

Adenosine-1,3-diazaphenoxazine Derivative for Selective Base Pair Formation with 8-Oxo-2'-deoxyguanosine in DNA [Journal of the American Chemical Society 2011, 133, 7272–7275 DOI: 10.1021/ja200327u]. Yosuke Taniguchi, Ryota Kawaguchi, and Shigeki Sasaki\*

Page 7273. The base pair combinations of G-C, A-T, oxo-G-C, and oxo-G-A in ODN1 and ODN2 in the original Table 1 should be reversed as shown below. The corresponding melting temperature in the text should also be corrected according to the data for these base pairs.

Table 1. Thermodynamic Parameters of the Duplex Formed between ODN1 and  ${\rm ODN2}^a$ 

X in	Y in	$T_{\mathrm{m}}$	$\Delta H^{\circ}$	$\Delta S^{\circ}$	$\Delta G_{310 ext{K}}^{\circ}$
ODN1	ODN2	(°C)	(kcal/mol)	$(cal\ K^{-1}\ mol^{-1})$	(kcal/mol)
С	G	44.1	$-100.7 \pm 6$	$-288 \pm 20$	$-11.3 \pm 0.2$
G	C	48.2	$-114.7 \pm 18$	$-328 \pm 55$	$-13.1\pm0.8$
T	A	40.9	$-90.3 \pm 7$	$-259\pm22$	$-10.1\pm0.2$
A	T	41.9	$-92.9\pm10$	$-266 \pm 38$	$-10.4 \pm 0.3$
G	A	33.1	$-94.6 \pm 4$	$-280\pm13$	$-7.8 \pm 0.04$
A	G	35.4	$-71.1\pm 8$	$-201\pm25$	$-8.7\pm0.2$
oxo-G	C	44.3	$-98.3\pm8$	$-281\pm23$	$-11.3\pm0.2$
oxo-G	A	39.8	$-91.4 \pm 3$	$-263 \pm 9$	$-9.7 \pm 0.05$
C	oxo-G	42.3	$-96.8 \pm 8$	$-278\pm25$	$-10.6\pm0.2$
A	oxo-G	39.2	$-96.1 \pm 7$	$-279\pm21$	$-9.6 \pm 0.1$

<sup>a</sup> The combination of dC and 8-oxo-dG showed a  $T_{\rm m}$  of 42.3 °C, which is 1.8 °C lower than that of the corresponding natural CG pair (44.1 °C). This is probably because Watson—Crick base-pair formation involving dC requires the unfavorable anti conformation of 8-oxo-dG. On the other hand, the dA-8-oxo-dG combination provided a  $T_{\rm m}$  value similar to that of the dA-dT base pair ( $T_{\rm m}$  = 39.2 vs 41.9 °C) and a higher  $T_{\rm m}$  value than that of the dA-dG pair ( $T_{\rm m}$  = 39.2 vs 35.4 °C), because dA forms the base pair with the favorable syn conformation of 8-oxo-dG.

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