



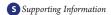
COMMUNICATION

pubs.acs.org/JACS

# A Programmed Polymer Folding: *Click* and *Clip* Construction of Doubly *Fused* Tricyclic and Triply *Fused* Tetracyclic Polymer Topologies

Naoto Sugai, Hiroyuki Heguri, Takuya Yamamoto, and Yasuyuki Tezuka\*

Department of Organic and Polymeric Materials, Tokyo Institute of Technology, O-okayama, Meguro-ku, Tokyo 152-8552, Japan



ABSTRACT: A tandem alkyne—azide addition, i.e., click, and an olefin metathesis condensation, i.e., clip, reactions in conjunction with an electrostatic self-assembly and covalent fixation (ESA-CF) process, have been demonstrated as effective means to produce constructions of programmed folding of polymers having doubly fused tricyclic and triply fused tetracyclic topologies. Thus, a series of cyclic poly-(tetrahydrofuran), poly(THF), precursors having an allyloxy group and an alkyne group (Ia), an allyloxy group and an azide group (Ib), and two alkyne groups (Ic) at the opposite positions was prepared by means of the ESA-CF method. The subsequent click reactions of Ia with a linear telechelic poly(THF) precursor having azide end groups (Id) and of Ib with Ic afforded a bridged dicyclic polymer (IIa) and a tandem spiro tricyclic precursor (IIb), respectively, both having two allyloxy groups at the opposite positions of the ring units. Finally, the intramolecular metathesis condensation reaction of IIa and of IIb in the presence of a Grubbs catalyst was performed to construct effectively a doubly fused tricyclic and a triply fused tetracyclic polymer topologies (III and IV), respectively.

Programmed folding of polymer molecules plays crucial roles in diverse events in biopolymer systems such as DNA packaging and protein three-dimensional (3D) structure formation. Moreover, a set of cyclic proteins, cyclotides, is increasingly gaining attention by their extraordinary stability and bioactivity ascribed to their fused multicyclic structures formed through the covalent folding by the intramolecular S-S linkage with cysteine residues.<sup>2</sup> On the other hand, the effective and programmed polymer folding by synthetic polymers into even primitive cyclic forms has been a formidable synthetic challenge until recently.<sup>3</sup> Remarkable progress has now been ongoing to produce a wide variety of *single* cyclic polymers based on (1) newly developed end-to-end prepolymer folding processes<sup>4</sup> and (2) alternative ring-expansion polymerization. By making use of newly prepared cyclic polymers having prescribed chemical structures, a wide variety of topology effects by cyclic polymers have now been unequivocally demonstrated.

For a synthetic approach to *multicyclic* polymer topologies consisting of the three subclasses of *fused, spiro,* and *bridged* forms we have developed an *electrostatic self-assembly and covalent fixation* (ESA-CF) protocol, in which linear and star telechelic precursors having cyclic ammonium salt groups carrying plurifunctional carboxylate counteranions were employed to form polymeric self-assemblies as key intermediates. The three

forms of dicyclic constructions, i.e.,  $\theta$ -shaped (fused), 8-shaped (spiro), and manacle-shaped (bridged), as well as a trefoil (spiro tricyclic) construction have been effectively produced through the covalent conversion of the electrostatic polymer self-assemblies. A variety of unprecedented spiro- and bridged-type tricyclic and tetracyclic polymer topologies have been constructed through an alkyne—azide click reaction by employing tailored single cyclic and dicyclic polymer precursors (kyklo-telechelics) obtainable also by the ESA-CF protocol, affording prepolymers having atom ring sizes as large as 300 members and having alkyne and/or azide groups at the prescribed positions in their cyclic structures.  $^{10}$ 

A class of *fused* multicyclic polymer topologies, in contrast to their *spiro* and *bridged* counterparts, is considered particularly intriguing in the context of programmed polymer folding. However, the production of *fused* multicyclic constructions with synthetic polymers has so far been limited to a few, including doubly *fused* tricycle topologies ( $\delta$ -graph) through the metathesis condensation with an 8-shaped precursor having allyl groups at the opposite positions of the two ring units obtainable by the ESA-CF protocol. To extend the current frontier of synthetic polymer chemistry, the construction of topologically significant polymers such as an  $\alpha$ -graph (doubly *fused* tricyclic), a  $K_{3,3}$  graph (triply *fused* tetracyclic), and a prisman graph (triply *fused* tetracyclic) has still been an ongoing challenge (Scheme 1).

Herein we report a tandem alkyne—azide addition, i.e., *click* reaction, and an olefin metathesis, i.e., *clip* reaction, in conjunction with the ESA-CF process as an effective means for producing constructions of programmed folding of polymers having well-defined and unprecedented *fused* multicyclic topologies, namely  $\gamma$ -graph (doubly *fused* tricyclic) and an unfolded tetrahedron graph (triply *fused* tetracyclic) constructions. <sup>13</sup> Notably, the combination of the *click* and *clip* reactions has also been successfully applied to produce a *catenated* multicyclic polymer of charm bracelet and daisy-chain architectures. <sup>14</sup>

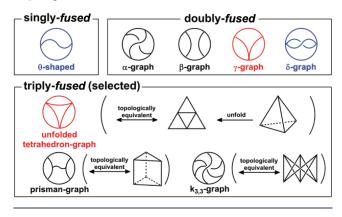
A series of single cyclic prepolymers (*kyklo*-telechelics) of symmetric and asymmetric functionalities at the opposite positions of the ring unit were prepared by the ESA-CF process by using a set of *N*-phenylpyrrolidinium-terminated poly(THF) prepolymers having an alkene group (1a) or an alkyne group (1b) at the center position of the chain, and carrying dicarboxylate counteranions having an alkyne (2a) or an azide (2b) group (Scheme 2). Thus, cyclic poly(THF) precursors having an allyloxy group and an alkyne group (Ia) and having an allyloxy group and an azide group (Ib), were synthesized, together with one having two alkyne groups (Ic). In addition, a linear poly(THF) having

Received: October 5, 2011

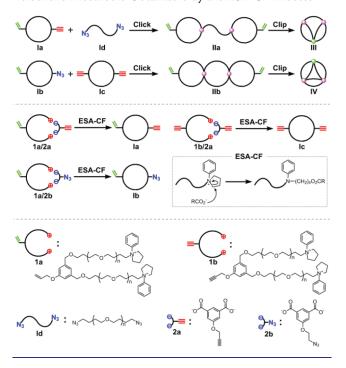
Published: November 10, 2011



Scheme 1. Graph Presentation of Fused Multicyclic Polymer Topologies



Scheme 2. Click and Clip Construction of Doubly and Triply Fused Multicyclic Polymer Topologies using a Series of Kyklo-Telechelic Precursors Obtainable by the ESA-CF Process



azide end groups (Id) was prepared by simply terminating a living polymerization of THF with tetrabutylammonium azide. (See Supporting Information (SI) for the preparation details.)

The subsequent click reaction <sup>10</sup> was performed in the presence of copper sulfate and sodium ascorbate by employing complementary sets of polymer precursors, namely cyclic (Ia)/linear (Id) and cyclic (Ib)/cyclic (Ic) combinations with a slight excess (5–10%) of the former precursors to produce a *bridged* dicyclic (manacle) polymer precursor (IIa) and a tandem *spiro* tricyclic counterpart (IIb), respectively, both having two allyloxy groups at the opposite positions of the ring units. These products, IIa and IIb, were finally isolated by preparative SEC in 63% and in 64% yield, respectively. The progress of the reaction and the subsequent purification process were monitored by SEC (Figure 1). In both systems, the SEC traces of both click

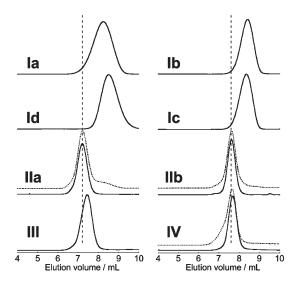
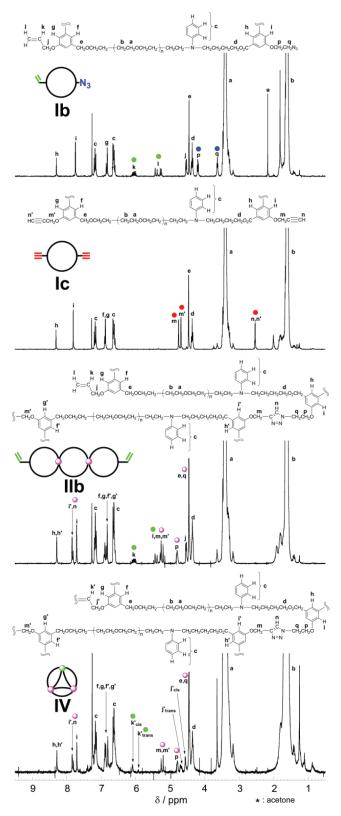


Figure 1. SEC traces of prepolymers, Ia, Ib, Ic, and Id, the click products, IIa and IIb, and the *fused*-type multicyclic polymer products, III and IV. Broken lines and solid lines show those obtained before and after fractionation, respectively. (THF was used as eluent at late flow 1.0 mL/min, with TSK G3000HXL for Ia, Id, IIa, and III, with TSK G4000HXL for Ib, Ic, IIb, and IV).

products shifted toward the higher molecular weight region with only traces at the lower molecular weight region due to the precursors charged slightly in excess (Figure 1, IIa and IIb, broken lines). After the purification by preparative SEC fractionation, the precursor fraction was completely eliminated as seen in Figure 1 (IIa and IIb, solid lines). The extent of the contraction of the 3D size for dicyclic IIa and tricyclic IIb was estimated by their  $M_{\rm p}({\rm SEC})/M_{\rm n}({\rm NMR})$  ratios of 0.82 and 0.63, respectively, with reasonable agreements with the previously reported values for the relevant polymers.  $^{10,11}$ 

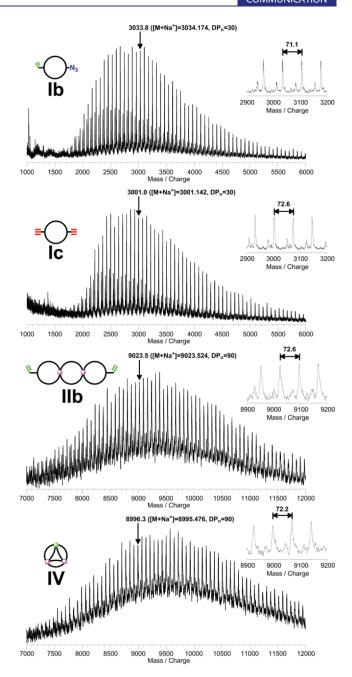
<sup>1</sup>H NMR spectra of **IIa** and **IIb** together with their polymer precursors (**Ia/Id** and **Ib/Ic**, respectively) are shown in Figure S1 in SI and in Figure 2, respectively. It is shown that the signals for the ethynyl protons (2.53 ppm in **Ia** and 2.54 and 2.55 ppm in **Ic**) and those of the azidomethylene protons (3.30 ppm in **Id** and 3.64 ppm in **Ib**) were replaced by the triazole proton signals at 7.65 ppm in **IIa** and 7.79−7.88 ppm in **IIb**, respectively, to confirm the effective click reaction. The signals for the allyloxy units (5.26−5.44 and 5.99−6.12 ppm in **IIa** and 5.23−5.47 and 5.96−6.15 ppm in **IIb**) are visibly intact during the click process. By comparison of the signal intensities of the main-chain protons with the linking-group protons,  $M_n$ 's (NMR) were determined to be 10 kDa and 11 kDa for **IIa** and **IIb**, respectively.

MALDI-TOF mass spectra of IIa and IIb together with those of their polymer precursors (Ia/Id and Ib/Ic, respectively) are shown in Figure S3 in SI and in Figure 3, respectively. A uniform series of peaks with an interval of 72 mass units (corresponding to the repeating THF units) were observed for all samples; moreover, each peak exactly matched the molar mass calculated from the chemical structure of the product. Thus, for IIa, the peak at m/z=7926.4, which is assumed to be the adduct with Na<sup>+</sup>, corresponds to IIa possessing the expected chemical structure with the degree of polymerization, DP<sub>n</sub>, of 85; (C<sub>4</sub>H<sub>8</sub>O) × 85 + C<sub>104</sub>H<sub>128</sub>N<sub>10</sub>O<sub>16</sub>, plus Na<sup>+</sup> = 7926.332. For IIb, the peak at m/z=9023.5, assumed to be the adduct with Na<sup>+</sup>, corresponds to IIb possessing the expected chemical structure with a DP<sub>n</sub> of 90; (C<sub>4</sub>H<sub>8</sub>O) × 90 + C<sub>148</sub>H<sub>180</sub>N<sub>12</sub>O<sub>24</sub>, plus Na<sup>+</sup> = 9023.524.



**Figure 2.** <sup>1</sup>H NMR (300 MHz) spectra of the *kyklo*-telechelic prepolymers, **Ib** and **Ic**, the click product, **IIb**, and the triply *fused* tetracyclic polymer product, **IV**. (CDCl<sub>3</sub>, 40 °C. The samples **IIb** and **IV** after the purification by means of the fractionation with preparative SEC.)

The subsequent intramolecular olefin metathesis reaction of **IIa** and **IIb**, possessing *bridged* dicyclic and tandem *spiro* tricyclic



**Figure 3.** MALDI-TOF mass spectra of *kyklo*-telechelic prepolymers, **Ib** and **Ic**, the click product, **IIb**, and the triply *fused* tetracyclic polymer product, **IV**. (Linear mode, matrix: dithranol with sodium trifluoroacetate.  $DP_n$  denotes the number of monomer units in the product.)

topologies and having allyloxy groups at the opposite positions of the ring units, were conducted under dilution (0.2 g/L) by repeated addition of a first-generation Grubbs catalyst into the reaction solution. SEC showed that the crude reaction products contained a noticeable portion of the intermolecular condensation products as seen by broken lines in SEC (Figure 1, IV, broken line). Hence, the product IV was first recovered by a column chromatography with silica gel, and finally isolated by the purification with preparative SEC fractionation technique (Figure 1, IV, solid line). The isolated yields of III and IV were 28% and 55%, respectively.

<sup>1</sup>H NMR spectra of the isolated III and IV are compared with their precursors IIa and IIb in Figure S2 in SI and in Figure 2,

respectively. It is shown that the signals for the allyloxy units in IIa and in IIb are completely replaced by those of the inner olefinic units at 6.07-6.12 (cis) and 5.90-5.95 (trans) ppm in III and at 6.04-6.15 (cis) and 5.90-5.96 (trans) ppm in IV. It is indicative of the metathesis condensation reaction proceeding effectively in both cases even under applied dilution.

MALDI-TOF mass spectra of III and IV are compared with their precursors IIa and IIb in Figure S3 in SI and in Figure 3, respectively, to confirm the successful construction of a doubly fused tricyclic and a triply fused tetracyclic polymer topologies. Those of III and IV showed a uniform series of peaks as in IIa and IIb. Thus, for III the peak at m/z = 7898.3, which is assumed to be the adduct with Na<sup>+</sup>, corresponds to III possessing the expected chemical structure with a DP<sub>n</sub> of 85;  $(C_4H_8O) \times 85 +$  $C_{102}H_{124}N_{10}O_{16}$ , plus Na<sup>+</sup> equals 7898.278. For **IV**, the peak at m/z = 8996.3, which is assumed to be the adduct with Na<sup>+</sup>, corresponds to IV possessing the expected chemical structure with a DP<sub>n</sub> of 90;  $(C_4H_8O) \times 90 + C_{146}H_{176}N_{12}O_{24}$ , plus Na<sup>+</sup> equals 8995.476. Since III and IV are produced from IIa and IIb by the elimination of an ethylene molecule, their molecular weights differ by 28 mass units. This was confirmed by the two sets of mass spectra as shown in Figure S3 in SI and in Figure 3.

The extent of the contraction of the 3D size for tricyclic III from dicyclic IIa and for tetracyclic IV from tricyclic IIb was estimated after the SEC fractionation by their  $M_{\rm p}({\rm SEC})/M_{\rm n}$ -(NMR) ratios, which showed noticeable reduction from 0.82 to 0.60 for III and from 0.63 to 0.52 for IV. Accordingly, the programmed polymer folding could produce unusually compact polymer conformation in their 3D structures.

In summary, a tandem alkyne—azide addition (i.e., a click reaction) and an olefin metathesis condensation (i.e., a clip reaction) in conjunction with an electrostatic self-assembly and covalent fixation (ESA-CF) process have been demonstrated upon the combination of NMR, SEC, and MALDI-TOF techniques as an effective means for producing constructions of programmed folding of polymers such as doubly fused tricyclic and triply fused tetracyclic polymer topologies.

#### ASSOCIATED CONTENT

Supporting Information. Experimental details for the synthesis, <sup>1</sup>H NMR spectra, MALDI-TOF MS spectra, and IR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

## **■ AUTHOR INFORMATION**

Corresponding Author ytezuka@o.cc.titech.ac.jp

# **■** ACKNOWLEDGMENT

We thank the JSPS Research Fellowship for the Young Scientists (N.S.); the Challenging Research Award, Tokyo Institute of Technology (T.Y.); Global COE Program (Education and Research Center for Material Innovation), MEXT, Japan (T.Y.); The Japan Securities Scholarship Foundation (T.Y.); The Kurata Memorial Hitachi Science and Technology Foundation (T.Y.); and the Yazaki Memorial Foundation for Science and Technology (T.Y.) for financial support. This study was also supported in part by grants from the Ministry of Education, Science and Culture, Japan through the Japan Society of Promotion of Sciences (23350050(Y.T.); 23685022(T.Y.); and 23106709(T.Y.)).

### ■ REFERENCES

- (1) For DNA packaging, see: (a) Osada, K.; Oshima, H.; Kobayashi, D.; Doi, M.; Enoki, M.; Yamasaki, Y.; Kataoka, K. *J. Am. Chem. Soc.* **2010**, 132, 12343–12348. For protein 3D structures:(b) Dobson, C. M. *Nature* **2003**, 426, 884–890. (c) Sosnick, T. R.; Hinshaw, J. R. *Science* **2011**, 334, 464–465.
- (2) (a) Craik, D. J. Science **2006**, 311, 1563–1564. (b) Craik, D. J. Trends Plant Sci. **2009**, 14, 328–335.
- (3) For recent developments in polymer folding with synthetic polymers, see: (a) Perrier, S. Nature Chem. 2011, 3, 194–196. (b) Schmidt, B. V. K. J.; Fechler, N.; Falkenhagen, J.; Lutz, J.-F. Nature Chem. 2011, 3, 234–238. For polymer folding into helix structures, see: (c) Hill, D. J.; Mio, M. J.; Prince, R. B.; Hughes, T. S.; Moore, J. S. Chem. Rev. 2001, 101, 3893–4012. (d) Yashima, E.; Maeda, K.; Iida, H.; Furusho, Y.; Nagai, K. Chem. Rev. 2009, 109, 6102–6211. (e) Gan, Q.; Ferrand, Y.; Bao, C.; Kauffmann, B.; Grélard, A.; Jiang, H.; Huc, I. Science 2011, 331, 1172–1175. For polymer folding by intramolecular cross-linking process, see: (f) Harth, E.; Van Horn, B.; Lee, V. Y.; Germack, D. S.; Gonzales, C. P.; Miller, R. D.; Hawker, C. J. J. Am. Chem. Soc. 2002, 124, 8653–8660. (g) Foster, E. J.; Berda, E. B.; Meijer, E. W. J. Am. Chem. Soc. 2009, 131, 6964–6966. (h) Zhang, K.; Lackey, M. A.; Cui, J.; Tew, G. N. J. Am. Chem. Soc. 2011, 133, 4140–4148.
- (4) (a) Yamamoto, T.; Tezuka, Y. In Complex Macromolecular Architectures: Synthesis, Characterization, and Self-Assembly; Hadjichristidis, N., Hirao, A., Tezuka, Y., Du Prez, F., Eds.; Wiley: Singapore, 2011; pp 3–19. (b) Laurent, B. A.; Grayson, S. M. Chem. Soc. Rev. 2009, 38, 2202–2213. (c) Yamamoto, T.; Tezuka, Y. Eur. Polym. J. 2011, 47, 535–541.
- (5) (a) Xia, Y.; Boydston, A. J.; Grubbs, R. H. Angew. Chem., Int. Ed. **2011**, 50, 5882–5885. (b) Shin, E. J.; Brown, H. A.; Gonzalez, S.; Jeong, W.; Hedrick, J. L.; Waymouth, R. M. Angew. Chem., Int. Ed. **2011**, 50, 6388–6391.
- (6) For recent developments in functional cyclic polymers, see: (a) Endo, K. Adv. Polym. Sci. 2008, 217, 121–183. (b) Yamamoto, T.; Tezuka, Y. Polym. Chem. 2011, 2, 1930–1941. (c) Deffieux, A.; Schappacher, M. Cell. Mol. Life Sci. 2009, 66, 2599–2602.
  - (7) Tezuka, Y.; Oike, H. J. Am. Chem. Soc. 2001, 123, 11570–11576.
- (8) (a) Oike, H.; Imaizumi, H.; Mouri, T.; Yoshioka, Y.; Uchibori, A.; Tezuka, Y. J. Am. Chem. Soc. **2000**, 122, 9592–9599.
- (9) Tezuka, Y.; Tsuchitani, A.; Yoshioka, Y.; Oike, H. Macromolecules 2003, 36, 65–70.
- (10) Sugai, N.; Heguri, H.; Ohta, K.; Meng, Q.; Yamamoto, T.; Tezuka, Y. J. Am. Chem. Soc. **2010**, 132, 14790–14802.
- (11) Tezuka, Y.; Fujiyama, K. J. Am. Chem. Soc. 2005, 127, 6266-6270.
- (12) For the K<sub>3,3</sub> graph, see: Chen, C.-T.; Gantzel, P.; Siegel, J. S.; Baldridge, K. K.; English, R. B.; Ho, D. M. *Angew. Chem., Int. Ed. Engl.* **1995**, 34, 2657–2660.
- (13) The doubly fused tricyclic (III) and triply fused tetracyclic (IV) topologies constructed in this study are expressed as  $\mathrm{III}_5(0,3)[0^{a,b}-0^a-0^b]$  and  $\mathrm{IV}_6(0,3)[0^{a,c}-0^{a,b}-0^{b,c}]$ , respectively, according to the systematic notation (ref 7).
- (14) (a) Clark, P. G.; Guidry, E. N.; Chan, W. Y.; Steinmetz, W. E.; Grubbs, R. H. *J. Am. Chem. Soc.* **2010**, *132*, 3405–3412. (b) Clark, P. G.; Day, M. W.; Grubbs, R. H. *J. Am. Chem. Soc.* **2009**, *131*, 13631–13633.
- (15) To optimize the process, four Ru catalysts, i.e., first- and second-generation Grubbs catalysts and first- and second-generation Hoveyda—Grubbs catalysts, were examined by using linear telechelic poly(THF)s, having two allyloxy or two butenoxy end groups. As a result, Hoveyda—Grubbs catalysts were not effective for allyloxy and butenoxy groups. However, both first- and second-generation Grubbs catalysts were reactive with allyloxy groups, rather than with butenoxy counterparts. Moreover, with a first-generation Grubbs catalyst, side reactions were suppressed better than with a second-generation Grubbs catalyst.