

Synthetic Efficiency in Enzyme Mechanisms Involving Carbocations: Aristolochene Synthase [*J. Am. Chem. Soc.* **2007**, *129*, 13008–13013]. Rudolf K. Allemann,* Neil J. Young, Shuhua Ma, Donald G. Truhlar,* and Jiali Gao*

Page 13009. It has been brought to our attention that one of the sentences in the Introduction to our article, about the gas-phase reactivity providing a standard agent with which the effects of the active site of aristolochene synthase could be evaluated, is similar to a sentence in a previous publication by Gutta and Tantillo,¹ about gas-phase reactivity providing a standard against which the effects of a surrounding environment can be evaluated. Unfortunately, this work by Gutta and Tantillo on the mechanism of pentalenene synthase was not referenced, for which we apologize.

Literature Cited

- (1) Gutta, P.; Tantillo, D. J. *J. Am. Chem. Soc.* **2006**, *128*, 6172–6179.

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Chemistry of Boryllithium: Synthesis, Structure, and Reactivity [*J. Am. Chem. Soc.* **2008**, *130*, 16069–16079]. Yasutomo Segawa, Yuta Suzuki, Makoto Yamashita,* and Kyoko Nozaki*

Supporting Information. The original CIF file contained incorrect information about compound **38b** (it was the data for **35a**·(THF)₂). The new Supporting Information contains the corrected CIF file.

Supporting Information Available: X-ray crystallographic data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Selective Ion Passage through Functionalized Graphene Nanopores [*J. Am. Chem. Soc.* **2008**, *130*, 16448–16449]. Kyaw Sint, Boyang Wang, and Petr Král*

Page 16449. An additional Supporting Information file was omitted.

Supporting Information Available: Details of the MD simulation methods, ab initio calculation methods, the details of the simulated system, partial charges and force field parameters for the water, graphene sheet, and nanopores, and complete refs 28 and 29. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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