

1 Electrostatics

1.1 Electric Field

1.1.1 Introduction

The fundamental problem that electrodynamics seeks to answer is the following: given a collection of electric charges q_1, q_2, q_3, \dots (the *source charges*), what force do they exert on another charge Q (the *test charge*)?

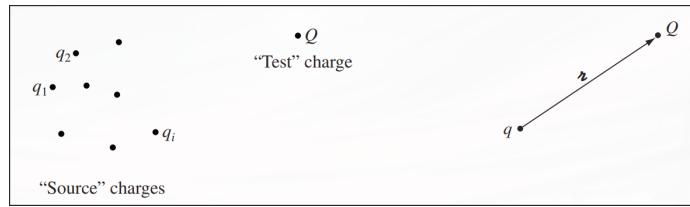


Figure 1: Source charges exerting a force on a test charge.

Electrostatics: Charges are at rest relative to one another.

Electrodynamics: Charges are in motion relative to one another.

These problems are solvable due to the **principle of superposition**, which states that the interaction between any two charges is unaffected by the presence of other charges. As a result, we can calculate the force due to each source charge independently and then add the results:

$$\mathbf{F} = \mathbf{F}_1 + \mathbf{F}_2 + \dots$$

While this principle simplifies the physics conceptually, real problems may still be complex because the positions, velocities, and accelerations of charges (both present and past) can matter. To build intuition, we begin with electrostatics.

1.1.2 Coulomb's Law

The force on a test charge Q due to a single point charge q , at rest and separated by a distance r , is given by **Coulomb's Law**:

$$\mathbf{F} = \frac{1}{4\pi\epsilon_0} \frac{qQ}{r^2} \hat{\mathbf{r}}$$

Here, ϵ_0 is the **permittivity of free space**:

$$\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{Nm}^2$$

It is often convenient to define the constant

$$k = \frac{1}{4\pi\epsilon_0} = 9.0 \times 10^9 \text{ Nm}^2/\text{C}^2$$

so Coulomb's Law can be written more compactly as

$$\boxed{\mathbf{F} = k \frac{qQ}{r^2} \hat{\mathbf{r}}}$$

1.1.3 The Electric Field

The **electric field** \mathbf{E} is a vector quantity defined at every point in space and depends on the configuration of source charges. Conceptually, it represents an entity that fills space around charges: when a test charge is placed at a given point, the electric field at that point determines the force it experiences.

If multiple point charges are present, the total force on a test charge Q is

$$\mathbf{F} = \mathbf{F}_1 + \mathbf{F}_2 + \dots = k \left(\frac{q_1 Q}{r_{1Q}^2} \hat{\mathbf{r}}_{1Q} + \frac{q_2 Q}{r_{2Q}^2} \hat{\mathbf{r}}_{2Q} + \dots \right)$$

Factoring out Q ,

$$\mathbf{F} = Q \mathbf{E}$$

where the electric field is defined as

$$\boxed{\mathbf{E}(\mathbf{r}) = k \sum_{i=1}^n \frac{q_i}{r_{iQ}^2} \hat{\mathbf{r}}_{iQ}}$$

Note that the electric field depends only on the source charges and the observation point, not on the test charge itself.

1.1.4 Charge Distributions

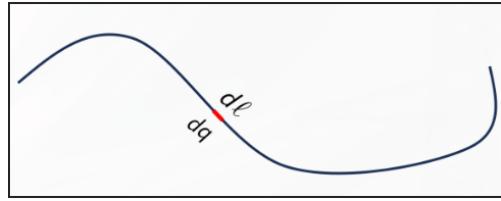
If the charge is distributed continuously over some region, the sum becomes an integral:

$$\boxed{\mathbf{E}(\mathbf{r}) = k \int \frac{1}{r_{qQ}^2} \hat{\mathbf{r}}_{qQ} dq}$$

In many physical situations, charge is distributed continuously rather than concentrated at a point. Such configurations are described using a **charge density**.

1. Linear Charge Distribution

Charge is distributed along a one-dimensional object (e.g., a thin wire) and is described by the **linear charge density**.



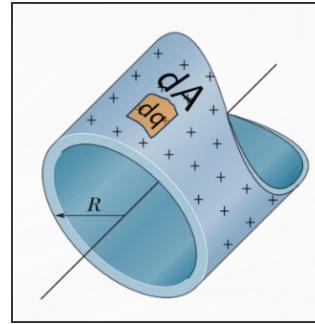
$$\lambda \equiv \frac{dq}{d\ell} \quad [\text{C m}^{-1}]$$

For an infinitesimal segment of length $d\ell$,

$$dq = \lambda d\ell$$

2. Surface Charge Distribution

Charge is distributed over a two-dimensional surface and is described by the **surface charge density**.



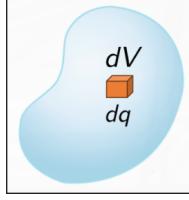
$$\sigma \equiv \frac{dq}{dA} \quad [\text{C m}^{-2}]$$

For an infinitesimal surface element dA ,

$$dq = \sigma dA$$

3. Volume Charge Distribution

Charge is distributed throughout a three-dimensional region and is described by the **volume charge density**.



$$\rho \equiv \frac{dq}{dV} \quad [\text{C m}^{-3}]$$

For an infinitesimal volume element dV ,

$$dq = \rho dV$$

Average Charge Density

In cases where the charge distribution is not uniform or when only the total charge over a finite region is known, it is useful to define an **average charge density**.

$$\bar{\lambda} = \frac{q}{L} \quad \bar{\sigma} = \frac{q}{A} \quad \bar{\rho} = \frac{q}{V}$$

where:

- q is the total charge,
- L is the total length,
- A is the total surface area,
- V is the total volume.

If there is uniform charge distribution, the charge density at any location is equivalent to the average charge density ($\rho = \bar{\rho}$).

1.2 Divergence and Curl of Electrostatic Fields

1.2.1 Field Lines, Flux, and Gauss's Law

The rest of electrostatics is assembling tools and tricks to avoid the involved integrals. The first of these tools is electric field lines. An **electric field line** is an imaginary curve drawn through a region of space so that its tangent at any point is in the direction of the electric field vector at that point.

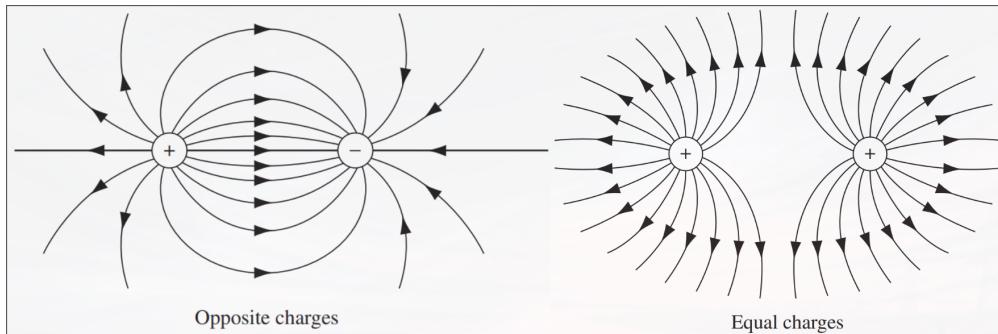


Figure 2: Field line for two opposite charges and two equal charges.

The magnitude of the field is indicated by the density of the field lines: it's stronger near the center point charge where the field lines are closer together. Field lines never intersect, because \mathbf{E} has a unique direction at each point. In terms of direction of the arrows, field lines begin on positive charges and end on negative ones. Field lines also cannot terminate in midair, though they may extend out to infinity.

2 Potentials

2.1 Electric Dipoles

Dipoles are a pair of point charges with:

- **equal** magnitude of charge q
- **opposite** signs
- separated by distance d

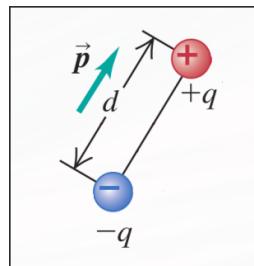


Figure 3: Diagram of a dipole.

Dipoles are defined by the dipole moment vector

$$\mathbf{p} = q\mathbf{r}_{-+} [\text{C m}]$$

where

$$\mathbf{r}_{-+} = \mathbf{r}_- - \mathbf{r}_+$$

$$d = |\mathbf{r}_{-+}|$$

The magnitude of \mathbf{p} can be found using the charge and distance

$$p = qd$$

Dipole moments are vectors, and they add accordingly: if you have two dipoles, \mathbf{p}_1 and \mathbf{p}_2 , the total dipole moment is $\mathbf{p}_1 + \mathbf{p}_2$. **Hint:** We can use this in the future to combine close charges and treat them as dipoles, simplifying our math.

Even though it is overall neutral, a water molecule's chemical bonds cause a displacement of charge. This forms a negative charge at the oxygen end, and a positive charge on the hydrogen, forming an electric dipole!

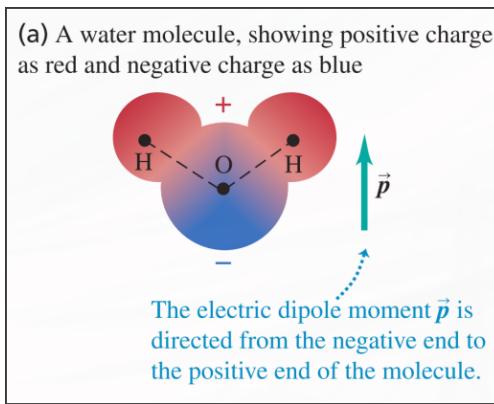


Figure 4: O-H bonds in water form dipoles.

2.1.1 Forces and Torque on Electric Dipoles

Force \mathbf{F}_+ and \mathbf{F}_- are not in the same direction which creates **torque**.

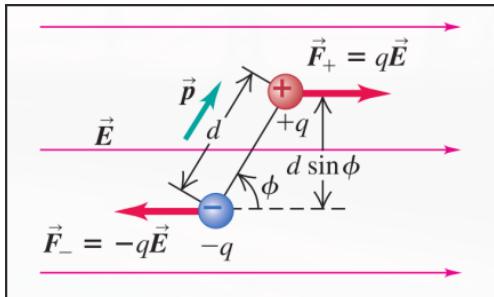


Figure 5: The net force on the electric dipole is zero, but there is a torque directed into the page.

According to the torque formula:

$$\begin{aligned}
 \tau &= \tau_+ + \tau_- \\
 &= \mathbf{r}_+ \times \mathbf{F}_+ + \mathbf{r}_- \times \mathbf{F}_- \\
 &= \mathbf{r}_+ \times q\mathbf{E} + \mathbf{r}_- \times (-q)\mathbf{E} \\
 &= q\mathbf{r}_{-+} \times \mathbf{E}
 \end{aligned}$$

$$\boxed{\tau = \mathbf{p} \times \mathbf{E} [\text{N m}]}$$

The magnitude of the torque is

$$\begin{aligned}
 |\tau| &= |\mathbf{p} \times \mathbf{E}| \\
 &= pE \sin(\phi_{p,E}) \\
 &= qd \sin(\phi_{p,E})
 \end{aligned}$$

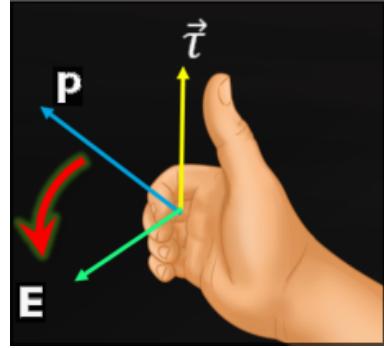


Figure 6: The direction of τ is found using the right-hand rule.

2.1.2 Potential Energy of Electric Dipoles

The potential energy of a dipole with moment \mathbf{p} placed in an external field \mathbf{E} , is the negative of the work done by \mathbf{E} on the dipole ($dU = -dW$). Therefore to find the potential energy of the dipole we need to find the work done by \mathbf{E} .

$$\begin{aligned}
 d\mathbf{W} &= d\mathbf{W}_+ + d\mathbf{W}_- \\
 &= \mathbf{F}_+ \cdot d\mathbf{r}_+ + \mathbf{F}_- \cdot d\mathbf{r}_- \\
 &= q\mathbf{E} \cdot d\mathbf{r}_+ + (-q)\mathbf{E} \cdot d\mathbf{r}_- \\
 &= q\mathbf{E} \cdot (d\mathbf{r}_+ - d\mathbf{r}_-)
 \end{aligned}$$

For an infinitesimal $d\mathbf{r}$, we can assume \mathbf{E} is constant, hence

$$\begin{aligned} W &= \int d\mathbf{W} = q\mathbf{E} \cdot \int (d\mathbf{r}_+ - d\mathbf{r}_-) \\ &= q\mathbf{E} \cdot (\mathbf{r}_+ - \mathbf{r}_-) \\ &= q\mathbf{E} \cdot \mathbf{r}_{-+} \\ &= \mathbf{p} \cdot \mathbf{E} \end{aligned}$$

Therefore

$$U = -\mathbf{p} \cdot \mathbf{E}$$