See discussions, stats, and author profiles for this publication at: https://www.researchgate.net/publication/23767593

ChemInform Abstract: Iridium-Catalyzed Asymmetric Hydrogenation of Cyclic Enamines

| ARTICLE in . | JOURNAL OF | THE AMERICAN | CHEMICAL | SOCIETY · | FEBRUARY 2009 |
|--------------|------------|--------------|-----------------|-----------|---------------|
|--------------|------------|--------------|-----------------|-----------|---------------|

Impact Factor: 12.11 · DOI: 10.1021/ja808358r · Source: PubMed

| CITATIONS | READS |
|-----------|-------|
| 82 | 39 |

4 AUTHORS, INCLUDING:



Jian-Hua Xie

Nankai University

104 PUBLICATIONS 2,512 CITATIONS

SEE PROFILE



Communication

Iridium-Catalyzed Asymmetric Hydrogenation of Cyclic Enamines

Guo-Hua Hou, Jian-Hua Xie, Pu-Cha Yan, and Qi-Lin Zhou

J. Am. Chem. Soc., 2009, 131 (4), 1366-1367 DOI: 10.1021/ja808358 Publication Date (Web): 08 January 2009

Downloaded from http://pubs.acs.org on February 25, 2009

More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Access to high resolution figures
- Links to articles and content related to this article
- · Copyright permission to reproduce figures and/or text from this article

View the Full Text HTML





Published on Web 01/08/2009

Iridium-Catalyzed Asymmetric Hydrogenation of Cyclic Enamines

Guo-Hua Hou, Jian-Hua Xie, Pu-Cha Yan, and Qi-Lin Zhou*

State Key Laboratory and Institute of Elemento-organic Chemistry, Nankai University, Tianjin 300071, China Received October 24, 2008; E-mail: qlzhou@nankai.edu.cn

The asymmetric hydrogenation of enamides, that is, N-acyl enamines, catalyzed by Rh and Ru complexes bearing chiral phosphine ligands is an elegant and efficient method for the enantioselective synthesis of chiral N-acyl secondary amines. ¹ In this hydrogenation reaction, the N-acyl group in the substrates is required to obtain high enantioselectivity by forming a chelate complex with the metal of the catalyst in the transition state.² Therefore, the chiral catalysts that proved to be successful in the asymmetric hydrogenation of enamides cannot be simply applied in the asymmetric hydrogenation of N,N-dialkyl enamines. Actually, the direct preparation of chiral tertiary amines by catalytic asymmetric hydrogenation of the N,N-dialkyl enamines remains a challenge and the reported efficient chiral catalysts are very limited.³ Recently, we demonstrated that the Rh-complex of chiral spiro phosphonite (R)-3 is a highly efficient catalyst for the asymmetric hydrogenation of enamines, (E)-1-(1-pyrrolidinyl)-1,2-diarylethenes, providing the corresponding tertiary amines with up to 99.9% ee.⁴ When we tried to apply the catalyst Rh/(R)-3 for the hydrogenation of cyclic N,N-dialkyl enamines 1a, we obtained very poor enantioselectivity (20% ee). For developing efficient methods for the preparation of optically active cyclic tertiary amines, we investigated iridium catalysts and found that the complex $Ir/(R_a, S, S)$ -4 is a highly efficient catalyst for the hydrogenation of enamines 1 (Scheme 1). In this Communication, we report the first highly enantioselective hydrogenation of cyclic N,N-dialkyl enamines by using chiral iridium complexes of spiro phosphoramidites⁵ and its application in the synthesis of optically active cyclic tertiary amines, including the natural product crispine A.

Scheme 1

Chiral cyclic tertiary amines are essential structural units in natural products and drugs.⁶ The asymmetric hydrogenation of cyclic N,N-dialkyl enamines provides a direct approach to the synthesis of optically active cyclic tertiary amines. During the development of efficient methods for the enantioselective synthesis of chiral alkaloids, we became interested in the asymmetric hydrogenation of cyclic enamines 1. The hydrogenation of 1a was first catalyzed by Ir/(R)-3 complex in THF and the cyclic tertiary amine 2a was isolated in quantitative yield with 71% ee (Table 1, entry 1). Systematic examination on the phosphorus ligands in the reaction showed that the monodentate spiro phosphoramidite ligand (R_a,S,S) -4 was most enantioselective (92% ee, entry 8). Ligands

 (S_a, S, S) - and (S_a, R, R) -Monophos-Pe⁷ also gave modest to good enantioselectivity (70% ee and 88% ee, respectively) (Table 1, entries 5 and 6). Other types of monodentate phosphorus ligands including the ligands SIPHOS, ShiP, and FuP and bidentate diphosphine ligands such as BINAP,8 SDP,9 Me-Duphos,10 Josiphos, 11 and Synphos, 12 showed very low enantioselectivities (entries 2-4, 10-14). The activity of the catalyst $Ir/(R_2,S,S)-4$ was remarkable, it can hydrogenate the enamine 1a under 1 atm of H₂ pressure, affording an even higher enantioselectivity (94% ee, entry 9). The catalyst loading can even be lowered to 0.1 mol % and the enantioselectivity is retained (entry 25). In addition to THF, dioxane and dimethoxyethane (DME) are also suitable solvents for the hydrogenation of enamine 1a (entries 23 and 24). The investigation of the effect of additives in the reaction showed that the addition of I2 is significant for obtaining full conversion and high enantioselectivity (entry 9 vs entries 15-18). The role of I_2 in this reaction is presumably the same as that in the Ir-catalyzed hydrogenation of imines.¹³ The oxidative addition of I₂ to the Ir(I) precursor generates Ir(III) complex, which has a higher activity and enantioselectivity in the hydrogenation of enamines.

The secondary interaction between metal and ligands in the catalysts has been reported in the Ir-catalyzed substitution and Rucatalyzed cyclopropanation using Monophos-Pe ligand. ¹⁴ We have

Table 1. Ir-Catalyzed Asymmetric Hydrogenation of Enamine **1a**, Optimizing Reaction Conditions^a

| entry | ligand | PH ₂ (atm) | solvent | additive | conv (%) ^b | ee (%) ^c |
|--------|----------------------------|-----------------------|------------|----------|--------------------------|------------------------|
| 1 | (R)- 3 | 50 | THF | I_2 | 100 | 71 (S) |
| 2 | (R)-SIPHOS | 50 | THF | I_2 | 100 | 44 (S) |
| 3 | (R)-ShiP | 50 | THF | I_2 | 100 | 18 (S) |
| 4 | (R)-FuP | 50 | THF | I_2 | 100 | 36 (S) |
| 5 | (S_a, S, S) -Monophos-Pe | 50 | THF | I_2 | 100 | 70(S) |
| 6 | (S_a, R, R) -Monophos-Pe | 50 | THF | I_2 | 100 | 88 (R) |
| 7 | $(R_{\rm a}, R, R)$ -4 | 50 | THF | I_2 | 100 | 57 (S) |
| 8 | (R_a, S, S) -4 | 50 | THF | I_2 | 100 | 92 (S) |
| 9 | (R_a, S, S) -4 | 1 | THF | I_2 | 100 | 94 (S) |
| 10 | (R)-SDP | 1 | THF | I_2 | 100 | 31 (R) |
| 11 | (S)-BINAP | 1 | THF | I_2 | 100 | 4 (S) |
| 12 | (R,S)-Josiphos | 1 | THF | I_2 | 100 | 60 (S) |
| 13 | (R,R)-Me-Duphos | 1 | THF | I_2 | 58 | 47 (S) |
| 14 | (R)-Synphos | 1 | THF | I_2 | 35 | 35 (S) |
| 15 | (R_a, S, S) -4 | 1 | THF | none | 5 | 10(S) |
| 16 | (R_a, S, S) -4 | 1 | THF | AcOH | 7 | 26 (S) |
| 17 | (R_a, S, S) -4 | 1 | THF | Et_3N | 4 | 14 (S) |
| 18 | (R_a, S, S) -4 | 1 | THF | Bu_4NI | 52 | 86 (S) |
| 19 | (R_a, S, S) -4 | 1 | CH_2Cl_2 | I_2 | 65 | 85 (S) |
| 20 | (R_a, S, S) -4 | 1 | Toluene | I_2 | 30 | 80 (S) |
| 21 | (R_a, S, S) -4 | 1 | Et_2O | I_2 | 22 | 86 (S) |
| 22 | (R_a, S, S) - 4 | 1 | MeOH | I_2 | 100 | 55 (S) |
| 23 | (R_a, S, S) -4 | 1 | Dioxane | I_2 | 100 | 92 (S) |
| 24 | (R_a, S, S) -4 | 1 | DME | I_2 | 100 | 94 (S) |
| 25^d | (R_a, S, S) -4 | 50 | THF | I_2 | 100 | 94 (S) |

^a Reaction conditions: 0.5 mol % [Ir(COD)Cl]₂, 2.2 mol % ligand, 5 mol % additive, [substrate] = 0.1 M, room temperature (15–20 °C). ^b Determined by GC. ^c Determined by chiral GC using a Supelco alpha-Dex 120 column (30 m × 0.25 mm × 0.25 μm). ^d S/C = 1000.

Table 2. The Asymmetric Hydrogenation of Enamines Catalyzed by $Ir/(R_a, S, S)-4^a$

| entry | R ¹ | R^2 | product | ee (%) ^b |
|----------|------------------------------------|-----------------|-----------|---------------------|
| 1 | C ₆ H ₅ | Me | 2a | 94 (S) |
| 2 | C_6H_5 | Et | 2b | 95 |
| 3 | C_6H_5 | ⁱ Pr | 2c | 96 |
| 4 | $4-MeC_6H_4$ | Me | 2d | 91 |
| 5 | 4-MeOC ₆ H ₄ | Me | 2e | 94 |
| 6 | $4-FC_6H_4$ | Me | 2f | 92 |
| 7 | 4-ClC ₆ H ₄ | Me | 2g | 95 |
| 8 | $4-BrC_6H_4$ | Me | 2h | 97 |
| 9 | $3-MeC_6H_4$ | Me | 2i | 89 |
| 10 | 3-MeOC ₆ H ₄ | Me | 2.j | 93 |
| 11 | $3-FC_6H_4$ | Me | 2k | 92 |
| 12 | 2-MeC ₆ H ₄ | Me | 21 | 87 |
| 13 | 2-ClC ₆ H ₄ | Me | 2m | 82 |
| 14 | 4-MeOC ₆ H ₄ | Et | 2n | 90 |
| 15 | $4-FC_6H_4$ | Et | 20 | 97 |
| 16 | 4-ClC ₆ H ₄ | Et | 2p | 94 |
| 17 | $4-BrC_6H_4$ | Et | 2q | 94 |
| 18^{c} | $^{n}C_{4}H_{9}$ | Me | 2r | 72 |

^a Reaction conditions are the same as those of Table 1, entry 9, yields >90%. b Determined by chiral GC (see Supporting Information). (S_a, R, R) -Monophos-Pe gave 66% ee. Very low ee values were obtained with other monodentate phosphorus ligands listed in Table 1.

tried to detect if there is a secondary interaction between iridium and ligands (R_a, S, S) -4 and (R_a, S, S) -Monophos-Pe. However, ¹H and ³¹P NMR analyses in the beginning or the end of the reaction showed no cyclometallated Ir-complex generated by cyclometalation at the methyl group of phosphoramidite ligand like in the Ircatalyzed substitution reaction. 14a,b

Encouraged by the promising result obtained in the hydrogenation of enamine 1a, a variety of dihydropyrrole substrates 1 were examined using catalyst $Ir/(R_a, S, S)$ -4. The alkyl group R^2 on the nitrogen atom in the substrates has no obvious influence on the reactivity and enantioselectivity of the reaction. The hydrogenations of enamines with N-Me (1a), N-Et (1b), and N-iPr (1c) produced the corresponding tertiary amines with comparable enantiomeric excesses (94-96% ee, Table 2, entries 1-3). The aryl group (\mathbb{R}^1) on the α -position of substrates 1 is necessary for obtaining high enantioselectivity. When the R¹ group was phenyl or a substituted phenyl, the hydrogenated products were isolated with high yields (>90%) with 82-97% ee (entries 1-17). However, the hydrogenation of substrate $1\mathbf{r}$, when R^1 was "Bu, afforded the amine $2\mathbf{r}$ with a lower ee value (72%, entry 18). The electronic property of substituents on the aryl ring of the substrate has very little effect to the enantiomeric excess of the product. The substituent at the ortho-position of the aryl ring of the substrate slightly diminished the enantioselectivity of the reaction. For example, the hydrogenations of 1-methyl-5-o-tolyl-2,3-dihydropyrrole (11) and 1-methyl-5-(o-chlorophenyl)-2,3-dihydropyrrole (1m) yielded the cyclic amines 21 and 2m in 87% ee and 82% ee, respectively (entries 12 and 13).

Figure 1

The catalyst $Ir/(R_a, S, S)$ -4 was also applied to the hydrogenation of enamines having a six-membered ring or a fused ring. In the hydrogenation of 1-methyl-6-phenyl-1,2,3,4-tetrahydropyridine (5), the cyclic tertiary amine 6 was provided in 21% ee. Other available monodentate phosphorus ligands listed in Table 1 were also tested and no more than 20% ee were obtained. In contrast, the hydrogenation of tricyclic enamine 7 yielded the corresponding amine 8 with 82% ee (Figure 1). This result prompted us to utilize this asymmetric hydrogenation reaction to synthesize the isoquinoline alkaloid crispine A,15 which was isolated from Carduus crispus, Linn. (welted thistle) and has a significant cytotoxic activity. 16 The tricyclic enamine 11 was conveniently synthesized from 2-(3,4dimethoxyphenyl)ethamine in 72% yield by a literature method. 15b The asymmetric hydrogenation of enamine 11 was performed under the optimized conditions using the catalyst $Ir/(S_{\omega}R,R)$ -4 to produce crispine A in 97% yield with 90% ee (Scheme 2).

Scheme 2

Acknowledgment. We thank the National Natural Science Foundation of China (Grant No. 20532010, 20572049, 20721062), the Major Basic Research Development Program (Grant No. 2006CB806106), and the "111" project (B06005) of the Ministry of Education of China for financial support.

Supporting Information Available: Experimental procedures, the characterizations of substrates and products, and the analysis of ee values of hydrogenation products. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) For reviews, see: Tang, W.-J.; Zhang, X.-M. Chem. Rev. 2003, 103, 3029. For reviews, see: Halpern, J. In Asymmetric Synthesis; Morrison, J. D.,
- Ed.; Academic Press: Orlando, FL, 1985; p 41. (3) (a) Lee, N. E.; Buchwald, S. L. J. Am. Chem. Soc. 1994, 116, 5985. (b) Tararov, V. I.; Kadyrov, R.; Riermeier, T. H.; Holz, J.; Börner, A. Tetrahedron Lett. 2000, 41, 2351. For asymmetric hydrogenation of unsubstituted enamines and *N*-arylenamines, see: (c) Hsiao, Y.; Rivera, N. R.; Rosner, T.; Krska, S. W.; Njolito, E.; Wang, F.; Sun, Y.; Armstrong, J. D., III; Grabowski, E. J. J.; Tillyer, R. D.; Spindler, F.; Malan, C. J. Am. Chem. Soc. 2004, 126, 9918. (d) Dai, Q.; Yang, W.; Zhang, X. Org. Lett. **2005**, 7, 5343
- (4) Hou, G.-H.; Xie, J.-H.; Wang, L.-X.; Zhou, Q.-L. J. Am. Chem. Soc. 2006, 128, 11774.
- (5) For Ir-catalyzed asymmetric hydrogenation of enamides, see: (a) Bunlaksananusorn, T.; Polborn, K.; Knochel, P. Angew. Chem., Int. Ed. 2003, 42, 3941. (b) Maire, P.; Deblon, S.; Breher, F.; Geier, J.; Böhler, C.; Rüegger, H.; Schönberg, H.; Grützmacher, H. Chem. Eur. J. 2004, 10, 4198. (c) Giacomina, F.; Meetsma, A.; Panella, L.; Lefort, L.; de Vries, A.; de Vries,
- J. G. Angew. Chem., Int. Ed. 2007, 46, 1497.

 (6) (a) Elliott, R. L.; Ryther, K. B.; Anderson, D. J.; Raszkiewicz, J. L.; Campbell, J. E.; Sullivan, J. P.; Garvey, D. S. Bioorg. Med. Chem. Lett. 1995, 5, 991. (b) Dunsmore, C. J.; Carr, R.; Fleming, T.; Tumer, N. J. J. Am. Chem. Soc. 2006, 128, 2224
- (7) Monophos-Pe = O, O'-(1,1'-dinaphthyl-2,2'-diyl)-N, N'-di(1-phenylethyl)-

- (8) BINAP = 2,2'-bis(diphenylphosphino)-1,1'-binaphthyl.
 (9) SDP = 7,7'-bis(diphenylphosphino)-1,1'-spirobiindane.
 (10) Me-Duphos = 1,2-bis(2,5-dimethylphospholano)benzene.
- (11) Josiphos = 1-[2-(diphenylphosphino)ferrocenyllethyldicyclohexylphosphine. (12) Synphos = (5,6),-(5',6')-bis(ethylenedioxy)-2,2'-bis(diphenylphosphino)-1,1'-biphenyl.
- (13) Xiao, D.; Zhang, X. Angew. Chem., Int. Ed. 2001, 40, 3425.
- (a) Kiener, C. A.; Shu, C.; Incarvito, C.; Hartwig, J. F. J. Am. Chem. Soc. **2003**, *125*, 14272. (b) Welter, C.; Dahnz, A.; Brunner, B.; Streiff, S.; Dübon, P.; Helmchen, G. *Org. Lett.* **2005**, *7*, 1239. (c) Mikhel, I. S.; Rüegger, H.; Butti, P.; Camponovo, F.; Huber, D.; Mezzetti, A. Organometallics 2008, 27, 2937.
- (15) (a) Wu, T. R.; Chong, J. M. J. Am. Chem. Soc. 2006, 128, 9646. (b) Szawkalo, J.; Czarnocki, S. J.; Zawadzka, A.; Wojtasiewicz, K.; Leniewski, A.; Drabowicz, J.; Czarnocki, Z. Tetrahedron: Asymmetry 2007, 18, 406.
- (a) Zhang, Q.; Tu, G.; Zhao, Y.; Cheng, T. Tetrahedron 2002, 58, 6795. (b) Knölker, H.-J.; Agarwal, S. Tetrahedron Lett. 2005, 46, 1173.

JA808358R