

Temperature Effect on Activation Rate Constants in ATRP: New Mechanistic Insights into the Activation Process [Macromolecules 2009, 42, 6050]. Florian Seeliger and Krzysztof Matyjaszewski*

Page 6053. Values of the rate constants in column 2 of Table 2 were erroneously referred to −20 °C instead of 20 °C. Table 2 with values of all rate constants at 20 °C should read as follows:

Table 2. Activation Rate Constants, Activation Free Energy (20 °C), and Eyring and Arrhenius Parameters for the Reactions of Alkyl Halides with Cu(I)Br/PMDETA in MeCN

	$k_{\rm act}(20 {}^{\circ}{\rm C}) ({\rm L \; mol^{-1} \; s^{-1}})$	ΔG^{\ddagger} (20 °C) (kJ mol ⁻¹)	$\Delta H^{\ddagger} (\text{kJ mol}^{-1})$	$\Delta S^{\ddagger} (J \text{ mol}^{-1} \text{ K}^{-1})$	$E_{\rm a}({\rm kJ~mol^{-1}})$	ln A
BrPN	$2.05 \times 10^{1 \ a}$	64.4	26.0 ± 0.9	-131 ± 4	28.1 ± 0.9	14.6 ± 0.4
BrAN	3.91	68.6	28.7 ± 0.6	-136 ± 2	30.9 ± 0.6	14.0 ± 0.2
MBriB	2.38	69.5	25.2 ± 0.9	-151 ± 3	27.5 ± 0.9	12.1 ± 0.4
EtBriB	1.75	70.2	25.3 ± 0.8	-153 ± 3	27.5 ± 0.8	12.0 ± 0.3
MBrP	1.34×10^{-1}	76.5	30.8 ± 0.6	-156 ± 2	33.2 ± 0.6	11.7 ± 0.2
ClPN	1.19×10^{-1}	77.2	32.9 ± 1.4	-151 ± 5	35.3 ± 1.3	12.2 ± 0.6
PEBr	9.73×10^{-2}	77.4	34.0 ± 1.0	-148 ± 3	36.4 ± 1.0	12.6 ± 0.4
ClAN	5.89×10^{-2}	78.8	33.4 ± 0.8	-155 ± 3	35.8 ± 0.8	11.7 ± 0.3
AllBr	3.80×10^{-2}	79.7	38.7 ± 1.3	-140 ± 4	41.1 ± 1.3	13.6 ± 0.5

^a Value was calculated from Eyring parameters and eq 3.

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