

Adiabatic and Non-adiabatic Concerted Proton—Electron Transfers. Temperature Effects in the Oxidation of Intramolecularly Hydrogen-Bonded Phenols [J. Am. Chem. Soc. 2007, 129, 9953–9963]. Cyrille Costentin, Marc Robert, and Jean-Michel Savéant\*

Page 9960. The standard free energy, enthalpy, and entropy of the reaction have been introduced in an unusual and confusing manner through

$$\Delta G^{0} = F(E_{A^{\bullet+}/A}^{0} - E_{2^{\bullet+}/2}^{0}) = \Delta H^{0} - T\Delta S^{0}$$

With the standard definitions,

$$\Delta G^0 = -F(E^0_{\Delta \bullet + I/\Delta} - E^0_{2 \bullet + I/2}) = \Delta H^0 - T\Delta S^0$$

The values found for  $\Delta H^0$  and  $\Delta S^0$  are consequently opposite to those originally reported:

$$\Delta H^0 = 0.103 \pm 0.059 \,\text{eV}$$

$$\Delta S^0 = 0.418 \pm 0.21 \text{ meV/K}$$

and eq 25 becomes

$$\ln\!\left(\frac{k}{\sqrt{T}}\right) = \ln\!\left(N_{\rm A}\chi d^2\sqrt{\frac{8\pi R}{M}}\right) + \frac{\Delta S^0}{2R} - \frac{\lambda/4 + \Delta H^0/2 + \Delta Z{\rm PE}}{RT} \quad (25)$$

With this double correction, the ensuing values of  $\lambda$  and  $\chi$  remain unchanged.

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**Polyvalent** Oligonucleotide Gold **Nanoparticle** Conjugates as Delivery Vehicles for Platinum(IV) Warheads [J. Am. Chem. Soc. 2009, 131, 14652-14653]. Shanta Dhar, Weston L. Daniel, David Α. Giljohann, Chad Α. Mirkin.\* and Stephen J. Lippard\*

Page 14653. The legend for Figure 2 should read as follows:

*Figure 2.* Cytotoxicity profiles of Pt-DNA-Au NP (red circles), cisplatin (black squares), and 1 (green triangles) in A549 cells and Pt-DNA-Au NP (black squares), cisplatin (red circles), and 1 (green triangles) with U2OS, HeLa, and PC3 cells.

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Noncovalent Cell Surface Engineering with Cationic Graft Copolymers [*J. Am. Chem. Soc.* **2009**, *131*, 18228–18229]. John T. Wilson, Venkata R. Krishnamurthy, Wanxing Cui, Zheng Qu, and Elliot L. Chaikof\*

A paper by Geert-Jans Boons and co-workers, which describes the synthesis and characterization of the cyclooctyne employed in this work, should have been cited in the main body of the text.

## **Literature Cited**

 Ning, X.; Guo, J.; Wolfert, M. A.; Boons, G.-J. Angew. Chem., Int. Ed. 2008, 47, 2253–2255.

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