

Liquid–Liquid Equilibria for Monoethylene Glycol + Water + Alkane Systems in the Range (273 to 313) K and Atmospheric Pressure. Philippe Sentenac, Charles Berro, Evelynne Rauzy,* Ilham Mokbel, and Jacques Jose, *J. Chem. Eng. Data* **2004**, *131*, 1577–1580.

The authors of “Liquid–Liquid Equilibria for Monoethylene Glycol + Water + Alkane Systems in the Range (273 to 313) K and Atmospheric Pressure” (*J. Chem. Eng. Data* **2004**, *131*, 1577–1580) declare that the experimental solubilities published there are completely erroneous. The reader is invited to consult another article (DOI: 10.1021/je900295u) that reports new data over the temperature range (283 to 323) K for two of the eight ternary systems appearing in *J. Chem. Eng. Data* **2004**, *131*, 1577–1580, namely, monoethylene glycol + water + *n*-hexane/2,2,4-trimethylpentane. In addition, the article reports data for four binary systems, namely, monoethylene glycol + *n*-hexane/2,2,4-trimethylpentane and water + *n*-hexane/2,2,4-trimethylpentane.

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Isothermal Binary Vapor–Liquid Equilibrium for 2-Methylpropane and *n*-Butane with 1,2-Ethanedithiol and 2-Methyl-2-propanethiol. Anne Penttilä,* Petri Uusi-Kyyny, Juha-Pekka Pokki, Minna Pakkanen, and Ville Alopaeus, *J. Chem. Eng. Data* **2010**, *55*, 291–296.

The temperature of the refractive index of 1,2-ethanedithiol and 2-methyl-2-propanethiol in Table 1 in the above paper was not reported. The temperature at which the refractive index was measured has been added to the heading of the table.

Table 1. Materials, Their Purities, and Refractive Indexes Measured at $T = 298.15$ K

component	supplier	GC analysis	refractive index		
		(mass fraction)	this work	ref 4	ref 5
2-methylpropane	Oy Aga Ab	0.9995			
<i>n</i> -butane	Oy Aga Ab	0.9995			
1,2-ethanedithiol	Fluka	0.9976	1.5562	1.5562	
2-methyl-2-propanethiol	Aldrich	0.9987	1.4200	1.42004	1.4200

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