

# Identifying High-Tc transmon candidates with minimal oxide thickness

Nathan Song, Tyrel McQueen

Department of Physics, U.C. Berkeley

Department of Chemistry, Johns Hopkins

What are transmons?

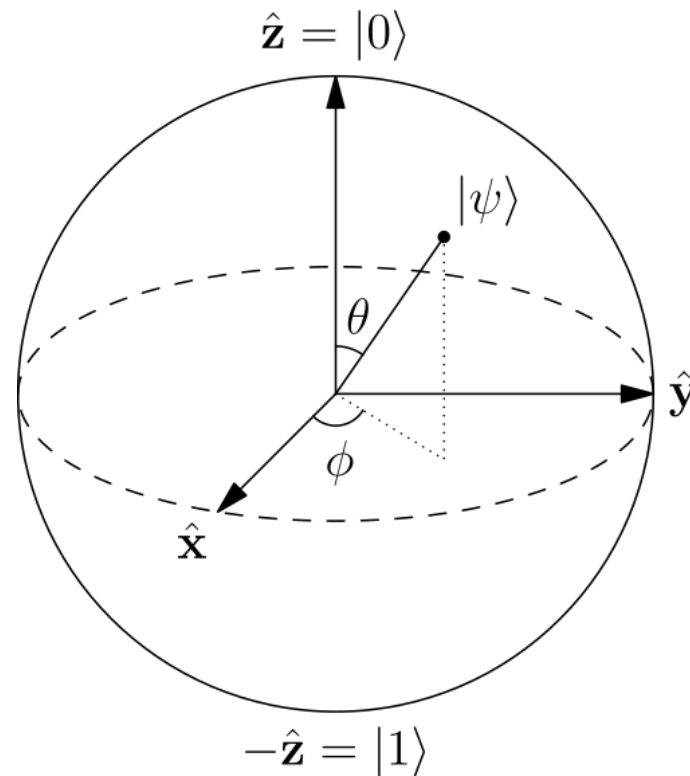
# Quantum Advantage

## Classical bit

● 0 state

● 1 state

## Qubit



### Probabilistic

Exact outcome *uncertain* until measurement / interaction

### Entanglement

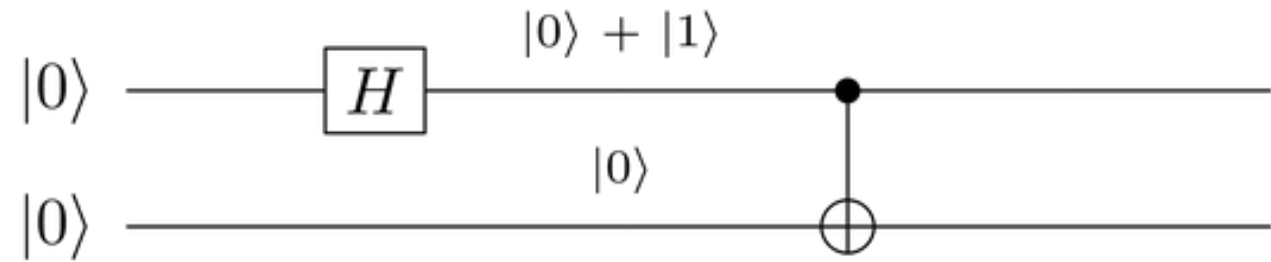
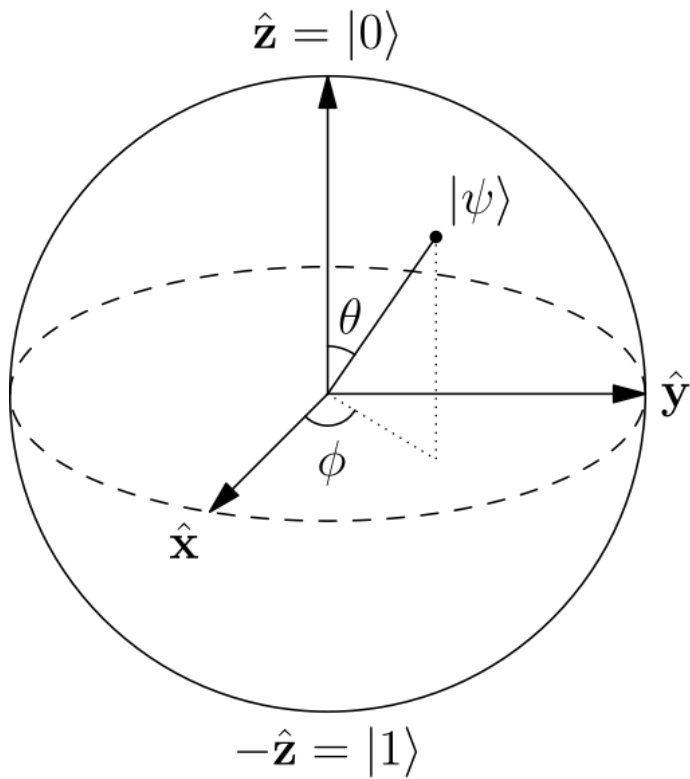
Qubit measurement outcomes become correlated after gates

### Quantum tunneling

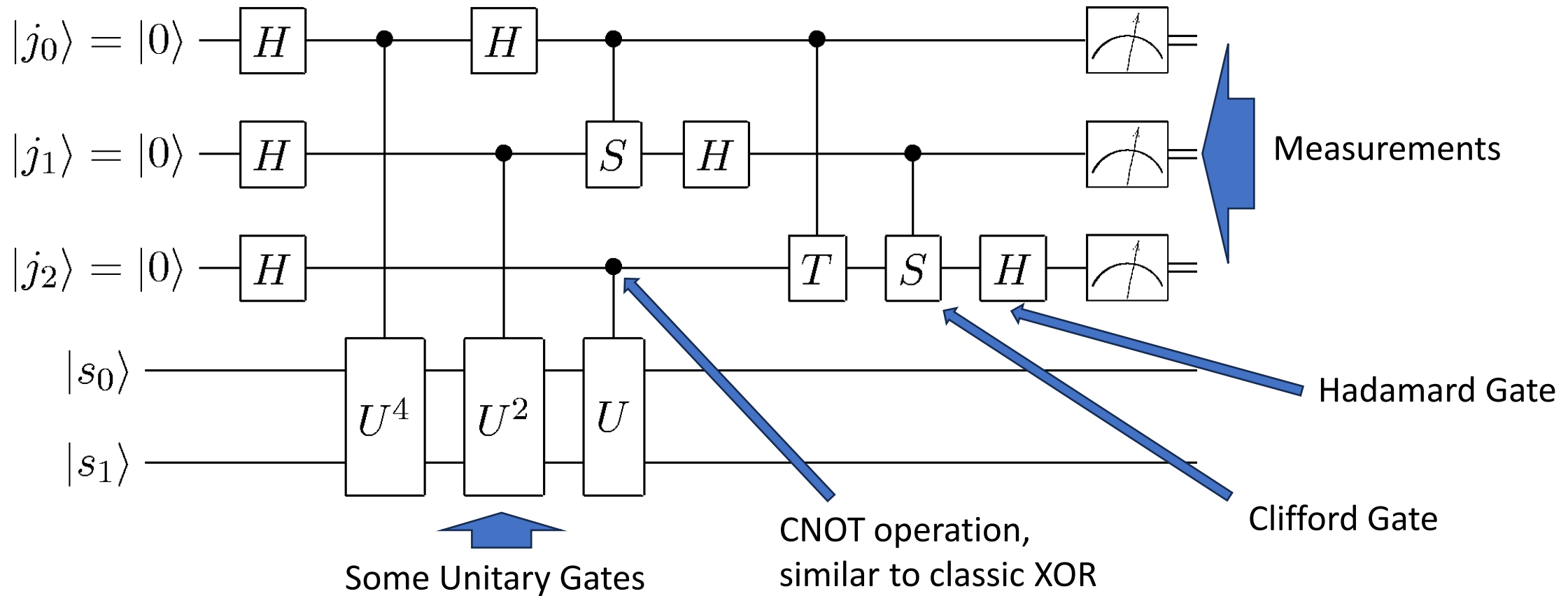
Quantum 'teleportation'

# Quantum Advantage

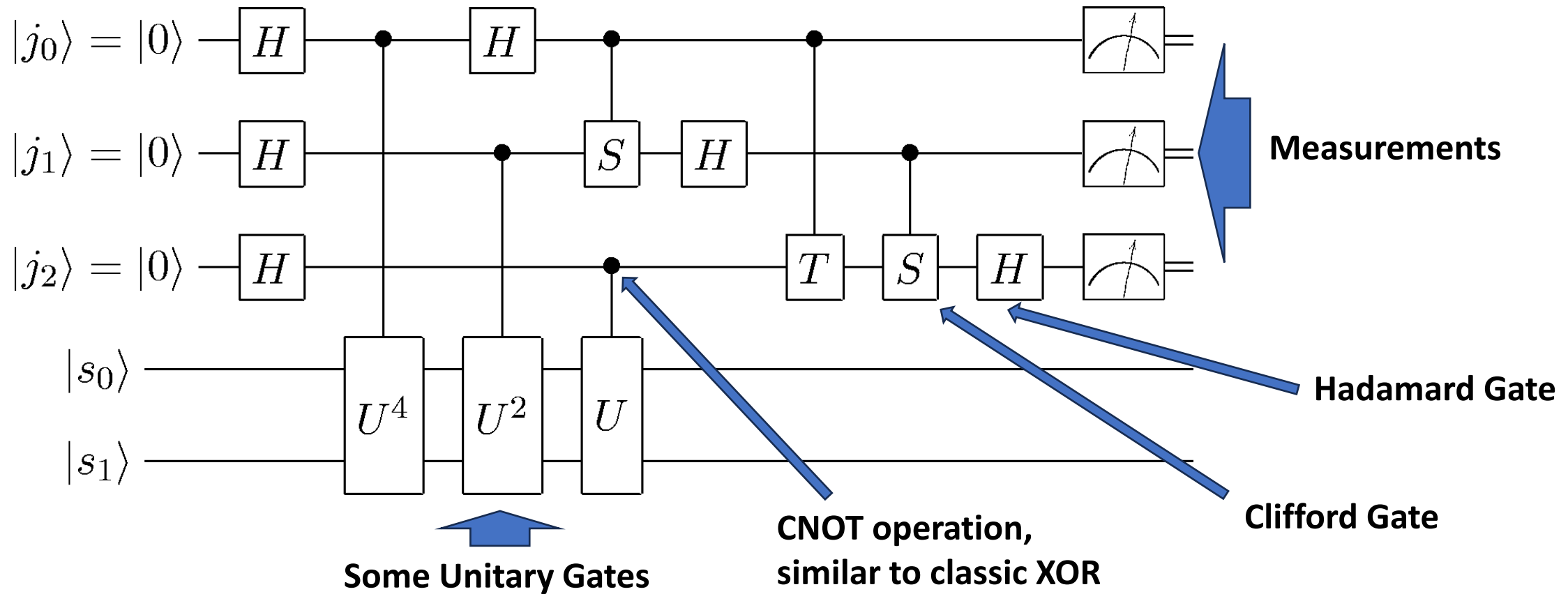
Qubit



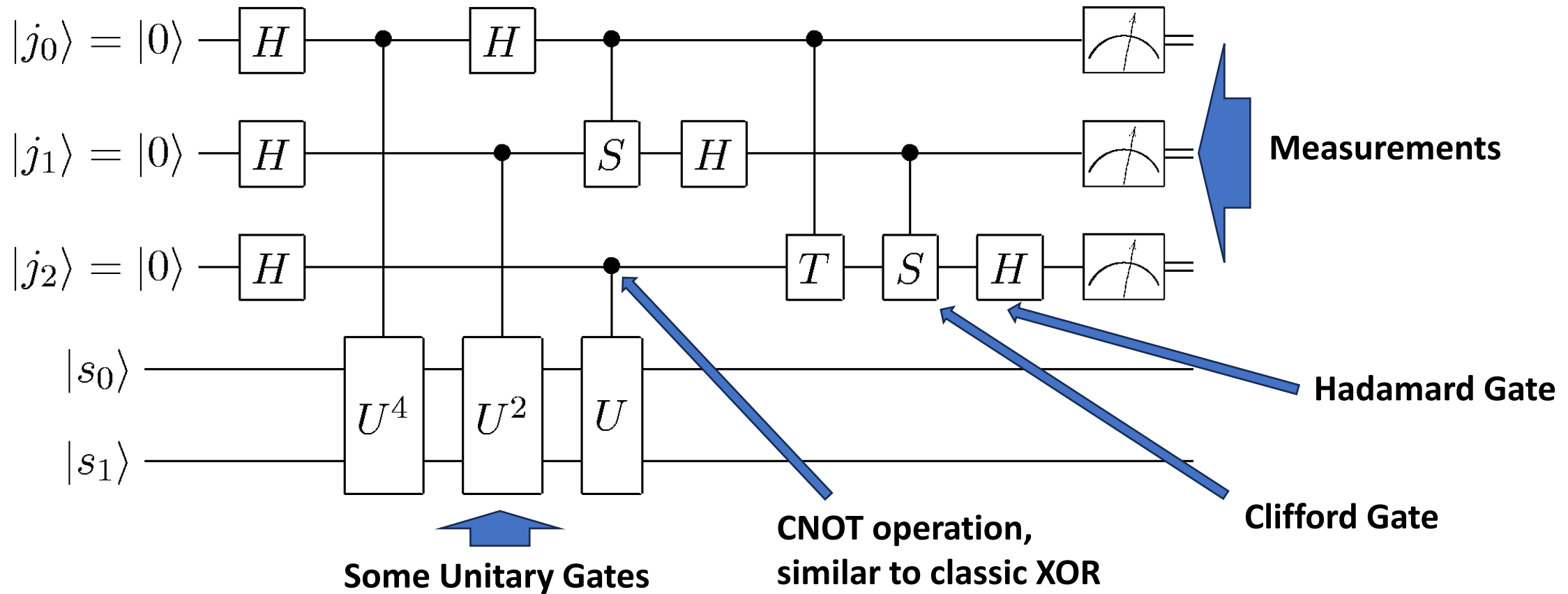
$$\frac{1}{\sqrt{2}} (|00\rangle + |11\rangle)$$



- Highly sensitive to external noise  
(insert transitions / arrows while explaining circuit complexity)



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(insert transitions / arrows while explaining circuit complexity)



Current record: 0.3 ms, Tantalum

Why are oxides important?



# Transmon Decoherence

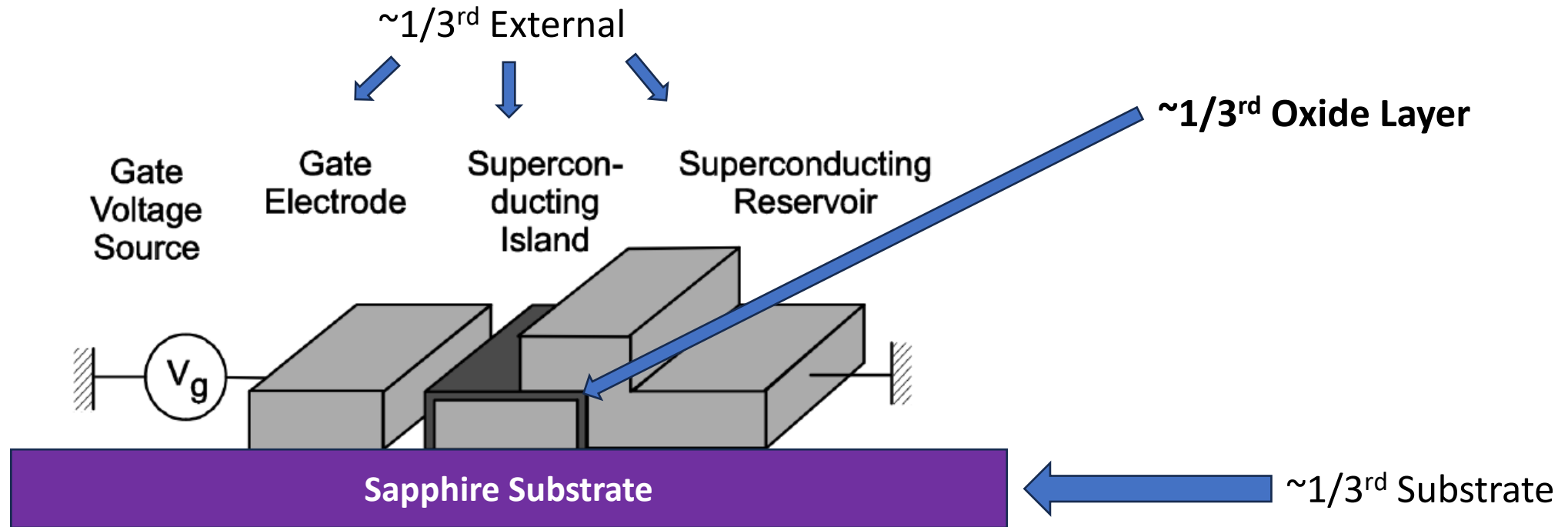


Image Credit: Sam Bader's thesis

Credit: McLellan, Dutta, et. al, Chemical profiles of the oxides on tantalum in state of the art superconducting circuits

# Transmon Decoherence

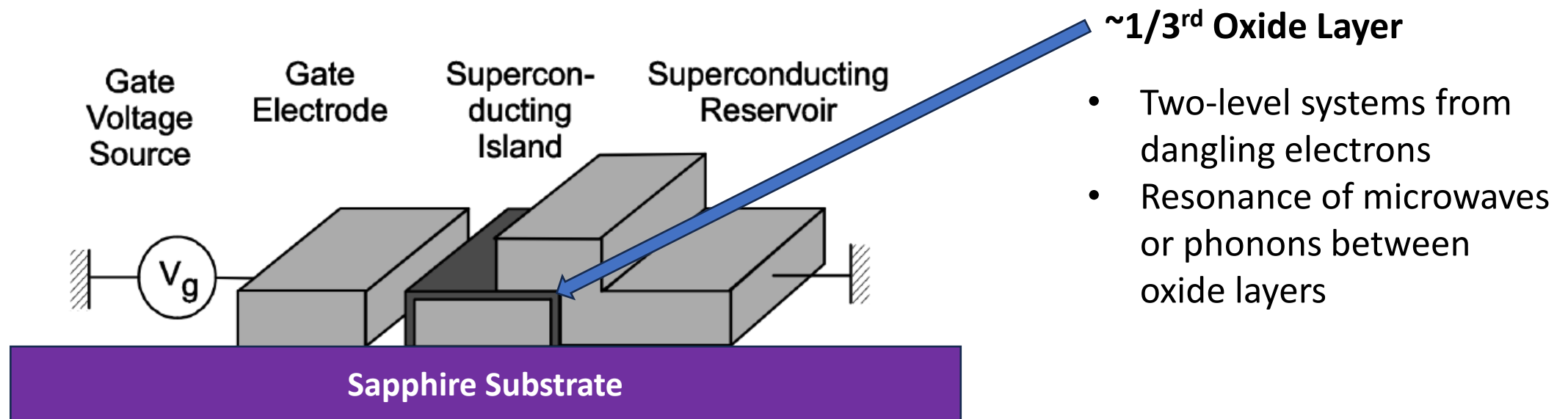


Image Credit: Sam Bader's thesis

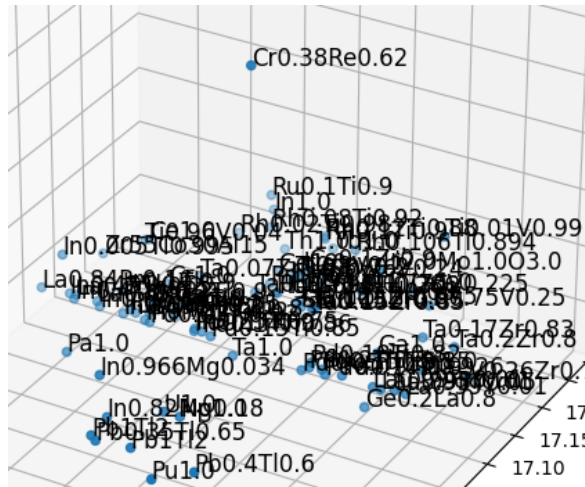
Credit: McLellan, Dutta, et. al, Chemical profiles of the oxides on tantalum in state of the art superconducting circuits

## Developed Metrics

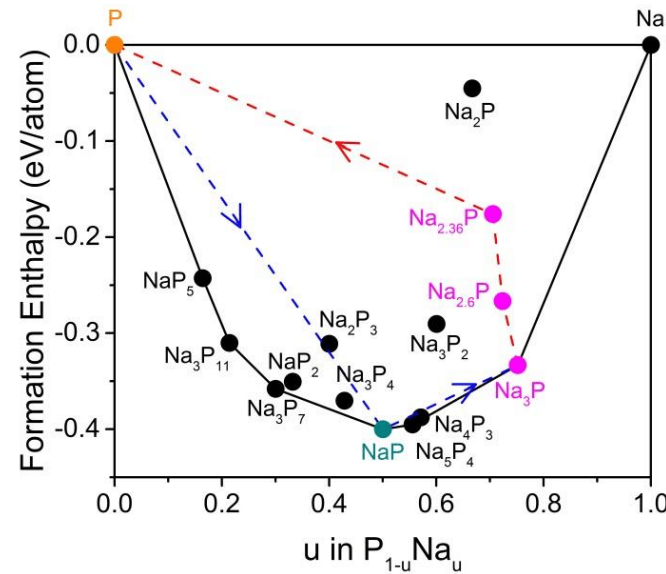
- Oxidation Metric
- % Removable Oxide
- Critical Temperature

# Oxidation Metric

How likely is a material to form an oxide?



1. Identify potential oxides



2. Solve for minimal energy

$\Delta H_{\text{Oxides}}$  = Net oxide energies

$\Delta H_{\text{Material}}$  = Material energy

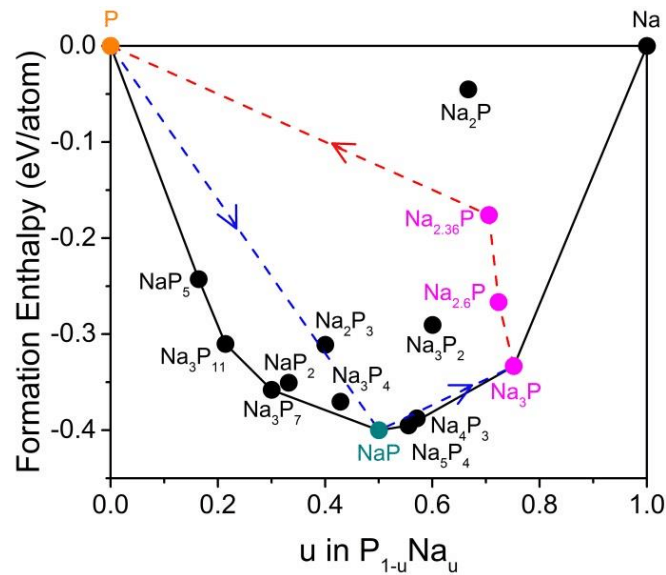
$\mu = \Delta H_{\text{Oxides}} - \Delta H_{\text{Material}}$

$metric = 1 - \mu / \Delta H_{\text{Oxides}}$

3. Compare oxide and material Formation Energies

## % Removable Oxide

How much of the oxides could we remove?



2. Solve for minimal energy

### Minimization Data

- Formation energy of oxide
- Optimal amount oxides

### Additional Data

- Melting temp of oxides
- Melting temp of material

*Courtesy of ASU's Hong Research Group*

for **oxide** in **oxides**:

if **oxide\_melt** < **material\_melt**

**% += oxide\_percentage**

3. Calculate % of removable oxide

# Example: Tantalum

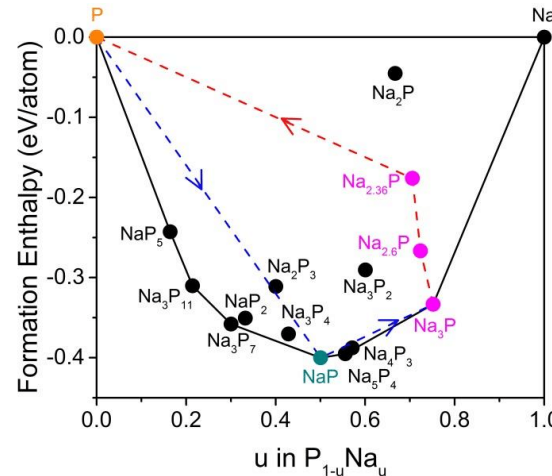
## Identify Ta-based oxides

Name	Formation E	Features
Ta2O5	$\Delta H_{\text{Ta2O5}}$	Mass
Ta2O3	$\Delta H_{\text{Ta2O3}}$	# Atoms
TaO2	$\Delta H_{\text{TaO2}}$	
Ta6O	$\Delta H_{\text{Ta6O}}$	
Ta3O8	$\Delta H_{\text{Ta3O8}}$	

### Additional Data

- Melting temp of oxides
- Melting temp of material

*Courtesy of ASU's Hong Research Group*



### ML Model

Nelder-mead algorithm

### Minimization Goal

Obtain ideal oxide composition

### Constraints

- Conservation of mass
- Positivity of material

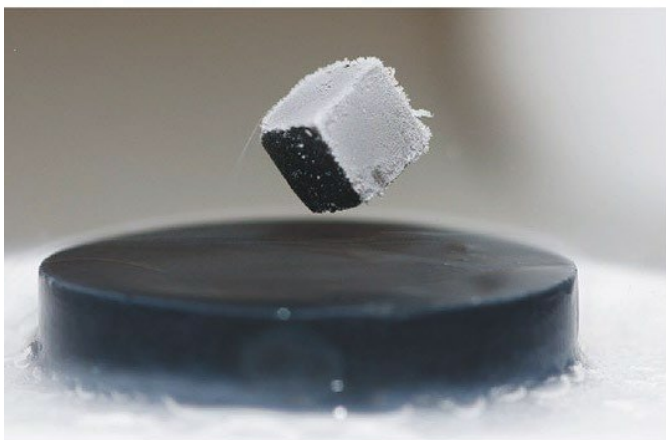
## Calculate metrics

% Total	Meltable?
55%	No
20%	Yes
10%	No
6%	Yes
9%	No

*\*Filler data*

**Net Oxide Formation E**  
**Net % Meltable Oxide**

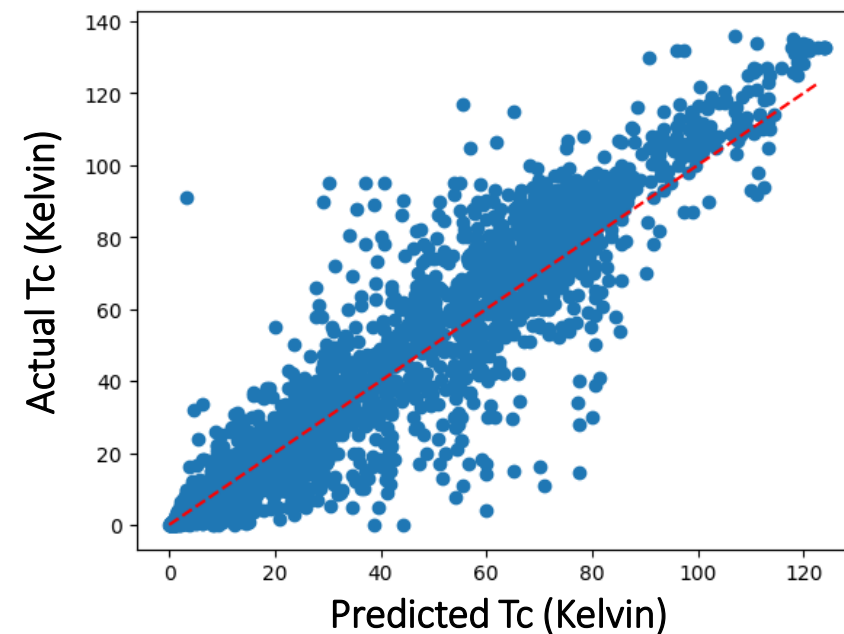
# Critical Temperature



1. Take known  $T_c$  from SuperCon database



2. Create dataset with elemental composition

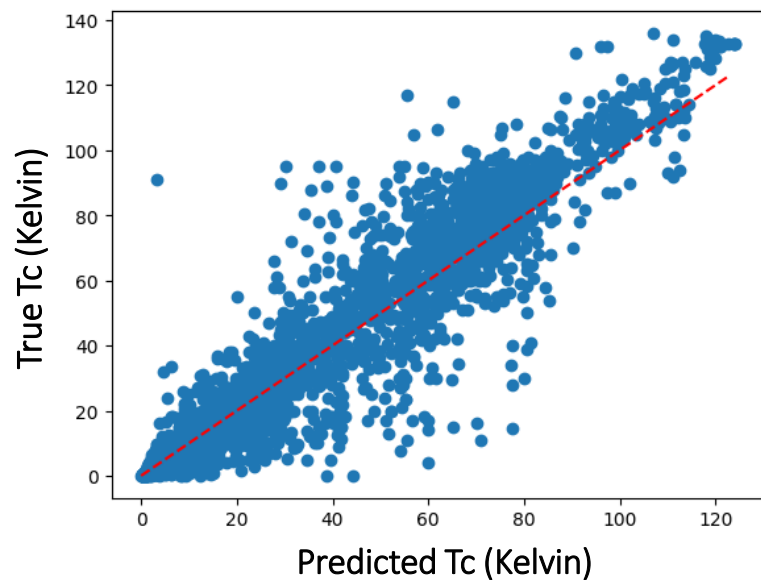


3. Predict  $T_c$  for entries from Materials Project

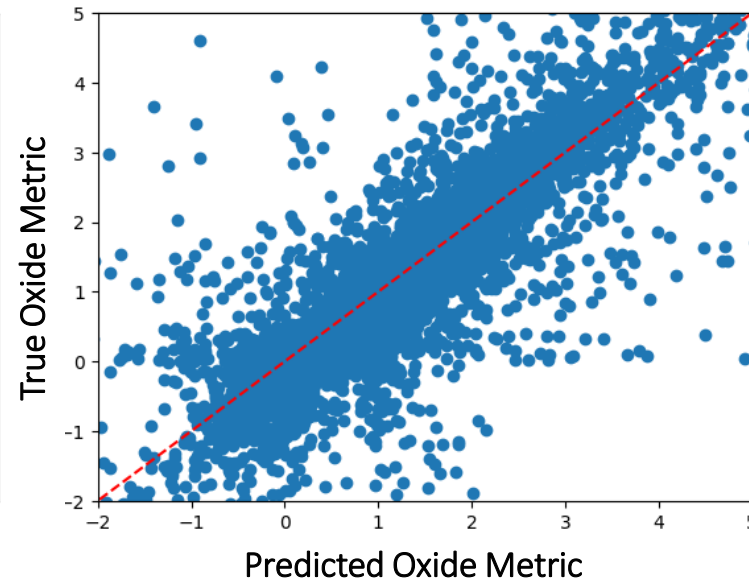
## Metric Predictability (w/ XG boost)

Y-axis: Actual values

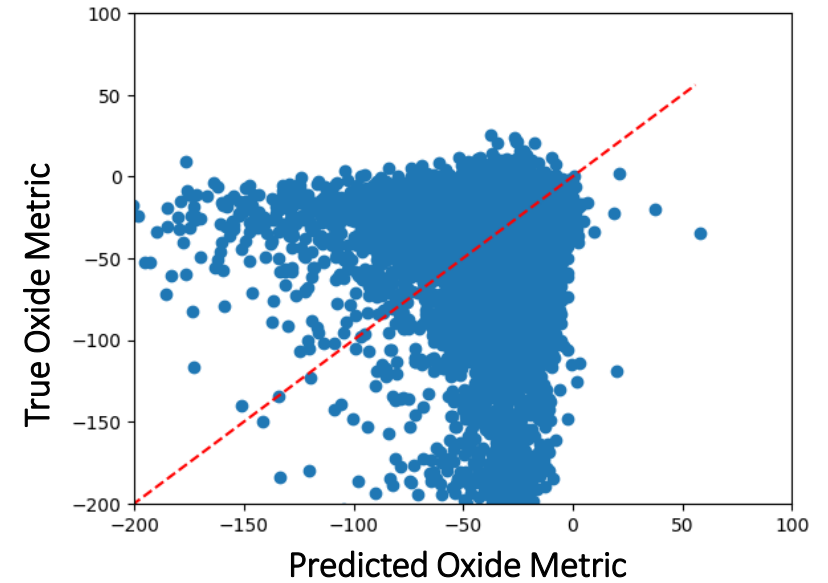
X-axis: Predicted values



1. Critical Temperature



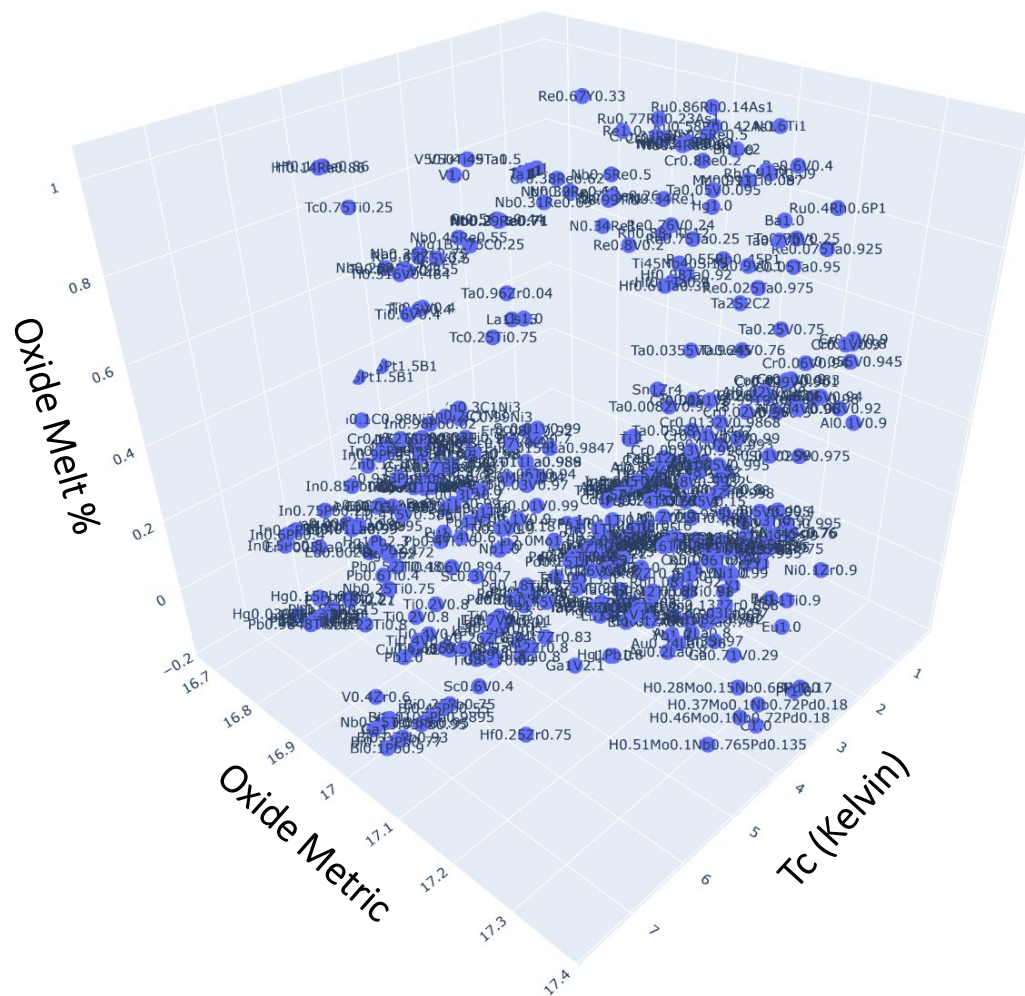
2. Simplified Oxidation Metric



3. True Oxidation Metric



# Material Metrics



## Material Analysis

- Identify material (Ex. Ta, Nb)
- Generate 'neighborhood'
- Identify patterns (Ex. Cuprates)
- Select high-Tc materials with thin oxide layers

Perform MD simulations for top material candidates

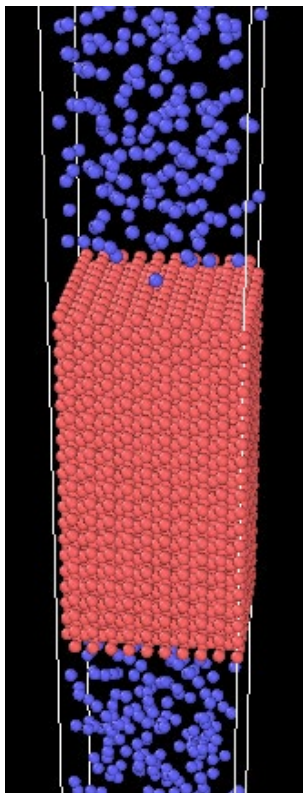
## Results

- Creation of metrics applicable to any material
- Prediction of metrics solely from elemental composition
  - Tc **predictable**, Oxidation Metric **complicated**

## Bonus:

LK-99 'Room-Tc Superconductor' -> **CuP6Pb9O25**

Predicted Tc -> **39.89267 Kelvin**



## Contributors

Tyrel McQueen

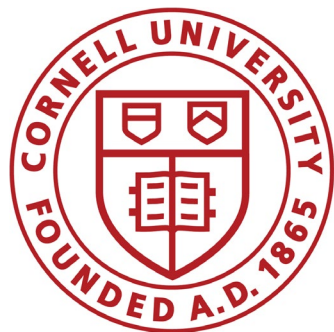
## Special Thanks

Jim Overhiser

Evan Crites

REU cohort

McQueen group



## Bonus+: First-Principles Tantalum Oxidation Simulations

- Force Field calculations for oxide growth  
*(In progress)*

# Transmon Decoherence

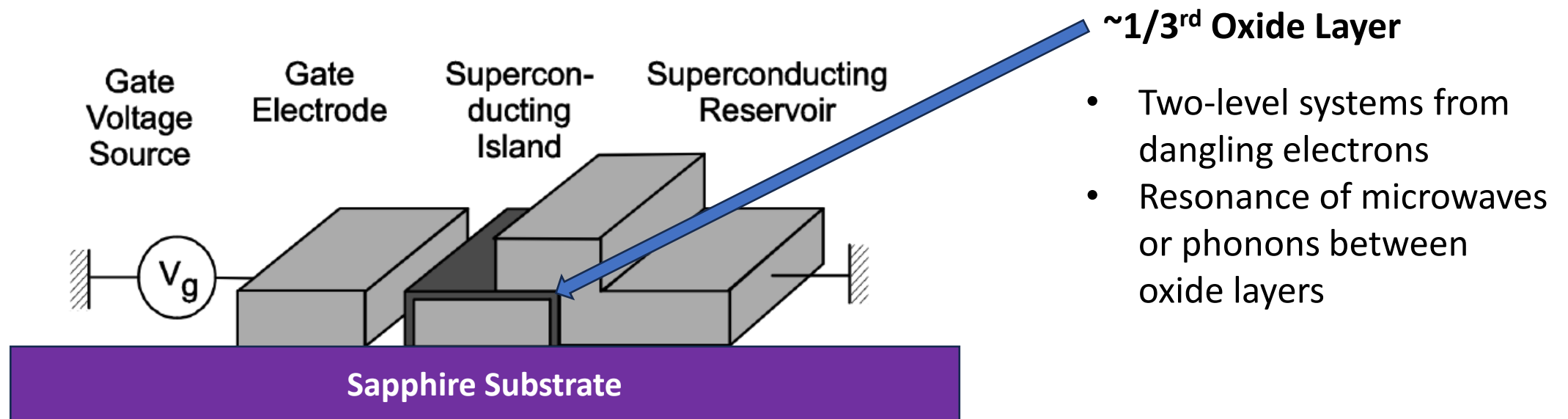
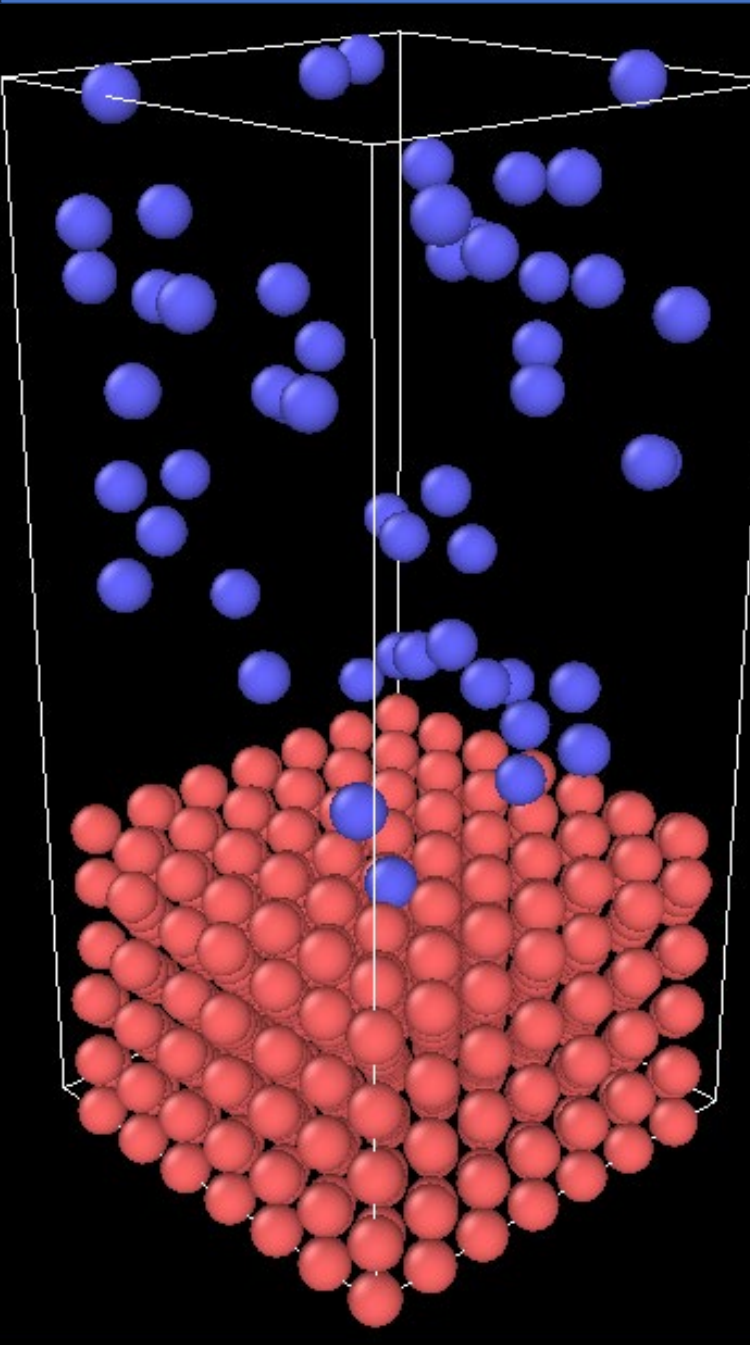


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## MD (Molecular Dynamics) for Ta-O interaction

Ta-Ta interaction – PINN potential

Bonded Ta-O interaction – M-BKS potential

$$\Phi_{ij}^{M-BKS} = q_i q_j / r_{ij} + A_{ij} \exp(-r_{ij} / \rho_{ij}) - C_{ij} / r_{ij}^6 \\ + D_{ij} (1 - \exp(-a_{ij}(r_{ij} - r_e)))^2,$$

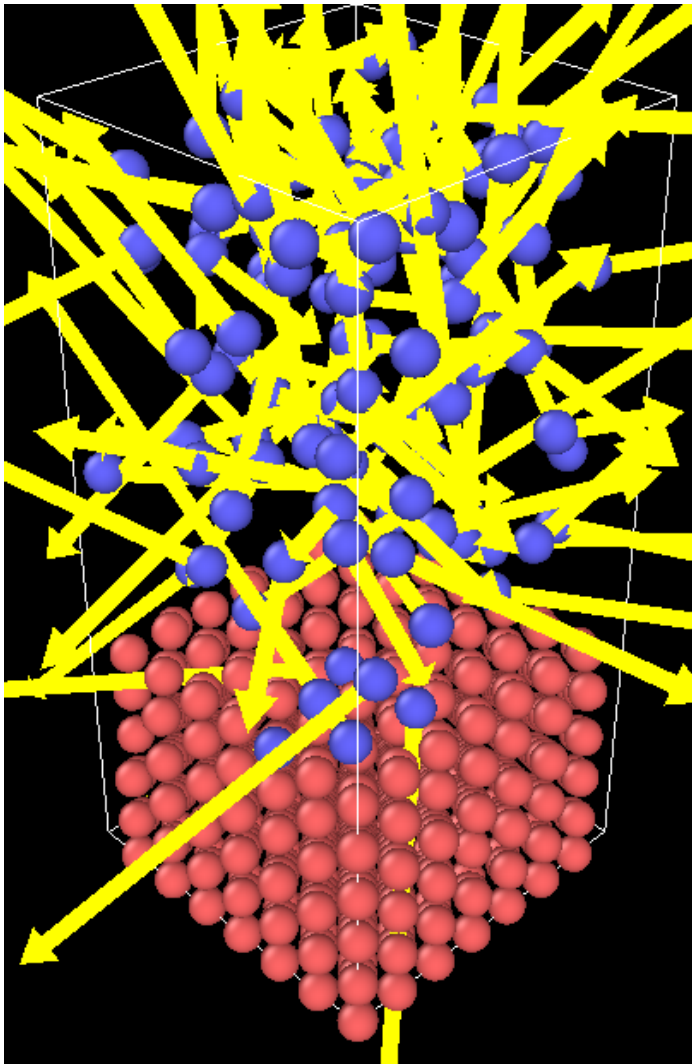
*In order:*

*Coulomb force, Pauli repulsion, Van-der Waals,  
Covalent bond (Morse)*

O-O interaction – Lennard-Jones potential

**MISSING:** Non-bonded Ta-O interaction





Film	Ta <sup>5+</sup>	Ta <sup>3+</sup>	Ta <sup>1+</sup>	Ta <sub>int</sub> <sup>0</sup>
Native	2.257 ± 0.023	0.370 ± 0.016	0.370 ± 0.017	0.368 ± 0.019
Ta treated in 10:1 BOE	1.853 ± 0.028	0.296 ± 0.022	0.302 ± 0.023	0.400 ± 0.021
Ta treated in triacid	4.826 ± 0.036	0.379 ± 0.016	0.545 ± 0.020	1.198 ± 0.027

### Current setup:

- Equilibrate system with Nose-Hoover over 5 ps to 300K
- Heating stage with Nose-Hoover over 15 ps to 600K
- Steady temperature for 20 ps

## How?

- Minimize oxide layers
- Remove oxides entirely
- **Understand why oxides decohere qubits**



## Why?

- Oxide interfaces have unmatched spin pairs
- Oxide layers cause interference
- Microwave loss amplified by oxide layers