

NTSE (Chemistry Sheet-1)

MATTER IN OUR SURROUNDINGS

1.	Which of the following statements is/are con						
	(A) Intermolecular forces of attraction in solids are maximum.						
	(B) Intermolecular forces of attraction in gases are minimum						
	(C) Intermolecular spaces in solids are mini(D) All of the above	mum					
2	The smell of perfume spreads out by a proc	ess known as -					
	(A) evaporation (B) diffusion	(C) condensa	ation (D) fus	sion			
3.	Densities of gases change with-		()				
	(A) temperature and volume	(B) pressure only					
	(C) temperature only	(D) pressure and ter	nperature				
4.	Carbogen is a mixture of -	(0) 00 00	(D) 00 0 0				
	(A) $CO \& CO_2$ (B) $CO \& O_2$	(C) CO ₂ & O ₂	(D) CS ₂ & O ₂	<u>}</u>			
5.	Pair of substances undergoing sublimation						
	(A) Bromine and Iodine		nd Nausadar				
^	(C) Sodium and Borax	(D) Naphthal	ene and Blue Vi	itriol			
Ь.	Dry ice is -	(P) solid carbon mor	aovido				
	(A) solid carbon - di - oxide(C) liquid carbon - di – oxide	(B) solid carbon mor(D) liquid oxygen	loxide				
7.	Boiling point of 10% aqueous solution of co		heric pressure is	3 -			
	(A) 100 ⁰ C	(B) > 100 ⁰ C					
	$(C) < 100^{\circ}C$	` ,	no di ot				
Ω	The change of state of substance from gas	(D) not possible to p	redict				
0.	(A) melting (B) boiling		(D) vaporization	on			
9.	Which of the following conditions is most far						
	(A) High pressure, low temperature	(B) Low pressure, lo	•				
	(C) Low pressure, high temperature	(D) High pressure, h	igh temperature	;			
10.	The boiling point of alcohol is 78 ⁰ C. What is	s this temperature is h	(elvin scale?				
	(A) 373 K (B) 351 K	(C) 341 K	(D) 78 K				
11.	Boiling point of a substance -						
	(A) decreases with decreasing the atmospheric pressure.						
	(B) decreases with increasing the atmospheric pressure.						
	(C) increases with decreasing the atmospheric pressure.(D) increases with increasing the atmospheric pressure.						
12.	Which of the following statements is not correct?						
	(A) A compound is a pure substance.						
	(B) A compound is homogeneous in nature.						
	(C) A compound always contains two or more elements.						
	(D) A compound can be separated into its constituent elements by some physical process.						
13.	If a substance can be separated into different kinds of matter by some physical process, then the						
	substance is - (A) an element (B) a molecule	e (C) a compo	und	(D) a mixture			
	TATALLERINELL (D) a MORCUR	: (C) a combo	นเน	TOTA HIIXIUITE	;		

14.	An element is made up of	only one kind of -						
	(A) atoms (B)	molecules	(C) compounds	(D) all of the	ese			
15.	What difference in the pro	perty of two miscil	ole liquids enables	s their separation by frac	tional distillation?			
	(A) Boiling point	(B) Melting po	int (C) Visco	sity (D) Density				
16.	The father of chemistry is	-						
	(A) Priestley (B)	Lavoisier	(C) Dalton	(D) Mendeleeff				
17.	A pure substance can only	y be-						
	(A) a compound		(B) an ele	ement				
	(C) an element or a compound (D) a heterogeneous mixture							
18.	Which one the following is	The state of the s						
		Ozone	(C) Carborundun	n D) Quicklime				
19.	Which one of the following							
	` ,	Ozone	(C) Silica	(D) Graphite				
20.	Which one of the following		orrect?					
	(A) All elements are homogeneous.							
	(B) Compounds always contain two or more different elements.							
	(C) A mixture is not always heterogeneous.							
	(D) Air is a heterogeneous mixture							
21.	One sample of air is found to have 0.03% carbon dioxide and another sample 0.02%. This illustrates							
	that -							
	(A) air is a compound.							
	(B) air is an element							
	(C) air does not follow the law of constant proportions.							
00	(D) air is a mixture							
22.	Which one of the following) is not mixture?	(D) 0 II I					
	(A) Distilled water		(B) Sugar dissolv	ed in water				
22	(C) Liquefied petroleum	ah aya ata yati a mwa	(D) Gasoline	uuraa an daaran ay mada 2				
23.	Which of the following is a characteristic property of both mixtures and compounds?							
	(A) Their properties are same as those of their components.							
	(B) Energy is releases when they are formed.							
	(C) Their masses are equal to the sum of the masses of their components.(D) They contain the components in fixed proportions.							
24	Whish of the following processes is a physical change?							
24 .			•	uoto				
	(A) Sodium dissolves in w	LET D	(B) Iron ru					
25	(C) Iron is magnetised (C) Magnesium wire is burnt A method of obtaining oxygen, which illustrates a physical change and does not involve a chemical							
25.	change is -							
	(A) electrolysis of water		(B) decor	nnosition HaOa				
	•			nposition H ₂ O ₂				
00	(C) heating potassium chlorida		` '	ng liquid air				
26.	Which one of the following	•	•	N = 1	(D) D ('			
	(A) Sublimation	(B) Combustion	n (C	c) Electrolysis	(D) Rusting			