

**NMAM INSTITUTE OF TECHNOLOGY, NITTE**  
*(An Autonomous Institution affiliated to VTU, Belagavi)*  
**First Semester B.E. (Credit System) Degree Examinations**  
**Make up Examinations - July - August 2021**

**20PH102 – ENGINEERING PHYSICS**

Duration: 3 Hours

Max. Marks: 100

**List of constants:** Velocity of light,  $c = 3 \times 10^8 \text{ ms}^{-1}$ , Planck's constant,  $h = 6.63 \times 10^{-34} \text{ Js}$ ,  
 Electron mass,  $m = 9.11 \times 10^{-31} \text{ kg}$ , Electron charge,  $e = 1.6 \times 10^{-19} \text{ C}$ ,  
 Permittivity of vacuum,  $\epsilon_0 = 8.85 \times 10^{-12} \text{ F/m}$ , Boltzmann constant,  $k = 1.38 \times 10^{-23} \text{ J/K}$ ,  
 Avogadro number,  $N_A = 6.023 \times 10^{26} / \text{kgmole}$ , Neutron mass  $m = 1.678 \times 10^{-27} \text{ kg}$

**Note: Answer Five full questions choosing One full question from each Unit.**

**Unit – I**

**Marks**   **BT\***   **CO\***   **PO\***  
                   6   L\*2        1    1,2

- |   |    |   |    |     |   |     |
|---|----|---|----|-----|---|-----|
| 1 | a) | What are matter waves? Mention their properties.  | 6  | L*2 | 1 | 1,2 |
| 4 | b) | Obtain the time independent Schrodinger wave equation for a particle in one dimensional potential well of infinite height and discuss about energy eigen values.                        | 10 | L2  | 1 | 1,2 |
| 4 | c) | An electron is bound in a one dimensional potential well of width 1Å, but of infinite wall height. Find its energy values in the ground state and also in the first two excited states. | 4  | L3  | 1 | 1,2 |
| 4 | a) | Derive an expression for group velocity on the basis of superposition of two travelling waves.  | 6  | L2  | 1 | 1,2 |
| 4 | b) | Derive the time independent Schrodinger wave equation.  | 10 | L2  | 1 | 1,2 |
| 4 | c) | Compare the momentum, and the kinetic energy of an electron with de Broglie wavelength of 1Å, with that of a photon with same wavelength.   | 4  | L3  | 1 | 1,2 |

**Unit – II**

6   L1        2    1,2

- |   |    |  |    |    |   |     |
|---|----|--|----|----|---|-----|
| 4 | a) | Explain the seven crystal systems with neat diagrams.  | 6  | L1 | 2 | 1,2 |
| 4 | b) | Describe the procedure to determine the Miller indices of crystal planes. Derive an expression for inter - planar spacing of a crystal.  | 10 | L2 | 2 | 1,2 |
| 5 | c) | Calculate the glancing angle of incidence of x-rays of wavelength of 0.58Å on the plane (1 3 2) of NaCl, which results in second order diffraction maxima taking the lattice spacing as 3.81Å. | 4  | L3 | 2 | 1,2 |
| 5 | a) | Explain the origin of continuous and characteristic x-ray spectrum.  | 6  | L2 | 2 | 1,2 |
| 5 | b) | Define packing factor and coordination number. Calculate the coordination number and packing factor for bcc and fcc structures.  | 10 | L2 | 4 | 1,2 |
| 5 | c) | Nickel has fcc structure with lattice constant 3.52Å. Calculate the inter-planar spacing for (a) (101) planes, and (b) (123) planes.   | 4  | L3 | 2 | 1,2 |

**Unit – III**

6   L2        3    1,2

- |   |    |   |    |    |   |     |
|---|----|---|----|----|---|-----|
| 5 | a) | Mention the assumptions and limitation of free electron theory.   | 6  | L2 | 3 | 1,2 |
| 5 | b) | What are relaxation time and collision time? On the basis of free electron theory of metals, obtain an expression for the electric conductivity of the metal.   | 10 | L2 | 3 | 1,2 |
| 5 | c) | Estimate the relaxation time of conduction electrons in silver from following data:<br>Resistivity = $1.6 \times 10^{-8} \text{ ohm. m}$ Density = $10.5 \times 10^{-3} \text{ kg/m}^3$ , Atomic weight = 107.88. | 4  | L3 | 3 | 1,2 |

20PH102

Make up – July – August 2021

6. a) Distinguish between Type I and Type II superconductors.  
 b) Discuss briefly the BCS theory of superconductivity. Explain any two applications of superconductivity.  
 c) Superconducting tin has a critical magnetic field of 217 gauss at 2 K. If the critical temperature for superconducting transition for tin is 3.7 K, find the critical, magnetic field at 3K.

6 L2 4  
 10 L2 3  
 4 L3 3

## Unit – IV

7. a) Describe an extrinsic and intrinsic semiconductor with neat diagram.  
 b) Define Fermi energy. Sketch the Fermi level in (i) intrinsic semiconductor (ii) n-type semiconductor and (iii) p-type semiconductor. Discuss the effect of temperature on Fermi level in intrinsic and n-type Semiconductor.  
 c) The Hall co-efficient of a specimen of a doped silicon is found to be  $3.66 \times 10^{-4} \text{ m}^3/\text{Coulomb}$ . The resistivity of the specimen is  $8.93 \times 10^{-3} \text{ ohm. m}$ . Find the mobility and density of the charge carrier, assuming single carrier conduction.

6 L2 4  
 10 L3 4

8. a) Explain the construction and working of a LED with neat diagrams.  
 b) Describe the formation of p-n junction. Explain its voltage current characteristics. What is the effect of temperature on conductivity of Intrinsic semiconductors?  
 c) Calculate the barrier potential existing at the junction of p-type silicon with an acceptor concentration of  $10^{23} \text{ m}^{-3}$  and n-type silicon with a donor concentration of  $10^{23} \text{ m}^{-3}$  if the intrinsic concentration at 300K is  $1.4 \times 10^{16} \text{ m}^{-3}$ .

4 L3 4  
 6 L2 4  
 10 L1 4

## Unit – V

9. a) Describe the construction and working of a He-Ne laser.  
 b) Describe the attenuation in the optical fiber. What are the advantages of optical communications over other conventional types of communication?  
 c) The ratio of the population of two energy levels is  $1.059 \times 10^{-30}$ . Find the wavelength of light emitted at 300K.
10. a) Explain the terms with neat energy level diagram. i) population inversion, ii) metastable state and iii) stimulated emission.  
 b) Describe the types of optical fibers and modes of transmission with suitable diagrams.  
 c) A step index optical fiber 63.5  $\mu\text{m}$  in core-diameter has a core of refractive index 1.53 and a cladding of index 1.39. Determine (i) the numerical aperture of the fiber and (ii) the critical angle for core-cladding interface.

6 L2 5  
 10 L3 5  
 4 L1 5  
 6 L2 5  
 10 L4 5  
 4 L1 5

BT\* Bloom's Taxonomy, L\* Level; CO\* Course Outcome; PO\* Program Outcome

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# NMAM INSTITUTE OF TECHNOLOGY, NITTE

(An Autonomous Institution affiliated to VTU, Belagavi)

First / Second Semester B.E. (Credit System) Degree Examinations

Make up/Supplementary Examinations – September 2021

## 20PH102 – ENGINEERING PHYSICS

Duration: 3 Hours

Max. Marks: 100

List of constants: Velocity of light,  $c = 3 \times 10^8 \text{ ms}^{-1}$ , Planck's constant,  $h = 6.63 \times 10^{-34} \text{ Js}$ ,  
Electron mass,  $m = 9.11 \times 10^{-31} \text{ kg}$ , Electron charge,  $e = 1.602 \times 10^{-19} \text{ C}$ ,  
Boltzmann constant,  $k = 1.38 \times 10^{-23} \text{ J/K}$ ,  
Avogadro number,  $N_A = 6.023 \times 10^{23} / \text{kg mole}$ .

Note: Answer any Five full questions.

Marks BT\* CO\* PO\*

- |   |    |  |    |           |   |      |
|---|----|--|----|-----------|---|------|
| 5 | a) | What are matter waves? Derive an expression for de Broglie wavelength of an electron of mass $m$ , accelerated by a potential of $V$ volts.  | 6  | L*2       | 1 | 1, 2 |
| 5 | b) | Derive Schrodinger's time independent one-dimensional wave equation for a particle of mass $m$ with energy $E$ .   | 10 | L3        | 1 | 1, 2 |
| 5 | c) | An electron is trapped in a one-dimensional region of length $1.2 \text{ \AA}$ . Calculate the energy required to excite the electron from the ground state to the first excited state?  | 4  | L3        | 1 | 1, 2 |
| 5 | a) | Explain: (i) Heisenberg's uncertainty principle, (ii) Probability density and (iii) Normalization of a wave function.  | 6  | L1        | 1 | 1, 2 |
| 5 | b) | What are Eigen values and Eigen functions? Discuss the wave function, probability density and energy Eigen values for a particle in an infinite potential well by considering its ground state and the first two excited states.                                     | 10 | L3        | 1 | 1, 2 |
| 5 | c) | Calculate the momentum and de Broglie wavelength associated with an electron subjected to a potential difference of $1.5 \text{ kV}$ .   | 4  | L3        | 1 | 1, 2 |
|   | a) | What are Miller Indices? Explain how the axial intercepts in a crystal plane are converted to Miller indices.  | 6  | L2        | 2 | 1, 2 |
|   | b) | Define Coordination number and Atomic packing factor. Calculate the atomic packing factor for base centered cubic (BCC) and face centered cubic (FCC) lattices by calculating number of atoms/unit cell and the relation between atomic radius and lattice constant. | 10 | L2,<br>L3 | 2 | 1, 2 |
|   | c) | Calculate the glancing angle for incidence of X-rays of wavelength $0.58 \text{ \AA}$ on the plane (132) of NaCl which results in 2 <sup>nd</sup> order diffraction maxima taking the lattice constant as $3.81 \text{ \AA}$ .                                       | 4  | L3        | 2 | 1, 2 |
|   | a) | What are X-rays? With the neat diagram of an X-ray tube explain the production of X-rays.  | 6  | L2        | 2 | 1, 2 |
|   | b) | Derive Bragg's law for the X-ray diffraction by crystal planes. Explain in detail the construction and working of Bragg's X-ray spectrometer.  | 10 | L3        | 2 | 1, 2 |
|   | c) | Iron crystallizes in BCC structure. Calculate the lattice constant. Given that, the atomic weight of iron is $55.85$ and density of iron is $7860 \text{ kg/m}^3$ .  | 4  | L3        | 2 | 1, 2 |

5.	a)	Mention the assumptions and limitation of classical free electron theory.	6	L1	3	
	b)	Obtain an expression for the electrical conductivity of a metal based on classical free electron theory.	10	L3	3	
	c)	Superconducting tin has a critical magnetic field of 217 gauss at 2 K. If the critical temperature for superconducting transition for tin is 3.7 K. Find the critical magnetic field at 3 K.	4	L3	3	uration st of c
6.	a)	Define Matthiessen's rule and explain the effect of temperature on electrical resistivity of metals.	6	L2	3	
	b)	Discuss critical magnetic field and Meissner effect in superconductors. Distinguish between Type-I and Type-II superconductors.	10	L3	3	
	c)	Find the temperature at which there is 2% probability that an energy level 0.3 eV above Fermi energy level is occupied.	4	L3	3	a) b)
7.	a)	What are direct and indirect band-gap semiconductors? Explain.	6	L2	4	
	b)	What are intrinsic semiconductors? Obtain an expression for the conductivity of an intrinsic semiconductor. Explain the effect of temperature on the resistivity of an intrinsic semiconductor.	10	L3	4	c)
	c)	An n-type semiconductor has a Hall coefficient of $3.66 \times 10^{-4} \text{ m}^3/\text{C}$ and its resistivity is found to be 2.12 ohm-m. Calculate charge carrier concentration and electron mobility at room temperature.	4	L3	4	a)
8.	a)	Distinguish between avalanche diode and Zener diode.	6	L2	4	b) c)
	b)	Explain Hall Effect. Obtain an expression for the Hall coefficient and carrier concentration of an n-type semiconductor. Mention any four applications of Hall effect.	10	L3	4	
	c)	Mobilities of electrons and holes in a sample of intrinsic germanium at 300K are $0.36 \text{ m}^2\text{V}^{-1}\text{s}^{-1}$ and $0.17 \text{ m}^2\text{V}^{-1}\text{s}^{-1}$ respectively. If the resistivity of the specimen is $2.12 \Omega\text{m}$ , compute the intrinsic carrier density.	4	L3	4	a) b) c)
9.	a)	Derive an expression for numerical aperture of an optical fiber in terms of refractive index of core and cladding.	6	L2	5	a)
	b)	What is a laser? Describe the construction and working of a He-Ne laser with necessary energy level diagram.	10	L3	5	b)
	c)	The core refractive index of the optical fiber is 1.40, its relative refractive index is 2.5%. Determine the numerical aperture and the critical angle.	4	L3	5	c)
10.	a)	Explain the basic components required for the construction of a laser?	6	L2	5	a) b)
	b)	Give general description of an optical fiber with the working principle. Explain in brief different attenuation mechanisms in an optical fiber.	10	L3	5	c)
	c)	Find the ratio of population of the two energy states, the transition between which emits photons of wavelength 694.3 nm, at 27°C.	4	L3	5	

BT\* Bloom's Taxonomy, L\* Level; CO\* Course Outcome; PO\* Program Outcome

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