



Correction to Intracellular Organic Matter from Cyanobacteria as a Precursor for Carbonaceous and Nitrogenous Disinfection Byproducts

Eric C. Wert* and Fernando L. Rosario-Ortiz

Environ. Sci. Technol. 2013, 47, 6332-6340; DOI: 10.1021/es400834k

In the section entitled "Nitrosamine Formation during Chloramination" the following statement was made with reference to NPYR and NPIP: "These nitrosamines contain the heterocyclic aromatic compounds pyrrole (five-member heteroring) and pyridine (six-membered heteroring), respectively". The structures of NPYR and NPIP do not contain the aromatic structures cited here (pyrrole and pyridine). Furthermore, the statement: "Although ozone reacts slowly with pyridine ($k = 0.5-3 \, \mathrm{M}^{-1} \, \mathrm{s}^{-1}$)59, hydroxyl radicals react rapidly with pyrrole ($k = 10^{10} \, \mathrm{L/mol/s}$) and pyridine ($k = 10^9 \, \mathrm{L/mol/s}$) functional groups found in NPYR and NPIP⁶⁰, possibly explaining the decrease in NPYR and NPIP during the $O_3/\mathrm{NH}_2\mathrm{Cl}$ process." should be disregarded as these aromatic structures are not present in NPYR and NPIP.

