

HPCSE I - Fall 2022

Tutorial: Diffusion

26 October 2022

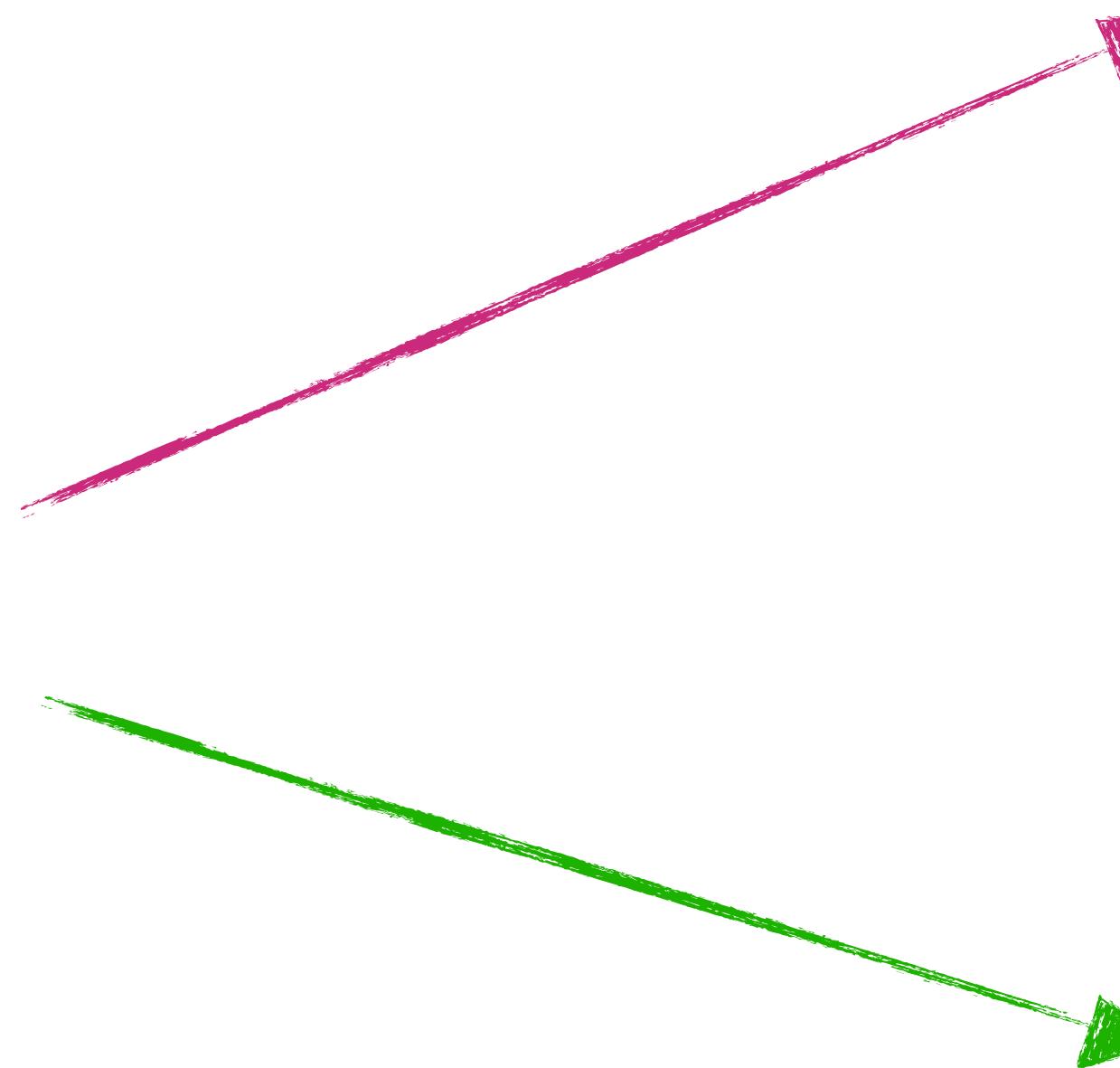
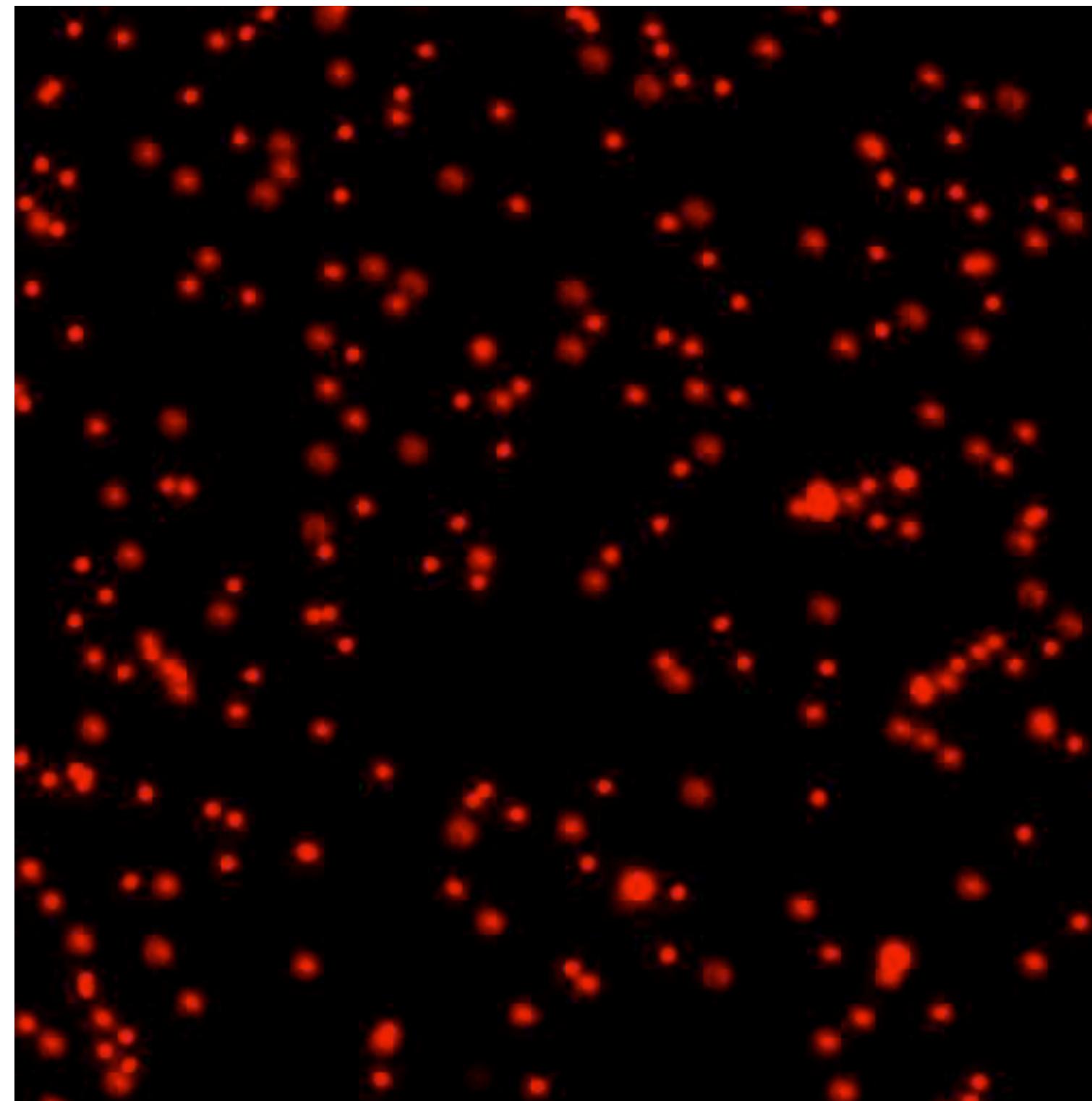
Athena Economides

Computational Science and Engineering Lab

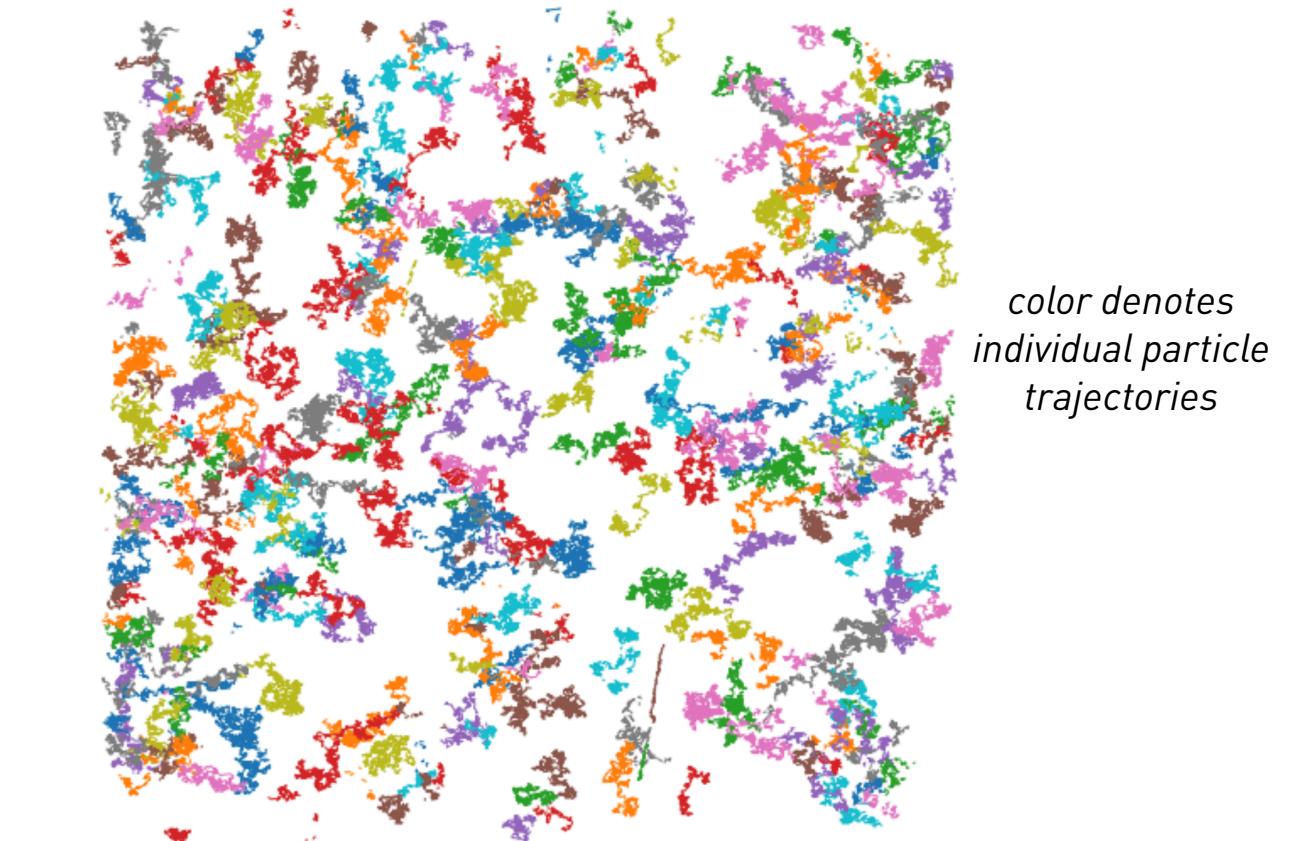
ETH Zürich

How to Quantify Diffusion?

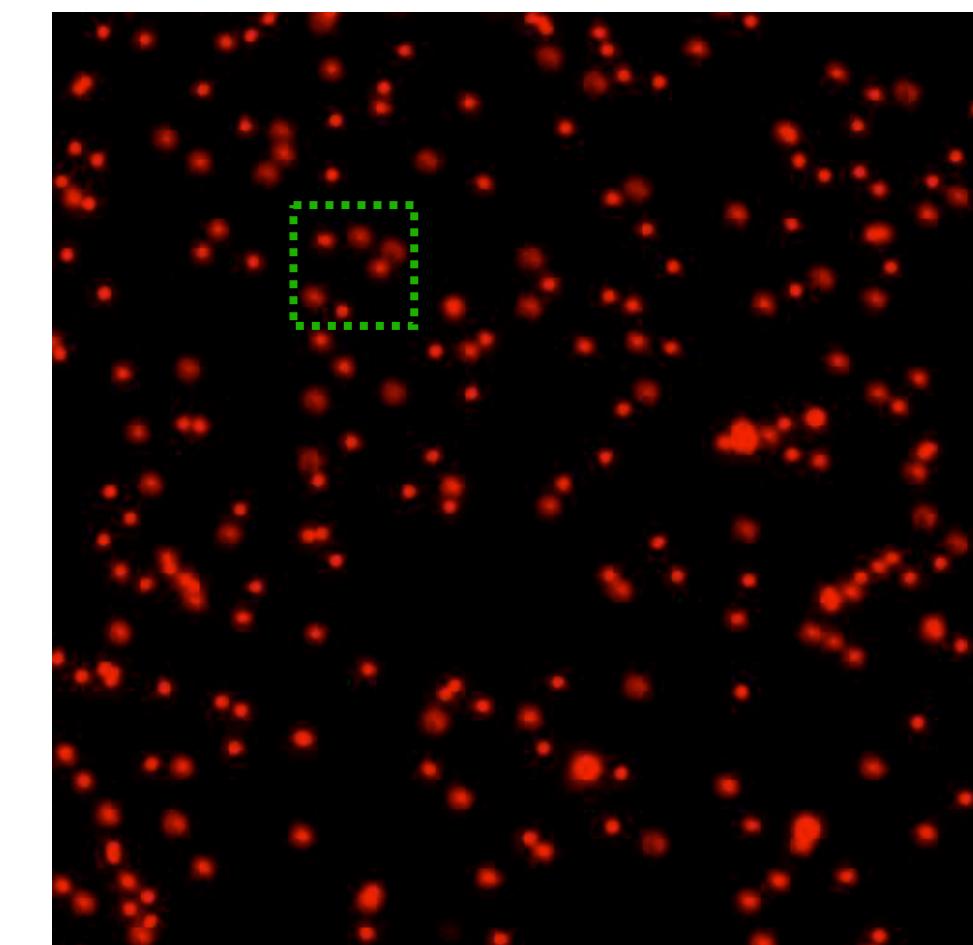
Brownian motion of particles



- **Particle-level:** Track individual **particle trajectory**, in time.
- **Outcome:** Estimation of diffusion coefficient, $D_{(2D)} = \frac{\langle x^2 \rangle}{4t}$



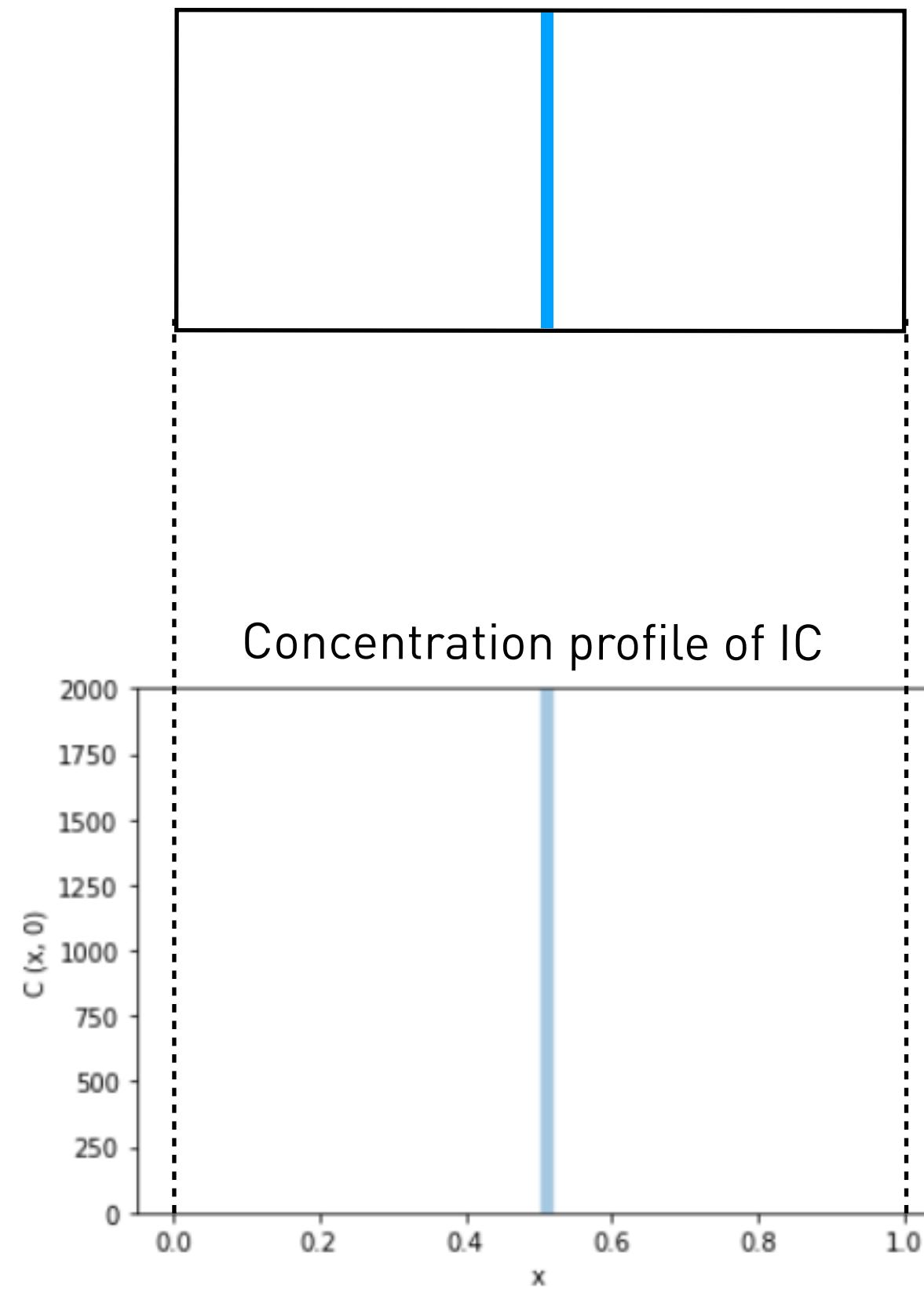
- **Macroscale-level:** Observe **concentration within a given volume**, in time.
- **Outcome:** Estimation of concentration field, $C(x, t)$, over time.



1D diffusion using Random Walks

Initial Condition (IC)

top-view of a plate with a dye placed on a line, at the middle of the plate

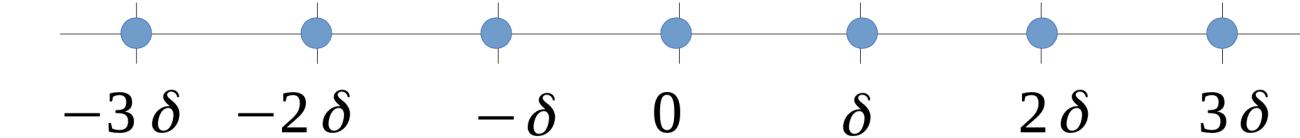


How would the dye "move" on the plate over time?

- How does the concentration profile of the dye change in time?

In the lecture:

A stochastic model of diffusion



- ASSUMPTIONS/RULES :
- 1) Each particle steps to the left or to the right once every τ , moving with a velocity $\pm U$ a distance $\delta = \pm U\tau$ (τ, δ are constants - depend on liquid, particle size, and T).
- 2) The probability of going to the left or to the right is 1/2. Successive steps are independent. The walk is not biased.
- 3) Each Particle moves independent of all other particles

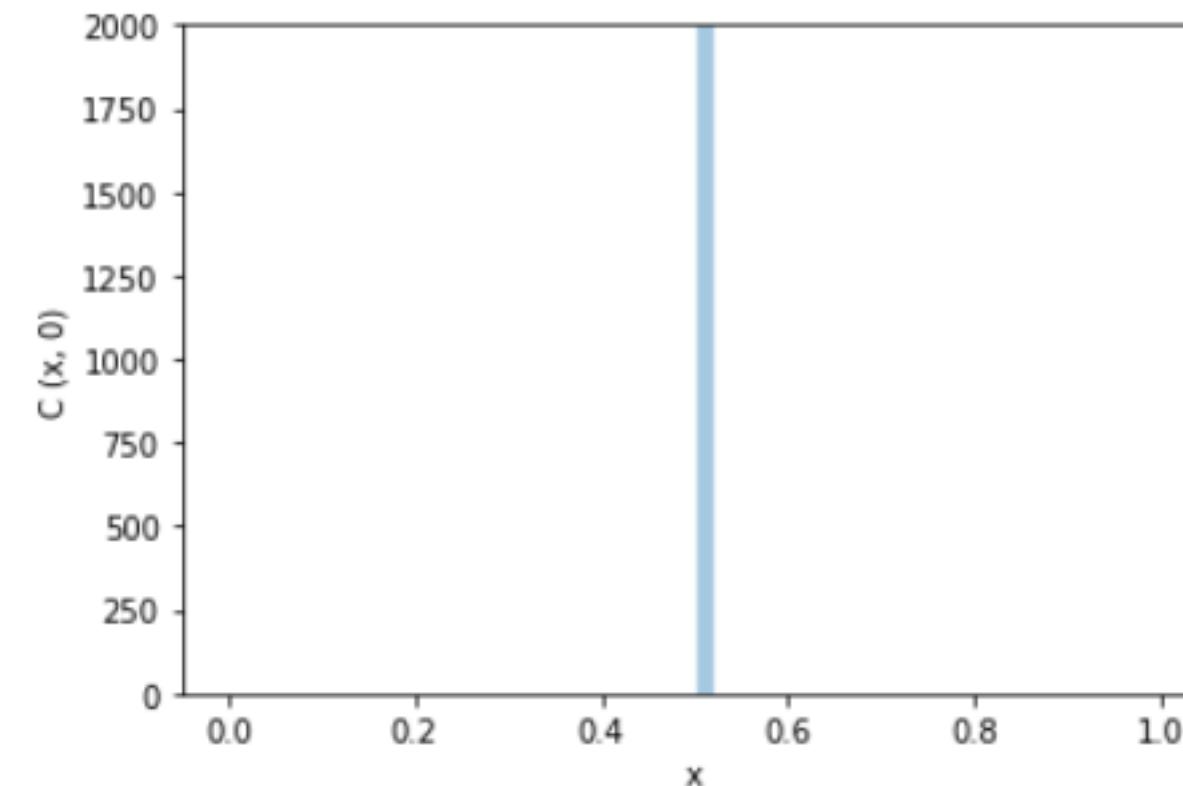
Let's simulate this!

- Simulate dye particles initially placed at the cyan line, in the middle of the plate.
- At each time-step: move each particle independently in space, with probability 0.5 to go left, and 0.5 to go right.
- Observe evolution of particles' position over time.
- Compute concentration of particles at each time step.

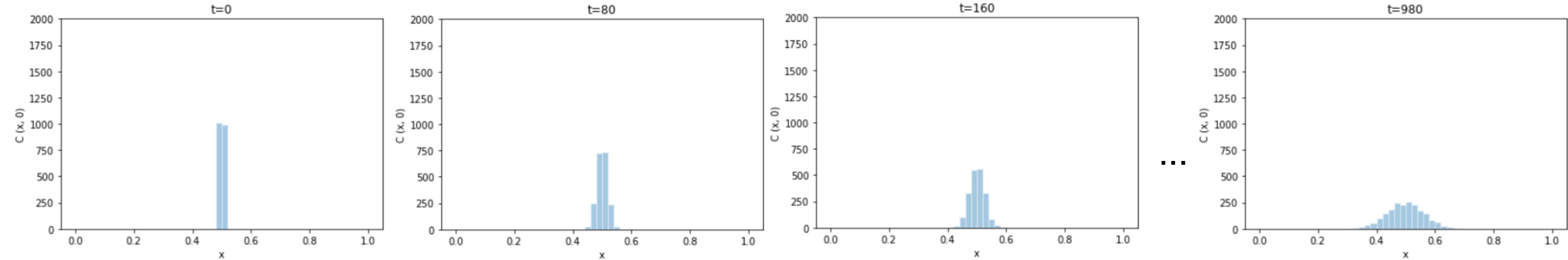
number of particles / volume of medium

1D diffusion using Random Walks

Initial Condition (IC)



Time evolution of particle concentration



... What happens at the Edge of the plate?? → Boundary Conditions (BC)

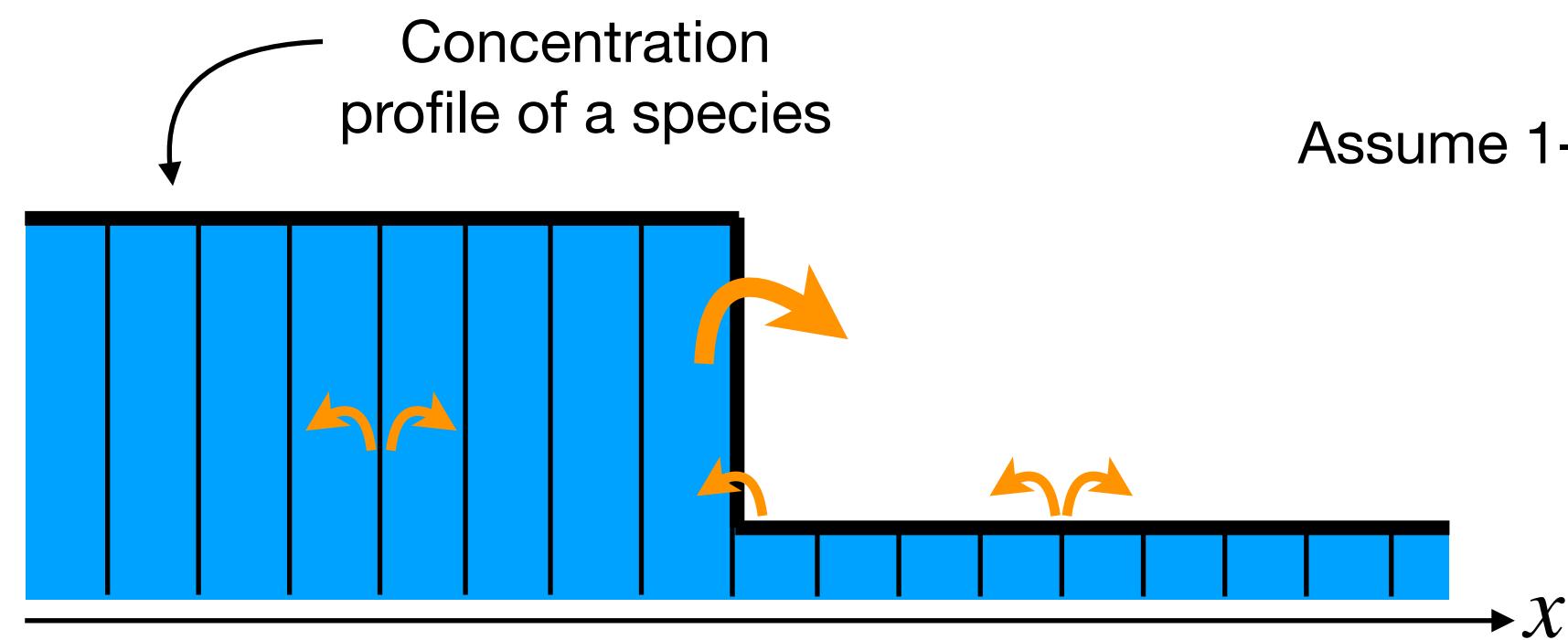
- Dye drops out of the plate (absorbing BC).
- Dye cannot leave the plate due to presence of walls (reflective BC).

1D diffusion using Random Walks

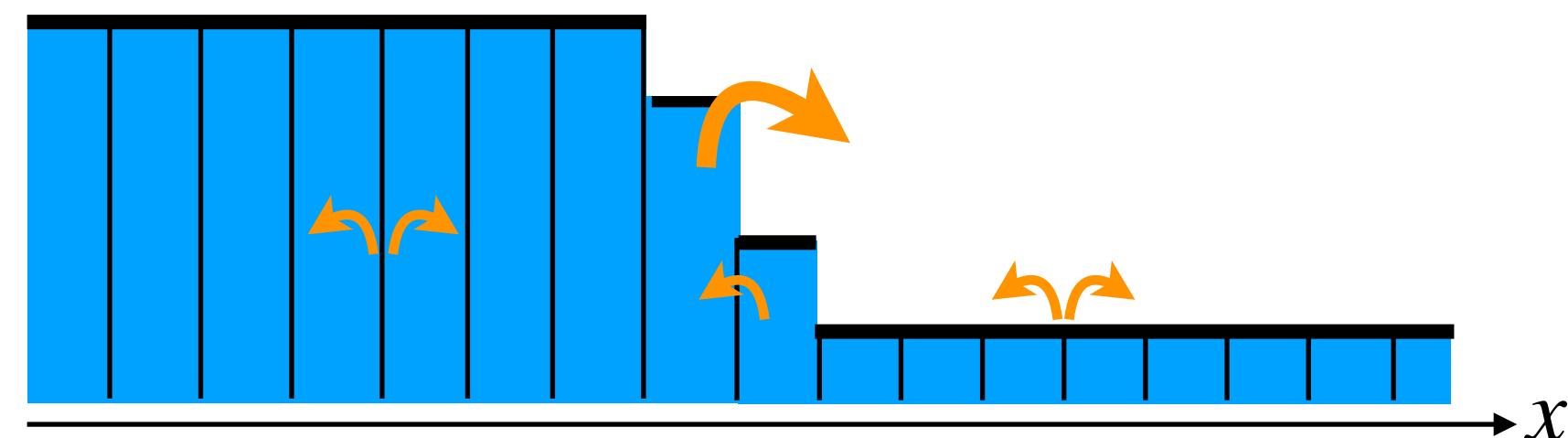
According to our assumptions, the motion of **one particle** is fully time-reversible and **random**. Yet when **particles move together**, they move in a **particular direction**.

- Why?
- ... i.e. What determines the direction of the macroscopically observed motion?

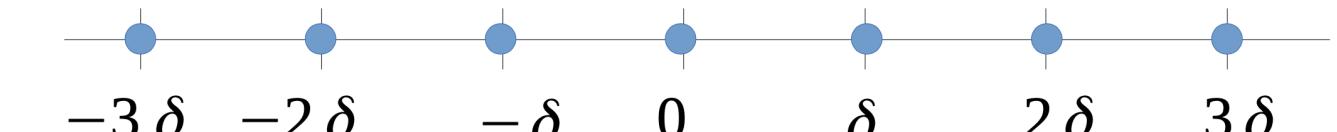
Particle-level picture:



...



A stochastic model of diffusion



- ASSUMPTIONS/RULES :
- 1)Each particle steps to the left or to the right once every τ , moving with a velocity $\pm U$ a distance $\delta = \pm U\tau$ (τ, δ are constants - depend on liquid, particle size, and T).
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Fick's 2nd Law

$$\frac{\partial c}{\partial t} = - \frac{\partial J_x}{\partial x} = D \frac{\partial^2 c}{\partial x^2}$$

1D diffusion using Random Walks

Conclusion 1: Average particle position remains constant.

1D Random Walk

- According to rule 1: $x_i(n) = x_i(n-1) \pm \delta$
- Mean displacement of N particles after n steps:

$$\langle x_i(n) \rangle = \frac{1}{N} \sum_{i=0}^N x_i(n)$$

$$\langle x_i(n) \rangle = \frac{1}{N} \sum_{i=0}^N [x_i(n-1) \pm \delta]$$

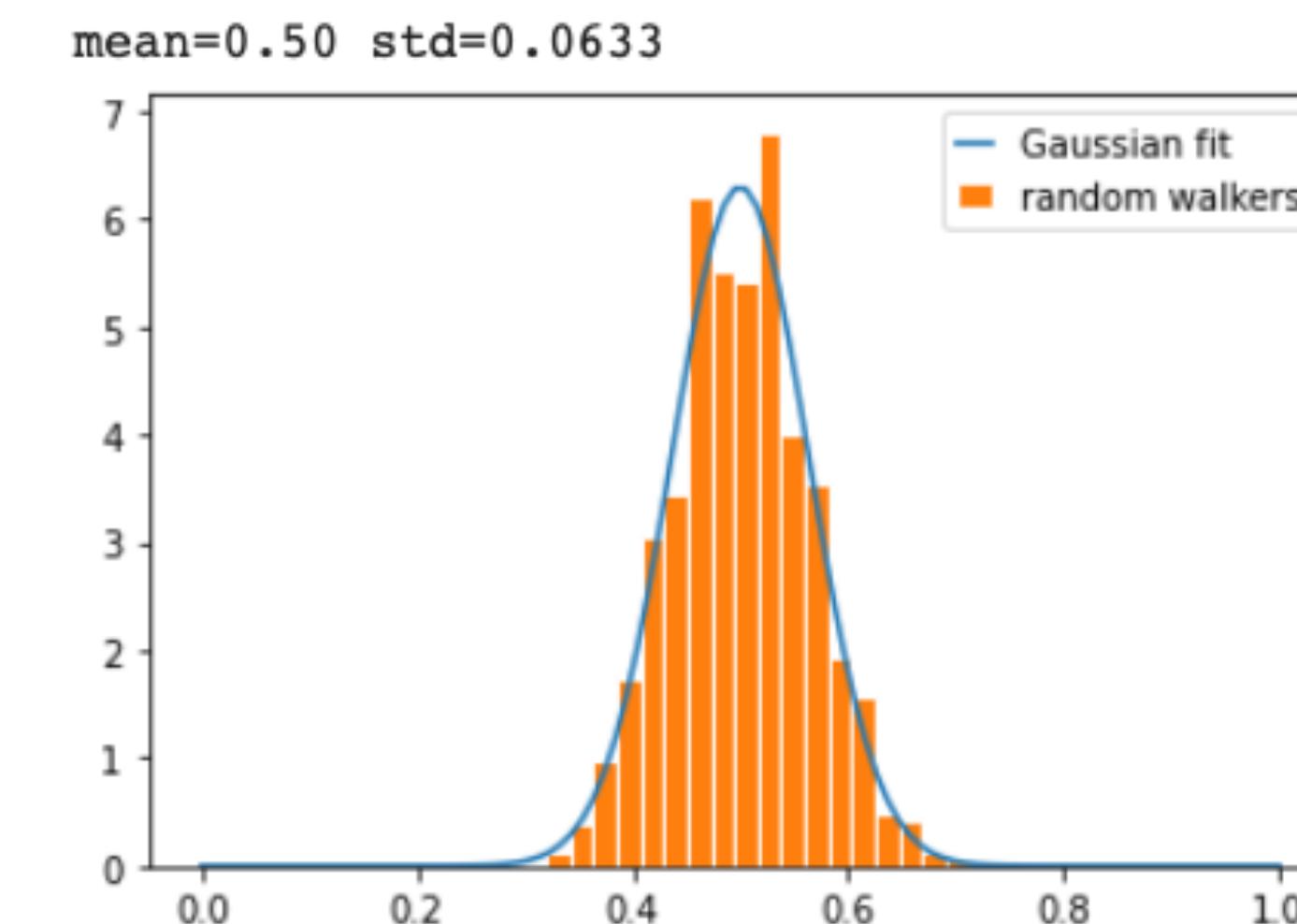
$$\langle x_i(n) \rangle = \frac{1}{N} \sum_{i=0}^N [x_i(n-1)]$$

$$\langle x_i(n) \rangle = \langle x_i(n-1) \rangle = \dots = \langle x_i(0) \rangle$$

- The average particle position remains constant.

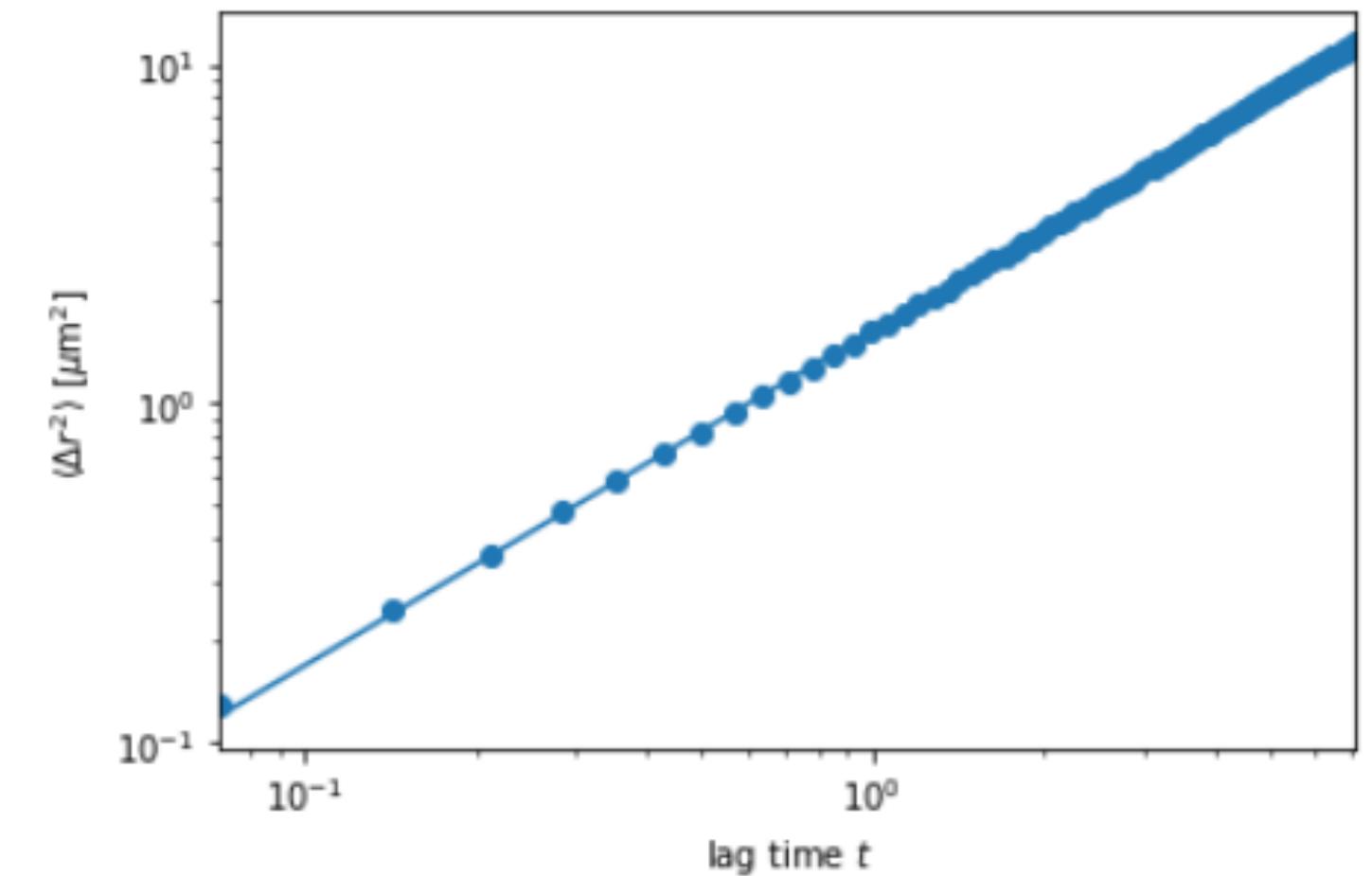
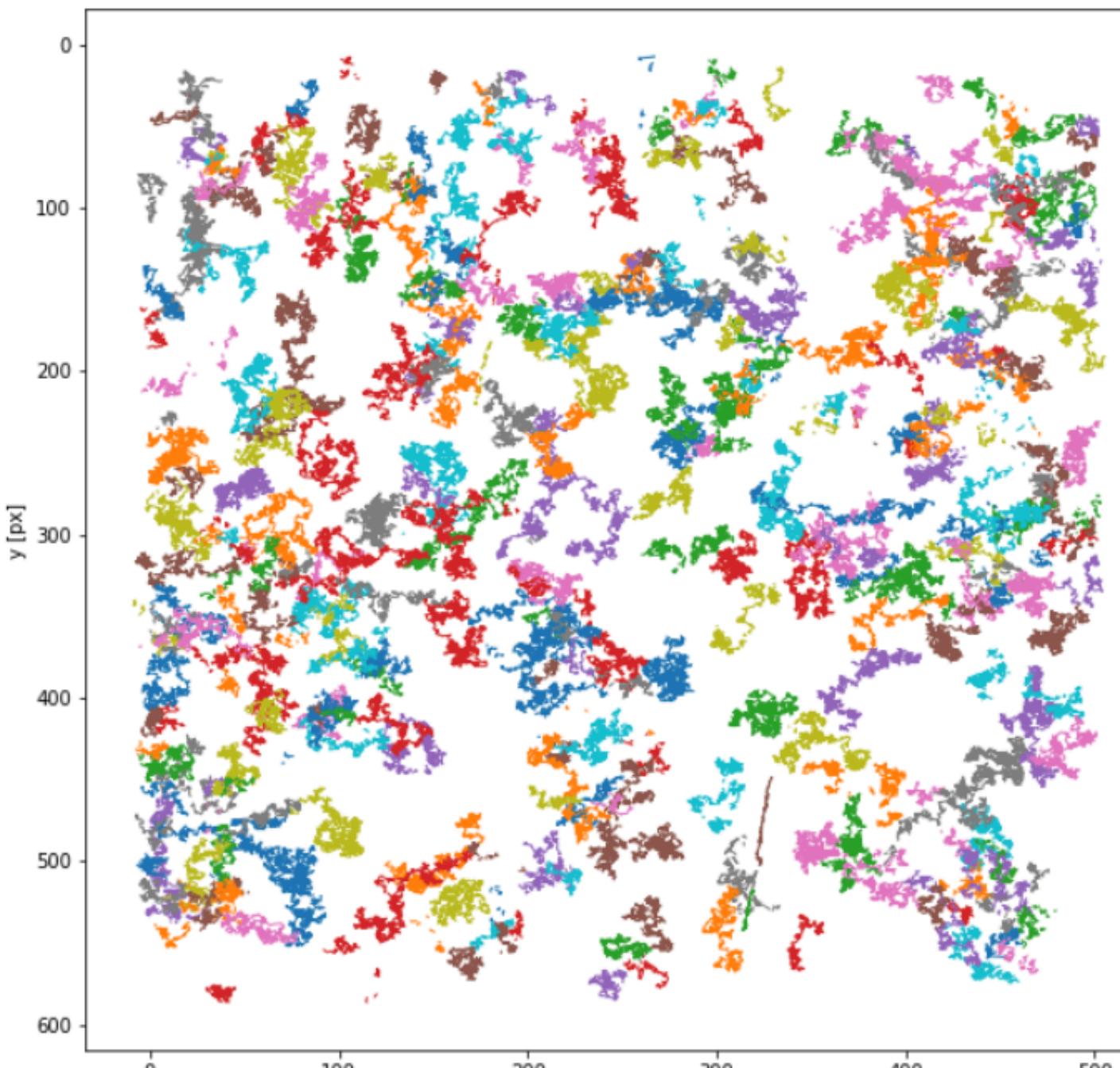
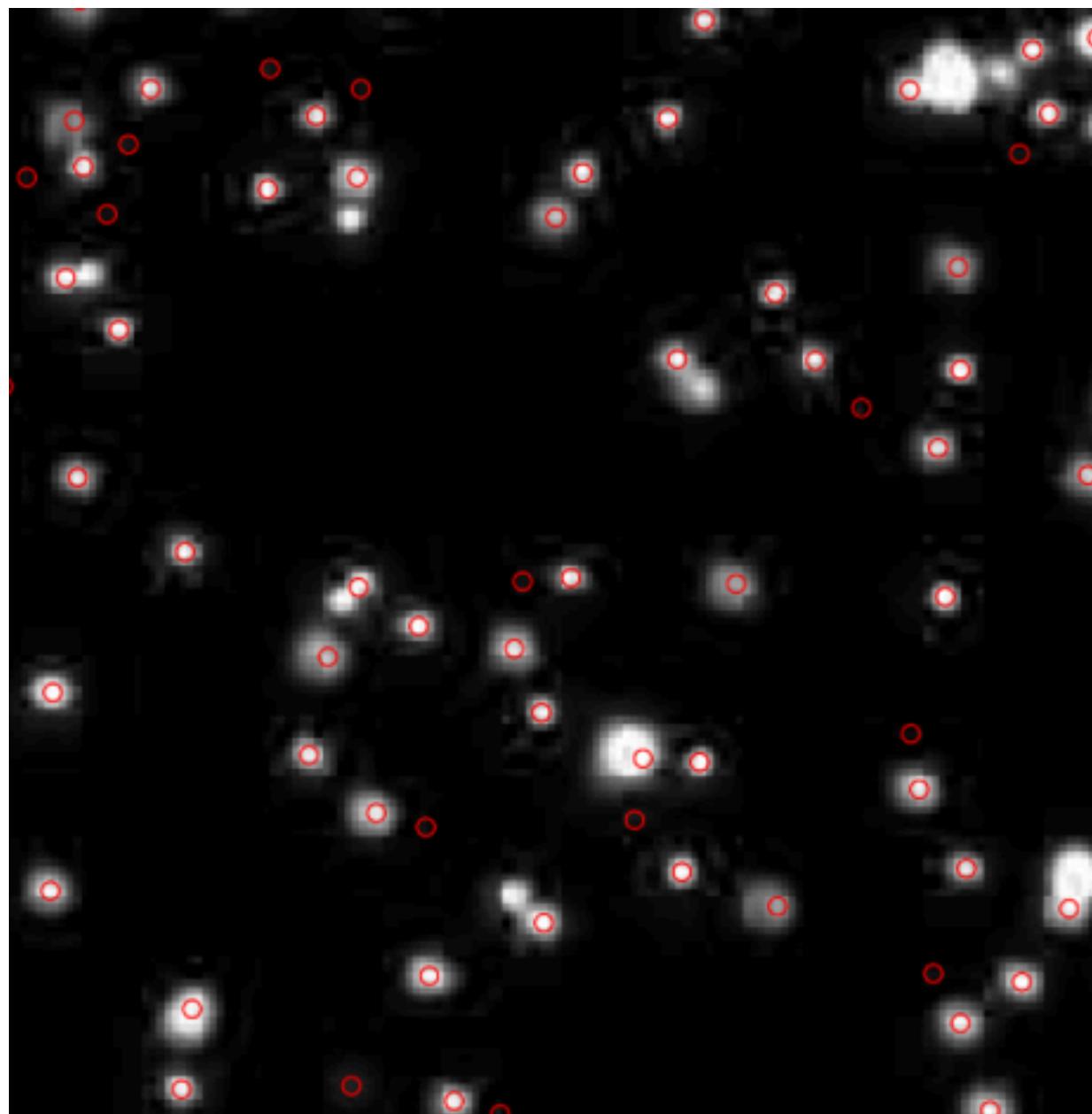
Let's verify this from our 1D particle simulation!

- We compare the distribution of the random walkers with the shape of a Gaussian probability density function (pdf).
- To do this, we use fit the x-positions of the random walkers at the final timestep, with a Gaussian pdf.
- We then plot together, the histogram of the x-positions of the random walkers, and the Gaussian pdf fit.



Computing the Diffusion coefficient

Track each particle's position over time



Ensemble mean
displacement over time

1D Random Walk (cont.)

- The particles spread:

$$\langle x_i^2(n) \rangle = \langle x_i^2(0) \rangle + n \delta^2$$

- The rate of the mean square displacement depends on the medium in which diffusion takes place and is linear in time:

$$n \delta^2 = \frac{t}{\tau} \delta^2 = \frac{\delta^2}{\tau} t$$

Define the diffusion constant (D):

$$D = \frac{\delta^2}{2 \tau}$$

Macroscopic viewpoint: The Diffusion Equation

For the concentration $C(x, t)$ we can describe the evolution in space $x \in \Omega \subseteq \mathbb{R}^2$ and time $t \in [0, T]$ as

$$\frac{\text{first order time derivative}}{\frac{\partial C(\mathbf{x}, t)}{\partial t}} = \frac{\text{second order spatial derivatives}}{D \nabla^2 C(\mathbf{x}, t)}$$

diffusion coefficient

where the diffusion constant D is a material property

Values of diffusion coefficients (gas)

Species pair (solute – solvent) ↴	Temperature (°C) ↴	D (cm ² /s) ↴	Reference ↴
Water (g) – air (g)	25	0.282	[3]
Oxygen (g) – air (g)	25	0.176	[3]

Values of diffusion coefficients (liquid)

Species pair (solute – solvent) ↴	Temperature (°C) ↴	D (cm ² /s) ↴	Reference ↴
Acetone (dis) – water (l)	25	1.16×10^{-5}	[3]
Air (dis) – water (l)	25	2.00×10^{-5}	[3]
Ammonia (dis) – water (l)	25	1.64×10^{-5}	[3]

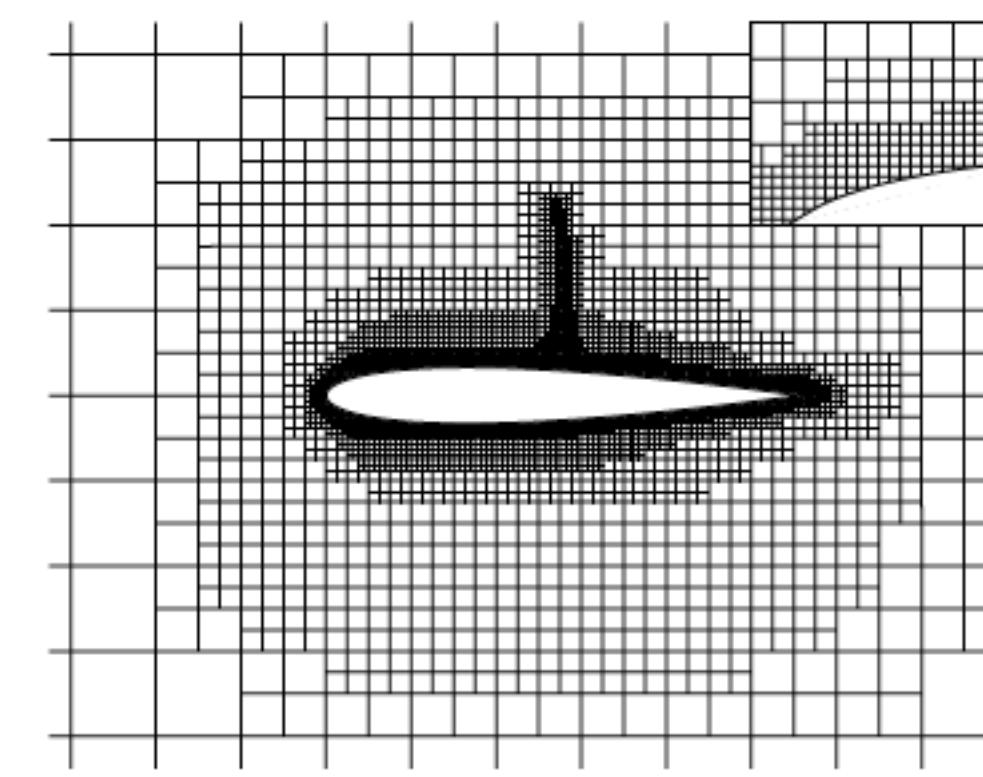
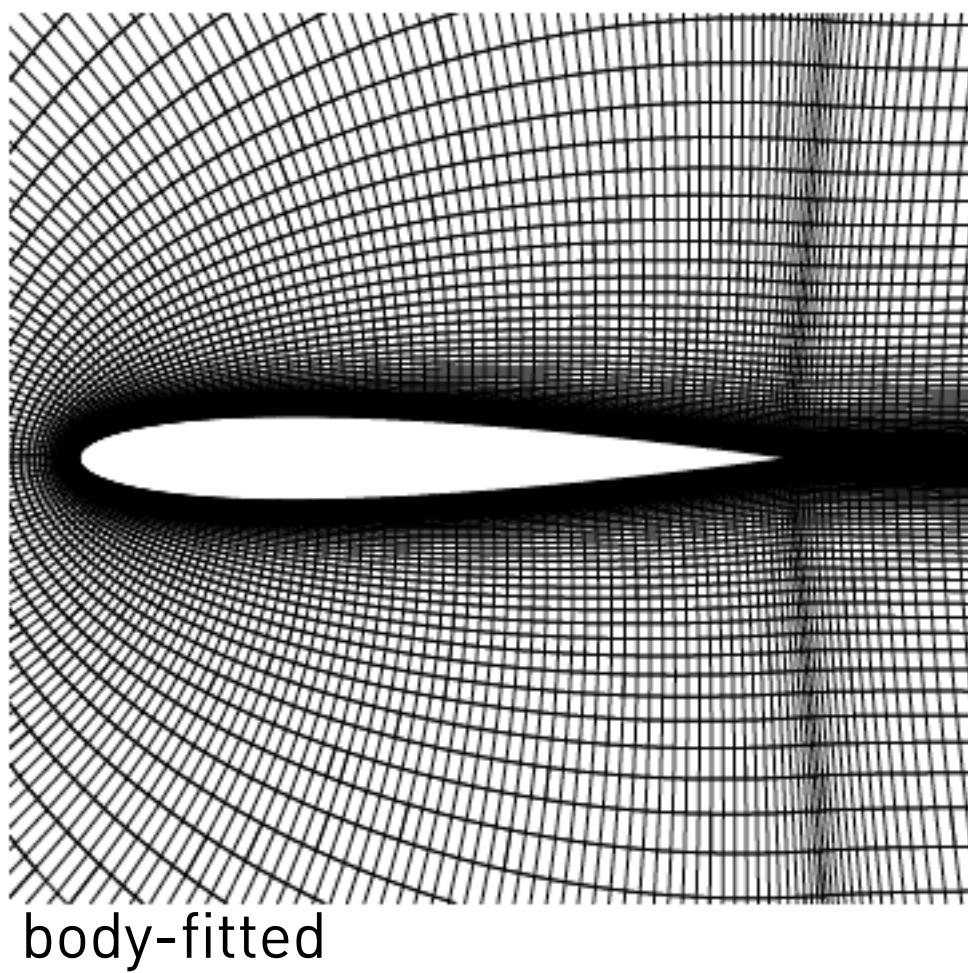
Values of diffusion coefficients (solid)

Species pair (solute – solvent) ↴	Temperature (°C) ↴	D (cm ² /s) ↴	Reference ↴
Hydrogen – iron (s)	10	1.66×10^{-9}	[3]
Hydrogen – iron (s)	100	124×10^{-9}	[3]
Aluminium – copper (s)	20	1.3×10^{-30}	[3]

Discretizing Space - Grids

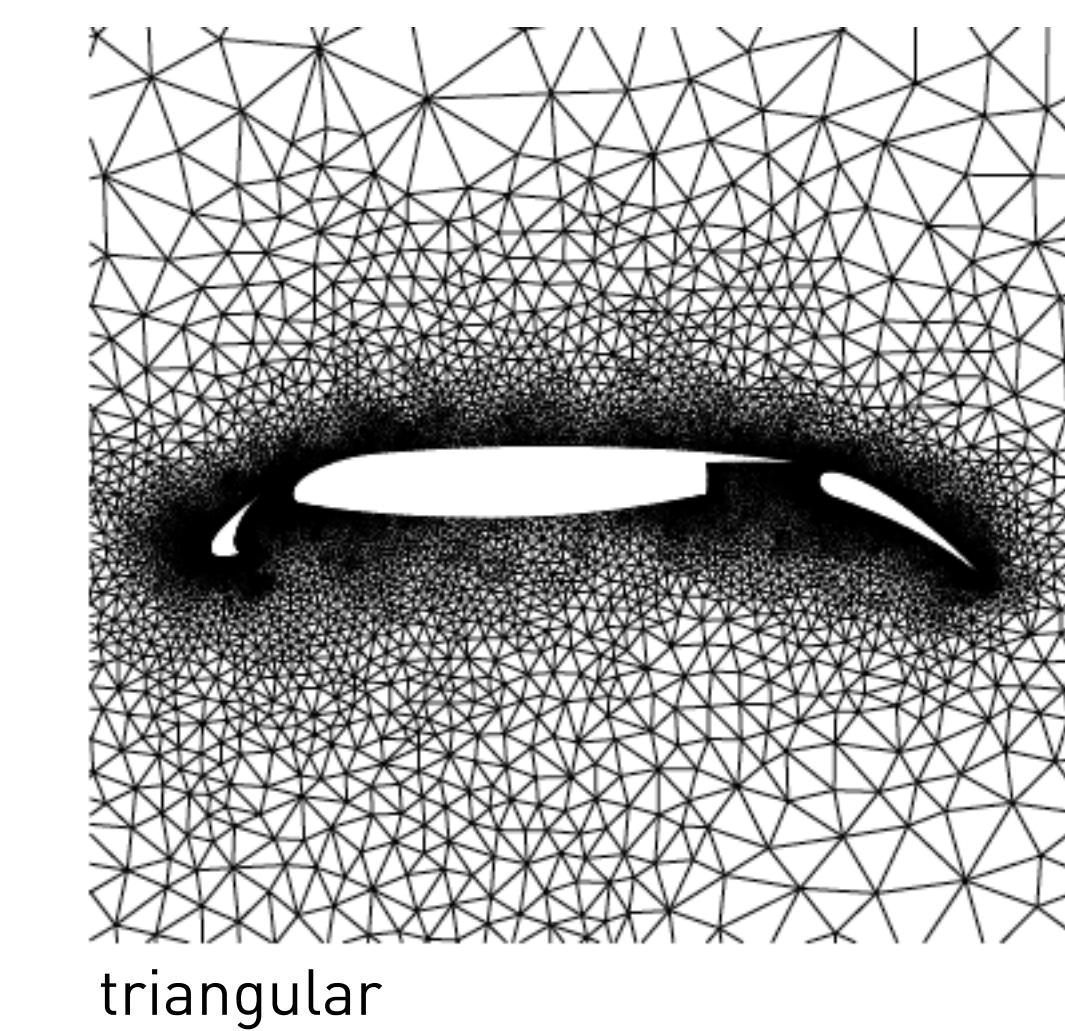
In order to solve the equations numerically we discretize them on a **grid**

structured



cartesian

unstructured



triangular

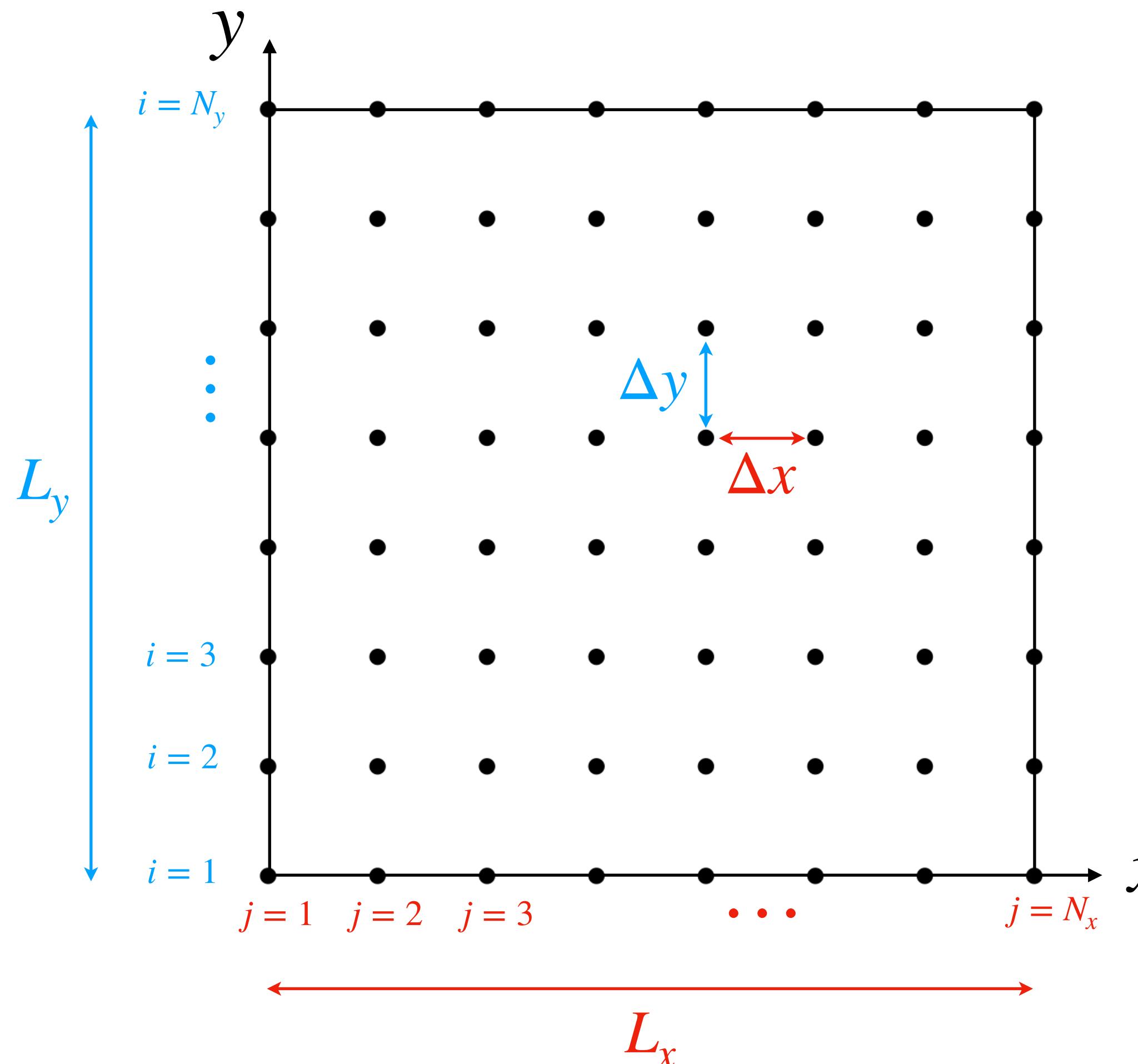
On the grid, the field $C(\mathbf{x}, t)$ takes discrete values $C_{i,j}^n \equiv C(x_i, y_j, t_n)$

Example:

- (uniform) Cartesian Grid $x_i = x_0 + i\Delta x$, $y_i = y_0 + j\Delta y$ and $t_n = t_0 + n\Delta t$

Discretizing a 2D Space - Grids

1. Spatial discretization



Grid spacing:

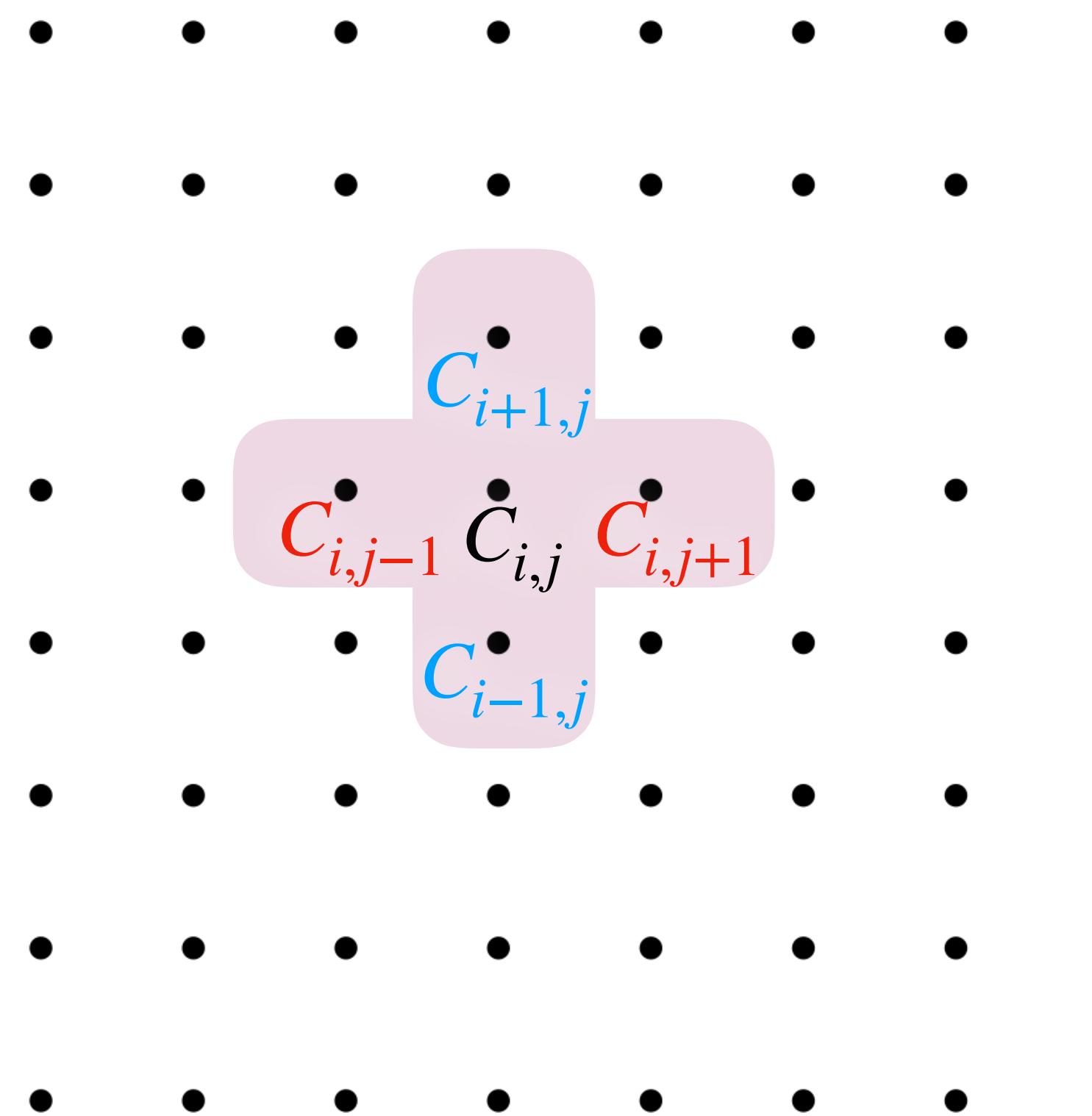
$$\Delta x = \frac{L_x}{N_x - 1}$$

$$\Delta y = \frac{L_y}{N_y - 1}$$

Spatial discretization: Central finite differences.

1. Spatial discretization: using central *finite difference* scheme.

$$\frac{\partial C(x, y, t)}{\partial t} = D \nabla^2 C(x, y, t) = D \left(\frac{\partial^2 C}{\partial x^2} + \frac{\partial^2 C}{\partial y^2} \right)$$


$$\approx D \left(\frac{C_{i,j-1} - 2C_{i,j} + C_{i,j+1}}{\Delta x^2} + \frac{C_{i-1,j} - 2C_{i,j} + C_{i+1,j}}{\Delta y^2} \right)$$

Time discretization: explicit Euler.

2. Time discretization: using *explicit Euler* scheme.

$$\frac{\partial C(x, y, t)}{\partial t} = D \nabla^2 C(x, y, t)$$

$$\frac{\partial C(x, y, t)}{\partial t} \approx \frac{C^{n+1} - C^n}{\Delta t}$$

time index

Combining all together:

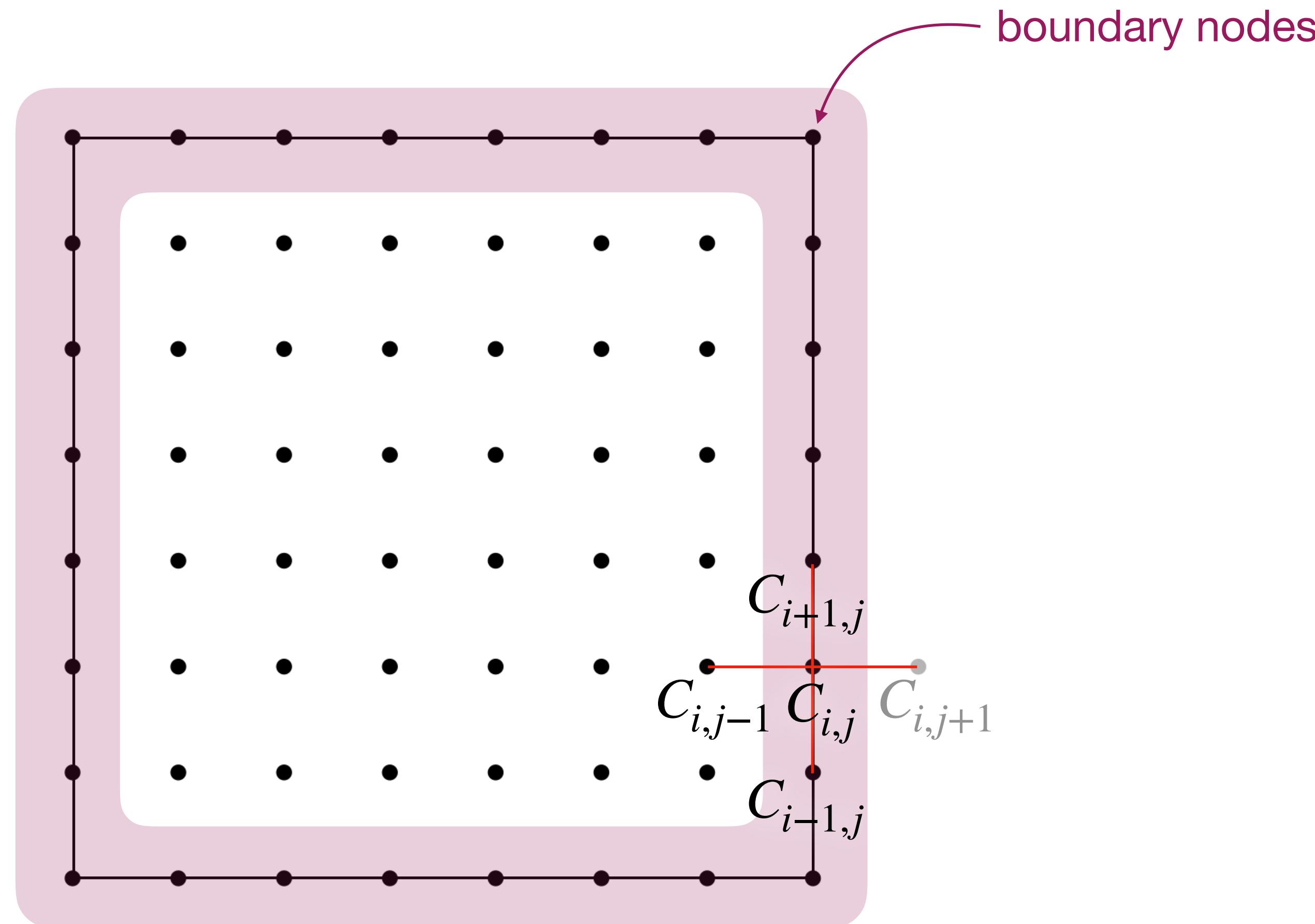
$$\frac{C_{i,j}^{n+1} - C_{i,j}^n}{\Delta t} = D \left(\frac{C_{i,j-1}^n - 2C_{i,j}^n + C_{i,j+1}^n}{\Delta x^2} + \frac{C_{i-1,j}^n - 2C_{i,j}^n + C_{i+1,j}^n}{\Delta y^2} \right)$$

taking: $\Delta x = \Delta y = \Delta r$

$$C_{i,j}^{n+1} = \dots (TODO)$$

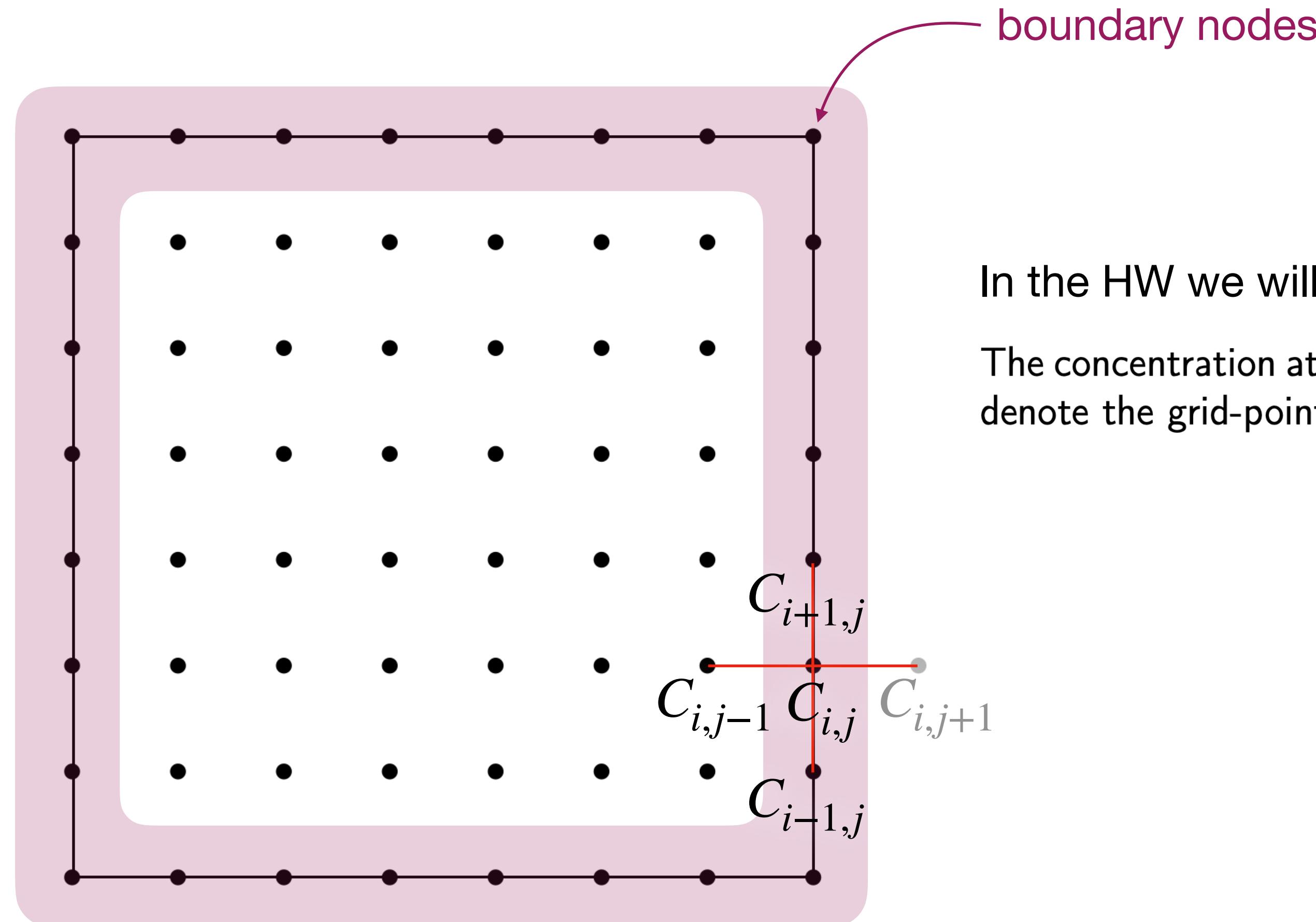
HW#4: Q3: Diffusion

3. Boundary nodes: application of the Boundary Condition



HW#4: Q3: Diffusion

3. Boundary nodes: application of the Boundary Condition

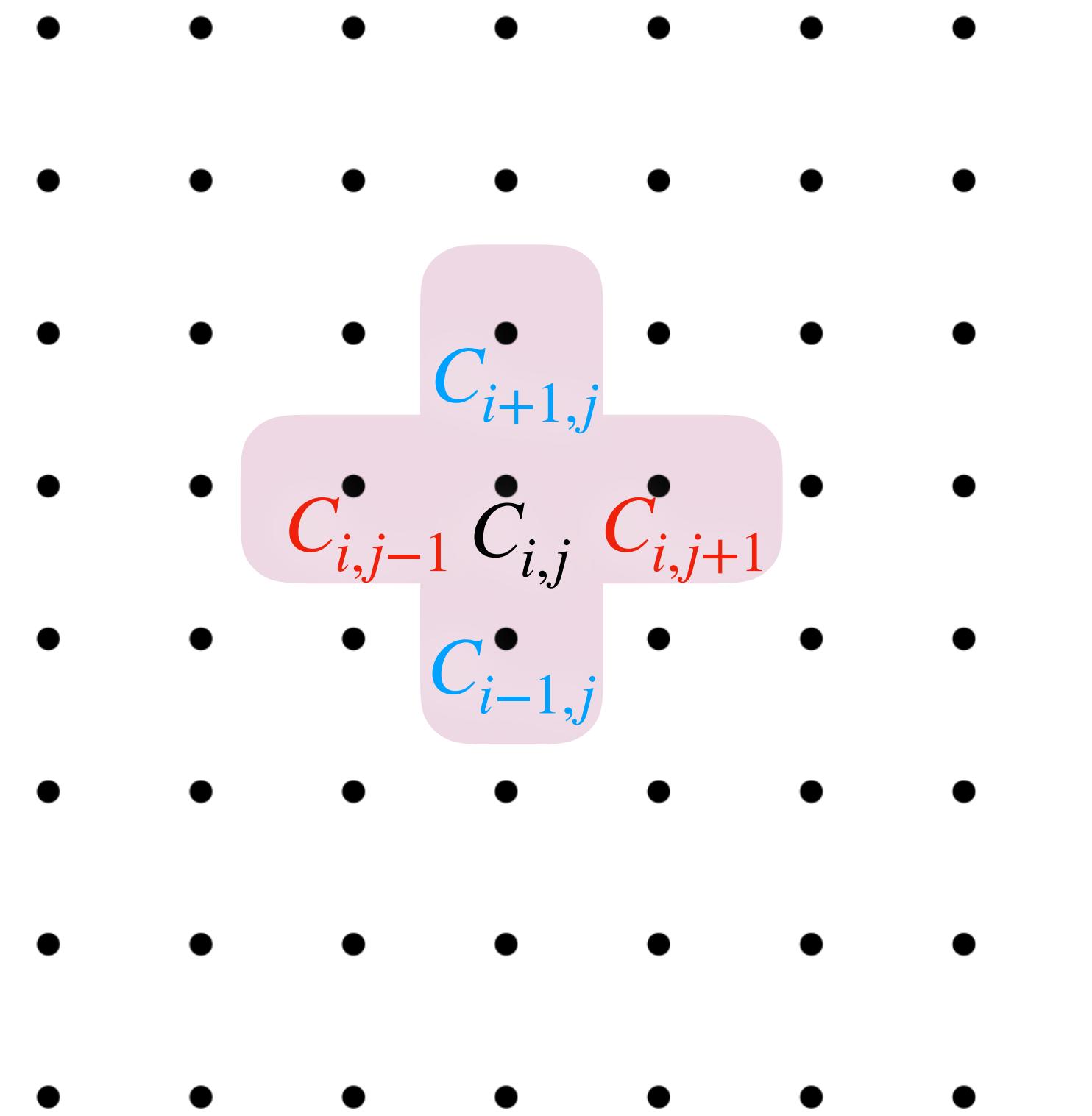


In the HW we will consider the following BC:

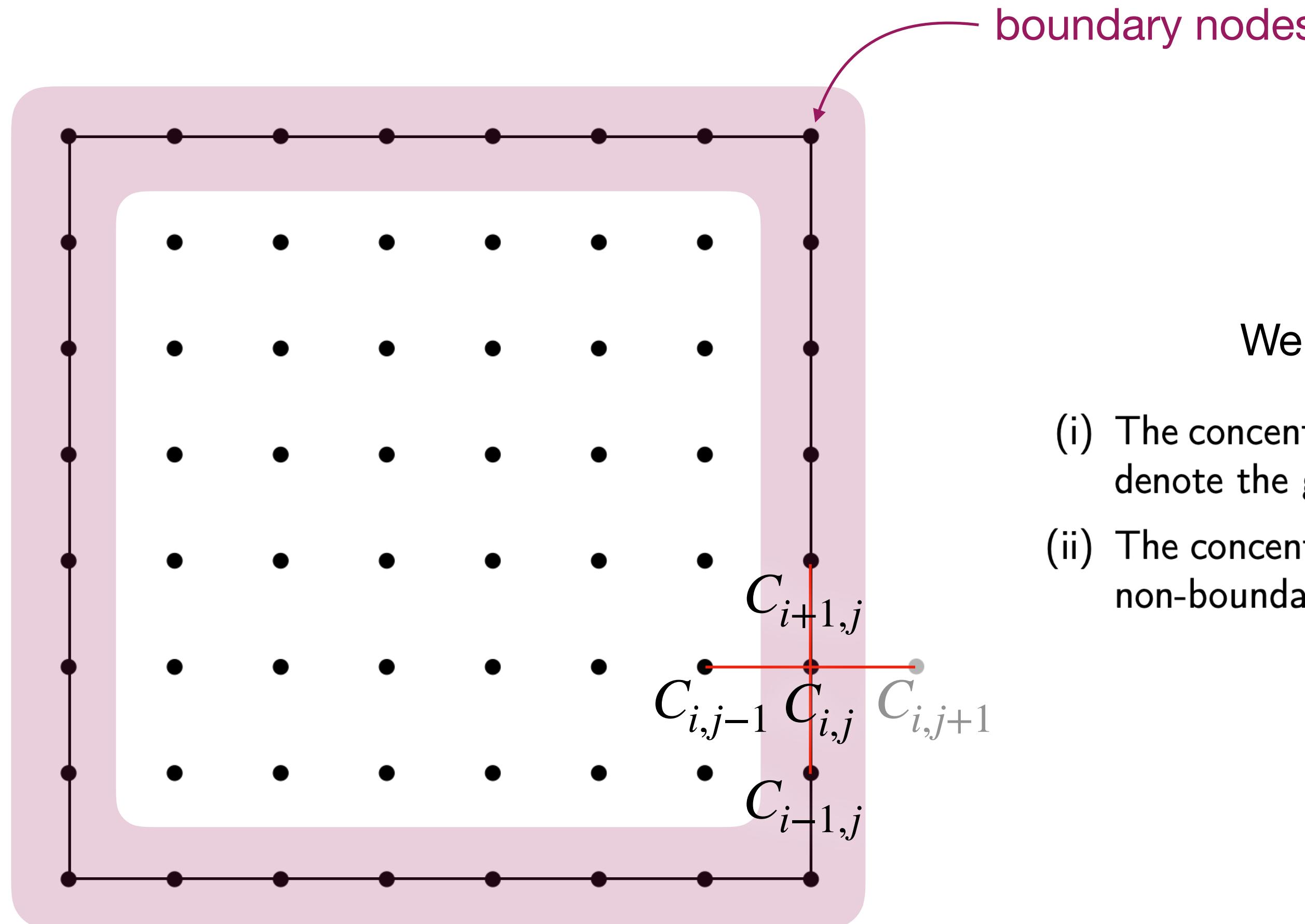
The concentration at all boundary grid-points is 0, i.e. $C(x_B, y_B, t) = 0$, where x_B, y_B denote the grid-points located on the domain boundary (perimeter).

Discretizing Derivatives - Finite Differences

$$\nabla^2 C(x, y, t) = \frac{\partial^2 C}{\partial x^2} + \frac{\partial^2 C}{\partial y^2} \approx \frac{C_{i,j-1} - 2C_{i,j} + C_{i,j+1}}{\Delta x^2} + \frac{C_{i-1,j} - 2C_{i,j} + C_{i+1,j}}{\Delta y^2}$$



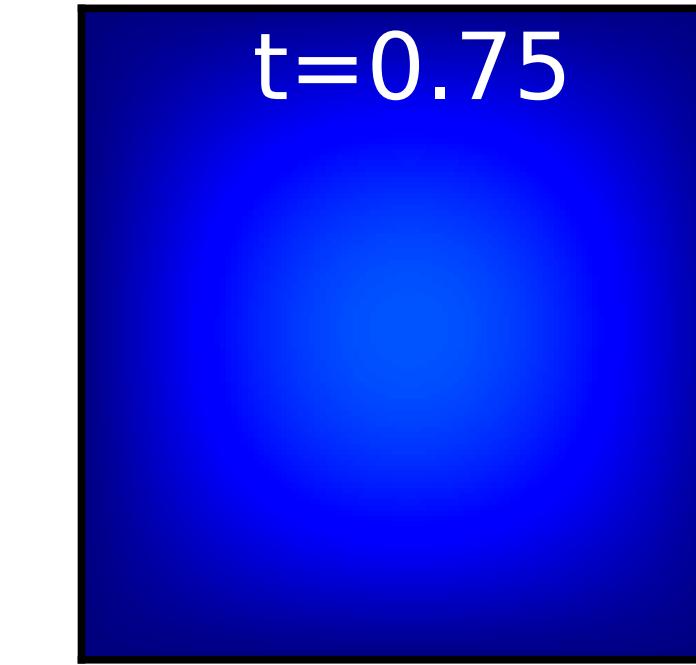
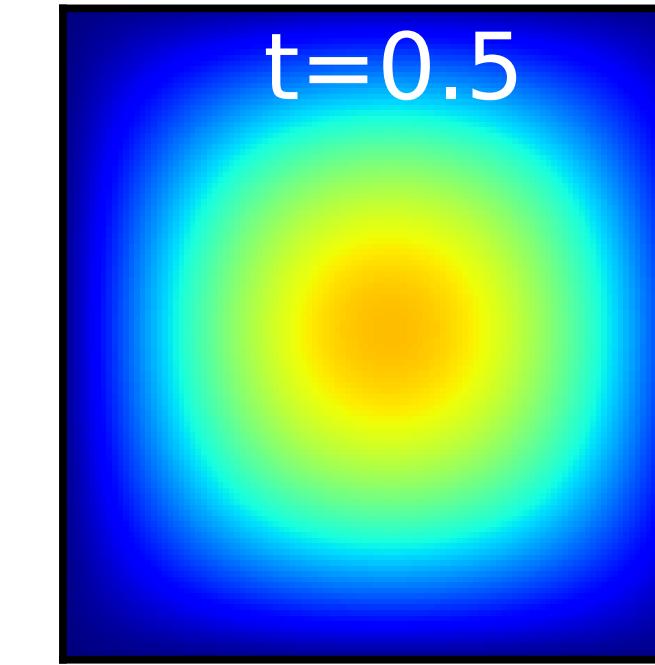
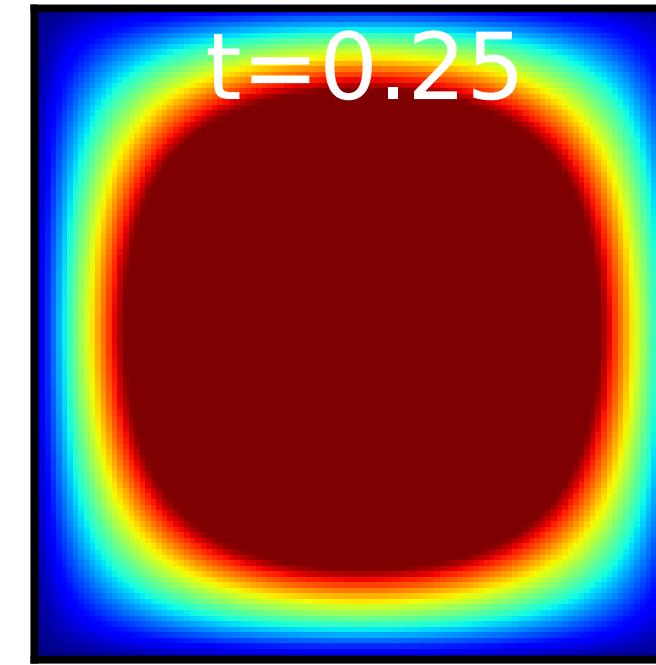
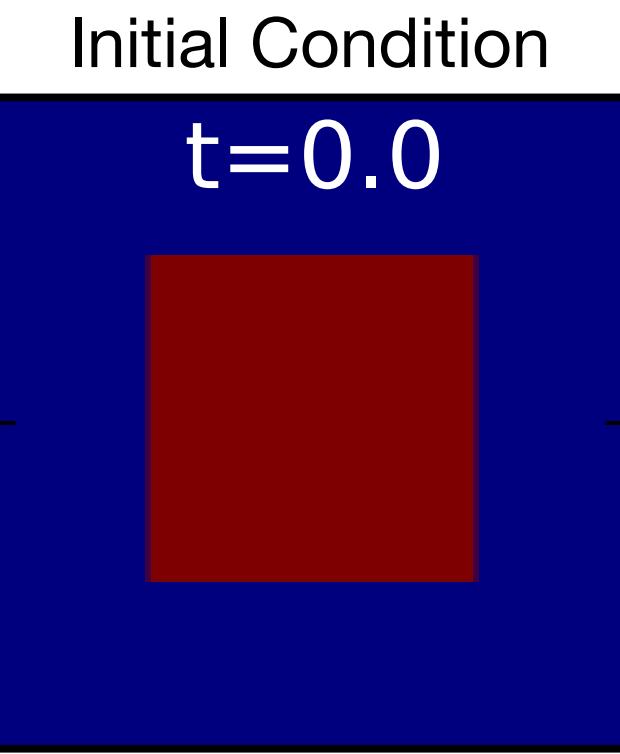
Discretizing Derivatives - Finite Differences



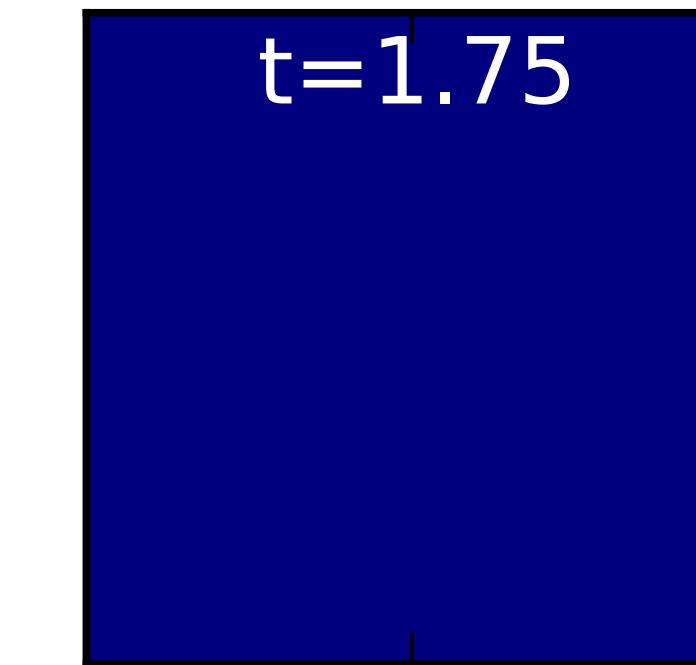
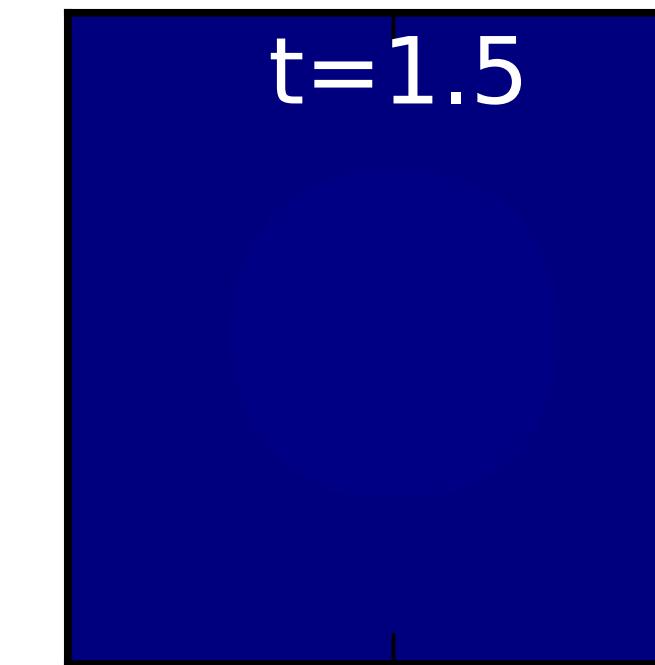
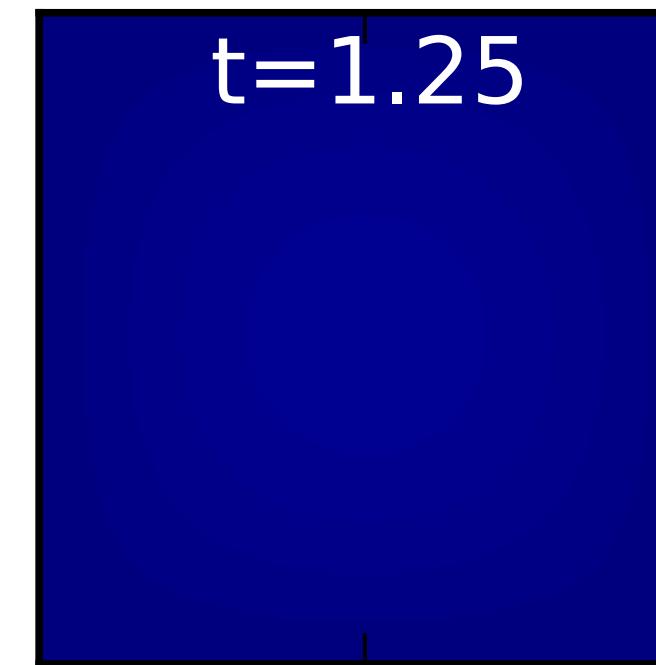
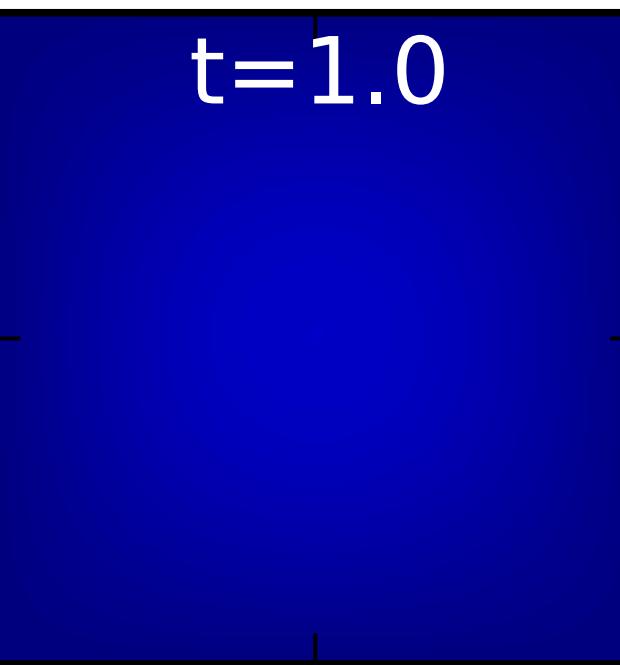
We will consider 2 cases, with 2 Boundary Conditions:

- (i) The concentration at all boundary grid-points is 0, i.e. $C(x_B, y_B, t) = 0$, where x_B, y_B denote the grid-points located on the domain boundary (perimeter).
- (ii) The concentration of a boundary grid-point is equal to the concentration of its closest non-boundary grid-point.

Diffusion with a square as Initial Condition



Using
absorbing
BCs !!



Fully homogeneous
material