

Correction to The Vaporization Enthalpies and Vapor Pressures of Two Insecticide Components, Muscalure and Empenthrin, by Correlation Gas Chromatography

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Te have detected several errors in referencing and transposition in Tables 8 and 9 of our manuscript

(4) van Genderen, A. C. G.; van Miltenburg, J. C.; Bolk, J. G.; van Bommel, M. J.; Ekeren, P. J.; van den Berg, G. J. K.; Oonk, H. A. J.

Table 8. Parameters of the Third Order Polynomial, Equation 5, and Predicted Normal Boiling Temperatures

	$A \cdot 10^{-8} / T^{3}$	$B \cdot 10^{-6} / T^{2}$	C/T	D	$T_{\rm nb}/{ m K}$ cal
run 1					
nonadecane	1.754	-2.5667	-315.614	6.741	605
eicosane	1.9005	-2.7565	-155.328	6.657	618
heneicosane	2.0495	-2.9483	11.927	6.562	631
docosane	2.195	-3.1366	171.15	6.478	644
Z 9-tricosene	2.3635	-3.3132	508.541	6.048	656
E 9-tricosene	2.3393	-3.2956	431.078	6.173	656
tetracosane	2.481	-3.5092	476.514	6.329	667
runs 3 and 4					
methyl dodecanoate	1.8735	-2.2983	8.943	6.730	537.5
empenthrin 1	2.2259	-2.7594	598.286	6.288	567.9
empenthrin 2	2.2612	-2.7886	703.209	6.144	569.2
methyl pentadecanoate	2.5324	-3.085	1314.607	5.460	587
methyl hexadecanoate	2.7423	-3.3388	1721.331	5.075	603
methyl octadecanoate	3.152	-3.8363	2504.867	4.344	635
ethyl octadecanoate	3.2679	-3.988	2696.756	4.202	644
methyl nonadecanoate	3.3509	-4.08	2879.859	4.001	650

(The Vaporization Enthalpies and Vapor Pressures of Two Insecticide Components, Muscalure and Empenthrin, by Correlation Gas Chromatography. J. Chem. Eng. Data 2013, 58 (12), 3513-3520). We have also added more detail. The changes are highlighted in bold. For convenience, the references cited are also included. We apologize for any inconvenience this may have caused.

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Table 9. A Summary of Liquid/Subcooled Liquid Vapor Pressures and Normal Boiling Temperatures and Comparison with Experimental or Estimated Values (in italics)

	•				
	$(10^4) \cdot p_{(1)} / \text{Pa} (298.15) \text{ K calc}$		$(10^4) \cdot p_{(1)}/P$		
	run 1	run 2	298.15 K, lit	$T_{\rm nb}/{ m K}$, calc	$T_{ m nb}/{ m K}$, lit
nonadecane	64.3 ± 5		64.5 ^a	605	604 ^a
eicosane	20.8 ± 1	20.9 ± 1	20.9 ^a	618	617 ^a
heneicosane	6.7 ± 0.2	6.6 ± 0.4	6.7 ^b	631	630 ^c
docosane	2.2 ± 0.2	2.2 ± 0.1	2.2^{b}	644	642 ^c
Z 9-tricosane,	1.2 ± 0.1	1.1 ± 0.1	85 ^d , 4.21 ^e , 47 ^k	449 ^f	453 ^f
E 9-tricosene	1.2 ± 0.1	1.0 ± 0.1	4.21 ^e	656	na ^g
tetracosane	0.23 ± 0.02	0.24 ± 0.2	0.24^{b}	667	664 ^h
	(10 ⁴)·p ₍₁₎ /Pa (298.15) K calc		$(10^4) \cdot p_{(1)}/P$		
	r	runs 3 and 4	298.15 K, lit	$T_{\rm nb}/{ m K}$, calc	$T_{ m nb}/{ m K}$, lit
methyl dodecanoate		6060 ± 3500	5850.i	537.5	540 ^j
empenthrin 1		594 ± 380	$141^{k,l}$, 230^l	567.9	568.7 ^m
empenthrin 2		601 ± 383	$141^{k,l}$, 230^l	569.2	568.7 ^m
methyl pentadecanoate		233 ± 155	279^{i}	418.9^{h}	414.2^{h}
methyl hexadecanoate		79 ± 55	71^i	455.6 ^j	458.2^{j}
methyl octadecanoate		9.2 ± 7.0	8.0^{i}	457 ⁿ	455.2 ⁿ
ethyl octadecanoate		4.3 ± 3.4	40 ^e	449.4°	443.2°
methyl nonadecanoate		3.1 ± 2.5	3.5 ⁱ	650	na ^g

^aReference 3. ^bReference 2. ^cReference 6. ^dReference 7. ^eReference 8. ^fBoiling temperature at p/Pa = 133, ref 9. ^gNot available. ^hBoiling temperature at p/Pa = 400; ref 10. ⁱReference 4. ^fBoiling temperature at p/Pa = 1.33; ref 10. ^kAt T/K = 296.8; ref 5. ^lReference 1. ^mReference 11. ⁿBoiling temperature at p/Pa = 533; ref 10. ^oBoiling temperature at p/Pa = 267, ref 12.

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