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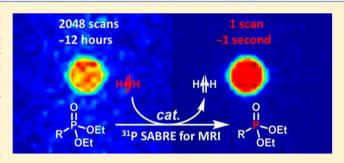
Improving the Hyperpolarization of ³¹P Nuclei by Synthetic Design

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Supporting Information

ABSTRACT: Traditional ³¹P NMR or MRI measurements suffer from low sensitivity relative to ¹H detection and consequently require longer scan times. We show here that hyperpolarization of ³¹P nuclei through reversible interactions with parahydrogen can deliver substantial signal enhancements in a range of regioisomeric phosphonate esters containing a heteroaromatic motif which were synthesized in order to identify the optimum molecular scaffold for polarization transfer. A 3588-fold ³¹P signal enhancement (2.34% polarization) was returned for a partially deuterated pyridyl substituted phosphonate ester. This hyperpolarization level is



sufficient to allow single scan ³¹P MR images of a phantom to be recorded at a 9.4 T observation field in seconds that have signalto-noise ratios of up to 94.4 when the analyte concentration is 10 mM. In contrast, a 12 h 2048 scan measurement under standard conditions yields a signal-to-noise ratio of just 11.4. ³¹P-hyperpolarized images are also reported from a 7 T preclinical scanner.

Phosphorus containing molecules are essential to the human body. 1-5 In addition, a number of widely prescribed pharmaceutical products contain one or more phosphorus atoms. 6-8 Due to 31P's high natural abundance, coupled with a wide chemical shift range that is indicative of molecular environment, ³¹P signal detection represents an exciting probe for magnetic resonance spectroscopy (MRS) and magnetic resonance imaging (MRI) studies in both inorganic chemistry and biochemistry. For example, recent studies have used ³¹P MRS for the *in vivo* monitoring of metabolites in breast cancer, of liver disease, schizophrenia, and tissue assessment. 13,14

The detection of ³¹P nuclei by magnetic resonance (MR), however, suffers from lower sensitivity than ¹H nuclei because of its lower magnetogyric ratio. 15 Consequently, high quality ³¹P NMR spectra or MR images require relatively large amounts of sample in conjunction with signal averaging. Hyperpolarization techniques are seen as a route to overcome this low sensitivity issue which arises from the small population difference (ca. 10⁻⁵) that exists between the Zeeman-split energy levels that are probed. Until recently, ³¹P hyperpolarization has received limited attention, and the development of a simple method to achieve it is desirable. 16,1'

Parahydrogen induced polarization (PHIP) is a popular method of hyperpolarization that utilizes molecular hydrogen as its source of polarization. By incorporating p-H₂ into an unsaturated molecule its singlet symmetry is broken, and this process unlocks its latent, and normally invisible, polarization. A recent report using PHIP for 13C detection, showed that labeled 1-13C-phospholactate-d₂ could be detected at just 5.74 mT.²⁴ PHIP has, however, been widely used to probe reaction mechanisms where it enables the detection of many metal hydride containing complexes. 25-27 In these studies magnetization transfer to ³¹P has proven to be a very important feature of the characterization of many reaction products.²⁸

A new form of PHIP, called Signal Amplification by Reversible Exchange (SABRE), has emerged as a tool for the rapid hyperpolarization of molecules without the need for changing their chemical identity.²⁹ SABRE utilizes a stable metal complex, such as [IrCl(COD)(IMes)] (1) (where IMes = 1,3-bis(2,4,6-trimethylphenyl)imidazolium and COD = cis,cis-1,5-cyclooctadiene), to form what is known as the polarization transfer catalyst (Scheme 1).

The role of catalyst 2 in Scheme 1 is to bring together p-H₂, now in the form of a pair of hydride ligands, and the hyperpolarization target (L). Polarization transfer catalysis (PTC) occurs during the short time period for which 2 maintains this ligand arrangement. Polarization is passed through the *I*-coupling network of the metal complex into the ligand rather than by changing the chemical identity of the ligand through formal p-H₂ incorporation.³⁰ H₂ loss and rebinding of the analyte are necessary for the delivery of high levels of analyte hyperpolarization in conjunction with high catalyst turnover. In this case, the p-H₂ refresh step involves the short-lived iridium dihydrogen dihydride complex 3.31

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Scheme 1. An Active SABRE Catalyst Can Employ Magnetic Inequivalence (e.g. 2_a) or Chemical Inequivalence (e.g. 2_b) at the Magnetization Transfer Stage^a

^aThe generation of hyperpolarized ligands (L) that return to bulk solution through ligand exchange enables the production of analytes that are highly visible to NMR and MRI.

To date, reported studies using the SABRE approach have focused on the polarization of $^1\mathrm{H}^{29,32-34}$ and $^{13}\mathrm{C}^{29,35}$ nuclei, although $^{15}\mathrm{N}^{36,37}$ $^{19}\mathrm{F}^{29}$ and $^{31}\mathrm{P}^{28,38}$ signals have been seen when polarized. Both of the $^{31}\mathrm{P}$ studies focused on PPh3, with Koptyug et al. reporting that the signal for free PPh3 can be enhanced by 2 orders of magnitude, and a single shot image recorded at 9.4 T. 38 Importantly, SABRE has been shown to be successful for a growing range of biologically relevant molecules; this has established the potential of the method to produce $^{1}\mathrm{H}$ and $^{13}\mathrm{C}$ hyperpolarized agents for subsequent MRI measurement. 29,32,35,39,40

Herein, we report the optimization of SABRE with respect to the hyperpolarization of ³¹P nuclei within heteroaromatic molecules. We rationalize how the chemical structure affects the observed polarization level and thereby outline a strategy that may subsequently find use in the production of future diagnostic agents for ³¹P MRI.

MATERIALS AND METHODS

All reactions utilizing air- and moisture-sensitive reagents were performed in dried glassware under an atmosphere of dry nitrogen. Dry solvents (THF, toluene and DCM) were obtained from a Braun MB-SPS-800. For thin-layer chromatography (TLC) analysis, Merck precoated plates (silica gel 60 F254, Art 5715, 0.25 mm) were used. Column chromatography was performed on Fluka Silica gel (60 Å, 220–440 mesh). 14 NMR, 13 C NMR, and 31 P NMR spectra were measured on Bruker 400 or 500 MHz spectrometers. Deuterated solvents (methanol- d_4 and chlorofrom-d) were obtained from Sigma-Aldrich and used as supplied.

The following compounds were synthesized according to literature procedures: $\mathbf{1}$, 41 d₁₀-diethyl phosphite, 42 4-pyridyl diphenylphosphine $\mathbf{L_{1}}$, 43 4-pyridyl diphenylphosphine oxide $\mathbf{L_{2}}$, 44 4-pyridylmethyl diethylphosphonate $\mathbf{L_{8}}$, 3-pyridylmethyl diethylphosphonate $\mathbf{L_{9}}$. Detailed synthetic procedures and characterization data can be found in the Supporting Information.

p-H₂ was produced by cooling H₂ gas over Fe₂O₃ at 30 K.²⁶ Samples involved the analysis of 5 mM concentrations of 1 and 5 or 6 equiv of substrate (L_1 - L_{12}) in methanol- d_4 (0.6 or 3.0 mL) solution that was located in either a 5 or 8 mm NMR tube fitted with a J. Young's Tap. The resulting solutions were degassed prior to the introduction of p-H₂ at a pressure of 3 bar. Samples were then shaken for 10 s in a specified fringe field

of either a 9.4 T Bruker Avance (III) NMR spectrometer or a 7 T Bruker BioSpec preclinical MRI scanner. They were then rapidly transported into the main magnetic field of the instrument for subsequent probing by NMR or MRI methods. The associated polarization transfer experiment data can be found in the Supporting Information.

RESULTS AND DISCUSSION

Examining 4-Substituted Pyridine Derivatives. Historically, *N*-heteroaromatic molecules have been shown to be excellent receptors for polarization transfer to ^{1}H and ^{13}C nuclei under SABRE. 29,31,33 We therefore synthesized a phosphine (\mathbf{L}_{1}), a phosphine oxide (\mathbf{L}_{2}), a phosphine sulfide (\mathbf{L}_{3}) and a phosphonate ester (\mathbf{L}_{4}) which incorporated a 4-substituted pyridine motif, as detailed in the Supporting Information (Figure 1). We selected the 4-substitution pattern

Figure 1. 1 H and 31 P signal intensity gains for analytes L_{1} - L_{4} under SABRE conditions using 1 as the catalyst.

for this study in order to minimize any steric interactions between the hyperpolarization target L and the IMes ligand in the catalyst 2. We expected this restriction to promote activity in the widest range of analytes possible. A series of samples were then prepared in methanol- d_4 that contained $\mathbf{L_1}$ - $\mathbf{L_4}$, at a concentration of 5 mM, and the catalyst precursor 1 at a 17% catalyst loading. These samples were probed after SABRE had taken place at 298 K on a 400 MHz NMR spectrometer by the application of a simple 90° *rf*-pulse. We will discuss the effects of SABRE on the pyridyl 1 H polarization, then the 31 P polarization, and finally the polarization received by the other substituent groups.

When this 90° pulse was applied to the ¹H nuclei of these samples, only L₃ proved to exhibit weak pyridyl proton polarization. The other three analytes produced good proton signal enhancements. For example, the two ¹H NMR signals of L₄, that were derived from the pyridine functionality, showed a 1499-fold total signal enhancement when the initial polarization transfer step was undertaken in a 45 G field. This high ¹H PTC was reflected in the fact that the corresponding ³¹P signal exhibited a similarly impressive 545-fold signal enhancement. For ³¹P, PTC was maximized through transfer at 0.5 G rather than the 45 G field observed for its ¹H nuclei. Figure 1 summarizes the corresponding ¹H and ³¹P signal intensity gains determined for L₁-L₄.

A series of control experiments were then undertaken on triphenylphosphine, triphenylphosphine oxide, and phenyl-diethylphosphonate in order to explore the role that the pyridine substituent might play in the hyperpolarization transfer process. These experiments revealed that no polarization transfer took place into the $^1\mathrm{H}$ or $^{31}\mathrm{P}$ nuclei of these materials under the same SABRE conditions as used to hyperpolarize \mathbf{L}_{1-4} described above. Based on this information it is clear that the pyridine motif is necessary for successful PTC under these conditions. This is achieved by the formation of an active SABRE catalyst as typified in Scheme 1. We note that this situation differs from that reported when the related [IrCl-

(H)₂(PPh₃)₃] complex was used to successfully polarize triphenylphosphine at temperatures between 333 and 353 K by Koptyug et al.³⁸ In this case the Ir–P bond proved sufficiently labile that at elevated temperatures SABRE was successful. However, when 1 reacts with an excess of PPh₃ a series of new and rapidly relaxing hydride ligand signals are observed at δ –7.33, –7.98, and –8.59 which are indicative of the formation of [Ir(H)₂(H₂)(IMes)(PPh₃)₂]Cl. The formation of such species precludes the observation of SABRE in these samples because of their role in the rapid quenching of p-H₂ (see the Supporting Information).⁴⁶

As a consequence of the formation of 2_a - L_4 , when the corresponding 1H NMR spectra are examined under SABRE conditions polarized hydride resonances appear as a singlet in the region around δ –22.15 for the pair of chemically equivalent hydride ligands (see the Supporting Information). Characteristic signals for L_4 , as a ligand in 2_a - L_4 , are also seen to be polarized in the associated ^{31}P NMR spectra. We illustrate selected regions of these hyperpolarized ^{1}H and ^{31}P NMR spectra after PTC at 0.5 G, in Figures 2 and 3 respectively to detail some of these effects.

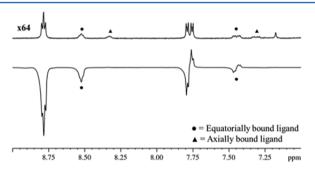


Figure 2. SABRE enhanced 1H NMR spectrum (lower) and corresponding thermal reference spectrum (upper, x64 vertical expansion) of 2_a - L_4 and L_4 (\bullet = equatorial ligand signal, \blacktriangle = axial ligand signal of 2_a - L_4).

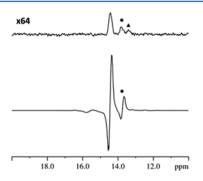


Figure 3. SABRE enhanced ³¹P NMR spectrum (lower) and thermal reference spectrum (upper, x64 vertical expansion) of 2_a - L_4 and L_4 (\bullet = equatorial ligand, \triangle = axial ligand of 2_a - L_4).

The hyperpolarized 1H NMR trace shown in Figure 2 confirms that both of the two pairs of equivalent ring protons of bound L_4 receive magnetization, while those of the axial group remain unaffected. This is manifested in the detection of emission signals for the free analyte at δ 8.76 and for the equatorial ligand at δ 8.50. The magnetic state that produces this emission effect is single spin order in nature and associated with an enhanced but inverted Zeeman population. ⁴⁷ In contrast, the corresponding pair of 3,5-ring protons in L_4

produce antiphase peaks that are separated by 13.5 Hz. This splitting corresponds to the value of ${}^3J_{\rm PH}$ in ${\bf L_4}$ and requires that a heteronuclear two-spin order proton-phosphorus term is created through SABRE. Application of a 90° rf-pulse to the 1H nuclei of this term creates the visible antiphase signal that is seen; the creation of homonuclear proton terms of this type is a well-established characteristic of PHIP. For ${\bf L_4}$, the total 1H signal enhancement was 1499-fold.

The hyperpolarized ³¹P NMR spectrum of the L₄ sample, collected as a single transient, also showed two antiphase peaks, at δ 14.44 and 13.73 for free and equatorially bound L₄ respectively (Figure 3). The signal for free L₄ showed a 545fold increase in size after PTC at 0.5 G relative to the corresponding thermally polarized trace. The separation between the point of maximum displacement and minimum displacement for the signal due to L₄ in this NMR trace is 27.0 Hz (smaller ⁴J_{PH} couplings are hidden within the peak envelope) and twice that of the separation shown in the 1H NMR spectrum. This difference confirms the creation of a proton-phosphorus two spin order term, which leads to a triplet when excited wherein the central feature has zero intensity (-1:0: +1). Related behavior in PHIP derived trihydride complexes have been described previously. 25,48-50 Application of the multinuclear OPSY protocol confirmed this deduction. 35,51

Interestingly, when $2a\text{-}L_4$ is probed by EXSY methods both the axial and equatorial L_4 ligands are observed to undergo exchange with the free ligand pool present in bulk solution. The exchange profile seen for axial- L_4 suggests that there is some incage recombination of this ligand, a process which is not observed for equatorial- L_4 (see the Supporting Information). The axial ligand loss rate is 0.645 \pm 0.010 s $^{-1}$ at 283 K and slightly more rapid than equatorial ligand loss rate which is 0.522 \pm 0.007 s $^{-1}$. The observed hyperpolarization seen in the equatorial- L_4 ^{1}H and ^{31}P signals is therefore a consequence of SABRE prior to ligand loss.

We note that pyridylphosphonates find widespread use within a biological setting as lysophosphatidic acid receptor antagonists, 52,53 nucleotide analogues, 54,55 and enzyme inhibitors. Importantly, related compounds have been shown to exhibit low toxicity in rat studies (LD $_{50}$ of 3.2 g/kg (oral)). As a consequence of these facts and the observation that L4 delivers high levels of visible 31 P hyperpolarization we further optimized the ability of pyridylphosphonates to respond to PTC. This was achieved by the synthesis of the phosphonate derivatives L_5 - L_{12} which are shown in Table 1, alongside the levels of 1 H and 31 P hyperpolarization that results from their analysis under SABRE.

Extension to 2 and 3 Substitution. In order to examine the effect that the regiochemistry of substitution played on the polarization level we replaced the 4- substitution pattern exhibited by L_1 - L_4 with 2- and 3- substitutions in the form of substrates L_5 and L_6 . Polarization transfer into 2-pyridylphosphonate L_5 proved to be unsuccessful, and this is due to the formation of a complex that yields a very broad hydride resonance at δ –31.3 which does not show PHIP. Upon cooling the solution to 233 K this hydride ligand signal separates into two resonances at δ –30.2 and δ –32.1 which confirms that this complex is a dihydride. The corresponding hydride ligand signals yield an average T_1 value of 0.56 s at 253 K and are therefore classically bonded and terminal in character. Nonetheless, the presence of this complex in solution provides a pathway to rapidly destroy the p- H_2 present,

Table 1. 1 H and 31 P NMR Total Signal Intensity Gains for L_{5} - L_{12} in a Methanol- d_{4} Solution under 3 bar of p- H_{2} Achieved through PTC by 1 under SABRE in the Specified Magnetic Field a

Substrate	¹ H ^a	³¹ P ^b	Substrate	¹ H ^a	31 P b
OEt OEt Ls	2	0	OEt OEt L6	2866	336 (0.44)
Eto P OEt OEt	3689	860 (1.12)		1353	48 (0.06)
N O III OEt OEt L9	1174	39 (0.051)	N OEt OEt L ₁₀	1005	60 (0.08)
N O OEt OEt L ₁₁	30	0	POEt OEt L ₁₂	224	87 (0.114)

 $^{\alpha}$ For $^{31}P,$ % polarization indicated in brackets. 5 mM concentrations of 1 with 5 equiv of L_7 and 6 of equivalents of $L_5\text{-}L_{12}$ were employed at (a) 45 G and (b) 0.5 G.

presumably via the formation of a dihydrogen-dihydride complex, with the result that there is very little observable polarization transfer into L_5 .

In contrast, 3-substitution proved to result in good PTC. The measured ¹H signal enhancement at 45 G for L₆ was 2866-fold, and the corresponding ³¹P signal enhancement was 336-fold. Clearly, the change to a 3-substitution pattern has improved the level of ¹H PTC relative to 4-substitution but surprisingly decreased the efficiency for ³¹P transfer. The synthesis of the bis-phosphonate L₇ was therefore undertaken, as this material contains two chemically equivalent ³¹P nuclei and one fewer pyridyl proton than the monosubstituted derivatives. We speculated that better ³¹P-PTC might result as a consequence of this change on the basis that each p-H2 molecule contains a specific amount of latent polarization, and consequently when shared with fewer acceptor sites a net gain in the level of observable transfer might result. The corresponding ¹H and ³¹P signal enhancements for L₇ were 2271- and 741-fold respectively when 6 equiv of the analyte was employed, and therefore this hypothesis is confirmed. However, one further and somewhat unexpected observation was made, the enhanced steric bulk of L7 results in a change in the active complex such that 2_b-L₇ results wherein one coordination site is occupied by chloride (see the Supporting Information). The two hydride signals for this complex appear at δ -23.74 and -24.06.⁵⁹ Because of this change in catalyst form we found that reducing the number of equivalents of L₇ to 5 led to an improvement in PTC, giving ¹H and ³¹P signal enhancements of 3689- and 860fold after transfer at 45 and 0.5 G respectively.

In order to further test the efficiency of polarization transfer to ^{31}P in these systems we added a series of spacers between the pyridyl carbon and the phosphorus center. In L_8 and L_9 this corresponds to the introduction of a $-CH_2-$ group, while for L_{10} it is -NMe- and for L_{11} it is -O-. These analytes were SABRE active, but while the pyridyl ring protons remained strongly enhanced the spacer acted to substantially reduce the levels of ^{31}P polarization. We also prepared a phosphonate using a pyrimidine motif (L_{12}) . This change again results in a

reduction in the number of proton environments able to accept magnetization. While successful, the levels of both ¹H and ³¹P signal enhancement were again lower than those achieved for the pyridylphosphonates.

We now address the potential of these substrates to act as MRI agents for which in-phase ^{31}P signals are desirable. As we have indicated, all of the described ^{31}P signals that were created through PTC in a 0.5 G appear in antiphase. While this effect can be refocused to produce an in-phase signal, we proposed that changing the polarization transfer field (PTF) might achieve a similar result. This would be achieved by changing the relative proportions of single spin (in-phase) and two spin (antiphase) components and result in a predominantly in-phase signal being obtained. A series of experiments were therefore performed in which the PTF was increased using a variable field polarizer 35 and the resulting ^{31}P signal monitored. We completed this process for both L_6 and L_7 (Figure 4).

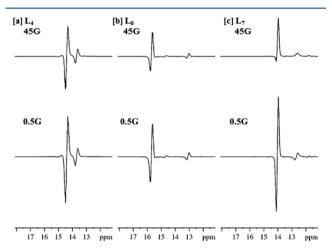


Figure 4. SABRE hyperpolarized ^{31}P NMR traces for L_4 [a], L_6 [b], and L_7 [c] that result after polarization transfer catalysis in a 45 or 0.5 G field.

Significant in-phase contributions were observed after transfer at 45 G without any significant reduction in the overall $^{31}\mathrm{P}$ signal enhancement (216-fold vs 336-fold for L_6 and 703-fold vs 860-fold for L_7 at 45 and 0.5 G respectively). While PTFs above 45 G did result in an increase in the relative single spin contribution, a sharp decline in the overall signal enhancement was noted. For L_6 the relative proportions of the single spin and two spin terms are 0.08:1 and 0.23:1 when the PTF is 0.5 and 45 G respectively. In contrast, when L_7 is examined, the relative proportions are 0.06:1 at 0.5 G and 0.88:1 at 45 G respectively. We conclude from this that L_7 is potentially the better MRI probe for these model studies. This is because the signal for L_4 proved to remain predominately antiphase at both 0.5 G (0.15:1) and 45 G (0.17:1), and it is therefore less suitable for such measurements.

Optimizing the Level of ³¹P **Hyperpolarization.** Further examination of the hyperpolarized ¹H NMR spectra revealed that the signal enhancements described so far also extend into the ethyl groups of the phosphonate. Typically these ¹H signal enhancements were 50-fold per ethyl group and reflect a relayed ¹H-³¹P-¹H transfer process. It was therefore reasoned that deuteration of the phosphonate ethyl esters would increase the level of retained ³¹P polarization as transfer to ²H is less efficient due to the large frequency difference that exists between them. Therefore, we synthesized **d**₁₀-**L**₄, **d**₁₀-**L**₆, and

 d_{20} - L_7 and analyzed them under analogous conditions to their protio-counterparts (Figure 5). The levels of hyperpolarization

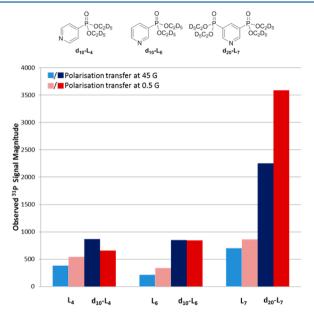


Figure 5. Structure of substrates d_{10} - L_4 , d_{10} - L_6 , and d_{20} - L_7 and their signal intensity gains, alongside those of their protio counterparts L_4 , L_6 , and L_7 , under SABRE after PTC at the specified PTF.

seen in the aromatic proton signals of d_{10} - L_4 , d_{10} - L_6 , and d_{20} - L_7 proved to remain comparable to those of their protio analogues, but the ^{31}P signals were found to dramatically increase in intensity.

Substrate d₂₀-L₇ yielded a ³¹P signal gain of 3588 (2.3% polarization) after PTC at 0.5 G. A 2251-fold signal enhancement (1.4% polarization) is, however, achieved after PTC at 45 G with significant in-phase character being visible in the detected signal. We note this is an order of magnitude larger than previously reported for ³¹P enhancements using this technique and can be achieved without the need for elevated temperatures during the polarization transfer step.³⁸ In a further development, we completed magnetization transfer by inserting the sample into a μ -magnetic shield, which attenuates the Earth's field by a factor of 3000-fold, prior to moving it into the 9.4 T magnet for examination. While this created in-phase ³¹P magnetization,³⁰ the observed signal enhancement decreased to 381-fold (0.497%) under these conditions. It would therefore appear that simply changing the polarization transfer field to a readily accessible value between 0.5 and 130 G reflects the optimal route to achieve in-phase signal. We also note that complexes of type 1 are known to catalyze hydrogen—deuterium exchange,⁶⁰ and, as such, there is the potential to detect 1-proton-PHIP in such analytes. 61,62 Hence, we prepared a sample of d_{20} - L_7 and 1 in methanol- d_4 under 3-bar of H_2 and followed it by ¹H NMR for a period of 10 days. At this point, 90% of the 2,6-protons had been replaced by deuterium. Subsequent SABRE catalysis led to a large reduction in the ³¹P polarization transfer level. If the 31P enhancements that were seen here were due to a 1-proton-PHIP, then the enhancement levels should remain comparable to those observed at day 1. This suggests that when such samples are examined they should be prepared shortly before their use.

Effect of Relaxation on Hyperpolarization Level. We have thus far exemplified the ability to hyperpolarize ³¹P, and

we next evaluated the spin—lattice relaxation times (T_1) of the analytes L_4 , d_{10} - L_4 , L_6 , d_{10} - L_6 , L_7 , and d_{20} - L_7 . The T_1 relaxation values of the Zeeman based ³¹P polarization are 8.3, 7.6, 9.2, 9.3, 5.3, and 6.0 s respectively in methanol- d_4 . For d_{20} - d_7 they increase to 6.5 s in ethanol- d_6 and decrease to 3.6 s in d_2 0. These d_1 0 values, although short, do offer the opportunity to obtain images using the SABRE technique.

Imaging at High Field. With a number of these substrates displaying favorable results we were hopeful that it would be possible to collect MRI with improved spatial resolution and short acquisition times. A series of hyperpolarized samples were therefore interrogated in a 9.4 T vertical bore scanner using the rapid acquisition with relaxation enhancement (RARE) pulse sequence. We employed an echo train length of 32 and a matrix size 32 \times 32 with zero filling to 128 \times 128. This resulted in a total single scan acquisition time of 500 ms. For comparison purposes, thermal images of L_6 , L_7 , and d_{20} - L_7 were recorded with 2048 averages, as shown in Figure 6. Under hyperpolarization conditions, the largest gains in signal-to-noise ratio (SNR) were observed for d_{20} - L_7 (94.4) and reflected a >8-fold improvement on that of the thermal control (11.4). Improvements in SNR of >4-fold were also seen for L_7 .

Similar images were also recorded on a 7 T BioSpec 70/30 preclinical scanner using d₂₀-L₇ as the imaging agent (Figures 6g and 6h). A >6-fold improvement in SNR was achieved in the single scan hyperpolarized image (SNR = 27.7) when compared to the corresponding 2048-average thermal average (SNR = 4.5). We note that insertion of the sample tube into the horizontal scanner caused more residual motion of the sample than was apparent in the analogous measurements on the vertical bore scanner. In addition, the time taken to transfer the sample to the magnet is ca. 5 s longer for the horizontal scanner, and the magnetic fields experienced by the sample during transfer differ. A combination of these effects could account for the reduction in observed SNR that is observed in the hyperpolarized images that were recorded on the BioSpec. Nonetheless, these results confirm the value of these signal enhancements.

CONCLUSIONS

We have demonstrated that the 1 H and 31 P nuclei of a series of pyridyl substituted phosphonate esters, a phosphine, and a phosphine oxide can be efficiently hyperpolarized using SABRE. Substitution at the 3-position delivered the greatest efficiency in polarization transfer catalysis to 1 H and 31 P, compared to substitution at the 2- and 4- positions. In the ligands that showed high performance, the phosphorus center was connected directly to the pyridyl ring such that a strong $^{3}J_{\rm PH}$ coupling of 13.5 Hz exists between protons in the pyridyl ring and the 31 P nucleus.

The phosphonate esters used in this study contained ethyl substituents. These were shown to receive polarization via a relay mechanism which first polarized the pyridyl protons, then the ³¹P nucleus, and finally the ethyl protons. The net gain for the ethyl protons was of the order of 50-fold per ethyl group in comparison to the 2–3 thousand level shown by the aromatic protons. The effect of leakage of the ³¹P magnetization to the ¹H nuclei of these ethyl groups was shown to reduce the level of retained ³¹P polarization.

When the ${}^{1}H$ and ${}^{31}P$ polarization levels are compared between the protio forms of L_4 and L_6 and their deuterated ethyl counterparts d_{10} - L_4 and d_{10} - L_6 , it could be seen that the

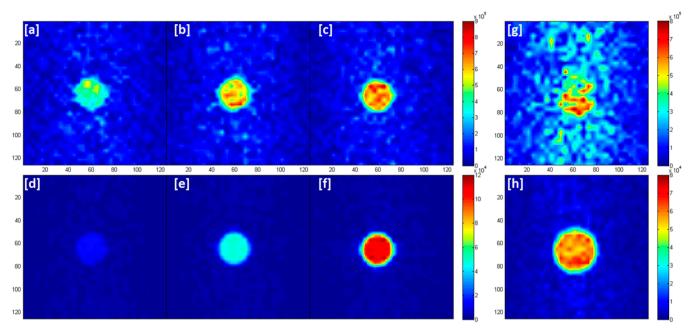


Figure 6. [a]-[c] Thermally polarized ^{31}P images of L_{6} , L_{7} , and d_{20} - L_{7} , respectively collected over 2048 averages. The flip angles used were 86° for L_{6} and 90° for L_{7} and d_{20} - L_{7} . The repetition time was 20 s. [d]-[f] Hyperpolarized images of L_{6} , L_{7} , and d_{20} - L_{7} , respectively collected in a single shot after PTC at 45 G. (TE/FOV/slice thickness were 3.2 ms/4 × 4 cm/5 mm respectively). [g] 2048-Average thermal image and [h] 1 scan hyperpolarized ^{31}P image of d_{20} - L_{7} collected on a 7 T BioSpec 70/30 preclinical scanner. (TE/FOV/slice thickness were 12 ms/3 × 3 cm/10 mm respectively). Samples employed 5 mM concentrations of 1 in methanol- d_{4} (3 mL) with L_{6} (6 equiv), L_{7} or d_{20} - L_{7} (5 equiv) and 3 bar of p-H₂. SNR values were calculated by dividing the mean signal intensity value in the region with signal by the standard deviation of the residual signal seen in an analogously sized noise region.

best levels of ${}^{1}H$ polarization result for three-substituted \mathbf{d}_{10} - \mathbf{L}_{6} . Furthermore, the ³¹P polarization levels are broadly similar for these two substrates at 1%. However, this raw observation masks the fact that for symmetrical d_{10} - L_4 antiphase signal dominates at all monitored polarization transfer fields. In addition, the T_1 values for d_{10} - L_4 are much less than those for four-substituted d_{10} - L_6 in the presence of 2. The result of the small T_1 of d_{10} - L_4 is that it must actually be better ³¹P polarized under SABRE than d_{10} - L_{6} ; we observe the magnetization ca. 3 s after polarization transfer has ceased and relaxation has reduced the hyperpolarized signal amplitude. These observations confirm that 3-substituted pyridines are more suitable candidates for the future rational design of MRI contrast agents that exploit SABRE. Deuterium incorporation into the ethyl groups has also been shown to increase the level of ³¹P signal enhancement.

The symmetric bis-3-5-substituted derivative L₇ was also synthesized, which proved to deliver strong ³¹P polarization. We have shown that this is a consequence of more efficient transfer into the three remaining pyridyl protons and that this synthetic strategy would enable the most atom efficient route to delivering such a pyridyl substituted agent. As a consequence, for d₂₀-L₇, the corresponding ³¹P signal gain exceeds 3500-fold (2.3% polarization) which is an order of magnitude larger than that obtained using other approaches. ^{28,38}

We also prepared systems where a C, N, or O based spacer was introduced to separate the ¹H and ³¹P functionalities. While the ¹H nuclei of the pyridyl groups in these analytes still retained high levels of hyperpolarization, the level of transfer into the ³¹P center was reduced by an order of magnitude. We concluded that the addition of a spacer was therefore undesirable.

In these studies, the process of SABRE was found to result in the creation of two types of ³¹P hyperpolarization. These are single-spin Zeeman magnetization and longitudinal two-spin heteronuclear H-P magnetization. When the corresponding terms were probed by a radio frequency pulse to ³¹P they yield in-phase and antiphase signals, respectively. As an in-phase signal is desirable when investigating these hyperpolarized substrates using traditional MRI sequences we demonstrated that if the magnetic field experienced by the sample at the point of polarization transfer catalysis is 45 G, optimal in-phase ³¹P signal enhancement results under SABRE in these systems. Interrogation of the resulting samples on a 9.4 T vertical bore scanner proved to allow images to be collected in a single scan at a 5 mM concentration. The SNR of a typical image was 94.4 and far exceeds that of a similar 12 h measurement under normal conditions which employed 2048 averages. Although the T_1 values for these ³¹P signals proved to be around 6 s, we have demonstrated that we can obtain good images on phantoms. We are working toward creating further phosphorus containing molecules that have longer T_1 values which might ultimately be used for in vivo applications. We expect that the methods of Levitt and Warren will be particularly relevant here.^{63–67}

ASSOCIATED CONTENT

S Supporting Information

Full synthetic procedures and spectroscopic data, NMR and MRI polarization transfer experiments, T_1 data and characterization data. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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