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Addition/Correction

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Corrections to Isomerization of an Enantiomerically Pure Phosphorus-Bridged [1]Ferrocenophane

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By calculating the equilibrium constant $K°$ from the $\Delta G°$ value (in kcal/mol), unfortunately, we used kJ instead of kcal for the RT term. The correct $K°$ value is 5.0×10^2 instead of the published value of 4.43. Second, the correct $\Delta G°$ value is −3.69 kcal/mol instead of the published value of −3.68 kcal/mol. These errors do not have a bearing on the conclusions of the publication, but the following statements need to be corrected.

(1) The second to last sentence of the abstract should be corrected to "According to DFT calculations, the equilibrium constant $K°$ for **4**-$\mathbf{C}_1 \rightleftharpoons$ **4**-$\mathbf{C}_s$ is 5.0×10^2 (ΔE^{SCF} = −3.74 kcal/mol; $\Delta H°$ = −3.81 kcal/mol; $\Delta G°$ = −3.69 kcal/mol)."

(2) The last sentence of the first paragraph in the right column on page 3510 should be corrected to "The equilibrium constant $K°$ was calculated as 5.0×10^2, showing that the equilibrium mixture at 25 °C should be dominated by isomer **4**-$\mathbf{C}_s$ (ΔE^{SCF} = −3.74 kcal/mol; $\Delta H°$ = −3.81 kcal/mol; $\Delta G°$ = −3.69 kcal/mol)."

(3) The $\Delta G°$ value at the end of the first paragraph of the Summary and Conclusions on page 3511 should be corrected to −3.69 kcal/mol.

The authors apologize to the readers for these errors.

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