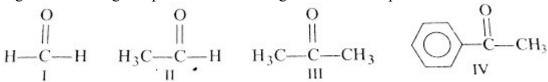


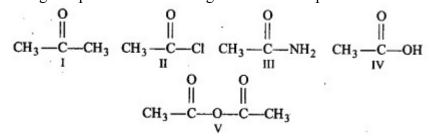
TOPIC: CARBONYL COMPOUNDS

DPP

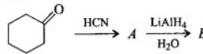
1. Arrange the following compounds in decreasing order of nucleophilic addition reaction



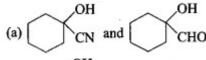
- 1) II > IV > III > I
- 2) I > II > III > IV
- 3) IV > III > II > I
- 4) II > III > IV > I
- 2. Arrange the following compounds in decreasing order of nucleophilic addition reaction



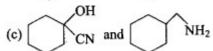
- 1) II > V > I > IV > III 2) III > IV > I > V > II 3) II > I > V > III > IV 4) IV > III > V > II > IV
- 3. In the given reaction sequence $C_6H_5CHO \xrightarrow{H_2N-OH} A \xrightarrow{P_2O_5} B$, A and B respectively are
 - 1) $C_6H_5 CH = N OH, C_6H_5CN$
- 2)
 - $\| C_6H_5 CH = N OH, C_6H_5C NH,$
- 3) $C_6H_5 CH = N OH, C_6H_5CHO$
- 4) $C_6H_5 CH = N OH, C_6H_5 COOH$



- 4. In the given reaction
- A and B will respectively be:

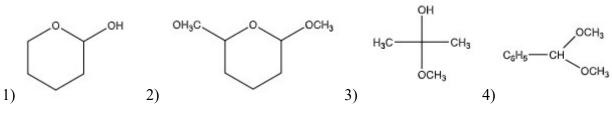


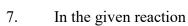
(b) CN and CH_2NH_2



(d) and NH

- 5. In the given reaction
- $\xrightarrow{1. \text{ HCHO/H}^{\oplus}} X$, X is
- (a) OH
- (b) OH OH
- (c) OH
- (d) OH
- 6. Which of the following structures contains a hemiacetal group?





$$\frac{\text{NBS}}{\text{ether}} \xrightarrow{\text{Hg}} \frac{\text{CO}_2}{\text{H}^{\oplus}/\text{H}_2\text{O}} (X)$$

, (X) will be



COOH

8. What are A, B and C in the given reaction?

in all cases

9.

in all cases

(B) and (C)

OH

$$\frac{\overset{\oplus}{\text{NaC}} \overset{\ominus}{=} \text{C--CH}_3}{\text{H}^{\oplus}/\text{H}_2\text{O}} X \xrightarrow{\text{H}_{,2}\text{Pd}-\text{BaSO}_4} Y$$

, Identify Structure of Y:

$$\begin{array}{c|c}
& \text{OH} \\
& \text{C-C} \\
& \text{CH}_3
\end{array}$$

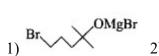
$$C = C < H$$

$$\begin{array}{c}
O \\
\hline
NaCN \\
HCI
\end{array}$$
 (X)

10.
$$\xrightarrow{\text{NaCN}} (X)$$
, (X) is:

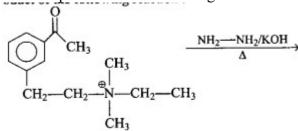
$$Br \xrightarrow{O} \frac{1. \text{ CH}_3\text{MgBr}}{2. \text{ H}_3^{\oplus}\text{O}} (X)$$

$$(X)$$
 will be





12. Find the product of the following reactions:



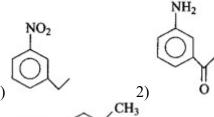
- 2)
- 3)
- 4) None of these

- $\xrightarrow{\text{Zn-Hg}} \text{Product}:$
- 3) но
- 4) CI

14.

15.

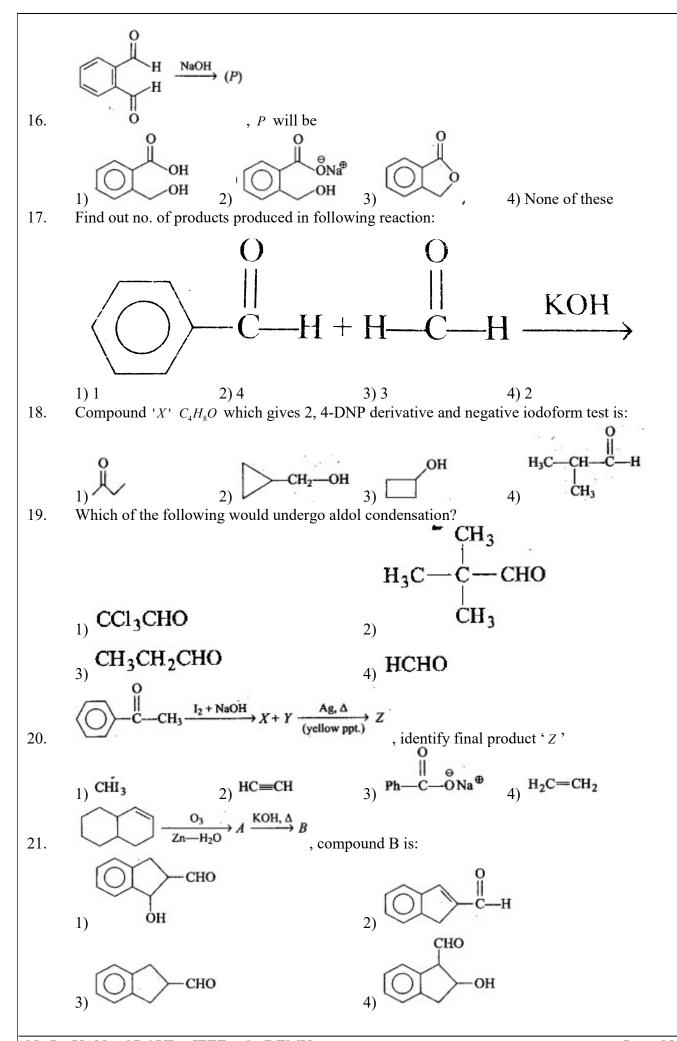
13.

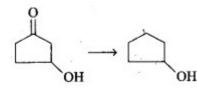


- N=0
- 4) NH₂

$$X \xrightarrow{\text{KOH}} CH_3$$

- , find out the structure of X
- 2)
- 3)
- 4) None of these





Above conversion can be achieved by

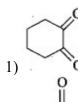
- 1) Wolff-Kishner reduction
- 2) Clemmensen reduction

3) LiAlH₄

4) $NaBH_4$

$$\frac{0}{2. \text{LiAlH}_4} \xrightarrow{\text{1. HCN}} (A);$$
3. NaNO₂ + HCl

, product A will be







4) None of these

$$C$$
— $CHBr_2$
 $NaOH(aq.)$
 $Product:$

24.

22.

23.

25. Give the correct sequence of reagents for the following conversion

$$Br \xrightarrow{O} \begin{array}{c} O \\ \parallel \\ C \xrightarrow{C} CH_3 \xrightarrow{?} HO \xrightarrow{CH_2 \xrightarrow{C} CH_2} \xrightarrow{O} \begin{array}{c} O \\ \parallel \\ C \xrightarrow{CH_3OH \text{ (excess)}} \end{array} \xrightarrow{Mg/\text{ether}} \begin{array}{c} O \\ \parallel \\ O \xrightarrow{H_2O/H_3^{\oplus}O} \end{array}$$

2) $\xrightarrow{\text{Mg/ether}}$ $\xrightarrow{\text{O}}$ $\xrightarrow{\text{H}_2\text{O/H}_3^{\oplus}\text{O}}$ $\xrightarrow{\text{CH}_3\text{OH/H}^{\oplus}}$

$$\xrightarrow{\text{CH}_3\text{OH (excess)}} \xrightarrow{\text{Mg/ether}} \xrightarrow{\text{O}} \xrightarrow{\text{H}_2\text{O/H}_3^{\oplus}\text{O}}$$

$$4) \xrightarrow{\text{H}_2\text{O/H}_3^{\oplus}\text{O}} \xrightarrow{\text{CH}_3\text{OH (excess)}} \xrightarrow{\text{Mg/ether}} \xrightarrow{\text{O}}$$

26. The product formed in the reaction is:

27. Find out number of substrates those cannot undergo Cannizzaro's reaction.

28. Examine the structural formulas of compounds given below and identify number of compounds which show positive iodoform test.

29. Of the following compounds, how many would give positive test with Tollen's reagent?

30. Consider the following reactions and identify how many reactions can give carbonyl compounds as major product.

CHEMISTRY

1-10	2	1	1	2	1	1	4	4	1	4
11-20	3	2	1	1	2	3	2	4	3	2
21-30	2	1	3	2	3	1	5	6	6	5

HINTS

- 1. Nucleophilic addition reaction of carbonyl group directly proportional to electron withdrawing group on carbonyl carbon
- 2. better leaving group favour to the nucleophilic addition reaction
 Better leaving group to be weak base.
- 3. $CHO \xrightarrow{NH_2OH} CH = N OH \rightarrow CN$
- 4. step-1: gives cynohydrine
 In that cynohydrine, CN group reduces to primary amine. (*CH*₂*NH*₂)
- 5. Formaldehyde react with acid gives carbo cation. $\binom{\oplus}{CH_2OH}$), carbocation is the electrophile.
- 6. Aldehydes are react with mono hydric alcohols gives hemiacetol. Hemiacetol containing *OH* and *OR* on same carbon. Ketones are react with dihydric alcohols gives cyclic ketal
- 7. Bromination carry at allyl position. Allyl bromides is converted to Grignard reagent. Grignard reagent react with *CO*₂ followed by acid hydrolysis. Gives carboxylic acid.
- 8. for the A, hydrogenation carry at double bond. For the B, $LiAlH_4$ unable to reduce C = C, for the C, Hydrogenation carry at C = C and Carbonyl. Order of the hydrogenation, alkyne > alkene > Carbonyl group.
- 9. step-1: nucleophilic addition reaction. Propynlide attacking at carbonyl carbon, carbonyl oxygen convert into the alcohol.
 - Step-2: C = C undergo mild reduction with lindar catalyst gives cis alkene.
- 10. Conceptual
- 11. Ketones are react with gridnard reagent followed by acid hydrolysis gives tert-alcohol. An alcohol react with bromide ion loss *HBr*
- 12. Wolf Kishner reduction carbonyl is reduced to CH_2 including carry the dehydro halogenation and Hoffman elimination.
- 13. In Clemennson reduction, carbonyl reduces to CH_2 , including OH is replaced by Cl and NO_2 reduces to NH_2
- 14. In Clemennson reduction, carbonyl reduces to CH_2 , including OH is replaced by Cl and NO_2 reduces to NH_2
- 15. Intra aldol condensation carbonyl group is reacted with α hydrogens
- 16. Intra molecular cannizaros
- 17. cannizaro and intra molecular cannizaro reaction
- 18. aceto group containing carbonyl compounds can give both iodoform and 2,4 DNP test
- 19. α hydrogen containing aldehydes and ketones involve aldol condensation.
- 20. aceto group containing carbonyl compounds can gives iodoform. Iodoform react with silver gives acetylene.
- 21. intra molecular aldol condensation

22.	Wolf Kishner reduction carbonyl is reduced to CH_2 including carry the dehydro hat Hoffman elimination	logenation and
23.	Conceptual	
24.	Intra molecular cannizaro	
25.	Conceptual	
26.	Conceptual	
27.	Without α – hydrogen containing aldehydes are involve in cannizaro	
28.	aceto group containing carbonyl compounds can gives haloform.	
29.	Aliphatic and aromatic aldehydes involve Tollen's reaction.	
30.	Conceptual	
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