CHEMISTRY OF MATERIALS

Addition/Correction

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Correction to Kinetic Study of the Electrochemical $FePO_4$ to $LiFePO_4$ Phase Transition

Jan L. Allen,* T. Richard Jow, and Jeffrey Wolfenstine

Chem. Mater. **2007**, *19*, 2108−2111. DOI: 10.1021/cm062963o

Pages 2110−2111. The reported value of the activation energy is incorrect owing to a calculation error. The correct activation energy determined from the slope of the curve in Figure 3 should be 29 kJ/mol not 13 kJ/mol as originally published. This correction does not affect the discussion/conclusion of the original article that the activation energy is low for a solid−solid transformation. However, while this activation still implies a low energy pathway for Li ion diffusion and it is lower than that reported by Takahashi et al. (ref 23 in the original article) for lattice Li ion diffusion in $LiFePO_4$ (39 kJ/mol), the corrected activation energy is less consistent with a phase boundary mechanism for Li ion diffusion suggested in the original article. A corrected Figure 3 is shown.

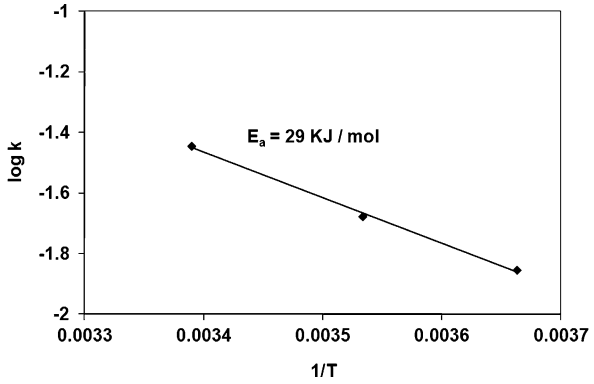

Figure 3. Activation energy was obtained from this Arrhenius plot of the electrochemical lithiation of $FePO_4$ to form $LiFePO_4$, using the values of the Avrami rate constant parameter obtained from Figure 2. The trend line is a linear best fit of the data points. The slope of the line was used to obtain the activation energy for this process.

Received: March 7, 2012
Accepted: March 9, 2012
Published: March 22, 2012