

Correction to “Evaporation Rates and Vapor Pressures of the Even-Numbered C₈–C₁₈ Monocarboxylic Acids”

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The vapor pressure for solid hexadecanoic (palmitic) acid was incorrectly reported in Table 1 as $1.3 (\pm 0.4) \times 10^{-7}$ Pa. The correct value is $1.3 (\pm 0.4) \times 10^{-6}$ Pa.

Table 1. Measured Vapor Pressures, Enthalpies, and Entropies of Sublimation or Vaporization for the Monocarboxylic Acids at 298 K

	p_s^0 (298 K) (Pa)	ΔH_{sub}^0 (kJ/mol)	ΔS_{sub}^0 [J/(mol·K)]
octanoic acid ^a (H ₃ C(CH ₂) ₆ COOH)	$2.5 (\pm 0.9) \times 10^{-1}$	113.3 ± 6	272.5 ± 19
decanoic (capric) acid ^a (H ₃ C(CH ₂) ₈ COOH)	$5.4 (\pm 1.1) \times 10^{-2}$	129.6 ± 5	314.6 ± 15
dodecanoic (lauric) acid ^a (H ₃ C(CH ₂) ₁₀ COOH)	$2.28 (\pm 0.5) \times 10^{-3}$	147.2 ± 4	347.2 ± 13
tetradecanoic (myristic) acid ^a (H ₃ C(CH ₂) ₁₂ COOH)	$7.0 (\pm 2.7) \times 10^{-5}$	168.6 ± 9	390.0 ± 31
hexadecanoic (palmitic) acid ^a (H ₃ C(CH ₂) ₁₄ COOH)	$1.3 (\pm 0.4) \times 10^{-6}$	193.8 ± 11	441.7 ± 36
octadecanoic (stearic) acid ^a (H ₃ C(CH ₂) ₁₆ COOH)	$9.5 (\pm 3.5) \times 10^{-8}$	204.1 ± 9	454.2 ± 31
octadecenoic (oleic) acid ^b (H ₃ C(CH ₂) ₈ =(CH ₂) ₈ COOH)	$1.9 (\pm 0.9) \times 10^{-6}$	135.6 ± 3	248.9 ± 8

^aSolid phase. ^bLiquid phase.