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Development of (*E*)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-5-nitro-*N*-phenylbenzamide, ML336: Novel 2-Amidinophenylbenzamides as Potent Inhibitors of Venezuelan Equine Encephalitis Virus

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Supporting Information

ABSTRACT: Venezuelan equine encephalitis virus (VEEV) is an emerging pathogenic alphavirus that can cause significant disease in humans. Given the absence of therapeutic options available and the significance of VEEV as a weaponized agent, an optimization effort was initiated around a quinazolinone screening hit 1 with promising cellular antiviral activity (EC₅₀ = 0.8 μ M), limited cytotoxic liability (CC₅₀ > 50 μ M), and modest in vitro efficacy in reducing viral progeny (63-fold at 5 μ M). Scaffold optimization revealed a novel rearrangement

affording amidines, specifically compound **45**, which was found to potently inhibit several VEEV strains in the low nanomolar range without cytotoxicity (EC₅₀ = 0.02–0.04 μ M, CC₅₀ > 50 μ M) while limiting in vitro viral replication (EC₉₀ = 0.17 μ M). Brain exposure was observed in mice with **45**. Significant protection was observed in VEEV-infected mice at 5 mg kg⁻¹ day⁻¹ and viral replication appeared to be inhibited through interference of viral nonstructural proteins.

INTRODUCTION

Alphaviruses are a group of infectious pathogens found worldwide for which no therapeutic agents have been successfully developed. Approximately 30 alphaviruses of the Togaviridae family have been identified, and almost a third contribute significantly to human disease. 1 Most members of the alphavirus genus are transmitted by infected mosquitoes. Depending on the virus, infection generally manifests in one of two forms.² Arthritogenic alphaviruses such as Chikungunya, Ross River, Mayaro, and Sindbis viruses cause fever, rash, and persistent arthralgia. Eastern, Western, and Venezuelan equine encephalitis viruses (EEEV, WEEV, and VEEV, respectively) are endemic to the American continent and produce a flulike illness in human populations that is characterized by fever, headache, myalgia, sore throat, vomiting, and virus-dependent propensity to progress to fatal encephalitis. 3,4 Encephalitic alphaviruses have garnered increased attention in recent years because of the ease

with which they can be weaponized as biothreats.⁵ For instance, VEEV was previously investigated as an incapacitating bioterror agent given that low levels of viral exposure caused illness and that the virus could be generated in high titer and stored indefinitely.^{6,7} These features, combined with the discovery that the virus could be aerosolized, are responsible for VEEV, along with EEEV and WEEV, being classified as Centers for Disease Control (CDC) and National Institute for Allergy and Infectious Diseases (NIAID) category B bioterrorism agents.^{8,9}

VEEV provides an attractive platform to investigate intervention of alphavirus infection. Virus is difficult to detect in blood and cerebral-spinal fluid (CSF) samples from patients infected with EEEV or WEEV; however, VEEV can be easily found in these and other biospecimens, thus facilitating the in

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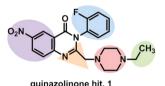
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vivo efficacy assessment of new antiviral agents. ¹⁰ Additionally, VEEV has a better safety margin than that of EEEV. While some strains of EEEV have a human mortality rate averaging 50%, fatality due to VEEV infection is rare (<1% overall). ^{4,11} Nonetheless, VEEV is a serious pathogen that warrants specific attention in its own right. Fatality rates in equines can reach up to 85%, and in human populations, up to 14% of infected patients experience chronic neurological complications that manifest as depression, behavioral disorders, convulsions, and impairment of coordinated muscular movement. ¹² On the basis of epidemiologic data, the neurological disease is more prevalent and severe in children. ¹³

While there are no FDA-approved human vaccines or therapeutics available for VEEV (or any alphavirus), advances in the development of agents targeting encephalitic alphaviruses are investigated and have recently been reviewed. To date, human VEEV vaccines have seen limited success due to insufficient protection, transient immunity, or the incidence of adverse reactions. ^{14–19} Several small molecule-derived anti-VEEV compounds have been reported, though most have uncharacterized mechanisms of action, demonstrate only weak in vitro inhibition, and do not appear to have been pursued further. 7,20-22 An exception includes the diazodiimine compound, BIOder, an inhibitor of the GSK-3 β host protein that governs proinflammatory responses.²³ While this proof-ofconcept compound showed only partial efficacy in preventing mice from succumbing to VEEV infection, it demonstrated the potential of using a small molecule inhibitor in this capacity. Given the health threat posed by alphaviruses and the absence of available countermeasures, a program aimed at identifying small molecule-derived VEEV inhibitors was undertaken by our team.

■ RESULTS AND DISCUSSION

Screening and Hit Identification. As part of the Molecular Libraries Probe Production Centers Network (MLPCN),²⁴ this team launched a screening campaign seeking new structural hits for development as potent, small molecule inhibitors of VEEV. The high throughput screen (HTS), a cell-based assay that measured the ability of a compound to inhibit a VEEV-induced cytopathic effect (CPE), was developed and validated employing TC-83 strain, an attenuated vaccine strain of wild type Trinidad donkey VEEV. 25-27 The use of the TC-83 strain was advantageous because of its high congruency to wild type virus and fewer regulatory restrictions in handling (BL-2 and nonselect agent); however, it retained high similarity to wild type with respect to genomic sequence, replication mechanism, and ability to obtain high viral titer (> 10×10^9 pfu/mL) compared to that observed with wild type virus.²⁶ A total of 348, 140 compounds from the Molecular Libraries Small Molecule Repository (MLSMR) were evaluated at a single concentration of 20 μ M, resulting in 3608 hits that inhibited a VEEV-induced CPE by >14%. Of these, 90 compounds were identified through subsequent antiviral dose-response experiments and cellular cytotoxicity (Vero 76 cells) evaluation that resulted in ≥30% inhibition, CPE EC₅₀ \leq 12.5 μ M and CC₅₀ \geq 25 μ M, respectively. The hit list was further refined to five compounds based on lack of reactive functionality, synthetic feasibility, PubChem promiscuity²⁸ analysis, acceptable aqueous solubility in PBS buffer for the most promising scaffolds, and validation of activity from resynthesized and purified powder samples. Validated hits were further assessed in an in vitro viral titer reduction assay to quantitatively determine their effect on the production of infectious progeny virus. The team drafted a set of project criteria that, if reached, would afford a soluble, cellular-acting, small molecule probe that was efficacious in potently attenuating a VEEV-induced CPE and reduced viral plaques by at least 1000-fold at a concentration of 5 μ M. These properties, coupled with a sufficient window for cytotoxicity and suitable physiochemical parameters, were expected to deliver an impactful probe compound worthy of in vivo efficacy studies. The quinazolinone chemotype, represented by hit compound 1, was an attractive starting point due to its preliminary activity and physiochemical profile. Improvements were explored by tuning various structural components (shaded regions, Figure 1).



CPE EC $_{50}$ = 0.8 μ M CC $_{50}$ > 50 μ M titer reduction (5 μ M) = 63-fold molecular weight = 411.43 g/mol cLogP = 3.4 tPSA = 90.96 PBS solubility = 122 μ g/mL

Figure 1. Shaded regions of hit compound 1 optimized by structure—activity relationships.

Quinazolinone Chemistry and SAR. The optimization process was driven by the TC-83 CPE-based dose response assay, cytotoxicity assessment, and titer reduction experiments. The possibility of performing any future in vivo studies was expected to pivot on the ability of an anti-VEEV compound to reach encephalitic brain tissue.²⁹ As such, careful attention was paid to balancing potency in both CPE-based and titer reduction assays with suitable physiochemical properties such as cLogP, polar surface area (tPSA), molecular weight, and solubility such that analogs were more likely to be CNS-permeable. 30,31 Generally, analogs of 1 were prepared in a three-step procedure that involved treating substituted anthranilic acids 2 with chloroacetyl chloride in the presence of triethylamine to afford 2-(2-chloroacetamido)-5-nitrobenzoic acid 3 (Scheme 1).32 Dehydrative amidation was carried out by treating intermediates 3 with a selected aniline and POCl₃. 33 The resulting chloromethylquinazolinones 4 were aminated with alkylpiperazines to give quinazolinones 5.34

Scheme 1. Synthetic Route for Assembly of Quinazolinone Analogs a

$$O_2N$$
 O_1
 O_2N
 O

"Reagents: (a) ClCOCH₂Cl, NEt₃, CH₂Cl₂, 0 °C to rt, 95%; (b) substituted aniline, POCl₃, acetonitrile, MWI, 150 °C, 56–70%; (c) KI, K₂CO₃, alkylpiperazine, acetonitrile, MWI, 80 °C, 25–80%.

Table 1. Structure-Activity Relationships Involving the Quinazolinone C6-Nitro Group

entry	compound	R ¹ N NEt		VEEV (TC-83) CPE (EC ₅₀ , μM) ^a	cytotoxicity (CC ₅₀ , μM) ^b	selectivity index (CC ₅₀ /EC ₅₀)	VEEV titer, c,d log reduction	cLogPe
1	1	NO ₂	2-F-phenyl	0.8	> 50.0	> 62.5	1.8	3.4
2	6	Н	2-F-phenyl	> 25.0	> 50.0	NA	NT	3.6
3	7	CF ₃	2-F-phenyl	> 25.0	> 50.0	NA	NT	4.5
4	8	NO_2	phenyl	0.3	> 50.0	> 166.7	> 7.8	3.3
5	9	I	phenyl	17.1	> 50.0	> 2.9	0.3	4.6
6	10	CN	phenyl	1.1	> 50.0	> 45.5	1.7	2.9
7	11	SO_3H	phenyl	> 25.0	> 50.0	NA	NT	-1.6
8	12	CO ₂ H	phenyl	> 25.0	> 50.0	NA	NT	1.2
9	13	NH-tetrazole	phenyl	> 25.0	> 50.0	NA	NT	3.0
10	14	CONH ₂	phenyl	> 25.0	> 50.0	NA	NT	2.4
11	15	CONMe ₂	phenyl	> 25.0	> 50.0	NA	NT	2.2
12	16	2-pyridyl	phenyl	> 25.0	> 50.0	NA	NT	4.1
13	17	3-pyridyl	phenyl	> 25.0	> 50.0	NA	NT	3.9

^aData were an average of >3 experiments. ^bData were an average of >2 experiments. ^cNT = not tested. ^dData were collected using 5 μM compound, and data were analyzed using Microsoft Excel 2010. ^eData were calculated using SYBYL 8.0, Tripos Associates, St. Louis, MO, 2010.

Table 2. Quinazolinone Structure-Activity Relationships Involving the N-Aryl and Piperazine Groups

entry	compound	$ \begin{array}{c cccc} O_2N_0 & & & & \\ & & & & & \\ \hline & & & & \\ \hline & & & & \\ \hline & & & & \\ \hline & & & & \\ \hline & & & & \\ \hline & & & & \\ \hline & & & & & $		VEEV (TC-83) CPE (EC ₅₀ , μM) ^a	cytotoxicity (CC ₅₀ , μM) ^b	selectivity index (CC ₅₀ / EC ₅₀) ^c	VEEV titer, d,e log reduction	cLogP ^f	
1	18	2-F-phenyl	CH ₂	N-methyl	0.8	> 50.0	> 62.5	7.1	2.9
2	19	2-F-phenyl	CH ₂	N-i- propyl	4.4	> 50.0	> 11.4	NT	3.7
3	20	2-F-phenyl	CH_2	N-phenyl	> 25.0	> 50.0	NA	NT	4.3
4	21	2-F-phenyl	CH_2	N-H	0.2	> 50.0	> 250.0	> 7.8	2.3
5	22	2-F-phenyl	CH_2	0	8.7	> 50.0	> 5.7	0.3	2.3
6	23	2-F-phenyl	CH_2	CH ₂	4.7	> 50.0	> 10.6	0.4	3.6
7	24	2-F-phenyl	CO	N-H	> 25.0	> 50.0	NA	NT	2.0
8	25	3-F-phenyl	CH ₂	N-ethyl	0.4	> 50.0	> 125.0	7.1	3.4
9	26	4-F-phenyl	CH_2	N-ethyl	0.8	> 50.0	> 62.5	5.9	3.4
10	8	phenyl	CH ₂	N-ethyl	0.3	> 50.0	> 166.7	> 7.8	3.3
11	27	methyl	CH ₂	N-ethyl	> 25.0	> 50.0	NA	NT	1.3
12	28	<i>i</i> -propyl	CH_2	N-ethyl	> 25.0	> 50.0	NA	NT	2.1
13	29	benzyl	CH_2	N-ethyl	> 25.0	> 50.0	NA	NT	3.0
14	30	Н	CH ₂	N-ethyl	> 25.0	> 50.0	NA	NT	1.5
15	31	phenyl	CH_2	N-H	0.2	> 50.0	> 250.0	> 7.8	2.1
16	32	2-MeO-phenyl	CH ₂	N-H	> 25.0	> 50.0	NA	NT	2.2
17	33	3-MeO-phenyl	CH_2	N-H	5.1	> 50.0	9.8	NT	2.2
18	34	4-MeO-phenyl	CH ₂	N-H	0.2	> 50.0	> 250	> 7.4	2.2
19	35	3-NMe ₂ -phenyl	CH_2	N-H	> 25.0	> 50.0	NA	NT	2.6
20	36	4-NMe ₂₋ phenyl	CH_2	N-H	> 25.0	> 50.0	NA	NT	2.6
21	37	3-thiophene	CH ₂	N-H	1.4	> 50.0	> 35.7	NT	1.8

^aData were an average of >3 experiments. ^bData were an average of >2 experiments. ^cNA = not applicable. ^dNT = not tested. ^eData were collected using 5 μM compound, and data were analyzed using Microsoft Excel 2010. ^fData were calculated using SYBYL 8.0, Tripos Associates, St. Louis, MO. 2010.

Survey of the scaffold began with determining the necessity of the C6-nitro functionality. Migration of the C6-nitro group to any other of the available C5-, C7-, or C8-positions afforded analogs without significant CPE assay potency (>25 μ M). Replacement of the nitro group was initially investigated on the parent scaffold bearing a 2-fluorophenyl moiety; however, once it was determined that removal of the fluorine substituent from the phenyl group improved potency 3-fold and substantially enhanced reduction in viral titer, subsequent generations of analogs were made with this alteration (Table 1, entries 1 vs 4). Substitution of the nitro group with a hydrogen atom or a

trifluoromethyl group resulted in loss of CPE assay activity (entries 2, 3). Several replacements were studied, including a carboxylic acid, tetrazole, sulfonic acid, and amide isosteres; however, these changes were not tolerated with the exception of nitrile analog 10 (entry 6), which lost some potency and was inferior in reducing viral titer compared to hit compound 1.

Given the potency dependency on the presence of the C6 nitro functionality, a majority of the SAR was investigated with this moiety preserved, and attention shifted to modifications of the *N*-ethyl portion of the piperazine appendage (X and Y, Table 2). Replacement of the *N*-ethyl group for *N*-methyl did not impact

Journal of Medicinal Chemistry

VEEV selectivity VEEV (TC-83) cytotoxicity $titer,^{c,d}$ cLogP^e entry compound index CPE (CC₅₀, (CC_{50}/EC_{50}) log $(EC_{50},$ $\mu M)^b$ reduction $\mu M)^a$ > 25.0 3.5 1 38 > 50.0 NA 2 39 3.6 > 50.0 > 13.9 NT 3.6 3 40 > 50.0 > 45.5 NT 2.2 1.1

Table 3. Quinazolinone Structure-Activity Relationships Involving the Alkylpiperazine Appendage

^aData were an average of >3 experiments. ^bData are an average of >2 experiments. ^cNT = not tested. ^dData were collected using 5 μM compound, and data were analyzed using Microsoft Excel 2010. ^eData were calculated using SYBYL 8.0, Tripos Associates, St. Louis, MO, 2010.

0.9

0.6

NHCH₂

'n

> 50.0

> 50.0

> 55.6

> 83.3

NT

NT

2.1

2.4

Scheme 2. Formation of Rearranged Amidines^a

4

41

42

$$\begin{array}{c} O_2N \\ O_$$

^aReagents: (a) CH₃NHCH₂CH₂NCH₃BOC, KI, K₂CO₃, CH₃CN, MWI, 80 °C, 10 min, 35%; (b) TFA, CH₂Cl₂, rt, 45 min, then aq NaHCO₃, 50%; (c) CH₃NHCH₂CH₂NHCH₃, K₂CO₃, CH₃CN, 50 °C, 2 h, 69%.

the CPE assay but profoundly improved the reduction of viral plaques (entry 1). The introduction of a larger N-isopropyl substituent adversely affected potency, suggesting a spacial constraint in the binding site (entry 2). The loss of potency associated with the N-phenyl analog 20 also supported a steric argument; however, the observed effect may have been due, at least in part, to the preference of a more basic amine capable of having beneficial interactions with proximal binding residues. To address this ambiguity, N-H analog 21 was prepared and found to be superior to any of its predecessors in the study with respect to CPE potency (EC₅₀ = 0.2 μ M) and titer reduction efficacy (>7.8 log). Furthermore, the analogs bearing nonbasic, N-ethyl replacements (e.g., morpholine 22, piperidine 23, and amide 24) were determined to possess significantly inferior activity profiles compared to N-H analog 21. These combined results suggested that there is a strong preference for a basic *N*-H moiety in this region of the scaffold.

In a parallel effort the effect of changing the 2-fluoropheny-lamide substituent was also surveyed (Table 2, R²). Compared to the hit 1, a dramatic increase in titer reduction was observed by incorporating a 3-fluorophenyl group (25, entry 8); however, simplifying to a nonsubstituted phenyl ring advantageously produced analog 8 with similar overall activity to analog 21. Exchange of the phenyl group for alkyl groups of various sizes or a hydrogen atom was not tolerated (27–30, entries 11–14). Other significant findings included the hybrid R²-phenyl-

containing *N*-H piperazine **31** and 4-methoxyphenyl derivative **34**, which both produced a favorable reduction in CPE and viral plaques that was comparable to analog **21**. Gratifyingly, this endeavor revealed that several analogs could be generated with improved activity as compared to the hit compound by modulating two distinct regions of the scaffold while advantageously adjusting molecular weight and cLogP.

More extensive study of the piperazine appendage included excision of the methylene linkage between the quinazolinone core to provide piperazine 38 and NH-piperidine analog 39 (Table 3), neither of which led to impressive CPE inhibition. Trading the piperazine group for a hexahydropyrimidine attachment or ring expanded 1,4-diazepane moiety only marginally affected CPE potency as compared to hit 1 (analogs 40 and 41, respectively). Nonetheless, the team pursued a series of open-chain, diamino analogs that are represented by n-propylamine 42. Interestingly, 42 showed respectable CPE inhibition (EC₅₀ = 0.6 μ M), but it was the only one of its kind that was isolated.

2-Amidinephenylbenzamide Chemistry. Examination of reactions employing a two-carbon-tethered, acyclic diamine revealed the formation of a rearranged (*E*)-amidine **45** that was structurally confirmed by ¹H and ¹³C NMR, NOESY, and 2D NMR spectroscopy (Scheme 2).³⁵ The reaction was initially carried out with mono-BOC protected diamines, thus providing a snapshot of the reaction pathway leading to the amidine

Table 4. Amidine Structure-Activity Relationships

entry	compound	R ⁵ 5 N N R ⁶ H 3C N N R ⁷			VEEV (TC-83) CPE potency ^a	cyto- toxicity ^b	selectivity index ^c	VEEV titer, d,e	$cLogP^f$	
		R ⁴	R ⁵	\mathbb{R}^6	\mathbb{R}^7	EC ₅₀ , μΜ	CC ₅₀ , μΜ	CC ₅₀ /EC ₅₀	log reduction	
1	45	Н	NO ₂	phenyl	CH ₃	0.03	> 50.0	> 1666.7	> 7.2	3.4
2	47	Н	NO ₂	phenyl	CH ₂ CH ₃	0.05	> 50.0	> 1000.0	> 7.2	4.0
3	48	Н	NO ₂	3-thiophene	CH ₃	0.5	> 50.0	100.0	NT	3.1
4	49	Н	NO ₂	2-CH₃O- phenyl	CH ₃	33.7	> 50.0	> 1.5	NT	3.4
5	50	Н	NO ₂	3-CH ₃ O- phenyl	CH ₃	8.1	> 50.0	> 6.2	NT	3.4
6	51	Н	NO ₂	4-CH ₃ O- phenyl	CH ₃	0.04	> 50.0	> 1250.0	> 7.2	3.4
7	52	Н	NO ₂	2-F-phenyl	CH ₃	0.2	> 50.0	> 250.0	> 3.8	3.6
8	53	Н	NO ₂	3-F-phenyl	CH ₃	0.1	> 50.0	> 500.0	NT	3.6
9	54	Н	NO ₂	4-F-phenyl	CH ₃	0.09	> 50.0	> 555.6	NT	3.6
10	55	Н	NO ₂	i-propyl	CH ₃	> 25.0	> 50.0	NA	NT	2.8
11	56	Н	NO ₂	benzyl	CH ₃	3.7	> 50.0	> 13.5	NT	3.8
12	57	Н	Н	phenyl	CH ₃	> 25.0	> 50.0	NA	NT	3.3
13	58	NO ₂	Н	phenyl	CH ₃	8.4	> 50.0	> 6.0	NT	3.4
14	59	F	Н	phenyl	CH ₃	> 25.0	> 50.0	NA	NT	3.6
15	60	Н	F	phenyl	CH ₃	6.4	> 50.0	> 7.8	NT	3.6
16	61	F	F	phenyl	CH ₃	4.4	> 50.0	> 11.4	NT	3.8
17	62	Н	CF ₃	phenyl	CH ₃	22.3	> 50.0	> 2.2	NT	4.5
18	63	Н	CN	phenyl	CH ₃	0.4	> 50.0	> 125.0	> 5.8	3.2
19	64	Cl	CN	phenyl	CH ₃	0.1	> 50.0	> 500.0	NT	3.8
20	65	F	CN	phenyl	CH ₃	0.4	> 50.0	> 125.0	NT	3.4
21	66	Н	CN	2-F-phenyl	CH ₃	3.7	> 50.0	> 13.5	NT	3.3
22	67	Н	CN	3-F-phenyl	CH ₃	0.8	> 50.0	> 62.5	> 1.4	3.3
23	68	Н	CO ₂ CH ₃	phenyl	CH ₃	2.0	> 50.0	> 25.0	NT	3.6

"Data were an average of >3 experiments. Data were an average of >2 experiments. NA = not applicable. NT = not tested. Data were collected using 5 μ M compound, and data were analyzed using Microsoft Excel 2010. Data were calculated using SYBYL 8.0, Tripos Associates, St. Louis, MO, 2010.

product by proceeding through intermediate 44; however, a onepot conversion was achieved with symmetrically substituted diamines to afford 45 in a reasonable 69% yield. Amidines were the only isolated products from the reaction, and the exclusive isolation of the (E)-configured double bond amidine was presumed to result from electronic and steric contributions in the transition state. Interestingly, amidines were generated when chloromethylquinolinone starting materials were treated with diamines separated by a two-carbon linkage, but a three-carbon tethered diamine produced the expected quinazolinone 42. It is reasoned that amidine formation is favored when the reaction proceeds through a six-membered, spirocyclic intermediate such as 46, resulting from an intramolecular attack of the terminal amine, followed by carbon-nitrogen bond cleavage. To our knowledge, this rearrangement has not been previously reported, and as such, a more comprehensive examination of the scope of this transformation is underway and will be disclosed in due course. $^{\rm 35}$

2-Amidinephenylbenzamide SAR. The conversion of quinazolinone intermediate **43** to amidine **45** represented a fortuitous scaffold modification, resulting in a nearly a 7-fold improvement in antiviral potency over the best quinazoline-based analogs while preserving the viral titer reduction capability and noncytotoxic attributes (**45**, Table **4**). Functional groups that were optimized for the quinazolinone scaffold showed a correlative enhancement of antiviral potency for the amidine series, though at an improved, low-nanomolar threshold. While the importance of the quinazolinone C6-nitro group to retaining potency translated to the corresponding C5-position of the amidine chemotype, it was notable that the nitro group could be migrated to the amidine C4 position (compound **58**) or replaced with a fluorine atom or nitrile group (**60** or **63**, respectively) without obliterating activity.

Table 5. Comparison of Cellular Data and Physiochemical Properties of Key Compounds

		comparative data ^a					
entry	parameter	compd 1	compd 45	compd 51			
1	CPE EC ₅₀ , TC-83 strain	$0.8~\mu\mathrm{M}$	0.03 μM	$0.04~\mu\mathrm{M}$			
2	CPE EC ₅₀ , V3526 strain	$0.3 \mu M$	$0.02~\mu\mathrm{M}$	$0.02~\mu\mathrm{M}$			
3	CPE EC ₅₀ , TrD strain	$0.6~\mu\mathrm{M}$	$0.04~\mu\mathrm{M}$	$0.05~\mu\mathrm{M}$			
	titer reduction, TC-83 strain						
4	5 μ M compd concentration	1.8 log (69-fold)	7.2 log (16 million fold)	7.2 log (16 million fold)			
5	$1 \mu \text{M}$ compd concentration	0.5 log (3-fold)	7.2 log (16 million fold)	7.2 log (16 million fold)			
6	0.5 μ M compd concentration	NT	5.8 log (630 000 fold)	NT			
7	dose response, EC ₉₀	NT	$0.17~\mu\mathrm{M}$	$0.17~\mu\mathrm{M}$			
	titer reduction, TrD strain						
8	5 μ M compd concentration	no measurable viral replication	no measurable viral replication	no measurable viral replication			
9	1 μ M compd concentration	2.4 log (251-fold)	no measurable viral replication	no measurable viral replication			
10	$0.5 \mu M$ compd concentration	NT	no measurable viral replication	no measurable viral replication			
11	Vero 76 cellular toxicity, CC ₅₀	>50.0 µM	>50.0 µM	>50.0 µM			
12	aqueous solubility, PBS buffer	$>122.0 \ \mu g/mL$	$40.4 \mu \text{g/mL}$	$105.6 \mu\mathrm{g/mL}$			
13	molecular weight ^b	411.4 g/mol	367.4 g/mol	397.4 g/mol			
14	topological polar surface area, tPSA ^b	91.0 $Å^2$	99.8 $Å^2$	109.0 Å ²			
15	$cLogP^{c}$	3.4	3.4	3.4			
16	hydrogen bond donors ^d	0	1	1			
17	hydrogen bond acceptors ^d	7	5	6			
18	heavy atoms ^d	30	27	29			
19	ligand efficiency, LE	0.28	0.39	0.36			
20	lipophilic ligand efficiency, LLE	2.7	4.1	4.1			

^aNT = not tested. ^bData were generated using CambridgeSoft ChemBioDraw, version 12. ^cData were calculated using SYBYL 8.0, Tripos Associates, St. Louis, MO, 2010. ^dPubChem calculated chemical and physical parameters.

Table 6. Summary of in Vitro Pharmacology Data for Compound 45 (ML336)

entry	assessment	result
1	aqueous solubility, $1 \times$ PBS, pH 7.4^a	40.4 μ g/mL, 110.0 μ M
2	aqueous solubility, VEEV CPE assay medium ^b	$13.1 \mu\text{g/mL}$ or $35.7 \mu\text{M}$
3	chemical stability with $5 \times DTT$, $8 h^c$	99.0% parent remaining
	aqueous stability, 48 h ^c	
4	PBS, pH 7.4	82.9% parent remaining
5	1:1 PBS/acetonitrile	95.9% parent remaining
6	plasma stability, % remaining after 3 h (mouse) ^a	65.4% parent remaining
	plasma protein binding, $\%$ bound ^a	
7	$1 \mu\mathrm{M} \; (\mathrm{mouse})$	85.0%
8	$10 \mu\mathrm{M} \; (\mathrm{mouse})$	77.0%
9	BBB PAMPA, pH 7.4 ^{a,d}	$13 \times 10^{-6} \text{ cm/s}$
10	hepatic microsomal stability, after 1 h (mouse) ^a	42.9% parent remaining
11	hepatocyte toxicity (Fa2N-4), LD_{50} (human) ^{a}	>25.0 µM
	CYP450 inhibition profile, percent inhibition at $10 \mu \text{M}^{a,e}$	
12	1A2 (human)	1.5%
13	2C9 (human)	7%
14	2C19 (human)	17%
15	2D6 (human)	28%
16	3A4 (human)	18%
17	PanLabs profiling panel: assessment of percent inhibition of 67 host derived targets at a 10 μ M compound concentration f	norepinephrine transporter, 91%

^aData collected by Arianna Mangravita-Novo at the Conrad Prebys Sanford Burnham Medical Research Institute. ^bCPE assay medium = high glucose DMEM with 10% FBS and 1× Pen/Strep. ^cData collected by Patrick Porubsky at the University of Kansas Analytical Chemistry Core,

Specialized Chemistry Center. ^dDonor and acceptor pH: 7.4. Controls: verapamil–HCl (highly permeable), 138; corticosterone (moderately permeable), 14; theophylline (poorly permeable), 0.32. ^eControl inhibitors: for CYP3A4, ketoconazole, 99% inhibition; for CYP2C9, sulfaphenazole, 96% inhibition; for CYP1A2, α-naphtoflavone, 84% inhibition; for CYP2C19, lansoprazole, 86% inhibition; for CYP2D6, quinidine, 88–98% inhibition. ^fEurofins PanLabs data; all targets were assessed in duplicate and complete profile is included in the Supporting Information.

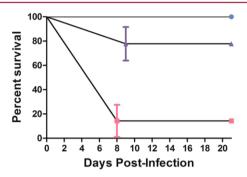
The result of this effort revealed a set of amidine-derived VEEV inhibitors with exceptional, low nanomolar antiviral potency, a robust selectivity window, and significant titer reduction efficacy. A subset of the most active representatives

from the amidine series and the quinazolinone series (generally, those with CPE EC₅₀ values of $\leq 1~\mu\text{M}$) were also evaluated in a CPE assay using wild type VEEV, Trinidad donkey (TrD) strain, to confirm activity against a clinically relevant virus. Gratifyingly,

Table 7. In Vivo Exposure Data for Compound 45 in Mice

			plasma data $(mouse)^e$		brain data (mouse) ^e	
compd 45 dose ^a (mg/kg)	$_{(\min)^b}^{\text{time}}$	number of mice per group ^c	average d concentration \pm standard deviation (μ M)	CV (%)	average d concentration \pm standard deviation (μM)	CV (%)
1	20	3	0.20 ± 0.03	15.1	ND	ND
	120	3	0.023 ± 0.007	30.8	0.19 ± 0.22	115.8
5	20	3	1.45 ± 0.26	17.6	ND	ND
	120	3	0.29 ± 0.01	4.7	0.21 ± 0.02	10.7
10	20	3	2.68 ± 0.36	13.5	ND	ND
	120	3	0.49 ± 0.25	49.9	0.35 ± 0.06	17.0

^aIntraperitoneal administration. ^bEach time point is an average of two LC/MC injections (<20% difference). ^cC57BL/6 mice were used. ^dData were an average of three independent experiments for each dose and time point (i.e., three mice in each cohort, including vehicle control, not shown). ^cCV = coefficient of variation; ND = not determined.



key	group description	#mice per group	#live mice at day 21	percent survival
•	∨ehicle (no ∨irus or compound treatment)	7	7	100%
	untreated VEEV infected mice	7	1	14%
	VEEV infected mice treated with 45 (5 mg/kg/day for 4 days)	7	5	71%

Figure 2. Survival data for VEEV infected C3H/HeN mice treated with compound **45** over a 21-day study compared to controls. Survival curves were generated using the Kaplan—Meier method⁴² and were compared using the log rank (Mantel—Cox) test; $X^2 = 11.70$, p = 0.0029 (2 degrees of freedom). The error bars show the standard error of the mean.

strong agreement was observed across TC-83 and TrD strains for both chemotypes, especially for the most promising amidines of interest (entries 1–3, Table 5).

Pharmacology of Compound 45. Compared to the hit quinazolinone 1, analogs 45 and 51 showed the best enhancement of CPE potency while also benefiting from decreased molecular weight and maintained cLogP. These compounds were also assessed in plaque reduction assays employing TC-83 or TrD strains at lower compound concentrations than the 5 μ M concentration used during the SAR evaluation phase (entries 4-10, Table 5). While all three compounds blocked detectable levels of viral replication of wild type TrD strain at 5 μ M, only the amidines were effective at 0.5 μ M, the lowest concentration tested initially. Assessment of compounds 45 and 51 in a dose response titer reduction assay revealed EC₉₀ = 0.17 μ M for both compounds, demonstrating that in the 150 nM range, only $\frac{1}{10}$ of virus was produced compared to the control, indicating efficient inhibition of virus replication. While rearrangement of the structural architecture from quinazolinones to amidines altered the number of hydrogen bond donors and acceptors, the cLogP values remained constant and improvements in ligand efficiency³⁶ and lipophilic ligand efficiency³⁷ were due to enhancements in potency alone. Analogs 45 and 51 were relatively evenly matched with respect to in vitro potency and efficacy; however, amidine 45 was slightly favored in terms of reduced molecular weight and topological polar surface area. Consequently, amidine 45 was selected as the lead probe compound, ML336, and the compound was characterized further in terms of its in vitro pharmacology to establish a baseline profile against which future analogs could be optimized (Table 6). Diminished solubility was observed in CPE assay medium compared to an assessment in PBS; however, the solubility concentrations were still well above (~1200- to 3700fold) the observed EC_{50} of the compound in the CPE assay (entries 1–2, Table 6). The compound showed exceptional chemical stability in the presence of an excess of dithiotreitol (DTT). Aqueous stability, assessed initially in PBS, showed some liability, though this was shown to be an effect of some solubility limitation in that particular medium (entries 4 and 5, Table 6). In mice, compound 45 demonstrated limited liabilities in plasma stability and moderate plasma protein binding (entries 6–8). A parallel artificial membrane permeability assay (PAMPA) was used as an in vitro model of passive, transcellular blood—brain barrier (BBB) permeability at biological pH. Compound 45 registered at the lower end of the moderately permeable range of this assay.

Cytochrome P450 induction was not assessed; however, only insignificant inhibition of 1A2, 2C9, 2C19, 2D6, and 3A4 was observed (<30% at 10 μ M, entries 12–16). Some metabolic liability was detected after 1 h exposure to mouse liver homogenates (~43% of 45 remained), though the compound showed no toxicity (>25 μ M LD₅₀) toward human hepatocytes. Lastly, compound 45 was evaluated in radioligand binding assays against a panel of 67 GPCRs, ion channels and transporters. At a single concentration of 10 μ M, 91% inhibition of the human norepinephrine transporter was observed; however, inhibition of all other targets in the collection was \leq 36%. Overall, this pharmacological profile obtained of compound 45 was encouraging in combination with the potency and toxicity data that the prototype compound may be useful in establishing baseline optimization parameters for in vivo assessment.

To that end, preliminary in vivo mouse pharmacokinetic (PK) data were collected for compound **45** in order to assess plasma and brain exposure relative to time and dose (Table 7). The concentration of compound **45**, delivered by intraperitoneal (ip) administration, was determined in plasma at 20 and 120 min

Table 8. Effect of Compound 45 on Wild Type and Mutant VEE Viruses

VEEV strain	protein segment of located mutation	mutation	compound 45 CPE potency, EC_{50} (μ M)	fold loss of potency (mutant EC_{50} /parental EC_{50})
TC-83	nsP2	none (parental strain)	0.03	
	nsP2	Y102C	17	567
	nsP2	D116N	19	633
	nsP4	Q210K	>50	>1667
V3526	nsP2	none (parental strain)	0.05	
	nsP2	Y102C	19	380
	nsP4	Q210K	66	1320

post-administration for each of three doses (1, 5, and 10 mg/kg) with three mice in each cohort. Additionally, the concentration of compound 45 in brain tissue was determined at the terminal end point (120 min) for each group. In accordance with the in vitro PK profile, the compound was readily depleted from circulation over the experimental window; however, significant exposure was observed for each dose in plasma and brain tissue. Moreover, >10-fold concentrations over the in vitro CPE assay EC $_{50}$ values were observed in both plasma and brain for the 5 and 10 mg/kg doses after 120 min.

These preliminary in vivo studies encouraged us to evaluate 45 in a lethal mouse model of VEEV infection. ²⁰ Single and multiple dose range finding studies were performed to define any potential toxicity of compound 45 in mice prior to conducting efficacy studies. The compound was administered ip twice daily to C3H/HeN mice in doses of 5 and 40 mg kg⁻¹ day⁻¹ for 4 days. For all mice, no apparent toxicity or adverse effects were observed in the mice as measured by body weight loss, lethargy, hunched posture, and ruffled fur. To test the in vivo efficacy of compound 45 in a 21-day lethal mouse model, compound 45 was administered ip at 2.5 mg/kg twice daily for 4 days to one group of VEEV (TC-83) infected mice (Figure 2).⁴¹ Mice receiving treatment were dosed with compound 4 h prior to intranasal exposure to VEEV TC-83. Compared to the infected, untreated group, significant survival rates (71%) were observed with VEEV-infected mice treated with compound 45.

Mechanism of Action Studies. To understand the regions of the viral protein that are important to the molecular action of the compound, an assay using hit quinazolinone 1-resistant mutant viruses was performed. Previously we have reported the isolation of mutant virus strains that are resistant to hit quinazolinone 1 with mutations of Y102C and D116N in nonstructural protein 2.²⁷ More recently, using the same method, we have isolated additional quinazolinone 1-resistant mutant VEEV TC-83 strains with a mutation of Q120K in nsP4 (Table 8). The alphavirus genome comprises two open reading frames, a structural and nonstructural segment. The structural segment encodes for proteins constituting the virion: (1) envelope proteins E1 and E2 and (2) core nucleocapsid protein C. The nonstructural segment encodes nonstructural proteins 1-4 (nsP1-nsP4), which are necessary for transcription and replication of viral RNA.43

Mutant viruses obtained from those studies⁴⁴ were used to examine the resistance profile of amidine **45**. To confirm that each of these sites conferred resistance, the amino acids were changed in VEEV V3526 by site-directed mutagenesis, and the EC₅₀ value for compound **45** was measured for each mutant virus (Table 8). The activity of compound **45** significantly decreased when it was tested against these mutant viruses compared to the parental VEEV strains, suggesting that compound **45** may target a critical function of nsP2/nsP4 in the VEEV replicase complex,

resulting in inhibition of viral RNA synthesis. At this point in time, a function has not been ascribed to the domain in the nsP2 or nsP4 that compound 45 targets, but more advanced experiments are underway to elucidate their role in the potent inhibition exhibited by these compounds.

CONCLUSION

With the emergence of alphavirus infections and the biohazards associated with their misuse has evolved an urgent need for effective countermeasures that have yet to be realized. Our efforts, directed at the identification and development of VEEV inhibitors, revealed a quinazolinone hit scaffold that was optimized through structural modification to afford novel, rearranged amidines with tremendous potential for further lead development and utility in dissecting novel mechanistic elements associated with alphavirus replication. The probe compound 45 showed potent inhibition of viral CPE for several strains of VEEV in the low nanomolar range without inducing cytotoxicity. These antiviral effects translated to impressive cellular efficacy in plaque reduction assays at submicromolar concentrations of compound, thus demonstrating the ability of amidine 45 to potently arrest viral replication. Gratifyingly, in vitro phamacokinetic data collected on amidine 45 predicted moderate brain exposure that was corroborated by preliminary pharmacokinetic studies in mice. These results helped to establish a dosing strategy for efficacy studies in which amidine 45 was shown to impart significant protection to VEEV-infected mice in a 21-day study. Mechanistically, these compounds appear to inhibit viral replication through a unique pathway involving interference with viral nonstructural proteins 2 and 4. Ultimately, the discovery of amidine 45 and its analogs represents an exciting advance in alphavirus research in that these agents offer a distinctive mechanistic intervention and a potential therapeutic and biodefense opportunity for VEEV. Future work centers on optimizing the structural amidine architecture to refine in vivo pharmacokinetic parameters, evaluating these and improved compounds in advanced animal efficacy models for VEEV, determining specific proteins of mechanistic relevance, and profiling these compounds against other viruses of therapeutic interest.

EXPERIMENTAL SECTION

Chemistry. Purity of all final compounds was confirmed by HPLC/MS analyis and determined to be ≥95%. 1 H and 13 C NMR spectra were recorded on a Bruker AM 400 spectrometer (operating at 400 and 101 MHz, respectively) or a Bruker AVIII spectrometer (operating at 500 and 126 MHz, respectively) in CDCl₃ (residual internal standard CHCl₃ = δ 7.26), DMSO- d_6 (residual internal standard CD₃SOCD₂H = δ 2.50), or acetone- d_6 (residual internal standard CD₃COCD₂H = δ 2.05). The chemical shifts (δ) reported are given in parts per million (ppm), and the coupling constants (J) are in hertz (Hz). The spin multiplicities are reported as s = singlet, br s = broad singlet, d = doublet, t = triplet, q =

quartet, p = pentuplet, dd = doublet of doublet, dt = doublet of triplet, td = triplet of doublet, tt = triplet of triplet, and m = multiplet. The LC-MS analysis was performed on an Agilent 1200 RRL HPLC system with photodiode array UV detection and an Agilent 6224 TOF mass spectrometer. The chromatographic method utilized the following parameters: a Waters Acquity BEH C-18 2.1 mm \times 50 mm, 1.7 μ m column; UV detection wavelength = 214 nm; flow rate = 0.4 mL/min; gradient = 5-100% MeCN over 3 min with a hold of 0.8 min at 100% MeCN; the aqueous mobile phase contained 0.15% NH₄OH. The mass spectrometer utilized the following parameters: an Agilent multimode source that simultaneously acquires ESI+/APCI+; a reference mass solution consisting of purine and hexakis(1H,1H,3Htetrafluoropropoxy)phosphazine; and a makeup solvent of 90:10:0.1 MeOH/H₂O/HCO₂H which was introduced to the LC flow prior to the source to assist ionization. Melting points were determined on a Stanford Research Systems OptiMelt apparatus. Microwave irradiated reactions were carried out using a Biotage Initiator Classic synthesizer. Flash chromatography separations were carried out using a Teledyne Isco CombiFlash Rf 200 purification system with silica gel columns. Mass-directed fractionation separations were carried out using an Agilent 1200 HPLC system with photodiode array UV detection and an Agilent 6120 mass spectrometer. The chromatographic method utilized a Waters XBridge C-18, 19 mm \times 150 mm, 5 μ m column; UV detection wavelength = 214 nm; flow rate = 20 mL/min; focused gradient = 5-100% MeCN; the aqueous mobile phase contained 0.15% $\mathrm{NH_{4}OH}.$ Quinazolinones were generally prepared according to Scheme 1 and by the protocols detailed for compound 1, below, unless otherwise specified. Amidines were prepared by either of two methods depicted in Scheme 2 and by the protocols detailed for compound 45, below, unless otherwise specified.

Synthesis of 2-((4-Ethylpiperazin-1-yl)methyl)-3-(2-fluorophenyl)-6-nitroquinazolin-4(3*H*)-one (1). Step 1: Synthesis of 2-(2-Chloroacetamido)-5-nitrobenzoic Acid (3). Triethylamine (2 mL, 14.4 mmol) was added to a solution of 2-amino-5-nitrobenzoic acid (2.280 g, 12.5 mmol) in dry $\mathrm{CH_2Cl_2}$ (34 mL) under an atmosphere of nitrogen. The mixture was lowered to 0 °C, and a solution of chloroacetyl chloride (1.09 mL, 13.7 mmol) in dry $\mathrm{CH_2Cl_2}$ (16 mL) was slowly added. The mixture was allowed to reach rt over 2 h. The solvent was removed in vacuo, and water (20 mL) was added. The product was filtered, rinsed with water (2 × 20 mL), and then rinsed with 5% ether/hexanes (3 × 30 mL) to give 3 (3.08 g, 95%) as a light-brown solid. $^1\mathrm{H}$ NMR (400 MHz, acetone- $^4\mathrm{G}$) δ 12.26 (s, 1H), 8.93–8.87 (m, 2H), 8.45 (dd, $^4\mathrm{G}$) = 9.3, 2.8 Hz, 1H), 4.42 (s, 2H). $^{13}\mathrm{C}$ NMR (101 MHz, acetone- $^4\mathrm{G}$) δ 168.5, 166.9, 146.8, 143.1, 129.9, 127.7, 121.0, 117.2, 44.1.

Step 2: Synthesis of 2-(Chloromethyl)-3-(2-fluorophenyl)-6-nitroquinazolin-4(3H)-one. A microwave vial was charged with 2-(2-chloroacetamido)-5-nitrobenzoic acid (1.93 g, 7.5 mmol) and capped. The atmosphere was evacuated under reduced pressure and flushed with Ar. Dry MeCN (14 mL) was added to the vial, followed by POCl₃ (1.5 mL, 16.1 mmol) and a solution of 2-fluoroaniline (0.97 mL, 10.1 mmol) in dry MeCN (4 mL). The mixture was heated in a microwave reactor at 150 °C for 15 min. The mixture was transferred to a larger flask and slowly quenched with saturated aqueous NaHCO₃ (80 mL). The resulting solid was filtered and rinsed with water (3 × 50 mL) to give 2-(chloromethyl)-3-(2-fluorophenyl)-6-nitroquinazolin-4(3H)-one (1.39 g, 56%) as a burnt-orange solid. 1 H NMR (400 MHz, CDCl₃) δ 9.14 (d, J = 2.5 Hz, 1H), 8.60 (dd, J = 8.9, 2.6 Hz, 1H), 7.92 (d, J = 8.9 Hz, 1H), 7.64–7.57 (m, 1H), 7.48–7.31 (m, 3H), 4.40 (d, J = 12.0 Hz, 1H), 4.23 (d, J = 12.0 Hz, 1H).

Step 3: Synthesis of 2-((4-Ethylpiperazin-1-yl)methyl)-3-(2-fluorophenyl)-6-nitroquinazolin-4(3H)-one (1). To a microwave vial, 2-(chloromethyl)-3-(2-fluorophenyl)-6-nitroquinazolin-4(3H)-one (560 mg, 1.7 mmol) was added and dissolved in dry MeCN (9 mL). Potassium carbonate (464 mg, 3.4 mmol), N-ethylpiperazine (0.30 mL, 2.4 mmol), and KI (106 mg, 0.64 mmol) were successively added, and the vial was capped. The mixture was heated in a microwave reactor at 80 °C for 10 min. The mixture was concentrated and the product was purified by flash chromatography (0–10% MeOH/CH₂Cl₂) to give 1 (333 mg, 48%) as a pale-orange solid. 1 H NMR (400 MHz, CDCl₃) δ 9.15 (d, J = 2.6 Hz, 1H), 8.58 (dd, J = 8.9, 2.7 Hz, 1H), 7.88 (d, J = 8.9

Hz, 1H), 7.55–7.49 (m, 1H), 7.37–7.28 (m, 3H), 3.44 (d, J = 13.7 Hz, 1H), 3.27 (d, J = 13.7 Hz, 1H), 2.44–2.16 (m, 10H), 1.04 (t, J = 7.2 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 160.8, 159.4, 157.1, 156.9, 151.1, 146.2, 131.7, 131.6, 129.8, 129.4, 128.9, 124.7, 124.6, 124.4, 124.3, 123.9, 121.5, 116.8, 116.6, 61.8, 53.1, 52.6, 52.3, 12.1. LC–MS: $t_{\rm R}$ = 3.18 min, purity = 98%. HRMS (m/z): calcd for C₂₁H₂₃FN₃O₃ (M + H)⁺ 412.1779; found 412.1808.

2-((4-Ethylpiperazin-1-yl)methyl)-6-nitro-3-phenylquinazolin-4(3*H***)-one (8). Following the same three step procedure used to synthesize 1, 2-(chloromethyl)-6-nitro-3-phenylquinazolin-4(3***H***)-one (168 mg, 0.53 mmol) was used to produce 8 (101 mg, 48%) as a paleyellow solid, mp 179–182 °C (dec). ¹H NMR (400 MHz, CDCl₃) \delta 9.11 (d, J = 2.6 Hz, 1H), 8.55 (dd, J = 9.0, 2.6 Hz, 1H), 7.87 (d, J = 9.0 Hz, 1H), 7.58–7.50 (m, 3H), 7.34–7.29 (m, 2H), 3.27 (s, 2H), 2.48–2.26 (m, 10H), 1.06 (t, J = 7.2 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) \delta 161.5, 157.1, 151.2, 146.0, 136.1, 129.7, 129.5, 129.3, 128.8, 128.6, 123.8, 121.6, 61.3, 53.0, 52.7, 52.4, 12.0. LC–MS: t_R = 2.90 min, purity = 98%. HRMS (m/z): calcd for C₂₁H₂₄N₅O₃ (M + H)⁺ 394.1874; found 394.1896.**

2-((4-Ethylpiperazin-1-yl)methyl)-4-oxo-3-phenyl-3,4-dihydroquinazoline-6-carbonitrile (10). Following the same procedure used to synthesize **1**, 2-(chloromethyl)-4-oxo-3-phenyl-3,4-dihydroquinazoline-6-carbonitrile (25 mg, 0.085 mmol) was used to produce **10** (20 mg, 63%) as a yellow solid, mp 136–142 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 8.59 (d, J = 2.0 Hz, 1H), 7.95 (dd, J = 8.5, 2.0 Hz, 1H), 7.83 (d, J = 8.5 Hz, 1H), 7.59–7.46 (m, 3H), 7.34–7.27 (m, 2H), 3.26 (s, 2H), 2.56–2.22 (m, 10H), 1.06 (t, J = 7.2 Hz, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 161.2, 156.5, 149.7, 136.5, 136.1, 132.6, 129.7, 129.6, 129.0, 128.8, 121.9, 118.1, 110.7, 61.2, 52.7, 52.6, 52.4, 11.8. LC–MS: $t_{\rm R}$ = 2.91 min, purity = 97%. HRMS (m/z): calcd for C₂₂H₂₄N₅O (M + H)⁺ 374.1975; found 374.1996.

3-(2-Fluorophenyl)-2-((4-methylpiperazin-1-yl)methyl)-6-nitroquinazolin-4(3*H***)-one (18). Following the same procedure used to synthesize 1 (except the reaction was heated at 150 °C for 15 min), 2-(chloromethyl)-3-(2-fluorophenyl)-6-nitroquinazolin-4(3***H***)-one (83 mg, 0.25 mmol) and 1-methylpiperazine (0.04 mL, 0.36 mmol) were used to produce 18 (31 mg, 31%) as a light-orange solid, mp 218–221 °C. ¹H NMR (400 MHz, CDCl₃) \delta 9.11 (d, J = 2.6 Hz, 1H), 8.55 (dd, J = 8.9, 2.6 Hz, 1H), 7.86 (d, J = 8.9 Hz, 1H), 7.55–7.47 (m, 1H), 7.36–7.24 (m, 3H), 3.42 (d, J = 13.7 Hz, 1H), 3.25 (d, J = 13.7 Hz, 1H), 2.38–2.18 (m, 11H). ¹³C NMR (101 MHz, CDCl₃) \delta 160.7, 159.4, 157.0, 156.9, 151.1, 146.1, 131.6, 131.6, 129.7, 129.3, 128.8, 124.7, 124.6, 124.4, 124.2, 123.8, 121.4, 116.7, 116.5, 61.7, 54.9, 53.0, 46.1. LC—MS: t_R = 2.20 min, purity = 96%. HRMS (m/z): calcd for C_{20}H_{21}FN_5O_3 (M + H) ^+ 398.1623; found 398.1654.**

3-(2-Fluorophenyl)-6-nitro-2-(piperazin-1-ylmethyl)quinazolin-4(3H)-one (21). Step 1: Synthesis of tert-Butyl 4-((3-(2-Fluorophenyl)-6-nitro-4-oxo-3,4-dihydroquinazolin-2-yl)methyl)piperazine-1-carboxylate. In a round-bottom flask, 2-(chloromethyl)-3-(2-fluorophenyl)-6-nitroquinazolin-4(3*H*)-one (250 mg, 0.75 mmol) was dissolved in dry MeCN (3.5 mL) under a nitrogen atmosphere. Potassium carbonate (124 mg, 0.90 mmol), 1-BOC-piperazine (279 mg, 1.50 mmol), and KI (87 mg, 0.52 mmol) were added to the flask, and the mixture was thermally heated at 80 $^{\circ}$ C for 4 h. The solvent was removed in vacuo, and water (10 mL) was added. The product was extracted with EtOAc (3 \times 10 mL), washed with saturated aqueous NH₄Cl (10 mL) and saturated aqueous NaHCO $_3$ (10 mL), and then dried with Na $_2$ SO $_4$. The product was purified by flash chromatography (0-50% EtOAc/ hexanes) to give the title compound (201 mg, 56%) as a white solid. ¹H NMR (400 MHz, CDCl₃) δ 9.09–9.04 (m, 1H), 8.51 (dd, J = 8.9, 2.7 Hz, 1H), 7.83-7.77 (m, 1H), 7.50-7.42 (m, 1H), 7.29-7.20 (m, 3H), $3.39 \text{ (d, } J = 13.7 \text{ Hz, } 1\text{H}), 3.24 - 3.09 \text{ (m, } 5\text{H}), 2.25 - 2.16 \text{ (m, } 2\text{H}),}$ 2.11-2.02 (m, 2H), 1.36 (s, 9H).

Step 2: Synthesis of 3-(2-Fluorophenyl)-6-nitro-2-(piperazin-1-ylmethyl)quinazolin-4(3H)-one (21). In a round-bottom flask, tertbutyl 4-((3-(2-fluorophenyl)-6-nitro-4-oxo-3,4-dihydroquinazolin-2-yl)methyl)piperazine-1-carboxylate (71 mg, 0.15 mmol) was dissolved in dry CH₂Cl₂ (2 mL), and TFA (0.07 mL, 0.91 mmol) was slowly added dropwise. The reaction mixture was stirred at rt for 2 h, then diluted with water (4 mL) and CH₂Cl₂ (4 mL). The reaction was

quenched to pH 10 using saturated aqueous Na₂CO₃ (1 mL). The product was extracted with CH₂Cl₂ (3 × 10 mL), dried with Na₂SO₄, and purified by flash chromatography (0–10% MeOH/CH₂Cl₂) to give **21** (25 mg, 44%) as a white solid, mp 202–204 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 9.15 (d, J = 2.0 Hz, 1H), 8.64–8.53 (m, 1H), 7.89 (d, J = 8.8 Hz, 1H), 7.59–7.49 (m, 1H), 7.41–7.25 (m, 3H), 3.52–3.37 (m, 2H), 3.27 (d, J = 13.6 Hz, 1H), 2.81–2.59 (m, 4H), 2.37–2.10 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 160.7, 159.4, 157.0, 156.9, 151.1, 146.1, 131.7, 131.6, 129.7, 129.3, 128.8, 124.7, 124.6, 124.4, 124.3, 123.8, 121.4, 116.7, 116.5, 62.4, 54.3, 45.8. LC–MS: t_R = 2.22 min, purity = 100%. HRMS (m/z): calcd for C₁₉H₁₉FN₅O₃ (M + H)⁺ 384.1466; found 384.1471.

2-((4-Ethylpiperazin-1-yl)methyl)-3-(3-fluorophenyl)-6-nitroquinazolin-4(3*H***)-one (25). Following the same procedure used to synthesize 1, 2-(chloromethyl)-3-(3-fluorophenyl)-6-nitroquinazolin-4(3***H***)-one (157 mg, 0.47 mmol) was used to produce 25** (155 mg, 80%) as a white solid, mp 187–190 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 9.10 (d, J = 2.6 Hz, 1H), 8.56 (dd, J = 9.0, 2.6 Hz, 1H), 7.89 (d, J = 9.0 Hz, 1H), 7.58–7.50 (m, 1H), 7.29–7.22 (m, 1H), 7.18–7.12 (m, 2H), 3.36–3.26 (m, 2H), 2.51–2.24 (m, 10H), 1.07 (t, J = 7.2 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 164.0, 161.6, 161.3, 156.6, 151.0, 146.1, 137.3, 137.2, 130.6, 130.5, 129.4, 128.8, 124.6, 124.6, 123.7, 121.5, 117.2, 117.0, 116.9, 116.7, 61.4, 53.1, 52.7, 52.3, 12.0. LC–MS: t_R = 2.93 min, purity = 98%. HRMS (m/z): calcd for $C_{21}H_{23}FN_3O_3$ (M + H)+ 412.1779; found 412.1808.

2-((4-Ethylpiperazin-1-yl)methyl)-3-(4-fluorophenyl)-6-nitroquinazolin-4(3*H***)-one (26). Following the same procedure used to synthesize 1, 2-(chloromethyl)-3-(4-fluorophenyl)-6-nitroquinazolin-4(3***H***)-one (169 mg, 0.51 mmol) was used to produce 26** as a white solid (160 mg, 77%), mp 196–199 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 9.10 (d, J = 2.6 Hz, 1H), 8.56 (dd, J = 9.0, 2.6 Hz, 1H), 7.88 (d, J = 8.9 Hz, 1H), 7.37–7.31 (m, 2H), 7.28–7.21 (m, 2H), 3.29 (s, 2H), 2.52–2.24 (m, 10H), 1.07 (t, J = 7.2 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 164.2, 161.7, 161.6, 156.9, 151.0, 146.0, 131.9, 131.8, 130.9, 130.8, 129.3, 128.7, 123.7, 121.5, 116.5, 116.3, 61.5, 53.1, 52.8, 52.3, 12.1. LC-MS: t_R = 2.98 min, purity = 98%. HRMS (m/z): calcd for C₂₁H₂₃FN₅O₃ (M + H)⁺ 412.1779; found 412.1802.

6-Nitro-3-phenyl-2-(piperazin-1-ylmethyl)quinazolin-4(3*H***)-one (31). Following the same procedure used to synthesize 21, compound 31 was obtained as a white solid (49 mg, 86%), mp 190–195 °C. ^{1}H NMR (400 MHz, DMSO-d_{6}) \delta 8.82 (d, J = 2.5 Hz, 1H), 8.61 (dd, J = 9.0, 2.7 Hz, 1H), 7.93 (d, J = 9.0 Hz, 1H), 7.58–7.49 (m, 5H), 3.25 (s, 2H), 2.70–2.62 (m, 4H), 2.23–2.15 (m, 4H). ^{13}C NMR (126 MHz, CDCl₃) \delta 161.6, 157.0, 151.2, 146.0, 136.1, 129.7, 129.5, 129.3, 128.8, 128.7, 123.9, 121.6, 62.0, 54.2, 45.9. LC–MS: t_{R} = 2.61 min, purity = 99%. HRMS (m/z): calcd for C_{19}H_{20}N_{3}O_{3} (M + H)^{+} 366.1561; found 366.1567.**

3-(4-Methoxyphenyl)-6-nitro-2-(piperazin-1-ylmethyl)-quinazolin-4(3*H***)-one (34). Following the same procedure used to synthesize 21**, compound 34 was obtained as a white solid (146 mg, 90%), mp 180–183 °C (dec). ¹H NMR (500 MHz, DMSO- d_6) δ 8.80 (d, J = 2.5 Hz, 1H), 8.59 (dd, J = 9.0, 2.7 Hz, 1H), 7.92 (d, J = 9.0 Hz, 1H), 7.41–7.37 (m, 2H), 7.10–7.06 (m, 2H), 3.82 (s, 3H), 3.20 (s, 2H), 2.58–2.52 (m, 4H), 2.16–2.10 (m, 4H). ¹³C NMR (126 MHz, DMSO- d_6) δ 161.2, 159.3, 157.9, 150.9, 145.2, 130.2, 129.1, 128.8, 128.5, 122.5, 121.2, 113.9, 61.7, 55.5, 53.5, 45.2. LC–MS: t_R = 2.73 min, purity = 99%. HRMS (m/z): calcd for $C_{20}H_{22}N_5O_4$ (M + H)+ 396.1666; found 396.1670.

6-Nitro-2-(piperazin-1-ylmethyl)-3-(thiophen-3-yl)-quinazolin-4(3*H***)-one (37). Following the same procedure used to synthesize 21** except during purification 5% NH₄OH was added to MeOH in the mobile phase, 37 was obtained as a white solid (79 mg, 58%), mp 205–208 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 9.12 (d, J = 2.6 Hz, 1H), 8.55 (dd, J = 8.9, 2.6 Hz, 1H), 7.86 (d, J = 8.9 Hz, 1H), 7.49 (dd, J = 5.1, 3.2 Hz, 1H), 7.42 (dd, J = 3.2, 1.3 Hz, 1H), 7.09 (dd, J = 5.1, 1.3 Hz, 1H), 3.33 (br s, 2H), 2.86–2.75 (m, 4H), 2.42–2.31 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 161.4, 157.2, 150.8, 145.9, 133.3, 129.2, 128.6, 127.0, 125.8, 123.7, 123.5, 121.3, 62.0, 54.5, 46.0. LC–MS: t_R = 2.62 min, purity = 100%. HRMS (m/z): calcd for $C_{17}H_{18}N_5O_3S$ (M + H)⁺ 372.1125; found 372.1138.

2-(4-Ethylpiperazin-1-yl)-6-nitro-3-phenylquinazolin-4(3H)-one (38). Step 1: Synthesis of 6-Nitro-3-phenylquinazoline-2,4-(1H,3H)-dione. To a stirred mixture of 2-amino-5-nitrobenzanilide (1.97 g, 7.66 mmol) in dry toluene (25 mL) was carefully added a solution of 20% (w/v) phosgene in toluene (15 mL, 30.3 mmol). CAUTION! Phosgene is highly toxic and should be handled with care. The mixture was heated at 95 °C for 19.5 h, then concentrated and purified by flash column chromatography (0–90% EtOAc/hexanes) to give the title compound (1.45 g, 67%) as a colorless solid, mp 297–299 °C (dec). ¹H NMR (400 MHz, acetone- d_6) δ 10.93 (s, 1H), 8.82 (d, J = 2.6 Hz, 1H), 8.53 (dd, J = 9.0, 2.6 Hz, 1H), 7.58–7.45 (m, 4H), 7.45–7.39 (m, 2H).

Step 2: Synthesis of 2-(4-Ethylpiperazin-1-yl)-6-nitro-3-phenylquinazolin-4(3H)-one (38). Phosphorus pentachloride (44 mg, 0.21 mmol), 6-nitro-3-phenylquinazoline-2,4(1H,3H)-dione (44 mg, 0.16 mmol), and phosphorus oxychloride (1.5 mL, 16 mmol) were heated in a sealed vial at 120 °C for 16 h. The mixture was concentrated and redissolved in dry CH₂Cl₂ (5 mL). N-Ethylpiperazine (142 mg, 1.24 mmol) was added, and the mixture was briefly stirred at rt. The mixture was reconcentrated and the product was purified by mass-directed fractionation to give 38 (19 mg, 31%) as a yellow solid, mp 179-182 °C (dec). ¹H NMR (400 MHz, DMSO- d_6) δ 8.71 (d, J = 2.6 Hz, 1H), 8.46 (dd, J = 9.0, 2.8 Hz, 1H), 7.61-7.44 (m, 6H), 3.24-3.13 (m, 4H), 2.26-2.16 (m, 2H), 2.16-2.05 (m, 4H), 0.92 (t, J = 7.1 Hz, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 161.6, 156.0, 152.6, 142.4, 137.1, 128.9, 128.7, 128.6, 128.4, 126.7, 123.1, 117.9, 51.3, 51.2, 48.2, 11.7. LC-MS: t_R = 3.29 min, purity = 100%. HRMS (m/z): calcd for $C_{20}H_{22}N_5O_3(M+H)^+$ 380.1717; found 380.1732.

6-Nitro-3-phenyl-2-(piperidin-4-yl)quinazolin-4(3H)-one (39). Step 1: Synthesis of tert-Butyl 4-((4-Nitro-2-(phenylcarbamoyl)phenyl)carbamoyl)piperidine-1-carboxylate. To a stirred mixture of 1-(tert-butoxycarbonyl)piperidine-4-carboxylic acid (213 mg, 0.93 mmol) in dry CH₂Cl₂ (10 mL) was added a catalytic amount of DMF (2 drops) at rt followed by the dropwise addition of oxalyl chloride (479 mg, 3.77 mmol). The mixture was stirred at rt for 2 h and concentrated. The remaining residue was redissolved in dry THF (5 mL) and MeCN (5 mL) and slowly added to a mixture of 2-amino-5-nitro-Nphenylbenzamide (211 mg, 0.82 mmol) and 95% (w/w) NaH (128 mg, 5.07 mmol) in dry THF (10 mL) at -78 °C. The mixture was allowed to warm to rt and stirred for 16 h. The reaction was quenched with water, extracted with EtOAc (3×), washed with brine, and dried with Na₂SO₄. The product was purified by reverse-phase chromatography (C-18, 10-100% MeCN/H₂O) to give the title compound (38 mg, 10%) as an off-white solid. ¹H NMR (400 MHz, acetone- d_6) δ 11.41 (s, 1H), 10.17 (s, 1H), 8.84 (d, J = 9.3 Hz, 1H), 8.73 (d, J = 2.6 Hz, 1H),8.35 (dd, J = 9.3, 2.7 Hz, 1H), 7.77–7.71 (m, 2H), 7.38 (tt, J = 7.6, 2.0 Hz, 2H), 7.18 (tt, J = 7.1, 1.1 Hz, 1H), 4.08 (d, J = 12.7 Hz, 2H), 2.91– 2.72 (m, 2H), 2.63 (tt, I = 11.5, 3.8 Hz, 1H), 1.97 - 1.88 (m, 2H), 1.67 -1.53 (m, 2H), 1.40 (s, 9H).

Step 2: Synthesis of 6-nitro-3-phenyl-2-(piperidin-4-yl)quinazolin-4(3H)-one (39). Triethylamine (10 mg, 0.10 mmol), chlorotrimethylsilane (9 mg, 0.08 mmol), and tert-butyl 4-((4-nitro-2-(phenylcarbamoyl)phenyl)carbamoyl)piperidine-1-carboxylate (38 mg, 0.08 mmol) were dissolved in dry MeCN (1.5 mL) and heated in a microwave reactor at 150 °C for 10 min. The mixture was purified by flash chromatography (0-60% EtOAc/hexanes) and redissolved in 3:1 CH₂Cl₂/TFA (2 mL). The reaction mixture was stirred at rt for 16 h and concentrated. The product was purified by mass-directed fractionation to give **39** (14.4 mg, 51%) as a yellow solid, mp 193–200 °C (dec). ¹H NMR (400 MHz, DMSO- d_6) δ 8.80 (d, J = 2.5 Hz, 1H), 8.59 (dd, J = 9.0, 2.7 Hz, 1H), 7.89 (d, I = 9.0 Hz, 1H), 7.65 - 7.49 (m, 6H), 2.90 (d, I =10.6 Hz, 2H), 2.41-2.26 (m, 1H), 2.19-2.02 (m, 2H), 1.81-1.62 (m, 4H). 13 C NMR (101 MHz, DMSO- d_6) δ 160.8, 151.5, 144.7, 136.6, 129.6, 129.4, 128.7, 128.5, 128.4, 122.4, 120.6, 45.5, 41.5, 30.8. LC-MS: $t_{\rm R} = 2.96$ min, purity = 100%. HRMS (m/z): calcd for $C_{19}H_{19}N_4O_3$ (M + H)+ 351.1452; found 351.1460.

2-((Methyl(3-(methylamino)propyl)amino)methyl)-6-nitro-3-phenylquinazolin-4(3H)-one, 2,2,2-Trifluoroacetic Acid Salt (42). Step 1: Synthesis of tert-Butyl Methyl(3-(methyl((6-nitro-4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl)methyl)amino)propyl)-carbamate. Following the same procedure used to synthesize 1, 2-

(chloromethyl)-6-nitro-3-phenylquinazolin-4(3*H*)-one (221 mg, 0.70 mmol) and *tert*-butyl methyl(3-(methylamino)propyl)carbamate (240 mg, 1.19 mmol) were used to produce the title compound (106 mg, 31%) as a pale-orange solid. ¹H NMR (400 MHz, CDCl₃) δ 9.08 (d, J = 2.6 Hz, 1H), 8.54 (dd, J = 9.0, 2.7 Hz, 1H), 7.89 (d, J = 8.9 Hz, 1H), 7.61–7.51 (m, 3H), 7.33–7.28 (m, 2H), 3.31 (s, 2H), 3.13–3.05 (m, 2H), 2.78 (s, 3H), 2.32–2.25 (m, 2H), 2.17 (s, 3H), 1.50 (p, J = 7.3 Hz, 2H), 1.42 (s, 9H).

Step 2: Synthesis of 2-((Methyl(3-(methylamino)propyl)amino)methyl)-6-nitro-3-phenylquinazolin-4(3H)-one, 2,2,2-Trifluoroacetic Acid Salt (42). In a round-bottom flask, tert-butyl methyl (3-(methyl) (6nitro-4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl)methyl)amino)propyl)carbamate (148 mg, 0.31 mmol) was dissolved in dry CH₂Cl₂ (5 mL), and TFA (2 mL, 26.1 mmol) was added dropwise. The reaction mixture was stirred at rt for 45 min, then diluted with water (10 mL) and CH₂Cl₂ (10 mL). The mixture was adjusted to pH 10 using saturated aqueous Na₂CO₃ (10 mL). The layers were separated, and the aqueous layer was extracted with CH_2Cl_2 (3 × 25 mL). The combined organic extracts were dried with Na2SO4. The product was purified through a reverse-phase, C-18 column (5-100% MeCN/H₂O, RediSep Rf Gold C18 column from Teledyne Isco) to give 42 (11 mg, 7%) as a white solid. ¹H NMR (500 MHz, CDCl₃) δ 10.51 (br s, 2H), 9.10 (d, I = 2.6Hz, 1H), 8.56 (dd, J = 9.0, 2.7 Hz, 1H), 7.84 (d, J = 9.0 Hz, 1H), 7.66– 7.54 (m, 3H), 7.37-7.32 (m, 2H), 3.39-3.33 (m, 2H), 3.25-3.19 (m, 2H), 2.79 (s, 3H), 2.71 (t, I = 5.6 Hz, 2H), 2.36 (s, 3H), 1.96 (p, I = 5.5Hz, 2H). 13 C NMR (126 MHz, CDCl₃) δ 162.8, 162.6, 162.3, 162.0, 160.7, 157.7, 150.8, 145.9, 135.1, 130.5, 130.2, 129.0, 128.2, 128.0, 123.9, 121.3, 120.2, 117.9, 115.6, 113.2, 59.6, 56.9, 50.4, 43.3, 33.0, 21.9. LC-MS: $t_R = 2.25$ min, purity = 97%. HRMS (m/z): calcd for $C_{20}H_{24}N_5O_3$ (M + H)+ 382.1874; found 382.1878.

(E)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-5-nitro-Nphenylbenzamide (45). Method A. Step 1: Synthesis of tert-Butyl Methyl(2-(methyl((6-nitro-4-oxo-3-phenyl-3,4-dihydroquinazolin-2yl)methyl)amino)ethyl)carbamate (44). To a microwave vial, 2-(chloromethyl)-6-nitro-3-phenylquinazolin-4(3H)-one 43 (1.400 g, 4.4 mmol) was added and dissolved in dry MeCN (18 mL). Potassium carbonate (1.226 g, 8.9 mmol), tert-butyl methyl(2-(methylamino)ethyl)carbamate (1.176 g, 6.3 mmol), and KI (0.280 g, 1.7 mmol) were successively added, and the vial was capped. The mixture was heated in a microwave reactor at 80 °C for 5 min. The mixture was concentrated and the product was purified by flash chromatography (0-10% MeOH/ CH₂Cl₂ followed by 0-80% EtOAc/hexanes) to give tert-butylmethyl(2-(methyl((6-nitro-4-oxo-3-phenyl-3,4-dihydroquinazolin-2yl)methyl)amino)ethyl)carbamate 44 (0.719 g, 35%) as a pale-orange solid. 1 H NMR (400 MHz, CDCl₃) δ 9.13 (d, J = 2.6 Hz, 1H), 8.56 (dd, J= 9.0, 2.7 Hz, 1H), 7.89 (d, J = 9.0 Hz, 1H), 7.60 - 7.52 (m, 3H), 7.32 -7.27 (m, 2H), 3.36 (s, 2H), 3.22-3.06 (m, 2H), 2.75 (br s, 3H), 2.48 (br s, 2H), 2.20 (s, 3H), 1.39 (s, 9H).

Step 2: Synthesis of (E)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-5-nitro-N-phenylbenzamide (45). To a stirred solution of tert-butyl methyl(2-(methyl((6-nitro-4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl)methyl)amino)ethyl)carbamate 44 (678 mg, 1.45 mmol) in dry CH2Cl2 (20 mL), TFA (9.5 mL, 124 mmol) was slowly added. The reaction mixture was stirred at rt for 45 min, then diluted with water (40 mL) and CH₂Cl₂ (40 mL). The reaction was quenched to pH 10 using saturated aqueous Na2CO3 (40 mL). The organic phase was separated, and the aqueous phase was extracted with CH_2Cl_2 (2 × 40 mL). The combined organic phase was concentrated and purified by flash chromatography (0-5% MeOH/CH₂Cl₂) to give 45 (265 mg, 50%) as a pale-yellow solid, mp 168-173 °C. ¹H NMR (500 MHz, CDCl₃) δ 10.99 (s, 1H), 9.15 (d, J = 2.8 Hz, 1H), 8.15 (dd, J = 8.8, 2.8 Hz, 1H), 7.65-7.60 (m, 2H), 7.38-7.33 (m, 2H), 7.12 (tt, J = 7.3, 1.2 Hz, 1H), 6.80 (d, J = 8.8 Hz, 1H), 3.47 (t, J = 5.7 Hz, 2H), 3.28 (s, 3H),3.13 (s, 2H), 2.69 (t, J = 5.7 Hz, 2H), 2.26 (s, 3H). ¹³C NMR (126 MHz, $CDCl_3$) δ 162.6, 156.4, 154.0, 142.8, 138.2, 129.1, 127.6, 126.4, 126.2, 124.2, 123.8, 120.2, 55.1, 51.8, 49.7, 45.3, 36.9. LC-MS: $t_R = 3.19 \text{ min}$ purity = 99%. HRMS (m/z): calcd for $C_{19}H_{22}N_5O_3(M+H)^+$ 368.1717; found 368.1718.

(E)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-5-nitro-N-phenylbenzamide (45). Method B. To a stirred solution of 2-

(chloromethyl)-6-nitro-3-phenylquinazolin-4(3H)-one (158 mg, 0.50 mmol) in dry MeCN (2 mL) were added K₂CO₃ (140 mg, 1.01 mmol) and N,N'-dimethylethylenediamine (0.08 mL, 0.74 mmol). The mixture was stirred at 50 °C for 2 h. The mixture was concentrated and the product was purified by flash chromatography (0–5% MeOH/EtOAc) to give 45 (126 mg, 69%) as a yellow-orange solid. Analytical data were consistent with data obtained from method A. LC–MS purity = 100%.

(*E*)-2-((1-Ethyl-4-methylpiperazin-2-ylidene)amino)-5-nitro-*N*-phenylbenzamide (47). Method A as given for 45. Pale-yellow solid (18 mg, 26%), mp 150–157 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 10.66 (s, 1H), 9.14 (d, J = 2.8 Hz, 1H), 8.16 (dd, J = 8.8, 2.8 Hz, 1H), 7.65–7.56 (m, 2H), 7.38–7.33 (m, 2H), 7.16–7.10 (m, 1H), 6.81 (d, J = 8.8 Hz, 1H), 3.78 (q, J = 7.1 Hz, 2H), 3.46 (t, J = 5.6 Hz, 2H), 3.12 (s, 2H), 2.68 (t, J = 5.6 Hz, 2H), 2.25 (s, 3H), 1.30 (t, J = 7.2 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 162.9, 155.9, 154.6, 142.9, 138.2, 129.2, 127.8, 126.6, 126.1, 124.5, 124.0, 120.7, 55.4, 52.0, 46.8, 45.3, 43.6, 12.2. LC–MS: t_R = 3.34 min, purity = 99.7%. HRMS (m/z): calcd for $C_{20}H_{24}N_5O_3$ (M + H)+ 382.1874; found 382.1894.

(E)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-5-nitro-*N*-(thiophen-3-yl)benzamide (48). Method B as given for 45. Tan solid (46 mg, 50%), mp 170–175 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 11.26 (s, 1H), 9.16 (d, J = 2.8 Hz, 1H), 8.16 (dd, J = 8.8, 2.8 Hz, 1H), 7.75 (dd, J = 3.2, 1.3 Hz, 1H), 7.28–7.25 (m, 1H), 7.02 (dd, J = 5.2, 1.4 Hz, 1H), 6.80 (d, J = 8.7 Hz, 1H), 3.47 (t, J = 5.6 Hz, 2H), 3.28 (s, 3H), 3.13 (s, 2H), 2.69 (t, J = 5.7 Hz, 2H), 2.26 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 162.1, 156.7, 153.9, 143.0, 135.9, 127.7, 126.4, 125.9, 124.7, 123.8, 121.1, 110.7, 55.2, 52.0, 49.8, 45.4, 36.9. LC–MS: t_R = 3.19 min, purity = 99%. HRMS (m/z): calcd for $C_{17}H_{20}N_5O_3S$ (M + H)⁺ 374.1281; found 374.1289.

(*F*)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-*N*-(4-methoxyphenyl)-5-nitrobenzamide (51). Method A as given for 45. Yellow solid (33 mg, 41%). 1 H NMR (400 MHz, CDCl₃) δ 10.90 (s, 1H), 9.14 (d, J = 2.8 Hz, 1H), 8.14 (dd, J = 8.7, 2.8 Hz, 1H), 7.56–7.51 (m, 2H), 6.92–6.87 (m, 2H), 6.79 (d, J = 8.8 Hz, 1H), 3.80 (s, 3H), 3.47 (t, J = 5.6 Hz, 2H), 3.26 (s, 3H), 3.14 (s, 2H), 2.69 (t, J = 5.6 Hz, 2H), 2.26 (s, 3H). 13 C NMR (101 MHz, CDCl₃) δ 162.5, 156.5, 156.4, 154.0, 142.9, 131.5, 127.6, 126.34, 126.29, 123.8, 121.8, 114.3, 55.6, 55.2, 51.9, 49.8, 45.4, 37.0. LC–MS: t_R = 3.17 min, purity = 99%. HRMS (m/z): calcd for C_{20} H₂₄N₅O₄ (M + H)⁺ 398.1823; found 398.1840.

(*E*)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-*N*-(2-fluorophenyl)-5-nitrobenzamide (52). Method A as given for 45. Yellow solid (71 mg, 89%). 1 H NMR (400 MHz, CDCl₃) δ 11.07 (s, 1H), 9.17 (d, J = 2.8 Hz, 1H), 8.53 (td, J = 8.1, 1.5 Hz, 1H), 8.14 (dd, J = 8.8, 2.8 Hz, 1H), 7.19–7.13 (m, 1H), 7.11–7.01 (m, 2H), 6.78 (d, J = 8.8 Hz, 1H), 3.46 (t, J = 5.6 Hz, 2H), 3.22 (d, J = 1.4 Hz, 3H), 3.14 (s, 2H), 2.68 (t, J = 5.7 Hz, 2H), 2.25 (s, 3H). 13 C NMR (101 MHz, CDCl₃) δ 163.0, 156.9, 154.8, 154.1, 151.7, 142.6, 128.0, 126.8, 126.7, 125.5, 124.8, 124.7, 124.34, 124.26, 124.0, 122.74, 122.72, 114.9, 114.7, 55.3, 51.9, 49.6, 45.3, 36.9, 36.8. LC—MS: $t_R = 3.37$ min, purity = 99%. HRMS (m/z): calcd for $C_{19}H_{21}FN_5O_3$ (M + H)+ 386.1623; found 386.1639.

(*E*)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-*N*-(3-fluorophenyl)-5-nitrobenzamide (53). Method A as given for 45. Paleyellow solid (12 mg, 39%), mp 165–169 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 11.06 (s, 1H), 9.14 (d, J = 2.8 Hz, 1H), 8.18 (dd, J = 8.8, 2.8 Hz, 1H), 7.57 (dt, J = 11.0, 2.3 Hz, 1H), 7.33–7.23 (m, 2H), 6.87–6.75 (m, 2H), 3.48 (t, J = 5.6 Hz, 2H), 3.29 (s, 3H), 3.13 (s, 2H), 2.69 (t, J = 5.7 Hz, 2H), 2.26 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 164.2, 163.0, 162.3, 156.7, 154.0, 143.1, 139.94, 139.85, 130.3, 130.2, 127.9, 126.7, 126.1, 123.9, 115.53, 115.50, 111.1, 111.0, 107.9, 107.7, 55.3, 52.0, 49.9, 45.5, 37.1. LC-MS: t_R = 3.33 min, purity = 97%. HRMS (m/z): calcd for $C_{19}H_{21}FN_5O_3$ (M + H)+ 386.1623; found 386.1630.

(*E*)-2-((1,4-Dimethylpiperazin-2-ylidene)amino)-*N*-(4-fluorophenyl)-5-nitrobenzamide (54). Method A as given for 45. Yellow solid (14 mg, 44%), mp 170–174 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 11.00 (s, 1H), 9.12 (d, J = 2.8 Hz, 1H), 8.14 (dd, J = 8.8, 2.8 Hz, 1H), 7.61–7.53 (m, 2H), 7.08–6.99 (m, 2H), 6.79 (d, J = 8.8 Hz, 1H), 3.47 (t, J = 5.6 Hz, 2H), 3.25 (s, 3H), 3.14 (s, 2H), 2.69 (t, J = 5.7 Hz, 2H), 2.26 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 162.8, 160.6, 158.2, 156.7, 154.0, 143.0, 134.40, 134.38, 127.7, 126.5, 126.1, 123.8, 122.0, 121.9, 116.0, 115.7, 55.3, 52.0, 49.8, 45.4, 37.0. LC–MS: t_R = 3.12

min, purity = 99%. HRMS (m/z): calcd for $C_{19}H_{21}FN_5O_3$ $(M + H)^+$ 386.1623; found 386.1631.

(*E*)-5-Cyano-2-((1,4-dimethylpiperazin-2-ylidene)amino)-*N*-phenylbenzamide (63). Method B as given for 45. Yellow solid (25 mg, 71%), mp 167–169 °C. ¹H NMR (400 MHz, CDCl₃) δ 10.90 (s, 1H), 8.08 (dd, J = 11.8, 9.3 Hz, 1H), 7.61–7.51 (m, 3H), 7.37–7.30 (m, 2H), 7.13–7.07 (m, 1H), 6.53 (dd, J = 11.2, 6.9 Hz, 1H), 3.40 (t, J = 5.6 Hz, 2H), 3.22 (s, 3H), 3.07 (s, 2H), 2.64 (t, J = 5.6 Hz, 2H), 2.25 (s, 3H). 13 C NMR (101 MHz, CDCl₃) δ 163.0, 156.4, 152.2, 138.4, 136.2, 134.6, 129.2, 126.8, 124.3, 124.2, 120.4, 119.1, 106.0, 55.2, 52.0, 49.8, 45.5, 36.9. LC–MS: t_R = 3.11 min, purity = 99.5%. HRMS (m/z): calcd for $C_{20}H_{22}N_5O$ (M + H)+ 348.1819; found 348.1820.

(*E*)-4-Chloro-5-cyano-2-((1,4-dimethylpiperazin-2-ylidene)-amino)-*N*-phenylbenzamide (64). Method B as given for 45. Yellow solid (37 mg, 97%), mp 167–171 °C (dec). ¹H NMR (400 MHz, CDCl₃) δ 10.80 (s, 1H), 8.57 (s, 1H), 7.61–7.56 (m, 2H), 7.38–7.32 (m, 2H), 7.15–7.10 (m, 1H), 6.84 (s, 1H), 3.46 (t, J = 5.6 Hz, 2H), 3.26 (s, 3H), 3.13 (s, 2H), 2.69 (t, J = 5.6 Hz, 2H), 2.29 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 162.2, 156.8, 152.8, 138.9, 138.2, 137.8, 129.2, 125.3, 124.5, 124.2, 120.4, 116.3, 106.6, 55.1, 51.9, 49.8, 45.4, 37.0. LC–MS: t_R = 3.34 min, purity = 97%. HRMS (m/z): calcd for C₂₀H₂₁ClN₅O (M + H)⁺ 382.1429; found 382.1450.

(*E*)-Methyl 4-((1,4-Dimethylpiperazin-2-ylidene)amino)-3-(phenylcarbamoyl)benzoate (68). Step 1: Synthesis of 2-(2-Chloroacetamido)-5-iodobenzoic Acid. Following the same procedure for 3, 2-amino-5-iodobenzoic acid (6.58 g, 25.0 mmol) afforded the title compound (8.47 g, 100%) as a lavender solid. ¹H NMR (400 MHz, acetone- d_6) δ 11.86 (br s, 1H), 8.54 (d, J = 8.9 Hz, 1H), 8.41 (d, J = 2.2 Hz, 1H), 7.97 (dd, J = 8.9, 2.2 Hz, 1H), 4.35 (s, 2H).

Step 2: Synthesis of 2-(Chloromethyl)-6-iodo-3-phenylquinazolin-4(3H)-one. Procedure as done for 1, step 2. White solid (0.95 g, 41%). 1 H NMR (400 MHz, CDCl₃) δ 8.62 (d, J = 1.9 Hz, 1H), 8.08 (dd, J = 8.6, 2.1 Hz, 1H), 7.62–7.55 (m, 3H), 7.51 (d, J = 8.6 Hz, 1H), 7.37–7.32 (m, 2H), 4.24 (s, 2H).

Step 3: Synthesis of tert-Butyl-(2-(((6-iodo-4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl)methyl)(methyl)amino)ethyl)(methyl)carbamate. Method A as given for 45. Yellow oil (0.434 g, 90%). 1 H NMR (400 MHz, CDCl₃) δ 8.61 (d, J = 1.9 Hz, 1H), 8.04 (dd, J = 8.6, 2.1 Hz, 1H), 7.57–7.46 (m, 4H), 7.29–7.26 (m, 2H), 3.28 (s, 2H), 3.19–3.03 (m, 2H), 2.77–2.67 (m, 3H), 2.43 (br s, 2H), 2.15 (s, 3H), 1.38 (s, 9H).

Step 4: Synthesis of Methyl 2-(((2-(((tert-Butoxycarbonyl)(methyl)-amino)ethyl)(methyl)amino)methyl)-4-oxo-3-phenyl-3,4-dihydroquinazoline-6-carboxylate. To a solution of tert-butyl-(2-(((6-iodo-4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl)methyl)(methyl)amino)ethyl)(methyl)carbamate (0.434 g, 0.79 mmol) in MeCN (4.4 mL) and MeOH (2.2 mL) were added Pd(OAc) $_2$ (18 mg, 0.08 mmol), DPPF (44 mg, 0.08 mmol), Et $_3$ N (0.11 mL, 0.79 mmol), and K $_2$ CO $_3$ (0.328 g, 2.37 mmol). The reaction mixture was purged with CO for 5 min, followed by stirring under 1 atm of CO (balloon) at 60 °C for 3 h. The reaction mixture was filtered through Celite and rinsed with EtOAc (2 × 10 mL). The filtrate was concentrated and purified by flash chromatography (EtOAc/hexanes). The title compound (0.248 g, 65%) was obtained as a yellow oil and used in the next step without further purification.

Step 5: Synthesis of (E)-Methyl 4-((1,4-Dimethylpiperazin-2-ylidene)amino)-3-(phenylcarbamoyl)benzoate (68). Method A as given for 45. White solid (81 mg, 73%), mp 154–158 °C (dec). $^1\mathrm{H}$ NMR (400 MHz, CDCl₃) δ 10.91 (s, 1H), 8.93 (d, J = 2.2 Hz, 1H), 8.00 (dd, J = 8.3, 2.2 Hz, 1H), 7.67–7.62 (m, 2H), 7.38–7.32 (m, 2H), 7.13–7.07 (m, 1H), 6.77 (d, J = 8.3 Hz, 1H), 3.91 (s, 3H), 3.42 (t, J = 5.6 Hz, 2H), 3.26 (s, 3H), 3.09 (s, 2H), 2.65 (t, J = 5.6 Hz, 2H), 2.22 (s, 3H). $^{13}\mathrm{C}$ NMR (126 MHz, CDCl₃) δ 166.9, 164.1, 156.1, 152.4, 138.8, 133.5, 132.8, 129.2, 125.8, 124.7, 124.0, 123.5, 120.3, 55.2, 52.1, 49.8, 45.5, 36.9. LC-MS: t_{R} = 3.15 min, purity = 100%. HRMS (m/z): calcd for $\mathrm{C}_{21}\mathrm{H}_{25}\mathrm{N}_{4}\mathrm{O}_{3}$ (M + H)⁺ 381.1921; found 381.1927.

Biology. CPE-Based Antiviral Assay. VEEV strains of TC-83, V3526, and TrD were amplified in BHK-21 cells. Anti-VEEV activity and cytotoxicity of compound were measured as described elsewhere. ²⁷ Briefly, for a dose—response study, test compounds were solubilized in DMSO at 20 mM first and then diluted further in DMSO by a 2-fold serial dilution of eight points. Compounds were diluted in cell growth

medium (EMEM with 10% FBS) and then added to white cell plates in which Vero76 cells were grown overnight (seeding density of 12 000 cells/well in a volume of 45 μL). After a 1 h incubation, the cells were challenged with 600 plaque forming unit of virus per well. The cells were incubated for 2 days. Then cell viability was measured with 90 μL per well of CellTiter-Glo reagent (Promega). For cytotoxicity, cell medium was used in place of virus and incubated for 3 days. Experiments were done in triplicate for efficacy test and duplicate for cytotoxicity test. For EC $_{50}$ or CC $_{50}$ calculation, relative cell viability compared to the noninfected cells controls was plotted using XLfit (IBDS) and EC $_{50}$ and CC $_{50}$ were calculated using four parameter logistic model or sigmoidal dose—response model.

Titer Reduction Assay. A microplaque assay was used. ²⁷ Vero 76 cells grown in six-well plates were infected with virus at an MOI of 0.05 and incubated in culture medium with or without compounds. At 40 h postinfection, the titer of progeny virus in the medium was measured as follows. Supernates from the six-well plate from each treatment were diluted in DMEM supplemented with 5% FBS using a liquid handler, epMotion 5070 CB (Eppendorf Inc.). Vero 76 cells grown overnight in 96-well plates were infected with 25 μ L of the serially diluted samples. The plates were incubated for 1 h at 37 °C, 5% CO₂. Wells were rinsed with 100 μ L of PBS and replenished with DMEM supplemented with 0.75% methylcellulose and 10% FBS and incubated at 37 °C, 5% CO₂ for 3 days. The microplaques were visualized by staining with 0.2% crystal violet in 4% paraformaldehyde and 20% ethanol.

In Vivo Antiviral Efficacy Studies. Twenty-one C3H/HeN mice in the 5- to 6-week-old range were obtained from Charles River Laboratories (Wilmington, MA) and randomly assigned and evenly distributed to one of three treatment groups: group 1, vehicle control (no virus or compound treatment); group 2, VEEV only (no compound treatment); group 3, VEEV + compound treatment (5 mg kg⁻¹ day⁻¹ compound 45). Mice were dosed ip twice per day with vehicle only (1% carboxymethly cellulose, 200 μ L, group 1) or compound 45 formulated in 1% carboxymethlycellulose (2.5 mg/kg for each dose given twice a day, 200 µL each dose, group 3). Treatments were administered for 4 days, beginning 4 h prior to virus challenge for group 3 (day 0 to day 3). Mice in groups 2 and 3 were infected intranasally with 10 times the LD₅₀ of TC-83 (day 0, virus diluted in 50 μ L of PBS). For the group 1 mice (vehicle control), PBS was used in place of virus. Mice were weighed from day 0 to day 21 and checked twice a day for mortality and morbidity. The median time-to-death for the group challenged with VEEV was 8 days. The p values were generated from comparisons of survival data using the log rank (Mantel-Cox) test using Prism 6 (GraphPad Software, Inc.).

ASSOCIATED CONTENT

Supporting Information

Analytical characterization and experimental details for synthesized intermediates, in vitro and in vivo pharmacology assay protocols, Eurofins Panlabs Hit LeadProfiling data for compound 45. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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ABBREVIATIONS USED

BBB, blood—brain barrier; cLogP, calculated partitian coefficient of a compound in octanol/water; CNS, central nervous system; CPE, cytopathic effect; DMEM, Dulbecco's modified Eagle medium; EMEM, Eagle's minimum essential medium; FBS, fetal bovine serum; HTS, high throughput screen; ip, intraperitoneal; MLPCN, Molecular Libraries Probe Production Centers Network; MWI, microwave irradiation; PBS, phosphate buffered saline; tPSA, topological polar surface area; TrD, Trinidad donkey strain; VEEV, Venezuelan equine encephalitis virus

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- (44) The mutant viruses were discovered by passing TC-83 in the presence of the hit compound 1. At passage 4, viruses that were resistant to the compound began to emerge. Studies of six resistant clones showed that their maximum virus titers were \sim 10-fold lower compared to the parental strain. This may indicate that the resistant viruses were less fitted forms compared to the parental strain (see ref 27, this paper). Currently studies to characterize the mutant viruses are in progress.

■ NOTE ADDED AFTER ASAP PUBLICATION

This paper was published ASAP on October 7, 2014. Table 6 was corrected and the revised version was reposted on October 9, 2014.