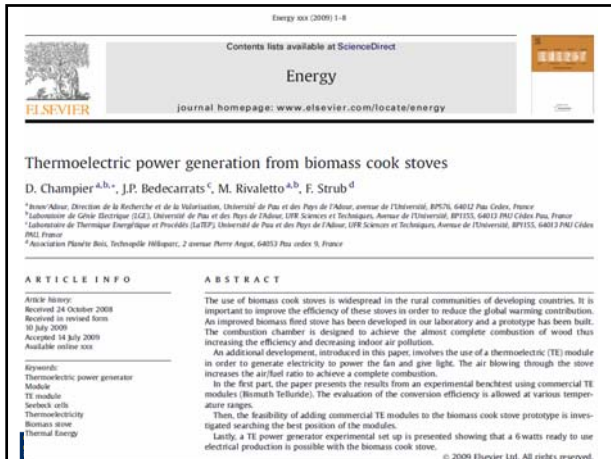


Turning heat into electricity  
Study finds topological materials could boost the efficiency of thermoelectric devices.

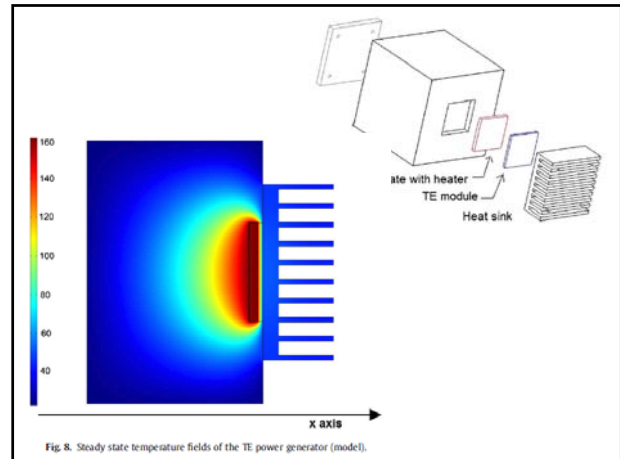
Jennifer Chu | MIT News Office  
January 16, 2018

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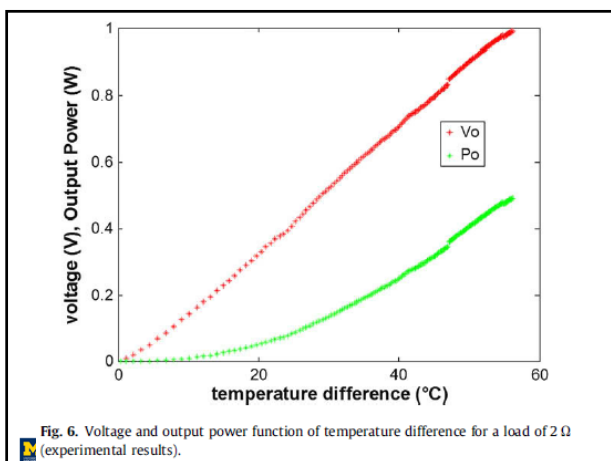
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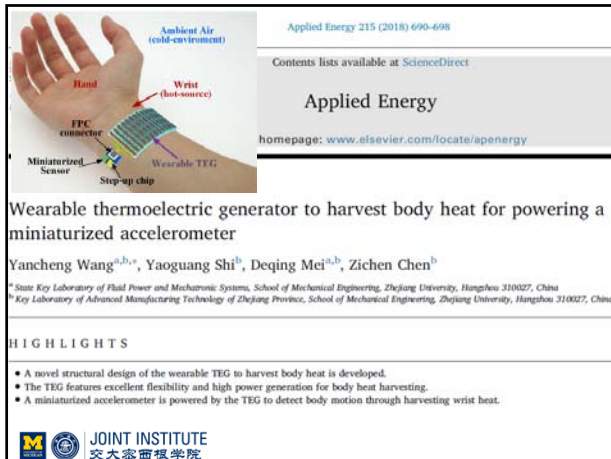
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Efficiency of TE power generation

$$\eta_{max} = \frac{T_h - T_c}{T_h} \frac{\sqrt{1 + ZT_{avg}} - 1}{\sqrt{1 + ZT_{avg}} + \frac{T_c}{T_h}}$$

- Figure of merit  $Z$  for TE material.  
 $Z = S^2 \sigma / k$  where  $S$ ,  $\sigma$ ,  $k$  are the Seebeck coefficient, electrical conductivity, thermal conductivity, respectively.

**Question:**  
 Calculate the maximum efficiency for an ideal TE material,  
 $ZT_{avg}=1$ ,  $T_h=250C$ ;  $T_c=20C$

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## Measuring Temperature

### Thermocouple

Voltage =  $f(\Delta T, \text{TC material})$

- Cold Junctions: A known ref temperature

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## Thermocouple

Characteristics of standard thermocouples

- Applications
- Cost
- Accuracy

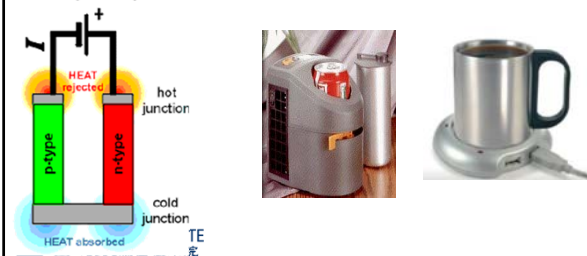
Type	Metals
E	Chromel vs. Constantan
J	Iron vs. Constantan
K	Chromel vs. Alumel
R	Platinum vs. Platinum 13% Rhodium
S	Platinum vs. Platinum 10% Rhodium
T	Copper vs. Constantan

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## The Peltier Effect

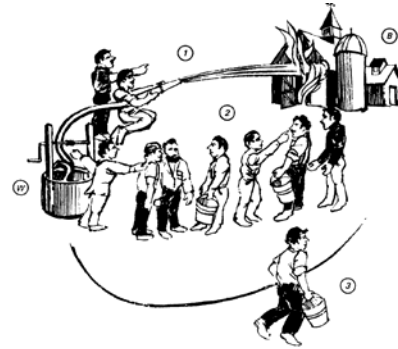
- In 1834, a French watchmaker and part time physicist named Jean Peltier described thermal effects at the junctions of dissimilar conductors when an electrical current flows between the materials.

Electrical power input



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## Introduction to Heat Transfer



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## Introduction to Conduction



Direct transfer of energy via

- lattice vibrations
- electron movement
- molecular collision

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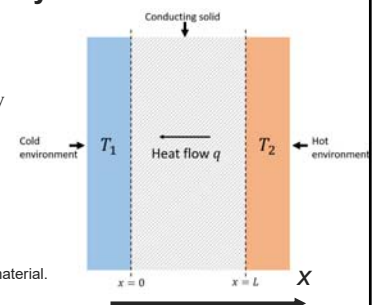
## Thermal Conductivity

### Fourier's law

- An observation for many materials.

$$q = -k (T_2 - T_1) / L$$

k: thermal conductivity,  
a physical property of the material.



**Question: What is the unit of k?**

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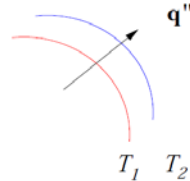
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## Fourier's law

$$q'' = -k \frac{\partial T}{\partial s}$$

$$\mathbf{q}'' = -k \nabla T$$



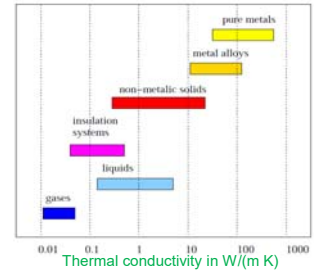
- proportional to the local gradient of temperature
- a vectorial quantity, which points in the direction opposite the temperature gradient, i.e. heat goes from hot to cold.
- heat flux lines are perpendicular to *isotherms*.



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## Thermal Conductivity

- Pure substances (gold, synthetic diamond carbon) have the highest conductivities, highly desirable for heat sink applications in computing. High conductivities are associated with regular lattices with little structural defects.



- Insulating materials are typically amorphous, such as glasses and ceramics.
- The thermal conductivity of solids and liquids can either increase or decrease with temperature.

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## Thermal Diffusivity

$$\alpha = \frac{k}{\rho c}$$

- $k$  thermal conductivity (W/(m·K))
- $c$  specific heat (J/(kg·K))
- $\rho$  density (kg/m<sup>3</sup>)

- Thermal diffusivity is the measure of thermal inertia.
- What does high thermal diffusivity mean?  
Heat moves rapidly through it because the substance conducts heat quickly relative to its volumetric heat capacity or 'thermal bulk'.

○ Copper at 25 °C	$1.11 \times 10^{-4}$
○ Iron	$2.3 \times 10^{-5}$
○ Glass, window	$3.4 \times 10^{-7}$
○ Wood (Yellow Pine)	$8.2 \times 10^{-8}$

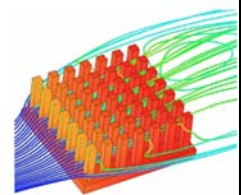
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## Extended Surfaces

- "Fins" create additional heat transfer area in a compact volume to enhance heat transfer.

Fins are used in:

- computer heat sinks, engine radiators, home heaters, heat exchangers, etc.



- The price to be paid?

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## Our core project

- How's your design going?  
Get it going and improve ver 2.0
- Any "spare tire plan" ?
- Teamwork !
- Understanding !
- A "team log book"



## Some Safety Rules (Again!)

- **Absolutely NO out-of-lab testing** (playing with alcohol burner lamp outside of the lab is forbidden and will lead to failure of this course)
- The lab supervisor/manager, technicians, and teaching assistants will ensure that you know of specific hazards and use personal protection equipment (PPE).
- Read and obey all operational signs and warnings.
- Power must be switched off whenever an experiment or project is being assembled or disassembled.
- Make measurements in live circuits with well-insulated probes and one hand behind your back. Do not allow any part of your body to contact any part of the circuit or equipment connected to the circuit.
- Never handle wet, damp or ungrounded electrical equipment.
- Avoid contact with the hot components.
- Never short-circuit a power source.
- When using a voltmeter or ammeter, begin with the highest range and work your way down to a suitable range.

## Review

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