Preparation of Novel Haloazide Equivalents by Iodine(III)-Promoted Oxidation of Halide Anions

Andreas Kirschning,* Md. Abul Hashem, † Holger Monenschein, Lars Rose, and Kai-Uwe Schöning

Institut für Organische Chemie der Technischen Universität Clausthal, Leibnizstrasse 6, 38678 Clausthal-Zellerfeld, Germany

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Cohalogenation of alkenes constitutes one of the most important classes of reaction used to form a carbon heteroatom bond in a regio-, chemo-, and stereoselective manner. Since the pioneering work of Hassner considerable attention has been given to the haloazidation of the alkenic double bond by using bromine azide **4a** or iodine azide **4b** as active reagent. This method constitutes a very useful procedure for introducing a nitrogen functionality into a carbon skeleton, leading to vinyl azides, amines, and heterocycles, particularly aziridines.

In continuation of studies devoted to ligand transfer reactions from iodine(III) onto halide ions, ⁸ we investigated the use of the azide group as a mobile ligand. This method would create haloazide-like species under much milder conditions, namely, in an organic solvent, than commonly applied. A two-phase system by the interaction of Br_2 or NBS with NaN_3 in the presence of $acid^{7.9}$ is often required for the preparation of bromine azide. Alternatively, the reagent system $NBS/TMSN_3$ in DME/H_2O has been developed. 10 Iodine azide has been generated from sodium azide and iodine chloride in polar solvents. 11 However, as a result of its explosive character, its use has often been hampered.

Thus, (diacetoxyiodo)benzene (1) was reacted with tetraethylammonium bromide (2a) in dichloromethane

 $^\dagger Present$ address: Department of Chemistry, Jahangivnagar University, Savar, Dhakar 1342, Bangladesh.

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Scheme 1

$$\begin{array}{c} \text{Et}_{4}\text{NX} \\ \text{PhI}(\text{OAc})_{2} & \xrightarrow{\textbf{2a},\textbf{b}} \left[\text{Et}_{4}\text{NX}(\text{OAc})_{2} \right] & \text{X-N}_{3} \\ \textbf{1} & \text{3a,b} & \text{4a,b} \end{array}$$

$$\begin{array}{c} \text{TMSN}_{3} \\ (2 \text{ equ.}) \downarrow & \text{Et}_{4}\text{NX} & & \text{TMSN}_{3} \\ & & & & \text{-25°C} & \text{R}^{1} \\ & & & & & \text{R}^{2} \end{array}$$

$$\left[\text{PhI}(\text{N}_{3})_{2} \right] & \xrightarrow{\textbf{2a},\textbf{b}} \left[\text{Et}_{4}\text{NX}(\text{N}_{3})_{2} \right] & \xrightarrow{\textbf{R}^{2}} \text{N}_{3} & \text{X}^{2} \end{array}$$

2a - 4a, 6a: X= Br; 2b - 4b, 6b: X= I

Table 1

Table 1				
	Halide X	Ratio a:I	Ratio a:b:ca	
			X (X_{X}
7	Br I	<i>rac</i> -12a,b ^c <i>rac</i> -13a,b ^d	>20:1 15:1	92 62
			$\sum_{\underline{x}} X$	N ₃
8	Br I	<i>rac-</i> 14a,b <i>rac-</i> 15a,b	> 20:1 2.5:1	67 52
BnO OBn BnO	BnO-	OBn BnO N ₃	OBn B	OBn X N ₃
9	Br I	16a,b,c 17a,b,c ^e	1.3:1.7:1 <1:10:0	74 49
Et ₃ SiO OR		$\begin{array}{c} \text{Et}_3\text{SiO} \\ \text{Et}_3\text{SiO} \\ \end{array} \begin{array}{c} \text{OR} \\ \text{Br} \\ N_3 \end{array}$	Et ₃ SiO	OR ON ON
10 (R=SiEt ₃) 11 (R=SiMe ₂ tBu)	Br Br	18a,b 19a,b	1:1.9 1:1.4	73 70

 $[^]a$ Determined from the $^1{\rm H}$ NMR spectra of the crude product. $^b{\rm Total}$ yields of chromatographically purified products. $^c{\rm See}$ ref 9a. $^d{\rm See}$ ref 2. $^c{\rm See}$ ref 17a.

at room temperature and presumably gave tetraethylammonium [di(acyloxy)bromate (I)] (3a) (Scheme 1). 12 Treatment of this solution with TMSN $_3$ followed by addition of alkenes 7–11 led to the corresponding bromo-azidation products 12, 14, 16, 18, and 19 (Table 1). From these observations it is reasonable to assume that either tetraethylammonium [bis(azido)bromate (I)] (6a) or bromine azide 4a is formed under these conditions (Table 1). When tetraethylammonium iodide (2b) was employed, the corresponding 1,2-iodo azides 13, 15, and 17 were generated instead, again presumably via the iodate(I)

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Scheme 2

complexes **3b**¹³ and **6b**. In this case, iodine azide **4b** may be involved in the addition reaction. The active agent can also be generated by reversing the sequence. In this alternative method, which we found gave yields slightly higher than those of the first route, the hypervalent azidoiodine(III) reagent 514 was formed first, followed by azide transfer onto the tetraalkylammonium halogenides **2a** and **2b**.

In analogy to bromate(I) **6a**, bisazido iodate **6b** promotes azidoiodination of alkenes such as cyclohexene 7 or indene 8, mostly in a highly anti-selective manner (Table 1). Sensitive β -iodo azides **13**² and **15** were isolated in good yield. Addition of these haloazide equivalents to carbohydrate-derived cyclic enol ethers **9–11** proceeded in a highly regioselective manner, yielding 2-deoxy-2haloglycosyl azides 16-19. The reduced anti-stereoselectivity observed for the addition of the bromate(I) reagent **6a** to glycals **9–11** indicates that the intermediate cyclic bromonium is less stable than the corresponding iodonium ion.3 Thus, the ring oxygen in these pyrans is sufficient to cause ring opening of the intermediate bromonium ion to the oxonium ion. In the case of indene **8**, the aromatic ring exerts a similar enhanced stabilization on the intermediate cation, which results in partial syn-addition.

In contrast to the regioselectivity observed in these examples, the nitrophenyl-substituted propene 20, (R)-

carvone 23, and the unsaturated Weinreb amide 26 were preferentially transformed into the anti-Markovnikov 1,2adducts 21 (ratio of regioisomer 21/22, 10:1; 71%), 23 (24/ **25**, 7.5:1; 78%), and **27** (other regioisomers not found) (Scheme 2).15 From mechanistic studies it is known that the regioselectivity observed in these examples does not result from a free radical pathway, as is the case for chlorine azide and to some extent bromine azide addition to multiple bonds.³ In fact, this same phenomenon has been observed before in 1,2-addition reactions to highly hindered terminal alkenes such as 3,3-dimethyl-1butene.^{2,16} Similarly to Hassner and co-workers, we noticed a reduced reaction rate for these alkenes compared to those from Table 1. The oxidation of carvone 23 reveals the pronounced chemoselectivity of the new reagent system, whereas the use of alkene 26 illustrates the remarkable mildness of the halogen-ate complexes by tolerating a wide range of functional groups. At higher temperature, however, the intermediate iodonium ion was opened at the more substituted position in an intramolecular mode, followed by α -iodination of the carbonyl group to quantitatively yield the light-sensitive isomerically pure lactone 30 isolable in 55% yield. From nuclear Overhauser effect (NOE) experiments, the configuration of the three stereogenic centers in 30 was proven. The strong relative NOEs between 4-H and the

⁽¹⁴⁾ Compound 5 can only be generated in situ; its precise structure is unknown: (a) Kirschning, A.; Domann, S.; Dräger, G.; Rose, L. *Synlett* **1995**, 767. (b) Magnus, P.; Lacour, J. *J. Am. Chem. Soc.* **1992**, 114, 767. (c) Magnus, P.; Lacour, J.; Evans, P. A.; Roe, M. B.; Hulme, C. J. Am. Chem. Soc. 1996, 118, 3406 and references therein.

⁽¹⁵⁾ The ¹³C NMR data (in CDCl₃) are diagnostic: 21, 58.2 ppm for CH_2-N_3 and 28.3 ppm for CH-I; 24, 62.9 ppm for CH_2-N_3 and 57.6 ppm for C-I; 27, 59.4 and 61.7 ppm for C-I, 62.6 and 62.9 ppm for

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protons of both methyl groups attached to C3 (12.5%) and C5 (13.4%) are a clear proof.

Formation of lactone $\bf 30$ can be rationalized by assuming that the carbonyl group acts as an intramolecular nucleophile that attacks the primary iodonium ion. After α -deprotonation, the cyclization product $\bf 28$ is further transformed into $\bf 29$. As the iodonium source is employed in excess, a second electrophilic addition initiates the final steps toward $\bf 30$.

In summary, we present novel reagent systems that are readily generated in situ from bisazido iodo benzene and tetraalkylammonium halides and that synthetically behave like haloazide equivalents. They can be employed in nonpolar organic solvents and promote azidohalogenation of alkenes, including highly functionalized members, under very mild conditions.

Experimental Section

General Methods. All temperatures quoted are uncorrected. Optical rotations were measured at 581 nm. 1 H and 13 C NMR spectra were measured with 400 MHz using tetramethylsilane as the internal standard. CDCl $_{3}$ is the solvent for all NMR experiments except where otherwise stated. All solvents used were of reagent grade and were further dried. Reactions were monitored by TLC on silica gel 60_{P254} and detected either by UV absorption or by staining with H_{2} SO $_{4}$ /4-methoxybenzaldehyde in ethanol. Preparative column chromatography was performed on silica gel 60 (230–400 mesh). Glycals 9–11 were synthesized according to references. 18,19 Addition products 12 , 9a 13 , 2 and 17 17 have been described in detail before. Cyclohexene 7 and indene 8 are commercially available.

General Procedure for the Azidohalogenation of Alkenes. A suspension of PhI(OAc)₂ (1 equiv) in dry CH₂Cl₂ (5-10 mL/ mmol) under nitrogen was cooled to −30 °C. TMSN₃ (2 equiv) was added, and stirring was continued for 30 min at -30°C. Then Et₄NI or Et₄NBr (0.75 equiv) was added in one portion, and the color of the solution turned to red-brown. After 15 min of stirring at ambient temperature, alkene (0.25 equiv for alkenes 11, 20, and 23; 0.4 equiv for alkene 26; and 0.5 equiv for alkenes 7-10) was added. The reaction was monitored by TLC and was terminated by addition of saturated NaHSO₃ solution. The aqueous phase was separated and extracted twice with CH₂Cl₂. The combined organic layers were dried (MgSO₄) and concentrated in vacuo to afford the crude product which, was purified by flash chromatography. (Note: for the labile iodine azide adducts it is essential to perform workup with NaHSO₃ and chromatographic purification as rapid as possible.)

trans/cis-2-Azido-cyclohexyl Bromide (12a,b). Cyclohexene 7 (500 mg, 6.08 mmol) was used to prepare the title compound 12a (1.14 g, 5.6 mmol, 92%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 80:1). Colorless oil; IR (film) 2099 cm⁻¹; 1 H NMR δ 3.87 (ddd, J = 4.2, 9.6, 11.2 Hz, 1H), 3.48 (ddd, J = 4.0, 9.4, 9.4 Hz, 1H), 2.36 (m, 1H), 2.17 (m, 1H), 1.80 (m, 3H), 1.40 (m, 3H); 13 C NMR δ 66.9 (d), 55.4 (d), 36.7, 32.2, 26.1, 24.1 (4t). Anal. Calcd for C₆H₁₀BrN₃: C, 35.31; H, 4.94; Br, 39.16; N, 20.59. Found C, 35.43; H, 4.83; Br, 39.19; N, 20.43. The *cis*-isomer 12b was not detected by 1 H NMR spectroscopy.

trans/cis-2-Azido-cyclohexyl Iodide (13a,b). Cyclohexene 7 (500 mg, 6.08 mmol) was used to prepare the title compounds 13a,b (948 mg, 3.77 mmol, 62%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 80:1). The ratio for

13a,b was determined to be 15:1 by 1H NMR spectroscopy. Colorless oil **13a**; IR (film) 2097 cm $^{-1}$; 1H NMR δ 3.96 (ddd, $J\!=\!4.0,\,10.0,\,11.0$ Hz, 1H), 3.51 (ddd, $J\!=\!4.0,\,9.5,\,9.5$ Hz, 1H), 2.44 (m, 1H), 2.17 (m, 1H), 2.01 (m, 1H), 1.88 (m, 1H), 1.64–1.20 (m, 4H); ^{13}C NMR δ 67.6 (d), 38.8 (t), 33.6 (d), 32.3, 27.5, 24.3 (3t). Anal. Calcd for C₆H₁₀IN₃: C, 28.70; H, 4.01; N, 16.74. Found C, 27.96; H, 4.44; N, 15.98.

trans-1-Azido-2-bromo-indane (14a). Indene **8** (500 mg, 4.3 mmol) was used to prepare the title compounds **14a** (687 mg, 2.88 mmol, 67%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 80:1). Colorless oil; IR (film) 2100 cm⁻¹; ¹H NMR δ 7.42–7.20 (m, 4H), 5.04 (d, J = 5.2 Hz, 1H), 4.40 (ddd, J = 5.2, 6.0, 6.8 Hz, 1H), 3.62 (dd, J = 6.8, 16.0 Hz, 1H), 3.26 (dd, J = 6.0, 16.0 Hz, 1H); ¹³C NMR δ 140.6 (s), 138.5 (s), 129.9 (d), 128.2 (d), 125.3 (d), 125.0 (d), 73.7 (d), 51.6 (d), 41.9 (t). Anal. Calcd for C₉H₈BrN₃: C, 45.40; H, 3.39; N, 17.65. Found C, 45.71; H, 3.53; N, 17.77. The *cis*-isomer **14b** was not detected by ¹H NMR spectroscopy.

trans/cis-1-Azido-2-iodo-indane (15a,b). Indene **8** (500 mg, 4.3 mmol) was used to prepare the title compounds **15a,b** (637 mg, 2.24 mmol, 52%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate **80**:1). The ratio for **15a,b** was determined to be 2.5:1 by ¹H NMR spectroscopy.

First fraction (*trans-***15a**): colorless oil; IR (film) 2104 cm $^{-1}$; 1 H NMR δ 7.49 $^{-}$ 7.28 (m, 4H), 5.11 (d, J=5.5 Hz, 1H), 4.36 (ddd, J=5.5, 6.0, 7.0 Hz, 1H), 3.64 (dd, J=7.0, 16.0 Hz, 1H), 3.36 (dd, J=6.0, 16.0 Hz, 1H); 13 C NMR δ 142.8 (s), 139.7 (s), 129.1 (d), 128.1 (d), 125.0 (d), 124.9 (d), 70.8 (d), 36.5 (t), 30.9 (d). Anal. Calcd for C₉H₈IN₃: C, 37.92; H, 2.83; N, 14.74. Found C, 38.49; H, 3.38; N, 15.11.

Second fraction (*cis*-**15b**): colorless solid, mp 53 °C; IR (film) 2103 cm $^{-1}$; 1 H NMR δ 7.45 $^{-}$ 7.18 (m, 4H), 5.52 (d, J=1.6 Hz, 1H), 4.55 (ddd, J=1.6, 2.0, 6.0 Hz, 1H), 3.40 (dd, J=6.0, 16.0 Hz, 1H), 2.92 (dd, J=2.0, 16.0 Hz, 1H); 13 C NMR δ 142.7 (s), 138.9 (s), 129.1 (d), 128.1 (d), 125.6 (d), 124.9 (d), 64.2 (d), 37.2 (d), 36.1 (t). Anal. Calcd for $C_9H_8IN_3$: C, 37.92; H, 2.83; N, 14.74. Found C, 37.55; H, 3.06; N, 15.07.

3,4,6-Tri-O-benzyl-2-bromo-2-deoxy- α -D-talo-pyranosyl Azide (16a), 3,4,6-Tri-O-benzyl-2-bromo-2-deoxy- β -D-galacto-pyranosyl Azide (16b), and 3,4,6-Tri-O-benzyl-2-bromo-2-deoxy- α -D-galacto-pyranosyl Azide (16c). Glycal 9 (1.2 g, 2.88 mmol) was used to prepare the title compounds 16a-c (1.15 g, 2.13 mmol, 74%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 20:1). The isomeric ratio (1.3:1.7:1) was determined by 1 H NMR spectroscopy.

First fraction (**16c**): colorless oil; $[\alpha]^{22}_{\rm D} = +126.2^{\circ}$ (c 1.0, CHCl₃); IR (film) 2117 cm⁻¹; ¹H NMR δ 7.44–7.21 (m, 15H), 5.52 (d, J=4.0 Hz, 1H), 4.87, 4.73, 4.70, 4.51, 4.49, 4.42 (6d, J=11.4 and 12.0 Hz, 6H), 4.55 (dd, J=4.0, 11.0 Hz, 1H), 4.14 (br t, J=6.0, 6.0 Hz, 1H), 3.94 (dd, J=1.0, 2.6 Hz, 1H), 3.77 (dd, J=2.6, 11.0 Hz, 1H), 3.55 (d, J=6.0 Hz, 2H); ¹³C NMR δ 138.4–127.8 (Ph), 90.6 (d), 78.3 (d), 74.7 (d), 71.9 (d), 75.1 (t), 73.6 (t), 73.2 (t), 68.3 (t), 64.6 (d). Anal. Calcd for $C_{27}H_{28}O_4BrN_3$: C, 60.23; H, 5.24; N, 7.80; Br, 14.84. Found C, 60.55; H, 5.11; N, 8.01; Br, 14.07.

Second fraction (**16a**): colorless oil; $[\alpha]^{27}_D = +36.2^{\circ}$ (c 1.0, CHCl₃); IR (film) 2116 (cm⁻¹); ¹H NMR δ 7.24–7.08 (m, 15H), 5.36 (d, J = 4.0 Hz, 1H), 4.82, 4.43, 4.40, 4.33, 4.26, 4.25 (6d, J = 11.2 and 12.0 Hz, 6H), 4.21 (dt, J = 2.8, 6.0, 6.0 Hz, 1H), 3.83 (d, J = 6.0 Hz, 2H), 3.70 (dd, J = 3.6, 4.0 Hz, 1H), 3.54 (dd, J = 2.8, 2.8 Hz, 1H), 3.50 (dd, J = 2.8, 3.6 Hz, 1H); ¹³C NMR δ 138.5–127.6 (Ph), 89.1 (d), 74.3 (d), 73.7 (d), 73.5 (d), 73.3, 72.1, 72.0 (t), 67.9 (t), 46.8 (d). Anal. Calcd for $C_{27}H_{28}O_4BrN_3$: C, 60.23; H, 5.24; N, 7.80; Br, 14.84. Found: C, 60.31; H, 5.30; N, 7.77; Br, 14.41.

Third fraction **(16b)**: colorless solid, mp 94 °C; $[\alpha]^{20}_{\rm D} = +6.7^{\circ}$ (c 1.02, CHCl₃); IR (KBr) 2119 (cm⁻¹); ¹H NMR δ 7.42–7.22 (m, 15H), 4.75 (d, J=9.6 Hz, 1H), 4.86, 4.71, 4.69, 4.55, 4.46, 4.43 (6d, J=11.4 and 12.0 Hz, 6H), 4.12 (dd, J=9.6, 10.6 Hz, 1H), 3.89 (br d, J=2.8 Hz, 1H), 3.69 (br dd, J=5.6, 7.0 Hz, 1H), 3.60 (dd, J=5.6, 9.2 Hz, 1H), 3.57 (dd, J=7.0, 9.2 Hz, 1H), 3.55 (dd, J=3.0, 10.6 Hz, 1H); ¹³C NMR δ 138.0–127.8 (Ph),

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91.0 (d), 82.6 (d), 76.0 (d), 73.1 (d), 74.8 (t), 73.7 (t), 73.2 (t), 68.2 (t), 51.8 (d). Anal. Calcd for $C_{27}H_{28}O_4BrN_3$: C, 60.23; H, 5.24; N, 7.80; Br, 14.84. Found: C, 60.25; H, 5.26; N, 7.81; Br, 14.20.

3,4,6-Tri-O-benzyl-1,2-dideoxy-2-iodo- α -D-talo-pyranosyl Azide (17a) and 3,4,6-Tri-O-benzyl-1,2-dideoxy-2-iodo- β -D-galacto-pyranosyl Azide (17b). Glycal 9 (250 mg, 0.62 mmol) was used to prepare the title compounds 17a,b (174 mg, 0.30 mmol, 49%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 8:1). Spectroscopic and analytical data were in accordance with those listed in ref 17. The isomeric ratio (<1: 10) was determined by 1 H NMR spectroscopy.

2-Bromo-2-deoxy-3,4,6-tri-*O*-(triethylsilyl)- α -D-galactopyranosyl Azide (18a) and 2-Bromo-2-deoxy-3,4,6-tri-*O*-(triethylsilyl)- β -D-galacto-pyranosyl Azide (18b). Glycal 10 (1.7 g, 3.48 mmol) was used to prepare the title compounds 18a,b (1.55 g, 2.54 mmol, 73%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 200:1). The isomeric ratio (1:1.9) was determined by 1 H NMR spectroscopy.

First fraction (**18a**): colorless oil; $[\alpha]^{22}_{D} = +132.9^{\circ}$ (c 1.08, CHCl₃); IR (film) 2116 (cm⁻¹); ¹H NMR (C_6D_6) δ 5.20 (d, J = 4.0 Hz, 1H), 4.57 (dd, J = 4.0, 10.4 Hz, 1H), 4.21 (d, J = 2.0 Hz, 1H), 4.12 (dd, J = 2.0, 10.4 Hz, 1H), 4.10 (m, 1H), 3.97 (dd, J = 7.2, 10.0 Hz, 1H), 3.88 (dd, J = 6.0, 10.0 Hz, 1H), 1.19 –1.04 (m, 27H), 0.88 –0.78 (m, 12H), 0.71 –0.63 (m, 6H); ¹³C NMR (C_6D_6) δ 91.1 (d), 75.1 (d), 72.9 (d), 72.0 (d), 61.7 (t), 51.7 (d), 7.2, 6.9 (q), 5.7, 5.6, 4.8 (t). Anal. Calcd for $C_{24}H_{52}BrN_3O_4Si_3$: C, 47.19; H, 8.58; N, 6.88; Br, 13.08. Found: C, 47.21; H, 8.56; N, 6.71; Br, 13.12

Second fraction (**18b**): colorless oil; $[\alpha]^{23}_D = +12.8^{\circ}$ (c 1.01, CHCl₃); IR (film) 2115 (cm⁻¹); ¹H NMR (C_6D_6) δ 4.35 (d, J = 8.8 Hz, 1H), 4.28 (dd, J = 8.8, 10.0 Hz, 1H), 4.01 (d, J = 2.0 Hz, 1H), 3.87 (dd, J = 7.2, 10.0 Hz, 1-H), 3.80 (dd, J = 5.6, 10.0 Hz, 1H), 3.55 (dd, J = 2.0, 10.0 Hz, 1H), 3.32 (dd, J = 5.6, 7.2 Hz, 1H), 1.03 – 0.85 (m, 27H), 0.75 – 0.43 (m, 18H); ¹³C NMR (C_6D_6) δ 91.1 (d), 78.1(d), 76.7 (d), 71.9 (d), 61.6 (t), 54.2 (d), 7.2, 7.1, 7.0 (q), 5.7, 5.5, 4.8 (t). Anal. Calcd for $C_{24}H_{52}BrN_3O_4Si_3$: C, 47.19; H, 8.58; N, 6.88; Br, 13.08. Found: C, 47.14; H, 8.67; N, 6.50; Br, 13.23.

2-Bromo-6-O-(tert-butyldimethylsilyl)-2-deoxy-3,4-di-O-(triethylsilyl)- α -D-galacto-pyranosyl Azide (19a) and 2-Bromo-6-O-(tert-butyldimethylsilyl)-2-deoxy-3,4-di-O-(triethylsilyl)- β -D-galacto-pyranosyl Azide (19b). Glycal 11 (366 mg, 0.75 mmol) was used to prepare the title compounds 19a,bd (320 mg, 0.52 mmol, 70%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 100:1). The isomeric ratio (1:1.4) was determined by 1 H NMR spectroscopy.

First fraction (19a): oil; $^1\mathrm{H}$ NMR δ 5.49 (d, J=3.8 Hz, 1H), 4.36 (dd, J=3.8, 10.2 Hz, 1H), 3.99 (br s, 1H), 3.86 (dd, J=2.2, 10.2 Hz, 1H), 3.85 (dt, J=1.0, 6.2 Hz, 1H), 3.69 (dd, J=6.2, 10.0 Hz, 1H), 3.65 (dd, J=6.2, 10.0 Hz, 1H), 0.99 (t, J=8.0 Hz, 9H), 0.96 (t, J=8.0 Hz, 9H), 0.89 (s, 9H), 0.75-0.61 (m, 12H), 0.07 (s, 3H), 0.06 (s 3H); $^{13}\mathrm{C}$ NMR δ 90.7 (d), 74.8 (d), 71.5 (d), 72.4 (d), 61.6 (t), 51.2 (d), 25.8 (q), 18.2 (s), 6.92 (q), 6.89 (q), 5.2 [t), 5.1 (t), -5.3 (q), -5.4 (q). Anal. Calcd for $\mathrm{C_{24}H_{52}-BrN_3O_4Si_3}$: C, 47.19; H, 8.58; N, 6.88; Br, 13.08. Found: C, 47.35; H, 8.77; N, 6.59; Br, 13.20.

Second fraction (19b): oil; $^1\mathrm{H}$ NMR δ 4.68 (d, J=9.6 Hz, 1H), 3.97 (dd, J=9.6, 9.6 Hz, 1H), 3.93 (d, J=2.3 Hz, 1H), 3.73 (dd, J=7.1, 10.1 Hz, 1H), 3.69 (dd, J=6.0, 10.1 Hz, 1H), 3.64 (dd, J=2.3, 9.6 Hz, 1H), 3.46 (br t, J=6.2 Hz, 1H), 0.98 (t, J=8.0 Hz, 9H), 0.96 (t, J=8.0 Hz, 9H), 0.88 (s, 9H), 0.78–0.56 (m, 12H), 0.07 (s, 6H); $^{13}\mathrm{C}$ NMR δ 91.0 (d), 78.0 (d), 76.3 (d), 71.3 (d), 61.3 (t), 54.0 (d), 25.8 (q), 18.2 (s), 6.9 (q), 5.2 (t), 51.1 (t), -5.3 (q), -5.4 (q). Anal. Calcd for $C_{24}\mathrm{H}_{52}\mathrm{BrN}_{3}\mathrm{O}_{4}\mathrm{Si}_{3}$: C, 47.19; H, 8.58; N, 6.88; Br, 13.08. Found: C, 47.08; H, 8.70; N, 6.51; Br, 13.11.

1,4-Dimethoxy-2-nitro-6-[3'-(1'-azido-2'-iodo)propane] (21) and 1,4-Dimethoxy-2-nitro-6-[3'-(2'-azido-1'-iodo)propane] (22). Alkene 20 (167 mg, 0.75 mmol) was used to prepare the title compounds 21 and 22 (208 mg, 0.53 mmol, 71%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 20:

1). The regiomeric ratio (10:1) was determined by ¹H NMR spectroscopy.

First fraction (**21**): oil, IR (film) 2111.0 (cm⁻¹); ¹H NMR δ 7.28 (d, J=3.6 Hz, 1H), 6.99 (d, J=3.6 Hz, 1H), 4.38 (dddd, J=5.6, 6.2, 6.2, 8.8 Hz, 1H), 3.85 (s, 3H), 3.82 (s, 3H), 3.76 (dd, J=5.6, 13.0 Hz, 1H), 3.63 (dd, J=6.2, 13.0 Hz, 1H), 3.37 (dd, J=6.2, 14.3 Hz, 1H), 3.12 (dd, J=8.8, 14.3 Hz, 1H); ¹³C NMR δ 154.9, 145.6, 143.8, 135.9 (s), 122.3, 108.5 (d), 62.6 (q), 58.2 (t), 56.0 (q), 38.8 (t), 28.3 (t). Anal. Calcd for C₁₁H₁₃O₄N₄I: C, 33.69; H, 3.34; N, 14.28. Found: C, 33.79; H, 3.34; N, 14.22.

Second fraction (**22**): oil, IR (film) 2105.6 (cm $^{-1}$); 1 H NMR δ 7.30 (d, J = 3.2 Hz, 1H), 7.09 (d, J = 3.2 Hz, 1H), 3.89 (s, 3H), 3.84 (s, 3H), 3.85 $^{-3}$.68 (m, 1H), 3.33 (dd, J = 5.4, 10.7 Hz, 1H), 3.26 (dd, J = 5.7, 10.7 Hz, 1H), 3.10 (dd, J = 5.6, 13.8 Hz, 1H), 2.84 (dd, J = 8.0, 13.8 Hz, 1H); 13 C NMR δ : 154.9, 145.6, 143.8, 133.4 (s), 122.4, 108.5 (d), 62.7 (q), 61.8 (t), 55.9 (q), 35.5 (t); 7.8 (t).

(2'RS,5R)-5-[2'-(1'-Azido-2'-iodo)propyl]-2-methyl-cyclohex-2-en-1-one (24) and (2'RS,5R)-5-[2'-(2'-Azido-1'-iodo)propyl]-2-methyl-cyclohex-2-en-1-one (25). (R)-Carvone 23 (112 mg, 0.75 mmol) was used to prepare the title compounds 24 and 25 (186 mg, 0.58 mmol, 78%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 20:1). The regiomeric ratio (7.5:1) was determined by ¹H NMR spectroscopy.

First fraction (24, 2 diastereomers 1:1): yellow oil; IR (film) 2103.9 (cm $^{-1}$); ^{1}H NMR δ 6.75 (m, 1H), 3.84 (m, 2H), 2.67–2.20 (m, 4H), 2.10 (s, 3H), 1.84–1.74 (m, 3H), 1.59–1.48 (m, 1H); ^{13}C NMR δ 198.1, 197.9 (s), 143.4, 143.4 (d), 135.4, 135.3 (s), 62.9 (t), 57.6 (s), 44.0, 43.9 (d), 42.9, 42.2, 30.6, 29.9 (t), 30.8, 30.6 (q), 15.5 (q). Anal. Calcd for $C_{10}H_{14}IN_{3}O$: C, 37.63; H, 4.42; N, 13.17. Found: C, 37.39; H, 4.24; N, 13.77.

Second fraction (**25**, 2 diastereomers 1:1): yellow oil; IR (film) 2105.6 (cm $^{-1}$); ^{1}H NMR δ 6.70 (m, 1H), 3.33 (s, 2H), 2.55 (m, 5H), 1.74, 1.46 (2 s, 6H); ^{13}C NMR δ 198.3, 198.2 (s), 143.8, 143.5 (t), 135.4, 135.2 (s), 63.2, 63.2 (s), 41.9, 41.8 (t), 38.9, 38.4, 26.9, 26.2 (t), 21.1, 21.0 (q), 15.5 (q), 13.1 (t). Anal. Calcd for $C_{10}H_{14}$ -IN $_3O$: C, 37.63; H, 4.42; N, 13.17. Found: C, 37.81; H, 4.19; N, 13.66.

(2R,3R,4RS)-5-Azido-3-(diphenylmethylsiloxy)-4-iodo-N-methoxy-N,2,4-trimethyl-pentyl Amide (27). Alkene 26 (100 mg, 0.26 mmol) was used to prepare the title compound 27 (87 mg, 0.16 mmol, 62%) via the general procedure described above. Purification was achieved by column chromatography (petroleum ether/ethyl acetate 1:1). The isomeric ratio (1:1) was determined by ¹H NMR spectroscopy.

First fraction: brown oil; $[\alpha]^{23}_D = -5.3^\circ$ (c 1.0, CHCl₃); IR (film) 2105 (cm⁻¹); ¹H NMR δ 7.8–7.2 (m, 10H), 4.20 (d, J = 4.4 Hz, 1H), 3.77 (s, 3H), 3.48 (dq, J = 4.4, 7.2 Hz, 1H), 3.40 (s, 2H), 3.17 (s, 3H), 1.81 (s, 3H), 1.12 (d, J = 7.2 Hz, 3H), 0.85 (s, 3H); ¹³C NMR δ = 175.4 (s), 136.4, 135.9 (s), 134.3–127.3 (d), 76.3 (d), 62.6 (t), 61.6 (q), 59.4 (s), 40.8 (t), 32.6 (q), 28.0 (q), 15.0 (q), -2.4 (q). LRMS (ESI) m/z 574.9 (M + Na⁺).

Second fraction: brown oil; $[\alpha]^{23}_{\rm D} = -25.6^{\circ}$ (c 1.765, CHCl₃); IR (film) 2103 (cm⁻¹); ¹H NMR δ 7.73–7.32 (m, 10H), 3.75 (s, 3H), 3.56 (d, J= 12.0 Hz, 1H), 3.43 (d, J= 7.2 Hz, 1H), 3.37 (m, 1H), 3.29 (d, J= 12.0 Hz, 1H), 1.74 (s, 3H), 1.08 (d, J= 7.6 Hz, 3H), 0.84 (s, 3H); ¹³C NMR δ 177.0 (s), 136.2, 136.1 (s), 134.6–127.8 (d), 76.1 (d), 63.0 (t), 61.7 (q), (s), 42.1 (t), 32.6 (q), 28.4 (q), 16.1 (q), -2.8 (q). LRMS (ESI) m/z 574.9 (M + Na⁺).

(3R,4R,5R)-3-Iodo-5-iodomethyl-4-(diphenylmethylsilyloxy)-dihydro-furan-2-one (30). To a solution of 6b [generated from 167 mg (0.52 mmol) of PhI(OAc)2, 120 mg (1 mmol) of TMSN₃, and 100 mg (0.39 mmol) of Et₄NI] was added alkene 26 (50 mg, 0.13 mmol), and stirring was continued for 2 h at room temperature. Quantitative formation of the title compound 30 was monitored by TLC (petroleum ether/ethyl acetate 3:1; $R_f = 0.57$). The reaction mixture was rapidly hydrolyzed at 0 °C with saturated NaHCO₃. Use of aqueous NH₄Cl, NaHSO₃, or Na₂S₂O₃ solutions led to complete decomposition of **30**. Extraction of the aqueous phase with CH₂Cl₂ (3×), drying (MgSO₄) of the combined washings, and concentration under reduced pressure afforded an oil. Finally, purification was achieved by column chromatography (petroleum ether/ethyl acetate 15:1) and afforded the title compound 30 (42 mg, 0.07 mmol, 55%). Brown oil; $[\alpha]^{23}_D = +8.9^{\circ}$ (c 0.975, CHCl₃); IR (film) 1777, 1428, 1379, 1257, 1122, 1086, 1228, 961, 883, 836, 793, 739, 723, 699 (cm $^{-1}$); $^{1}\mathrm{H}$ NMR δ 7.72-7.33 (m, 12H), 4.27 (dd, J= 0.8, 11.6 Hz, 1H), 3.30 (d, J= 11.6 Hz, 1H), 3.24 (s, 1H), 1.81 (s, 3H), 1.26 (d, J= 0.8 Hz, 3H), 0.79 (s, 3H); $^{13}\mathrm{C}$ NMR δ 173.1 (s), 134.6, 134.5, 130.9, 130.8, 128.3, 128.2 (t), 133.9, 133.8 (s), 83.2 (s), 81.0 (t), 38.3 (s), 30.2 (q), 28.6 (q), 8.9 (t), -2.0 (q).

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Additions and Corrections

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Eric D. Soli, Amy S. Manoso, Michael C. Patterson, Philip DeShong,* David A. Favor, Ralph Hirschmann, and Amos B. Smith, III. Azide and Cyanide Displacements via Hypervalent Silicate Intermediates.

Page 3171. Reference 17 should be a citation to the studies of glycosyl azides from the Györgydeák group at Lajos Kossuth University, Debrecen, Hungary reported in: Györgydeák, Z.; Szilágyi, L.; Paulsen, H. *J. Carbohydr. Chem.* **1993**, *12*, 139–163. The paper cited in ref 17 in the manuscript should have been included in ref 18 instead.

We apologize for having failed to include this citation.

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