

Electromagnetism:

One of the 4 fundamental forces in nature:

Gravity Newton (1687) + Einstein (1915)

EM Maxwell (1860)

Weak Fermi (1930)

Strong Feynman, Gross, Politzer, Wilczek(1970)

- Gravity always around but hard to think about.
- EM effects much more evident: lighting, magnets, sparks from clothing,...
- Known since at least 3000bc (Egyptians) via electric eels
- Electricity comes from the greek word Elecktron (Amber) "from the sun"



Electromagnetism, electron, electronics,...



Piece of Amber

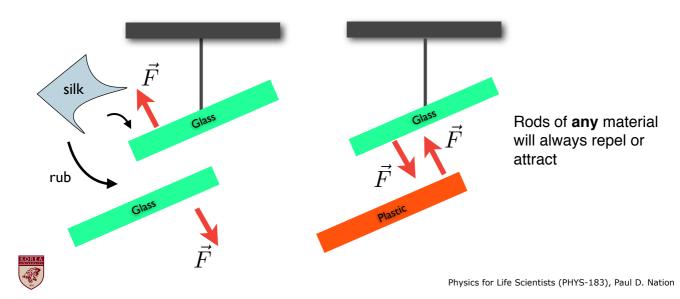
- Connection between electricity and magnetism not until 1820 from Oerstad.
- Now everything takes advantage of electromagnetism
 - Entire output of the world connected to EM



Electric Charge

- Every object contains electric charge.
- Charge is a fundamental property of the material out of which everything is made, just like mass.
- Everyday objects have vast amounts of charge
- The properties of charges first determined experimentally

Simple experiment:



- Situation can be described by two different types of charge that are related by a minus sign:



Charges with same sign repel; opposite charges attract

- Choice of +/- for charges is arbitrary. Determined by Benjamin Franklin (\$100 guy)

Positive (+) charges

Charges usually come in equal amounts (no <u>net</u> charge)

- Objects with no net charge are called neutral
- Have seen that objects can have net charge; determined types of charge
- How are charges distributed over objects?



Conductors & Insulators

- Materials can be divided into 4 categories depending on how easily charges move:

Conductors: Charges move easily. i.e. metals, saltwater, people,...

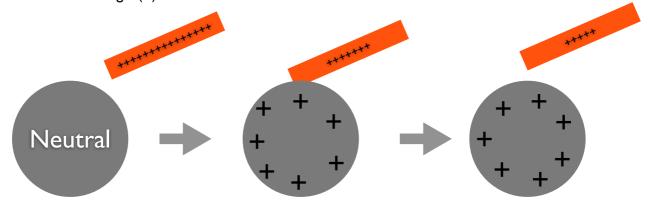
Insulators: Charges cannot move. i.e. rubber, glass, plastics, pure H2O

Semiconductors: In between conductors and insulators. i.e. silicon, germanium

Superconductors: Charges move perfectly. i.e mercury@2k, <u>aluminum@1.2k</u>

Conductors

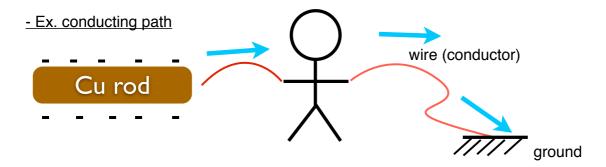
- Charges move freely + like charges repel.
- Add some charge (+) onto a conductor



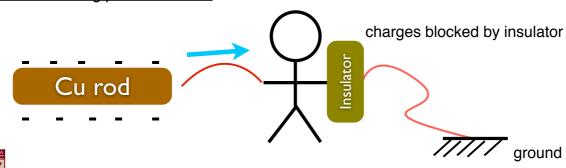


Charge spreads out over conductor to reach equilibrium

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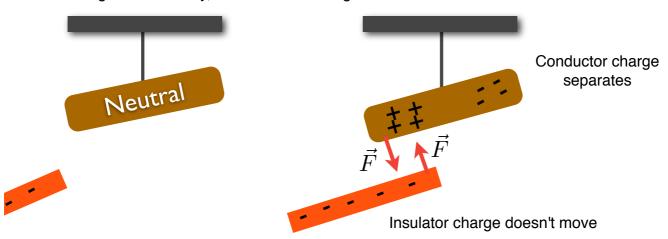


- Charge moves through conductors due to repelling force to ground
- Earth is a huge conductor... can always dump charge there.
 - Any conducting path connected to the Earth is called **grounded**
- Ex. conducting path + insulator

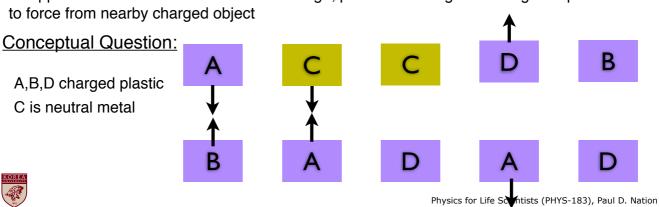




- Since charges move freely, can **induce** a charge on a conductor:

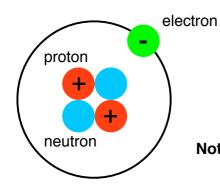


- Copper rod is neutral but had induced charge; positive and negative charges separated due



Where do charges come from?

- Recall that charge is a fundamental property of matter.



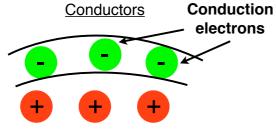
+ charges = protons

- charges = electrons

1st guess:

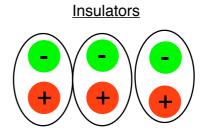
- Protons are the charges that move around

Not True! Only electrons are free to move in most materials



fixed in place

Electrons do not belong to single atom, spread out over surface



- Electrons tightly bound to nucleus, cannot move.
- None or few conduction electrons



Only way to get positive charge is to remove electrons such that:

of protons > # of electrons

Positive charge implies deficit of electrons

- Electrons are fundamental particles; they can not be subdivided (Thompson 1897)



Charge comes only in discrete units

charge
$$\rightarrow q=n\cdot e \qquad n=\pm 1,2,3\dots$$
 "elementary charge" $e=1.602\times 10^{19}{\rm C}$ Coulombs

- Elementary charge e is a fundamental constant

$$e = 1.602 \times 10^{-19} \text{C}$$
 1C $\approx 6.25 \times 10^{18} \text{e}$

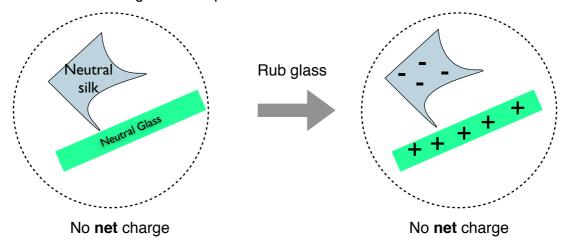


 $7.5 imes 10^{18}\,$ Grains of sand on the entire Earth!

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Charge is a conserved quantity.

- Let's return to the silk + glass example.



- Charge is **not** created, just moved around
- True for any physical process.

Conservation of charge: The **net** charge of <u>any</u> isolated system cannot change.

- 4th conservation law encountered: Energy, Momentum, Angular Momentum, Charge
- Key word here is "net": Can create +/- charges in pairs; net charge is still zero.



Coulombs Law:

- We know everything there is to know about individual charges
- How do charges interact?
- Saw that charges exert forces on each other:

Opposite charges attract Like charges repel

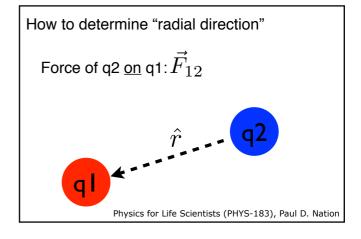
- Force between non-moving charges is called **Electrostatic Force**.
- Determined experimentally (like most of EM) by Coulomb (1785)

Coulomb's Law: Given two particles with charges q_1 and q_2 , the resulting force is:

$$\vec{F} = k \frac{q_1 q_2}{r^2} \hat{r} \, \text{\swarrow}^{\text{in radial direction}}$$

"electrostatic constant"

$$k = 9 \times 10^9 \mathrm{C}^2/\mathrm{N} - \mathrm{m}^2$$





- Check for consistency:

$$q_1 = e \quad ; \quad q_2 = -e$$

$$ec{F}_{12}$$
 \hat{r} $ec{r}$ $ec{r}$ $ec{F}_{12} = rac{-ke^2}{r^2}\hat{r}$

$$\vec{F}_{12} = \frac{-ke^2}{r^2}\hat{r}$$

$$q_1 = e \quad ; \quad q_2 = e$$

$$\vec{F}_{12} = \frac{ke^2}{r^2} \hat{r}$$

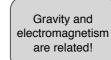


Connection between EM and Gravity:

$$ec{F}_{\mathrm{EM}} = k rac{q_1 q_2}{r^2} \hat{r}$$
 Coulomb's Law

$$ec{F}_{
m G} = G rac{m_1 m_2}{r^2} \hat{r}$$
 Newton's Law

- -Both equations have same form:
 - 1/r^2 spatial dependence
 - interactions correspond to products of particle properties; charges for Coulomb, and masses for Newton's law.
 - Both laws has a fundamental constant



- Amazing! Why should Gravity and EM follow same force law?
 - Hints at a deeper connection
 - This connection used by Einstein in formulating relativity



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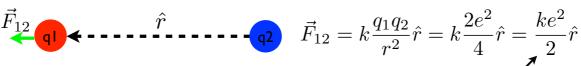
- Ex. force of gravity vs. EM force:
- Let us compare the force of gravity between an electron and a proton to the EM force.
- What is your guess? Which is stronger?

$$\frac{|F_{EM}|}{|F_G|} = \frac{k}{G} \frac{q_1 q_2}{m_1 m_2} = \frac{k}{G} \frac{e^2}{m_p m_e} \qquad \begin{array}{l} m_p = 1.67 \times 10^{-27} \mathrm{Kg} \\ m_e = 9.11 \times 10^{-31} \mathrm{Kg} \\ G = 6.67 \times 10^{-11} \mathrm{m}^3/\mathrm{Kg} - \mathrm{s}^2 \\ \frac{|F_{EM}|}{|F_G|} \approx \frac{9 \cdot 10^9 * 2 \cdot 10^{-38}}{7 \cdot 10^{-11} \times 2 \cdot 10^{-27} \times 9 \cdot 10^{-31}} = \frac{1}{7} \frac{10^{-29}}{10^{-68}} \approx \boxed{10^{39}}$$

- EM force is huge compared to gravity!
- As mentioned, objects nearly neutral, and mass of Earth is huge -> not noticed so much.

Ex. force of between two charges

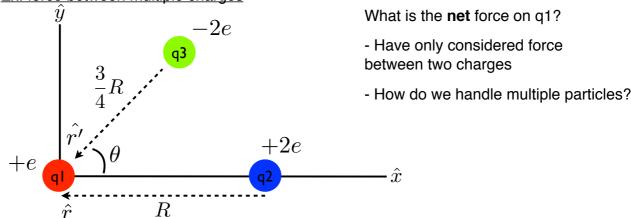
What is force on q1 from q2 when q1=e and q2=2e and separated by a distance of 4m?



in \hat{r} direction since both charges same sign



Ex. force between multiple charges



- Because we know that $ec{F}_{EM}$ has the same form as $ec{F}_{G}$ we can borrow the concept of superposition:

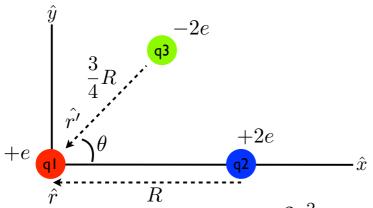
The **net** force from a collection of charges on charge q1 is given by:

$$\vec{F}_{1,\text{net}} = \sum_{i=2}^{N} F_{1i}$$



Superposition principle is not obvious. But it is confirmed by experiments.

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$$\vec{F}_{1,\text{net}} = \vec{F}_{12} + \vec{F}_{13}$$

- Find F12 (easy part first):
$$\vec{F}_{12}=krac{2e^2}{R^2}\hat{r}=-krac{2e^2}{R^2}\hat{x}$$

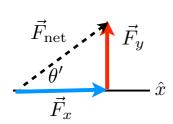
- F13:
$$\vec{F}_{13}=-k\frac{2e^2}{\left(\frac{3}{4}R\right)^2}\hat{r'}$$
 r' is not along x or y directions; must decompose into x and y pieces

$$\vec{F}_{13} = k \frac{16(2e^2)}{9R^2} \cos\theta \hat{x} + k \frac{16(2e^2)}{9R^2} \sin\theta \hat{y}$$

$$\vec{F}_{1,\text{net}} = k \frac{(2e^2)}{R^2} \left[\frac{16}{9} \cos \theta - 1 \right] \hat{x} + k \frac{16(2e^2)}{9R^2} \sin \theta \hat{y}$$



-What is the angle θ' of the resulting force?



$$\tan \theta' = \frac{F_y}{F_x} \qquad \theta' = \arctan \frac{F_y}{F_x}$$

$$\frac{F_y}{F_x} = \frac{16 \sin \theta}{9 (16/9 \cos \theta - 1)}$$

$$\frac{F_y}{F_x} = \frac{\sin \theta}{\cos \theta - \frac{9}{16}}$$

