DOI: 10.1002/chem.201805746



■ Fused-Ring Systems



Pd-Catalyzed Annulation of 1-Halo-8-arylnaphthalenes and Alkynes Leading to Heptagon-Embedded Aromatic Systems

Jianming Yan⁺, Md. Shafigur Rahman⁺, and Naohiko Yoshikai^{*[a]}

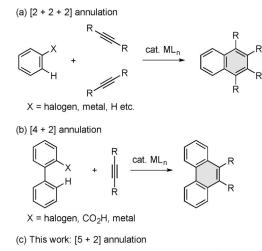
Abstract: A palladium-catalyzed heptagon-forming annulation reaction between 1-halo-8-arylnaphthalene and diarylacetylene is reported. The reaction is promoted using a catalytic system comprised of Pd(OAc)2, moderately electron-deficient triarylphosphine P(4-ClC₆H₄)₃, and Ag₂CO₃ to afford benzo[4,5]cyclohepta[1,2,3-de]naphthalene derivatives in moderate to good yields, in preference to fluoranthene as a competing byproduct. Twofold annulation can also be achieved to access a novel heptagon-embedded polycyclic aromatic hydrocarbon compound.

Polycyclic aromatic hydrocarbons (PAHs) are quintessential structural elements in functional molecules for optoelectronic applications, and therefore the development of efficient methods for the construction of PAHs from readily available starting materials represents an important subject in synthetic chemistry.[1] In this context, transition metal-catalyzed annulation reactions of aromatic substrates with alkynes have been extensively explored as a means to achieve extension of the π -system,^[2] among which those involving C-H activation have attracted significant attention due to the greater availability of the starting materials. Thus, [2+2+2] annulation reactions of a monofunctionalized benzene derivative with two molecules of alkyne have been developed for the synthesis of naphthalene, anthracene, and related acenes (Scheme 1 a).[3] Meanwhile, [4+2] annulation reactions of a 2-functionalized biaryl with an alkyne have been achieved to enable expedient synthesis of phenanthrene-type hexagonal polyaromatic compounds (Scheme 1b).[4,5]

Recently, PAHs that contain a seven-membered ring comprised of all sp²-carbon atoms have gained increasing attention for their warped structures that could lead to unique optoelectronic properties. [6,7] However, the synthesis of such heptagon-

- [a] Dr. J. Yan, * M. S. Rahman, * Prof. N. Yoshikai Division of Chemistry and Biological Chemistry School of Physical and Mathematical Sciences Nanyang Technological University Singapore 637371 (Singapore) E-mail: nvoshikai@ntu.edu.sa
- [+] These authors contributed equally to this work.
- Supporting information and the ORCID identification number(s) for the author(s) of this article can be found under: https://doi.org/10.1002/chem.201805746.

Part of a Special Issue in honor of the 20th IUPAC International Symposium on Organometallic Chemistry directed towards Organic Synthesis (OMCOS 20). To view the complete issue, visit Issue 40.



cat. Pd-PAr₃ Ag₂CO₃ X = Br or I

Scheme 1. Transition-metal-catalyzed alkyne annulations for π -extension.

embedded PAHs usually requires multiple steps, and direct construction of heptagons from simple and readily available fragments has been rare. Recently, Lin, Kwong, and co-workers disclosed a Pd-catalyzed three-component annulation reaction of aryl iodide, 8-bromo-1-naphthoic acid, and norbornadiene to afford a heptagon-containing benzo[4,5]cyclohepta[1,2,3de]naphthalene derivative, in which naphthalene and benzene rings are linked by a cis-ethenyl (C₂H₂) bridge.^[8] Except for this report, annulation approaches to heptagons have been elusive. [9] Herein, we disclose a palladium-catalyzed heptagonforming annulation reaction between 1-halo-8-arylnaphthalene and diarylacetylene. The reaction affords benzo[4,5]cyclohepta[1,2,3-de]naphthalenes bearing a variety of aryl groups on the 7- and 8-positions, and thus complements the scope of Lin/Kwong's annulation.

The present study commenced with a screening of reaction conditions for the annulation of 1-bromo-8-phenylnaphthalene (1 a) and diphenylacetylene (2 a) (Table 1). The reaction in the presence of Pd(OAc)₂ (10 mol%) and Ag₂CO₃ (1 equiv) in dichloroethane at 80 °C afforded, after 18 h, the desired heptagon product 3 aa in a low yield of 18% (Table 1, entry 1). The addition of triphenylphosphine (30 mol%) decreased the yield of 3aa while affording a trace amount of fluoranthene 4 (entry 2).[10] The use of less electron-donating triphenylarsine



Table 1. Optimization of reaction conditions. ^[a]				
Pd(OAc) ₂ (10 mol%) Ph ligand (30 mol%) Ag ₂ CO ₃ (1 equiv) Ph DCE, 80 °C, 18 h				
1a	2a		3aa	4
Entry	Ligand ^[b]	Conv. of 1 a [%] ^[b]	3 aa [%] ^[b]	4 [%] ^[b]
1	None	45	18	0
2	PPh₃	58	8	3
3	AsPh ₃	100 (60)	59 (29)	20 (21)
4	$P(4-CIC_6H_4)_3$	100 (100)	78 (75)	3 (4)
5	$P(3-CIC_6H_4)_3$	100	77	3
6	$P(4-FC_6H_4)_3$	100	65	0
7	$P(4-CF_3C_6H_4)_3$	55	37	4
8	$P(C_6F_5)_3$	29	15	0
9	$P(4-MeOC_6H_4)_3$	20	0	1
10	P(2-furyl) ₃	85	38	2
11	SPhos	83	10	3
12 ^[c]	$P(4-CIC_6H_4)_3$	100	72	4
13 ^[d]	$P(4-CIC_6H_4)_3$	40	30	0

[a] The reaction was performed using 0.1 mmol of 1a and 0.2 mmol of 2a. [b] Determined by GC using n-tridecane as an internal standard. The conversion and yield in the parentheses were determined at the reaction time of 1.5 h. [c] AgOAc was used instead of Ag_2CO_3 . [d] The reaction was performed in the absence of Ag_2CO_3 .

promoted full conversion of 1a and improved the yield of 3aa to 59%, albeit with increased amount (20%) of 4 (entry 3). At the reaction time of 1.5 h, the yields of 3 aa and 4 reached 29 and 21%, respectively, indicating that the formation of 4 stopped at the early stage. Moderately electron-deficient triarylphosphines, such as $P(4-CIC_6H_4)_3$, $P(3-CIC_6H_4)_3$, C(4-FC₆H₄)₃, were found to promote the annulation to afford 3 aa in up to 78% yield, whereas suppressing the fluoranthene formation (entries 4-6). Note that the reaction using P(4-CIC₆H₄)₃ reached full conversion at 1.5 h (entry 4). The annulation reaction became sluggish with more electron-deficient phosphines, such as P(4-CF₃C₆H₄)₃ and P(C₆F₅)₃ (entries 7 and 8), and no annulation was observed with electron-rich P(4-MeOC₆H₄)₃ (entry 9). Other monophosphines, such as tri(2-furyl)phosphine and SPhos, did not improve the reaction (entries 10 and 11). The reaction using AgOAc instead of Ag₂CO₃ afforded 3 aa in a slightly lower but comparable yield (entry 12), whereas removal of Ag₂CO₃ resulted in sluggish conversion (entry 13).

Having established the optimal catalytic system, various diarylacetylenes were subjected to the annulation reaction with 1 a (Table 2). Besides 2 a (Table 2, entry 1), diarylacetylenes bearing electron-donating groups, such as methoxy, methyl, and silyl groups or electron-withdrawing groups, such as halogen and trifluoromethyl groups at the *para* positions, participated in the reaction to afford the corresponding products 3 ab-3 ag in moderate to good yields (Table 2, entries 2–7). Diarylacetylenes bearing *meta*-substituents also afforded the desired products 3 ah and 3 ai in moderate yields (entries 8 and 9). Di(2-methoxyphenyl)acetylene smoothly participated in the reaction to afford 3 aj in 77 % yield (entry 10), whereas di(1-

www.chemeurj.org

Table 2. Annulation of 1 a with various diaryl alkynes. [a] Pd(OAc)₂ (10 mol%) P(4-CIC₆H₄)₃ (30 mol%) Ag₂CO₃ (1 equiv) DCE, 80 °C, 18 h Àr 2a-2k 3aa-3ak Yield [%] Entry 1 3 aa 79 75 (68)^[b] 2 4-MeOC₆H₄ 3 ab 3 4-MeC₆H₄ 3 ac 63 4 $4-Me_3SiC_6H_4$ 58 3 ad 5 4-CIC₆H₄ 53 3 ae 4-FC₄H₄ 60 6 3 af 7 4-F₃CC₆H₄ 3 ag 55 8 3,4-(MeO)₂C₆H₃ 3 ah 57^[c] 9 3-MeC₆H₄ 3 ai 67 10 2-MeOC₆H₄ 3 aj 77 70 11 2-thienvl 3 ak

[a] Unless otherwise noted, the reaction was performed using 0.1 mmol of 1a and 0.2 mmol of 2 according to the conditions in Table 1, entry 4. [b] The yield of a 0.5 mmol-scale reaction is shown in the parentheses. [c] The reaction was performed on a 0.6 mmol scale.

naphthyl)acetylene reacted rather sluggishly (<5% GC yield) presumably due to steric hindrance. Di(2-thienyl)acetylene was also amenable to the annulation, affording the product **3 ak** in good yield (entry 11). Unfortunately, other types of alkynes, such as 5-decyne, 1-phenyl-1-butyne, and diethyl acetylenedicarboxylate, failed to undergo the annulation with **1 a** under the present conditions.

Next, a variety of 1-halo-8-arylnaphthalenes were tested for the annulation with 2a (Scheme 2). The substrate bearing 4methoxyphenyl and iodo groups (1 b) smoothly underwent the annulation to afford the desired product 3 ba in 66% yield. The nonplanar structure of 3 ba was unambiguously confirmed by X-ray crystallographic analysis (see the box in Scheme 2).[11] Likewise, naphthalene substrates bearing different para-substituted phenyl groups (1 c-1 f) furnished the desired annulation products 3 ca-3 fa in moderate to good yields. The substrate bearing a meta-tolyl group (1 g) underwent exclusive annulation on the less hindered aryl C-H bond, thus affording 3 ga as the single regioisomer. In contrast, the substrate bearing a 2naphthyl group (1 h) afforded a mixture of the regioisomeric products 3 ha and 3 ha' with low regioselectivity, which would reflect the difficulty in controlling the regioselectivity of naphthalene C-H palladation.[12] The ortho-tolyl-substituted substrate reacted rather sluggishly (see 3ia), presumably due to steric clash between the methyl group and the naphthyl ring. The reaction tolerated the substrate bearing 4-pyridyl group, albeit with low reaction efficiency (see 3 ja).

The present catalytic system also proved to be effective for twofold annulation between a teraryl substrate **5** bearing two 1-bromonaphthyl moieties and **2a**, which afforded the double heptagon product **6** in a respectable yield of 47% (Scheme 3 a). X-ray crystallographic analysis of **6** revealed its highly nonplanar and *Ci*-symmetric structure and the lack of

Scheme 2. Pd-catalyzed annulation of various 1-halo-8-arylnaphthalenes $(1\,b-1\,j)$ with diphenylacetylene $(2\,a)$. X=Br except for $1\,b$ (X=I). The products $3\,ha$ and $3\,ha'$ were obtained as a mixture, and their ratio was determined by 1H NMR (see the Supporting Information for more details).

www.chemeurj.org

Scheme 3. Attempts on twofold annulation.

significant π – π stacking in the crystal packing. In contrast, we could not detect a twofold annulation product in the reaction between 1,5-dibromo-9,10-diarylanthracene **7** and **2a**. The reaction gave multiple products, among which only an unusual 1:3 coupling product **8** could be isolated and unambiguously characterized by X-ray analysis (Scheme 4b). Thus, one of the two reaction sites of **7** underwent the expected heptagon formation, whereas the other site reacted with two molecules of diphenylacetylene to give rise to the peculiar bridged polycyclic moiety through cleavage of the C¹–C³a bond of the anthracene moiety.

Scheme 4. Dehydrogenative cyclization leading to an extended conjugated system.

Exposure of the heptagon products **3 aa** and **3 ab** to DDQ/ methanesulfonic acid, which is among the typical conditions for the Scholl reaction, [13] did not cause expected dehydrogenative C—C coupling between the aryl substituents or between the aryl substituent and the benzo[4,5]cyclohepta[1,2,3-de]-naphthalene core. Instead, oxidative fragmentation and ring contraction took place to afford 4H-benzo[de]anthracen-4-one derivatives **9 a** and **9 b** in moderate yields. [11,14] Further exploration of transformations of the present heptagon compounds is under way.

UV/Vis absorption spectra of selected mono-heptagon compounds in CH₂Cl₂ featured the longest wavelength absorption maximum (λ_{max}) around 350–360 nm, showing moderate impact of the peripheral substituents on λ_{\max} (see Figures S7 and S8, Supporting Information). None of them showed visible fluorescence in dilute solution presumably due to nonradiative decay of the excited state through facile rotation of the arylaryl bonds. On the other hand, the bis-heptagon compound 6 showed λ_{max} of 284, 320, and 380 nm and moderate fluorescence at 558 nm (Φ = 0.38) (Figure S9). All the heptagon compounds are more distinctly emissive in the solid state, with $\lambda_{\rm em}$ of 489, 496, and 508 nm for 3ab, 3fa, and 6, respectively (Figure S10). The aggregation-induced emission (AIE)[15] behavior of 3ab was also observed using THF/water mixture samples (Figure S11). This fluorescence behavior could be ascribed to the nonplanar geometry that prevents π – π stacking as well as the aryl-aryl bonds that freely rotate in solution but are rotationally restricted in the aggregated state.

Proposed reaction pathways of the present annulation reaction are shown in Scheme 5. Oxidative addition of the C–Br bond of 1 to Pd⁰, followed by halide abstraction by Ag₂CO₃, ^[3a]

Scheme 5. Proposed reaction pathways for the Pd-catalyzed annulation. Relative free energies (in parentheses) and activation energies (kcal mol $^{-1}$) were calculated for a model system with [Pd] = PdP(4-ClC $_6$ H $_4$) $_3$ and X = OAc.

would generate a naphthylpalladium species **A** (X=carbonate). Insertion of the alkyne **2** into **A** and subsequent intramolecular C—H palladation of the alkenylpalladium species **B**, presumably via the concerted metalation—deprotonation mechanism, would lead to an eight-membered palladacycle **C**. Finally, reductive elimination of **C** would furnish the heptagon product **3** while regenerating Pd⁰. The species **A** could also undergo intramolecular C—H palladation to give a six-membered palladacycle **D**. Reductive elimination of **D** would afford fluoranthene **4**, while **D** might also undergo alkyne insertion into either of the Pd—C bonds, leading to **C** or its isomer.

The feasibility of the key steps in the above catalytic cycle was probed by DFT calculations starting from a model 8-phenylnaphthylpalladium complex A bearing P(4-ClC₆H₄)₃ and acetate ligands (see the Supporting Information for detail). Insertion of diphenylacetylene into A required an activation energy (ΔG^{\dagger}) of 33.6 kcal mol⁻¹, which was slightly lower than that of intramolecular C–H palladation of **A** (35.6 kcal mol⁻¹). The alkyne insertion was exergonic by 3.2 kcal mol⁻¹, and the subsequent C-H palladation of B and reductive elimination of C occurred with lower barriers. The formation of **D** was endergonic by 21.9 kcal mol⁻¹. Though reductive elimination of **D** was feasible ($\Delta G^{\dagger} = 8.5 \text{ kcal mol}^{-1}$), alkyne insertion into **D** required an exceedingly higher barrier ($\Delta G^{\dagger} = 30.1 \text{ kcal mol}^{-1}$). Thus, **D** is unlikely to be responsible for the product **3**, and the selectivity between 3 and 4 would be determined by the competition between alkyne insertion and C-H palladation pathways of A. In agreement with this implication, the reaction of pentadeuterated substrate [D₅]-1a with 2a using AsPh₃ or P(4-ClC₆H₄)₃ resulted in a markedly diminished yield of **4** (Scheme 6), compared with the reaction of parent 1 a (Table 1, entries 3 and 4).

In summary, we have developed a palladium-catalyzed annulation reaction between 1-halo-8-arylnaphthalenes and diarylacetylenes. The reaction represents a rare example of direct

Scheme 6. Reaction of pentadeuterated substrate [D₅]-1 a.

heptagon formation from simple aromatic precursors, enabling facile preparation of PAHs with warped geometries. Further extension of the scope of heptagon-forming annulation and its application to the synthesis of novel heptagon-embedded PAHs are underway.

Acknowledgements

This work was funded by the Ministry of Education (Singapore) and Nanyang Technological University (RG 114/15 and MOE2016-T2-2-043). We thank Drs. Yongxin Li and Rakesh Ganguly (Nanyang Technological University) for assistance with X-ray crystallographic analysis, and Prof. Qichun Zhang and Dr. Changjiang Yao (Nanyang Technological University) for assistance with measurements of solid-state fluorescence.

Conflict of interest

The authors declare no conflict of interest.

Keywords: alkynes \cdot annulation \cdot C—H activation \cdot palladium \cdot polycyclic aromatic hydrocarbons

- For selected reviews, see: a) J. E. Anthony, Angew. Chem. Int. Ed. 2008, 47, 452–483; Angew. Chem. 2008, 120, 460–492; b) X. Feng, W. Pisula, K. Müllen, Pure Appl. Chem. 2009, 81, 2203–2224; c) L. Chen, Y. Hernandez, X. Feng, K. Mullen, Angew. Chem. Int. Ed. 2012, 51, 7640–7654; Angew. Chem. 2012, 124, 7758–7773; d) A. Narita, X. Feng, K. Müllen, Chem. Rec. 2015, 15, 295–309; e) Y. Segawa, H. Ito, K. Itami, Nat. Rev. Mater. 2016, 1, 15002.
- [2] H. Ito, K. Ozaki, K. Itami, Angew. Chem. Int. Ed. 2017, 56, 11144–11164; Angew. Chem. 2017, 129, 11296–11317.
- [3] a) S. Kawasaki, T. Satoh, M. Miura, M. Nomura, J. Org. Chem. 2003, 68, 6836-6838; b) K. Ueura, T. Satoh, M. Miura, J. Org. Chem. 2007, 72, 5362-5367; c) T. Uto, M. Shimizu, K. Ueura, H. Tsurugi, T. Satoh, M. Miura, J. Org. Chem. 2008, 73, 298-300; d) T. Fukutani, K. Hirano, T. Satoh, M. Miura, J. Org. Chem. 2011, 76, 2867-2874; e) L. Adak, N. Yoshikai, Tetrahedron 2012, 68, 5167-5171; f) L. Ilies, A. Matsumoto, M. Kobayashi, N. Yoshikai, E. Nakamura, Synlett 2012, 23, 2381-2384; g) A. Bej, A. Chakraborty, A. Sarkar, RSC Adv. 2013, 3, 15812-15819; h) K. Komeyama, T. Kashihara, K. Takaki, Tetrahedron Lett. 2013, 54, 5659-5662; i) M. V. Pham, N. Cramer, Angew. Chem. Int. Ed. 2014, 53, 3484-3487; Angew. Chem. 2014, 126, 3552-3555.
- [4] a) G. Wu, A. L. Rheingold, S. J. Geib, R. F. Heck, Organometallics 1987, 6, 1941–1946; b) R. C. Larock, M. J. Doty, Q. Tian, J. M. Zenner, J. Org. Chem. 1997, 62, 7536–7537; c) C. Wang, S. Rakshit, F. Glorius, J. Am. Chem. Soc. 2010, 132, 14006–14008; d) A. Matsumoto, L. Ilies, E. Nakamura, J. Am. Chem. Soc. 2011, 133, 6557–6559; e) T. Nagata, K. Hirano,





- T. Satoh, M. Miura, *J. Org. Chem.* **2014**, *79*, 8960–8967; f) T. Nagata, T. Satoh, Y. Nishii, M. Miura, *Synlett* **2016**, *27*, 1707–1710; g) J. Yan, N. Yoshikai, *Org. Lett.* **2017**, *19*, 6630–6633.
- [5] For [4+2] annulation reactions using 2,2'-difunctionalized biaryls, see: a) Y. Wu, F. Wu, D. Zhu, B. Luo, H. Wang, Y. Hu, S. Wen, P. Huang, Org. Biomol. Chem. 2015, 13, 10386 10391; b) Y. Shoji, N. Tanaka, S. Muranaka, N. Shigeno, H. Sugiyama, K. Takenouchi, F. Hajjaj, T. Fukushima, Nat. Commun. 2016, 7, 12074; c) K. Ozaki, K. Murai, W. Matsuoka, K. Kawasumi, H. Ito, K. Itami, Angew. Chem. Int. Ed. 2017, 56, 1361 1364; Angew. Chem. 2017, 129, 1381 1384; d) K. Kanno, Y. Liu, A. Iesato, K. Nakajima, T. Takahashi, Org. Lett. 2005, 7, 5453 5456.
- [6] a) K. Kawasumi, Q. Y. Zhang, Y. Segawa, L. T. Scott, K. Itami, *Nat. Chem.* 2013, 5, 739–744; b) K. Kato, Y. Segawa, L. T. Scott, K. Itami, *Chem. Asian J.* 2015, 10, 1635–1639; c) K. Kawai, K. Kato, L. Peng, Y. Segawa, L. T. Scott, K. Itami, *Org. Lett.* 2018, 20, 1932–1935.
- [7] a) K. Y. Cheung, X. Xu, Q. Miao, J. Am. Chem. Soc. 2015, 137, 3910-3914;
 b) X. Gu, H. Li, B. Shan, Z. Liu, Q. Miao, Org. Lett. 2017, 19, 2246-2249;
 c) S. H. Pun, C. K. Chan, J. Luo, Z. Liu, Q. Miao, Angew. Chem. Int. Ed. 2018, 57, 1581-1586; Angew. Chem. 2018, 130, 1597-1602.
- [8] W. C. Fu, Z. Wang, W. T. K. Chan, Z. Lin, F. Y. Kwong, Angew. Chem. Int. Ed. 2017, 56, 7166–7170; Angew. Chem. 2017, 129, 7272–7276.
- [9] N. Chernyak, D. Tilly, Z. Li, V. Gevorgyan, Chem. Commun. 2010, 46, 150–152.

- [10] a) M. Smet, J. Van Dijk, W. Dehaen, Synlett 1999, 495–497; b) M. Bown, C. J. Dunn, C. M. Forsyth, P. Kemppinen, T. B. Singh, M. A. Skidmore, K. N. Winzenberg, Aust. J. Chem. 2012, 65, 145–152; c) J. E. Rice, Z.-W. Cai, J. Org. Chem. 1993, 58, 1415–1424; d) M. Yamaguchi, M. Higuchi, K. Tazawa, K. Manabe, J. Org. Chem. 2016, 81, 3967–3974.
- [11] CCDC 1831779 (3 ba), 1831782 (6), 1871891 (8), and 1871892 (9 b) contain the supplementary crystallographic data for this paper. These data are provided free of charge by The Cambridge Crystallographic Data Centre.
- [12] S. R. Neufeldt, M. S. Sanford, Acc. Chem. Res. 2012, 45, 936-946.
- [13] M. Grzybowski, K. Skonieczny, H. Butenschon, D. T. Gryko, Angew. Chem. Int. Ed. 2013, 52, 9900–9930; Angew. Chem. 2013, 125, 10084–10115.
- [14] See Scheme S1 in the Supporting Information for a possible reaction pathway.
- [15] J. Mei, N. L. C. Leung, R. T. K. Kwok, J. W. Y. Lam, B. Z. Tang, Chem. Rev. 2015, 115, 11718–11940.

Manuscript received: November 16, 2018 Accepted manuscript online: November 20, 2018 Version of record online: December 27, 2018

Chem. Eur. J. 2019, 25, 9395 - 9399