

# Additions and Corrections

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**M.-L. Hung, M. L. McKee, and D. M. Stanbury\*:** Large Reorganizational Energies in Electron Transfer: Oxidation of Hydroxylamine by Hexachloroiridate(IV).

Pages 5108–5112: In calculating  $E^\circ$  for the  $\text{NH}_2\text{OH}^+/\text{NH}_2\text{OH}$  redox couple (page 5111) an incorrect value for  $\Delta_f G^\circ$  for  $\text{NH}_2\text{OH}(\text{aq})$  was used because of a sign error in the reference for that quantity (ref 39, *Standard Potentials*). The correct value for  $\Delta_f G^\circ$  should be  $-23.4 \text{ kJ mol}^{-1}$ , as is evident from the original source (Latimer's *Oxidation Potentials*) and from the corresponding values for  $\Delta_f H^\circ$  and  $S^\circ$ . Correcting for this sign error leads to  $E^\circ = 0.91 \text{ V}$  for the  $\text{NH}_2\text{OH}^+/\text{NH}_2\text{OH}$  redox couple (a 0.5 V change). The corresponding value for  $K_1$  is corrected to 0.6. This  $10^8$ -fold reduction in  $K_1$  leads to an equivalent increase in the derived value for  $k_{11}$ , the self-exchange rate constant for the  $\text{NH}_2\text{OH}^+/\text{NH}_2\text{OH}$  redox couple. The derived value for  $\lambda_{11}$  should be reduced correspondingly. Full discussion of the significance of these corrections is reserved for a future paper. We thank Sergei Lyman (Brookhaven National Labs) for pointing out this error.

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