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Correction to *P*, *ρ*, *T* Measurements and Isobaric Vapor—Liquid Equilibria of the 1,3,3-Trimethyl-2-oxabicycle[2,2,2]octane + Propan-1-ol Mixture: Cubic and Statistical Associating Fluid Theory-Based Equation of State Analysis [*Journal of Chemical & Engineering Data* **2010**, *55*, 5932–5940 DOI: 10.1021/je100577v]. Marcos Torcal, Sandra García-Abarrio, Juan I. Pardo,* Ana M. Mainar, and José S. Urieta

In the original document (*J. Chem. Eng. Data* **2010**, *55* (12), 5932–5940), page 5937, Table 7 and page 5938, Table 9 contained errors in the data concerning 1,8-cineole. The corrected data are reported.

Table 7. Coefficients for the Equations of Vapor Pressure. The Equation of Antoine Provides $P^{\rm sat}/{\rm kPa}$ Using $T/{\rm K}$. Capital Letters: Equation of Antoine; Lower Case Letters: Wagner Equation

		A/a	B/b	C/c	d	$T_{\rm min}/{\rm K}$	$T_{\rm max}/{ m K}$
	1,8-cineole	13.9128	3523.60	-70.03		326.65	449.15
	1-propanol^a	-8.53706	1.96214	-7.6918	2.945		$T_{\rm c}^{\ a}$
а	Ref 14.						

Table 9. Properties and Parameters of the Pure Components Used for the Application of the EOS

PR and PT		$T_{\rm c}/{ m K}$		$P_{\rm c}/{ m MPa}$		ω
1,8-cineole propan-1-ol		661.12 ^a 536.98 ^c		3.019 ^a 5.175 ^c		0.338^{b} 0.629^{c}
SAFT	m	$v^{\circ \circ}/\mathrm{L} \cdot \mathrm{mol}^{-1}$	$u^0/k/K$	К	$\varepsilon/k/K$	range T/K
1,8-cineole propan-1-ol	5.2966 3.240 ^d	0.015698 0.0120^d	250.71 225.68 ^d	$0 \\ 0.01968^d$	$0 \\ 2619^d$	$326.65 - 449.15$ $293 - 493^d$
PC-SAFT	m_i	$\sigma_i/ ext{Å}$	$\varepsilon_i/k/{ m K}$	$k^{A_iB_i}$	$arepsilon^{A_iB_i}/k/{ m K}$	range T/K
1,8-cineole propan-1-ol	3.4254 2.9997 ^e	4.0597 3.2522 ^e	291.37 233.40 ^e	0 0.015268 ^e	0 2276.8^e	326.65-449.15 240-537 ^e

^a Joback's method, ref 35. ^b Lee–Kesler method, ref 36. ^c Ref 13. ^d Ref 38. ^e Ref 42.

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