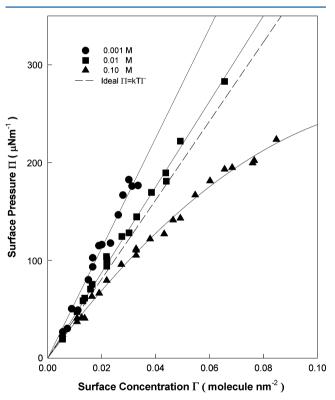


## Correction to "Thermodynamics of Ionized Monolayers: Surface Manometry on Very Low Density Spread Monolayers of Sodium Octadecyl Sulfate at the Air/Water Interface and Analysis of Ionic Double Layer Contributions to the Isotherms"

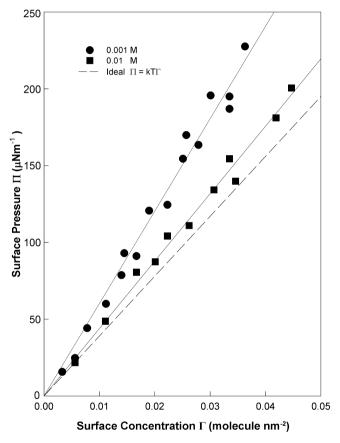
S. R. Middleton, N. R. Pallas, J. Mingins, and B. A. Pethica\*

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In the original manuscript, the first two Figures showing the experimental monolayer isotherms are given with surface pressure scales incorrect by a factor of 10. The surface pressure ranges for Figures 1 and 2 are 0-300 and  $0-250~\mu\text{N m}^{-1}$ , respectively. The correct surface pressures are uniformly used and discussed in the text. The corrected Figures are given here.



**Figure 1.**  $\Pi$ – $\Gamma$  isotherms at 20 °C for dilute monolayers of sodium octadecyl sulfate spread at the interface between water-vapor saturated air and aqueous NaCl solutions at concentrations of  $10^{-3}$ ,  $10^{-2}$ , and  $10^{-1}$  M. The straight lines shown for  $10^{-2}$  and  $10^{-1}$  M NaCl are statistical fits through the origin. The curve shown for  $10^{-1}$  M NaCl is to guide the eye. Also shown is the isotherm  $\Pi = kT\Gamma$  for an ideal two-dimensional gas.



**Figure 2.**  $\Pi$ – $\Gamma$  isotherms at 10 °C for dilute monolayers of sodium octadecyl sulfate spread at the interface between water-vapor saturated air and aqueous NaCl solutions at concentrations  $10^{-1}$  and  $10^{-2}$  M. The straight lines are statistical fits through the origin. Also shown is the isotherm  $\Pi = kT\Gamma$  for an ideal two-dimensional gas.