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Richard D. Harcourt*: Correction to “Valence Bond Structures for Three-Electron Three-Center and Four-Electron Three-Center Bonding Units: Some Further Examples”[†]

doi: 10.1021/jp911294x. (a) The entries for Table 4 of the subject paper were calculated using carbon $2p\pi$ AOs. With hydrogen $1s$ AOs, as described in the paper, the entries are the following.

TABLE 4: Cyclic H_6

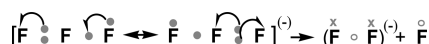
	W(Kekulé)	W(Dewar)	energy (au)	l in $r + ly$ or $y + lr$
$1 \leftrightarrow 2$	0.446	0.036	−3.1773	
$3 \leftrightarrow 4$	0.004	0.163	−3.2246	0.8
3	1/6		−3.2026	1.2
4		1/6	−3.2240	0.8

(b) In ref 1, VB studies have been reported for two exit channels for dissociation of X_3^- (X = halogen or hydrogen). With $X \equiv R \equiv Y$, the $X_3^- \rightarrow X_2 + X^-$ exit channel occurs¹ for $X = Cl, Br, I$, and H . It is accommodated by



from the ground-state generalized Reactant-like Complex \leftrightarrow Product-like Complex (or RC \leftrightarrow PC) resonance in the subject paper's Scheme 5.

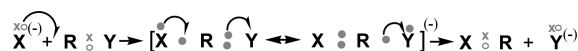
The F_3^- exit channel is primarily¹ $F_3^- \rightarrow F_2^- + F$. It is accommodated by



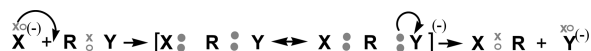
from the excited-state (*) PC \leftrightarrow RC resonance in the subject paper's Scheme 5. (Note: RC \equiv (PC)*; PC \equiv (RC)*.)

In ref 2, VB studies of H_3^- with F_3^- are also reported, and their electronic structures are contrasted.

(c) In ref 2, a preference is made for use of



rather than



as a VB formulation for the generalized $X^{(-)} + R:Y \rightarrow X:R + Y^{(-)}$ reaction. These formulations involve one-electron and concerted electron pair delocalizations for Reactants \rightarrow RC and PC \rightarrow Products, respectively.

References and Notes

- (a) Braida, B.; Hiberty, P. C. *J. Phys. Chem. A* **2008**, *112*, 13045. (b) Braida, B.; Hiberty, P. C. *J. Am. Chem. Soc.* **2004**, *126*, 14890.
- Harcourt, R. D. *Int. J. Quantum Chem.* **1996**, *60*, 553. See ref 20d of subject paper for increased-valence revision.

10.1021/jp1039145

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[†] Part of the “Klaus Ruedenberg Festschrift”.