Assignment 6

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1 Introduction

In this assignment a tubelight is simulated. Whenever a tubelight is used the electrons move from the cathode with zero energy at the cathode to the anode. The electric field in between accelerates the electrons and thus energizes it. Therefore if some electron crosses a threshold energy, then it can excite the atom which it hits and thus it can emit light. But there is some probability with which the electron hits an atom.

1.1 Assumptions in the simulation:

- 1:The electron comes to rest (i.e. it loses all of it's energy) after collision and is made to accelerate again starting with zero velocity.
- 2:The electron acceleration due to the electron field is equal to $1m/s^2$.
- 3:The time between each update of displacement, velocity and acceleration is 1 sec.
- 4:Five new electrons with standard deviation of 2 are added into the tubelight after every turn of the simulation.

2 Import Libraries

```
In [1]: from pylab import *
    import sys
```

3 Take input from user or use a set of predefined parameters

```
[100,5,500,5,1,2],[100,5,500,10,0.25,2],
[100,5,500,10,0.50,2],[100,5,500,10,1,2]]
```

4 The loop of the simulation

Here a function is defined which runs through the main loop of the simulation. Here the algorithm used is as defined in the problem.

* Note: The algorithm is explained in the comments

```
In [3]: def loop(params, quadratic):
            n=params[0]
            M=params[1]
            nk=params[2]
            u0=params[3]
            p=params[4]
            Msig=params[5]
            xx=np.zeros((n*M)) # electron position
            u=np.zeros((n*M)) #electron velocity
            dx=np.zeros((n*M)) #displacement in current turn
            I = []
            V = LJ
            X = \Gamma I
            for i in range(1,nk):
                ii=where(xx>0) #get the indices of positions greater than zero
                dx[ii]=u[ii]+0.5 #increase the displacement
                xx[ii]+=dx[ii] #increase the position
                u[ii] += 1 #increase the velocity
                reached=where(xx[ii]>n)#contains the indices
                #set position, velocities, displacements to zero
                xx[ii[0][reached]]=u[ii[0][reached]]=dx[ii[0][reached]]=0
                kk=where(u>=u0)
                11=where(rand(len(kk[0]))<=p)</pre>
                kl=kk[0][11] #contains the indices
                #of energetic electrons that suffer collision
                u[kl]=0 # reset the velocity after collision
                if quadratic==False:
                    rho=rand(len(kl)) #get random number
                if quadratic==True:
                    rho=power(0.5,len(kl)) # a quadratic probability distribution
                xx[kl]=xx[kl]-dx[kl]*rho #qet the actual value of x where it collides
                I.extend(xx[kl].tolist())
                m=int(rand()*Msig+M) #qet the (random)number of new electrons to be added
                empty=where(xx==0) #get empty spaces where electrons can be injected
                nv=(min(n*M-len(empty),m)) #if no empty spaces are left
                xx[empty[:nv]]=1 #inject the new electrons
```

```
u[empty[0][:nv]]=0 #with velocity zero
dx[empty[0][:nv]]=0 #and displacement zero
X.extend(xx.tolist())
V.extend(u.tolist())
return X,V,I
```

5 The functions to plot the graphs

```
plot_no_of_elec plots the number of electrons vs x.
plot_intensity plots the intensity vs x.
plot_intensity_map plots the relative brightness of the tubelight in grayscale.
plot_phase plots the phase space of the electrons.
In [6]: def plot_no_of_elec(X,u0,p):
        #plot the number of electrons vs x
            figure(1)
            hist(X,bins=np.arange(0,101,0.5),rwidth=0.8,color='g')
            title('Number of Electrons vs $x$ with $u_0=$%f and p=%f'%(u0,p))
            xlabel('$x$')
            ylabel('Number of electrons')
            show()
        def plot_intensity_map(I,u0,p):
        #plot the intensity map
            histogram_=hist(I,bins=np.arange(0,101,1),rwidth=0.8,color='r')
            x=histogram_[1][1:]
            y=histogram_[0]
            fig, (ax) = plt.subplots(nrows=1, sharex=True)
            extent = [x[0]-(x[1]-x[0])/2., x[-1]+(x[1]-x[0])/2.,0,1]
            intensity=ax.imshow(y[np.newaxis,:], cmap="gray", aspect="auto", extent=extent)
            ax.set_yticks([])
            ax.set_xlim(extent[0], extent[1])
            plt.title('Intensity map with u_0=\% and p=\%'\%(u0,p))
            plt.xlabel('$x$')
            plt.colorbar(intensity)
            plt.tight_layout()
            show()
        def plot_intensity(X,V,u0,p):
            #plot the histogram of intensity
            figure(0)
            histogram=hist(I,bins=np.arange(0,101,0.5),rwidth=0.8,color='r')
            title('Intensity histogram with u_0=\% and p=\% '\%(u0,p))
            xlabel('$x$')
            ylabel('Intensity')
            show()
```

```
return histogram

def plot_phase(X,V,u0,p):
#plot the phase space
   figure(2)
   plt.plot(X,V,'bo')
   title('Electron Phase Space with $u_0=$%f and p=%f'%(u0,p))
   xlabel('$x$')
   ylabel('Velocity-$v$')
   show()
```

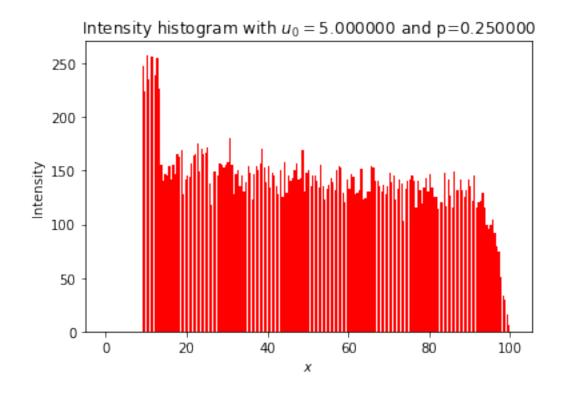
6 Plot of the graphs with uniform probability of electron.

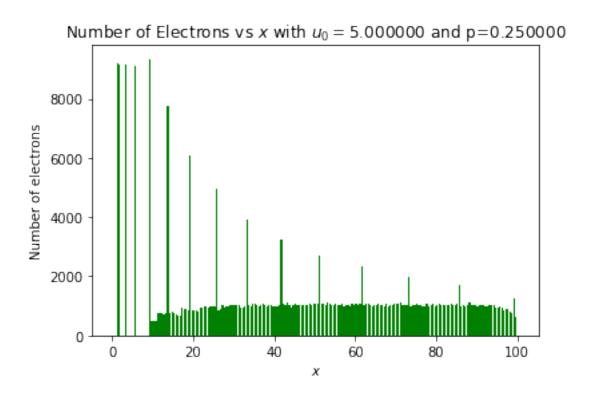
In the intensity vs x graph the intensity reaches a maximum at around x=15 and stays like that for around 4-5 bins and then it decreases. This is because of the fact that the electron comes to rest after collision. So it has to gain energy again from zero to be able to excite the atom for emitting light.

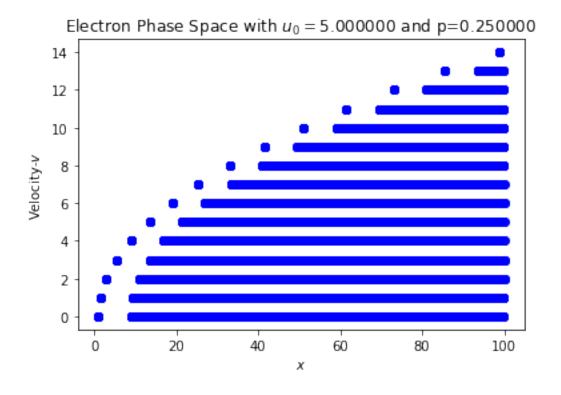
In the electron phase space the graph shows the allowed velocities at a particular value of x thus we can say that the velocities are quantized.

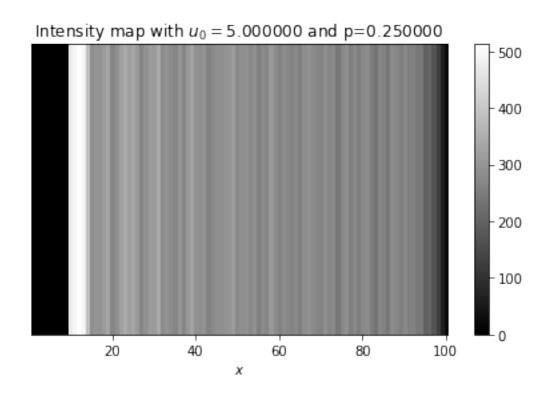
The number of electrons vs x graph shows the number of electrons which got excited at that value of x.

```
In [7]: param=[100,5,500,5,0.25,2]
     X,V,I=loop(param,False)
     histo=plot_intensity(X,V,param[3],param[4])
     plot_no_of_elec(X,param[3],param[4])
     plot_phase(X,V,param[3],param[4])
     plot_intensity_map(I,param[3],param[4])
```









7 Table of Population vs xpos

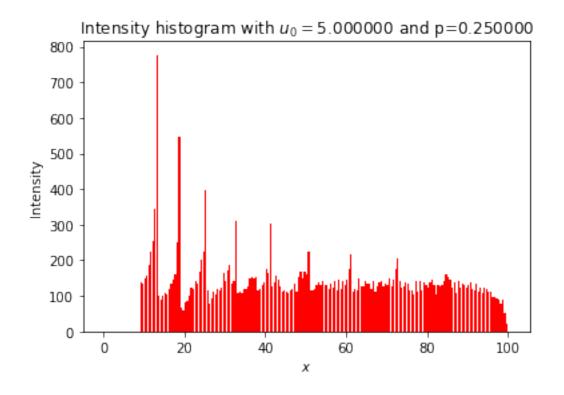
Sr.No.	xpos	Population
0	0.0	0.0
2	1.0	0.0
4	2.0	0.0
6	3.0	0.0
8	4.0	0.0
10	5.0	0.0
12	6.0	0.0
14	7.0	0.0
16	8.0	0.0
18	9.0	248.0
20	10.0	258.0
22	11.0	256.0
24	12.0	239.0
26	13.0	227.0
28	14.0	141.0
30	15.0	146.0
32	16.0	142.0
34	17.0	147.0
36	18.0	163.0
38	19.0	128.0
40	20.0	145.0
42	21.0	157.0
44	22.0	165.0
46	23.0	149.0
48	24.0	166.0
50	25.0	172.0
52	26.0	118.0
54	27.0	149.0
56	28.0	157.0
58	29.0	153.0
142	71.0	146.0
144	72.0	133.0
146	73.0	138.0
148	74.0	133.0
150	75.0	142.0
152	76.0	140.0
154	77.0	141.0
156	78.0	119.0
158	79.0	143.0
160	80.0	147.0
162	81.0	126.0
164	82.0	114.0
166	83.0	121.0
168	84.0	117.0

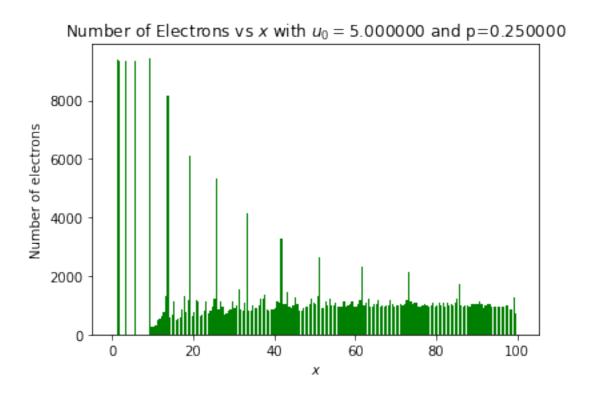
```
85.0
                   127.0
170
172
      86.0
                   149.0
174
      87.0
                  132.0
176
      88.0
                  132.0
      89.0
178
                  132.0
180
      90.0
                   135.0
182
      91.0
                   146.0
      92.0
184
                   120.0
186
      93.0
                  129.0
188
      94.0
                   99.0
190
      95.0
                   100.0
192
      96.0
                   92.0
194
      97.0
                   74.0
      98.0
                    34.0
196
198
      99.0
                    16.0
200
     100.0
                     0.0
```

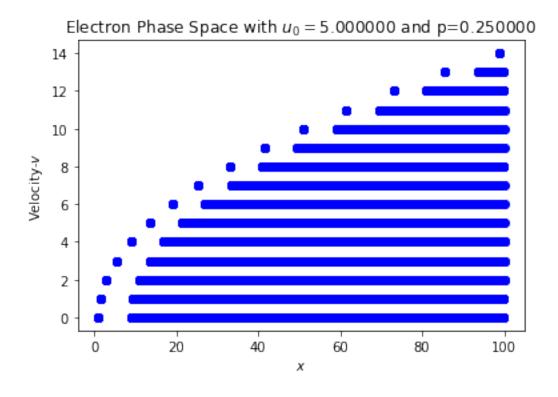
Note: For the purpose of showing the table every alternate element is shown in the table.

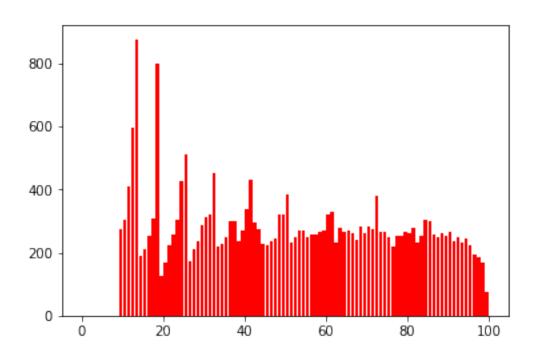
8 Plot of the graphs with quadratic probability of electron.

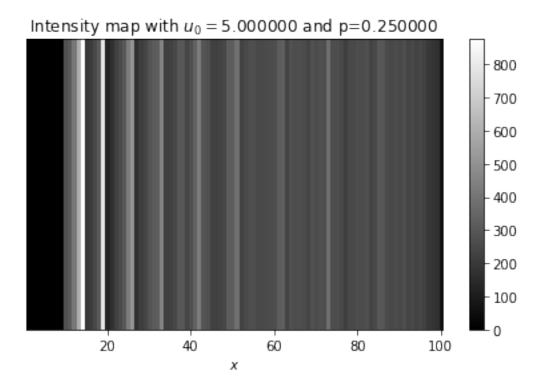
In the intensity vs x graph the intensity increases and then suddenly drops. This is different than the previous model as the electron getting excited is more towards the right side as it's velocity increases.











9 Iterating through the different parameters

The variation in the graphs when the cutoff velocity and probability are captured here.

One specific observation is that as the probability increases the graphs become more variate i.e. the superposition of the probability graphs become more seperated.

Also the maximum intensity also increases as the electrons get inonized more often.

Also as the cutoff velocity (i.e. the gas inside the tubelight is changed) is increased they initial excitation happens at a higher value of x.This is because the electron has to travel longer distances to be able to reach cutoff frequency.

