1. A classical gas in three dimensions is constrained by a wall to move in the $x \geq 0$ region. A potential

$$V(x) = \frac{\alpha x^2}{2}$$

attracts the atoms to the wall. The atoms are free to move in an area A in the y-z directions. If the gas is at uniform temperature T, show using the grand canonical ensemble that the number of particles varies as

$$N(x) = N\sqrt{\frac{2\alpha\beta}{\pi}}e^{-\frac{\alpha\beta x^2}{2}}$$

By considering a slab of gas between x and x + dx, show that locally the gas continues to obey the ideal gas law. Hence determine the pressure that the gas exerts on the wall.

- 2. Determine the density of states for non-relativistic particles in 1 and 2 dimensions. How does the density behave for particles on a plane and or for particles on a line?
- 3. Often particles are trapped in a quadratic potential. In d-dimensions, the potential energy felt by a single particle at a position \vec{r} is

$$V(\vec{r}) = \frac{1}{2} \sum_{i=1}^d \omega_i^2 r_i^2$$

with $r_1 = x$, $r_2 = y$, and so on. Compute the density of states, g(E) in 2 and 3 dimensions assuming that the energy is large enough that the spectrum may be treated as a continuum.

Tip: First determine G(E), the number of states with energy less than E.