

See discussions, stats, and author profiles for this publication at: <https://www.researchgate.net/publication/5314070>

Fluorous Synthesis of Hydantoin-, Piperazinedione-, and Benzodiazepinedione-Fused Tricyclic and Tetracyclic Ring Systems

ARTICLE *in* EUROPEAN JOURNAL OF ORGANIC CHEMISTRY · MAY 2006

Impact Factor: 3.07 · DOI: 10.1002/ejoc.200600077 · Source: PubMed

CITATIONS

34

READS

27

5 AUTHORS, INCLUDING:



[Wei Zhang](#)

University of Massachusetts Boston

188 PUBLICATIONS 3,709 CITATIONS

[SEE PROFILE](#)



[Steven J Geib](#)

University of Pittsburgh

300 PUBLICATIONS 8,327 CITATIONS

[SEE PROFILE](#)

Published in final edited form as:

European J Org Chem. 2006 ; (9): 2055–2059.

Fluorous Synthesis of Hydantoin-, Piperazinedione-, and Benzodiazepinedione-Fused Tricyclic and Tetracyclic Ring Systems

Wei Zhang^[a], Yimin Lu^[a], Christine Hiu-Tung Chen^[b], Dennis P. Curran^[b], and Steven Geib^[b]

[a] *Fluorous Technologies, Inc. University of Pittsburgh Applied Research Center, 970 William Pitt Way, Pittsburgh, Pennsylvania 15238, USA, Fax: +412-628-3053, E-mail: w.zhang@fluorous.com*

[b] *Department of Chemistry, University of Pittsburgh, Pittsburgh, Pennsylvania 15260, USA*

Abstract

Fluorous proline derivatives generated from one-pot, three-component [3+2] cycloaddition of azomethine ylides are employed for different post-condensation reactions to form hydantoin-, piperazinedione-, and benzodiazepinedione-fused tricyclic and tetracyclic ring systems. The high synthetic efficiency is achieved by conducting fast microwave reactions and easy fluorous-solid phase extractions for reaction mixture purifications. Methods developed for these novel drug-like heterocyclic compounds can be applied to diversity-oriented library synthesis.

Keywords

Fluorous synthesis; Microwave reaction; Solid-phase extraction; [3+2] Cycloaddition; Diversity-oriented synthesis

Introduction

Fluorous synthesis employs perfluoroalkyl (Rf) chains as “phase tags”¹ to improve the efficiency of reaction mixture purifications.² This technology shares the characteristics of solution-phase synthesis, which has homogenous reaction environment,³ easy intermediate analysis,⁴ and good compatibility to other synthetic techniques such as microwave⁵ and multicomponent reactions.⁶ Compared to its counterpart solid-phase synthesis, fluorous synthesis requires less development time and has the capability to explore new reactions on fluorous support directly.⁷ As a “beadless” synthetic technology, fluorous synthesis has been applied to parallel and mixture synthesis⁸ of small molecules, peptides,⁹ and oligosaccharides.¹⁰

We have recently developed several methods for synthesis of heterocyclic systems by using an orchestrated sequence of microwave-assisted fluorous multicomponent reaction (F-MCR) and fluorous-solid phase extraction (F-SPE) to speed up reactions and simplify purifications.^{6,11} Reported in this paper are approaches to three novel triaza tricyclic and tetracyclic ring systems **2–4** (Scheme 1). Proline derivatives **1** generated from one-pot, three-component [3+2] cycloaddition¹² of azomethine ylides are further converted to hydantoin-, piperazinedione-, and benzodiazepinedione-fused compounds **2–4**, respectively. Each of these

Supporting Information General experimental procedures and analytical data for representative intermediates and all final products are provided.

three heterocyclic scaffolds has four stereocenters on the central pyrrolidine ring and up to four points of diversity (R^1 to R^4). Compound **2** has a similar ring skeleton as tricyclic thrombin inhibitors.¹³ The structure of compound **3** is partially related to diketopiperazine-based inhibitors of human hormone-sensitive lipase.^{14,15} Compound **4** contains a privileged benzodiazepine moiety which has a wide range of pharmaceutical utilities.¹⁶

Results and Discussion

Preparations of fluorous amino esters **5** and one-pot, three-component 1,3-dipolar cycloaddition reactions were conducted by following established procedures.^{6a,b} Thus a mixture of 1.0 equiv of a fluorous aminoester, 1.2 equiv of a benzaldehyde, 1.5 equiv of an *N*-alkylmaleimide, and 3 equiv of Et₃N in DMF was heated under microwave at 130 °C for 20 min to afford proline derivative **1** (Scheme 2).^{17,18} Since the fluorous amino ester **5** was used as the limiting agent, only the desired product **1** was expected to be fluorous. The crude product was loaded on a FluoroFlash cartridge. The non-fluorous components such as unreacted aldehyde, *N*-alkylmaleimide, and Et₃N salt were eluted out with a fluorophobic solvent (80:20 MeOH-H₂O). Fluorous compound **1** was collected by eluting with MeOH, a more fluorophilic solvent. After F-SPE purification, the purity of the product is usually greater than 90% by ¹H NMR analysis (Figure 1). Bicyclic prolins **1** with different R^1 – R^3 substitution groups were synthesized in 75–90% yields. The stereochemistry of compound **1a** was established based on the literature information^{17c,17g} and confirmed by single-crystal X-ray diffraction (Figure 2, left). No evidence shows the racemization of the amino acid **5** during the cycloaddition.

With the key intermediates **1** in hands, we then performed post-condensation reactions to generate different heterocyclic ring systems. The reaction of **1** with 5 equiv of a phenylisocyanate or a phenylthioisocyanate in the presence of catalytic amount of *N,N*-4-dimethylaminopyridine (DMAP) in toluene gave urea or thiourea **6**. After F-SPE purification, compound **6** was mixed with K₂CO₃ and heated under microwave at 100 °C for 5 min. Fluorous tag cleavage and hydantoin ring formation produced tricyclic compound **2** (Scheme 3). Four analogs of **2** were produced in 75–85% yields. After F-SPE followed by HPLC purifications, the products had greater than 95% purities. The stereochemistry of compound **2a** was confirmed by single-crystal X-ray diffraction (Figure 2, right).

In the synthesis of piperazinedione-fused tricyclic compounds **3a** and **3b** (Scheme 4), direct *N*-acylations of **1a** with α -aminoacids or α -aminoacid chlorides were attempted, but reactions gave products in very low yields (10–25%). Acylation of **1a** with chloroacetyl chloride followed by chlorine displacement with BuNH₂ or 3,5-dimethylaniline gave compounds **8a** and **8b** in 92% and 90% yields, respectively. The detag/cyclization reactions were promoted by 1,8-diazabicyclo[4.3.0]non-5-ene (DBU) under microwave irradiation at 180 °C for 15 min to give product **3a** in 45% yield. However, under the same conditions, only a very small amount of **3b** (<5%) was detected from the reaction mixture by LCMS.

Synthesis of benzodiazepine-fused tricyclic compounds **4a–c** were accomplished by a three-step reaction sequence (Scheme 5). *N*-acylation of **1** with 2-nitrobenzoyl chloride gave acylation product **9**. We have found that the *N*-acylation reaction was sensitive to the R^1 substitution; only small R^1 groups such as H and Me gave products in good yields. Compounds **9** were then reacted with zinc dust in acetic acid under sonication to reduce the nitro group and form **10**. The cyclative tag cleavage of compounds **10** with DBU produced tricyclic compound **4a–c** in 45–58% yields.

Conclusion

In summary, we have developed synthetic routes to three triaza tricyclic and tetracyclic rings systems using the common intermediates generated by [3+2] cycloaddition of azomethine ylides. Microwave-assisted fluororous synthesis speeds up reactions and simplifies product purifications. These heterocyclic compounds with ring skeleton, stereochemistry, and substitution variations are good candidates for diversity-oriented synthesis.

Supplementary materials

Refer to Web version on PubMed Central for supplementary material.

Acknowledgements

This work was supported by the National Institute of General Medical Sciences SBIR Grants (2R44GM062717-02 and 2R44GM067326-02). We thank Professor Peter Wipf and Dr. John Hodges for helpful suggestions and discussions.

References

1. Curran DP. *Angew Chem Int Ed Eng* 1998;37:1175–1196.
2. Selected reviews and monographs on fluororous synthesis: Gladysz JA, Curran DP, Horvath IT, *Handbook of Fluororous Chemistry* Wiley-VCH, Weinheim, 2004b) Zhang W. *Chem Rev* 2004;104:2531–2556. [PubMed: 15137799] c) Zhang W. *Curr Opin Drug Disc Dev* 2004;7:784–797. d) Pozzi G, Shepperson I. *Coord Chem Rev* 2003;242:115–124. e) Zhang W. *Tetrahedron* 2003;59:4475–4489. f) Dobbs AP, Kimberley MR. *J Fluorine Chem* 2002;118:3–17. g) Tzschucke CC, Markert C, Bannwarth W, Roller S, Hebel A, Haag R. *Angew Chem Int Ed* 2002;41:3964–4000. h) Barthel-Rosa LP, Gladysz JA. *Coordination Chem Rev* 1999;190–192:587. i) Horvath IT. *Acc Chem Res* 1998;31:641–650. j) Studer A, Hadida S, Ferritto SY, Kim PY, Jeger P, Wipf P, Curran DP. *Science* 1997;275:823–826. [PubMed: 9012347] k) Horvath IT, Rabai T. *Science* 1994;266:72–76. [PubMed: 17814001]
3. Chen CHT, Zhang W. *Mol Diversity* 2005;9:353–359.
4. a) Curran DP. *Synlett* 2001:1488–1496. Curran, DP. *Handbook of Fluororous Chemistry*. Gladysz, JA.; Curran, DP.; Horvath, IT., editors. Wiley-VCH; Weinheim: 2004. p. 101
5. a) Zhang W, Nagashima T, Lu Y, Chen CHT. *Tetrahedron Lett* 2004;45:4611–4613. b) Zhang W, Chen CHT, Lu Y, Nagashima T. *Org Lett* 2004;6:1473–1476. [PubMed: 15101770] c) Zhang W, Lu Y, Chen CHT. *Mol Diversity* 2003;7:199–202. d) Olofeeson K, Kim SY, Larhed M, Curran DP, Hallberg A. *J Org Chem* 1999;64:4539–4541. e) Larhed M, Hoshino M, Hadida S, Curran DP, Hallberg A. *J Org Chem* 1997;62:5583–5587.
6. a) Lu Y, Zhang W. *Mol Diversity* 2005;9:91–98. b) Zhang W, Chen CHT. *Tetrahedron Lett* 2005;46:1807–1810. [PubMed: 18079977] c) Lu Y, Zhang W. *QSAR Comb Sci* 2004;23:827–835. d) Zhang W, Tempest P. *Tetrahedron Lett* 2004;45:6757–6760.
7. Zhang W, Lu Y, Geib S. *Org Lett* 2005;7:2269–2272. [PubMed: 15901186]
8. Selected papers on fluororous mixture synthesis: Zhang Q, Curran DP. *Chem Eur J* 2005;11:4866–4880. b) Zhang W. *Arkivoc* 2004:101–109. [PubMed: 18490966] c) Zhang W, Luo Z, Chen CHT, Curran DP. *J Am Chem Soc* 2002;124:10443–10450. [PubMed: 12197746] d) Luo Z, Zhang Q, Oderaotshi Y, Curran DP. *Science* 2001;291:1766–1769. [PubMed: 11230688]
9. a) Goto K, Miura T, Hosaka D, Matsumoto H, Mizuno M, Ishida H-k, Inazu T. *Tetrahedron* 2004;60:8845. b) Montanari V, Kumar K. *J Am Chem Soc* 2004;126:9528–9529. [PubMed: 15291542] c) Mizuno M, Goto K, Miura T, Hosaka D, Inazu T. *Chem Commun* 2003:972. d) de Visser PC, van Helden M, Filippov DV, van der Marel GA, Drijfhout JW, van Boom JH, Noort D, Overkleeft HS. *Tetrahedron Lett* 2003;44:9013–9016. e) Filippov DV, van Zoelen DJ, Oldfield SP, van der Marel GA, Overkleeft HS, Drijfhout JW, van Boom JH. *Tetrahedron Lett* 2002;43:7809–7812.
10. a) Manzoni L, Castelli R. *Org Lett* 2004;6:4195–4198. [PubMed: 15524441] b) Jing Y, Huang X. *Tetrahedron Lett* 2004;45:4615–4618. c) Miura T, Goto K, Hosaka D, Inazu T. *Angew Chem Int Ed* 2003;42:2047–2051. d) Manzoni L. *Chem Commun* 2003:2930–2931. e) Palmacci ER, Hewitt MC, Seeberger PH. *Angew Chem Int Ed* 2001;40:4433–4437. f) Curran DP, Ferritto R, Hua Y. *Tetrahedron Lett* 1998;39:4937–4940.

11. Nagashima T, Zhang W. *J Comb Chem* 2004;6:942–949. [PubMed: 15530122]
12. Recent reviews and monographs on [3+2] cycloadditions of azomethine ylides: Coldham I, Hufton R. *Chem Rev* 2005;105:2765–2809. [PubMed: 16011324] b) Ruck-Braun K, Freysoldt THE, Wierschem F. *Chem Soc Rev* 2005;34:507–516. [PubMed: 16137163] c) Harju K, Yli-Kauhaluoma J. *Mol Diversity* 2005;9:187–207. Padwa, A.; Pearson, WH., editors. *Synthetic Application of 1,3-Dipolar Cycloaddition Chemistry toward Heterocycles and Natural Products*. Wiley; Hoboken: 2003. e) Najera C, Sansano JM. *Curr Org Chem* 2003;7:1105–1150. Albers, M.; Meyer, T. *Handbook of Combinatorial Chemistry*. Nicolaou, KC.; Hartwig, W., editors. 1. Wiley-VCH; Weinheim: 2002. p. 453 f) Kantorowski EJ, Kurt MJ. *Mol Diversity* 1996;2:207–216.
13. Olsen J, Seiler P, Wagner B, Fisher H, Tschopp T, Obst-Sander U, Banner DW, Kansy M, Muller K, Diederrich F. *Org Biomol Chem* 2004;2:1339–1352. [PubMed: 15105924]
14. Slee DH, Bhat AS, Nguyen TN, Kish M, Lundeen K, Newman MJ, McConnell SJ. *J Med Chem* 2003;46:1120–1122. [PubMed: 12646020]
15. For related tricyclic piperazinedione ring systems, see Furutsuka K, Hayashi H, Shiono Y. *J Nat Prod* 1999;62:315–317. [PubMed: 10075772] b) Roe JM, Webster RAB, Ganesan A. *Org Lett* 2003;5:2825–2827. [PubMed: 12889884] c) Ley SV, Cleator E, Hewitt PR. *Org Biomol Chem* 2003;1:3492–3494. [PubMed: 14599007] and references cited therein.
16. a) Kamal A, Reddy KL, Devaiah V, Shankaraiah N, Reddy DR. *Mini-Rev Med Chem* 2006;6:53–68. [PubMed: 16457632] b) Horton DA, Bourne GT, Smythe ML. *Chem Rev* 2003;103:893–930. [PubMed: 12630855] c) Grieder A, Thomas AW. *Synthesis* 2003:1707–1711. d) Boojamra C, Burow KM, Thompson LA, Ellman JA. *J Org Chem* 1997;62:1240–1256. and references cited there in
17. Solid-supported [3+2] cycloaddition reactions of azomethine ylides: Komatsu M, Okada H, Akaki T, Oderaotashi Y, Minakata S. *Org Lett* 2002;4:3505–3508. [PubMed: 12323055] b) Ganguly AK, Seah N, Popov V, Wang CH, Kuang R, Saksena AK, Pramanik BN, Chan TM, McPhail AT. *Tetrahedron Lett* 2002;43:8981–8983. c) Hoveyda HR, Hall DG. *Org Lett* 2001;3:3491–3494. [PubMed: 11678690] d) Barrett AGM, Boffey RJ, Frederiksen MU, Newton CG, Roberts RS. *Tetrahedron Lett* 2001;42:5579–5581. e) Dondas HA, Grigg R, MacLachlan WS, MacPherson DT, Markandu J, Sridharan V, Suganthan S. *Tetrahedron Lett* 2000;41:967–970. f) Peng G, Sohn A, Gallop MA. *J Org Chem* 1999;64:8342–8349. [PubMed: 11674757] g) Bicknell A, Hird NW. *Bioorg Med Chem Lett* 1996;6:2441–2444. h) Hamper BC, Dukeshere DR, South MS. *Tetrahedron Lett* 1996;37:3671–3674. i) Hollinshead SP. *Tetrahedron Lett* 1996;37:9157–9160.
18. Fluorous silane-assisted [3+2] cycloaddition of azomethine ylides Komatsu M, Okada H, Yokoi S, Minakata S. *Tetrahedron Lett* 2003;44:1603–1606.

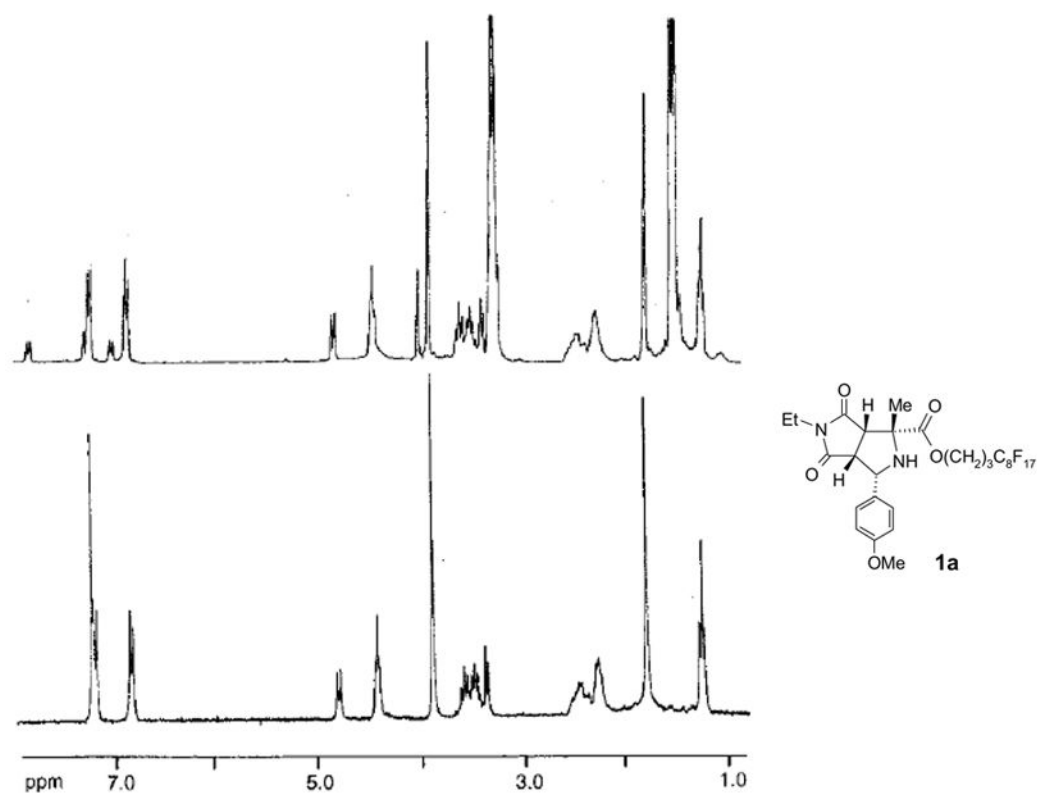


Figure 1.
¹H NMR (in CDCl₃) analysis of compound **1a**, before (top) and after (bottom) F-SPE.

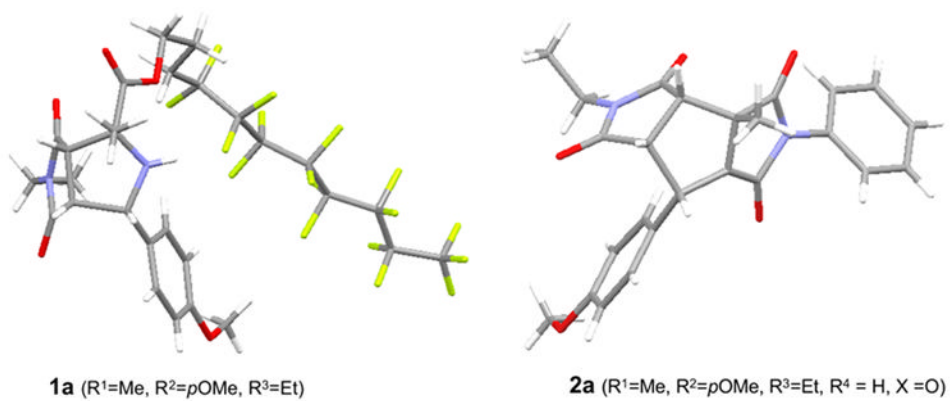
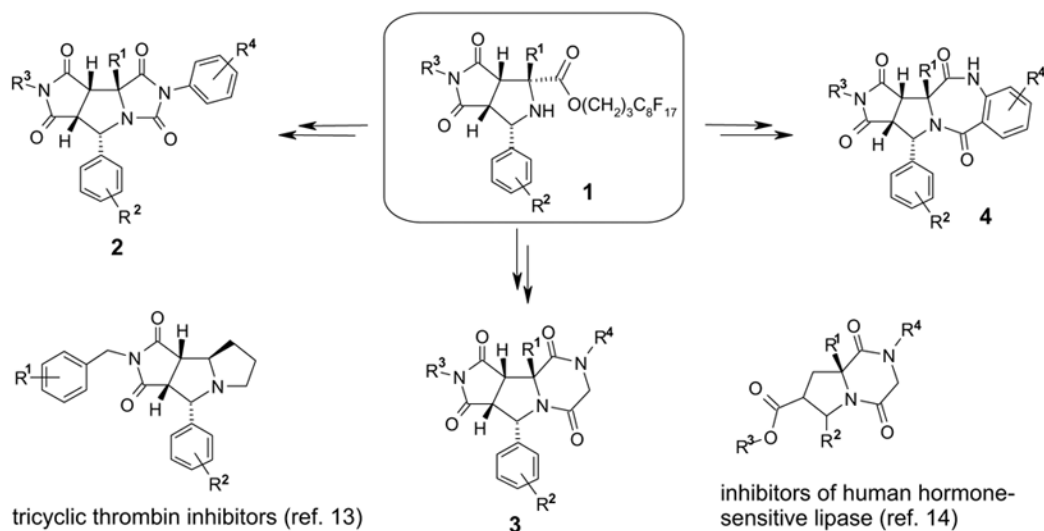
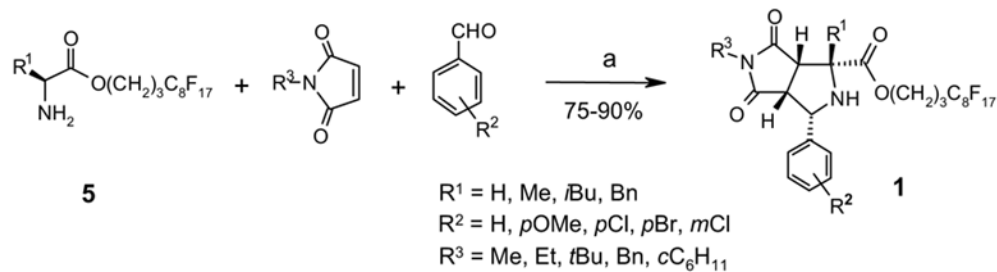


Figure 2.
Single-crystal X-ray structures of compounds **1a** and **2a**

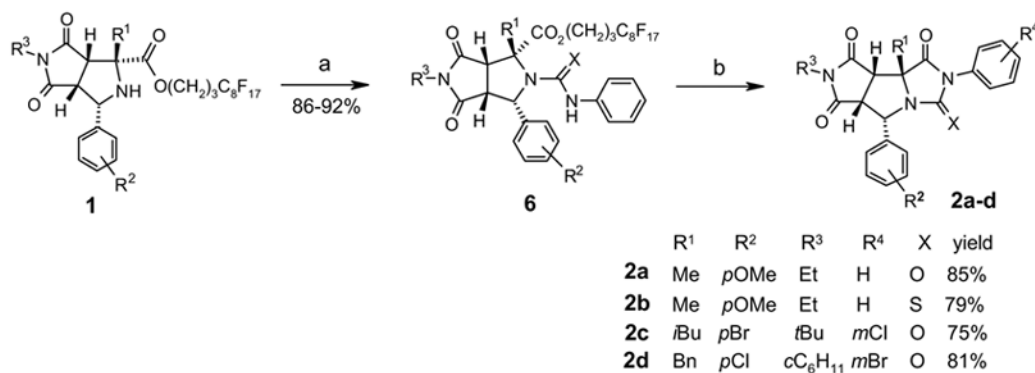


Scheme 1.
Fluorous Synthesis of Heterocyclics **2–4**

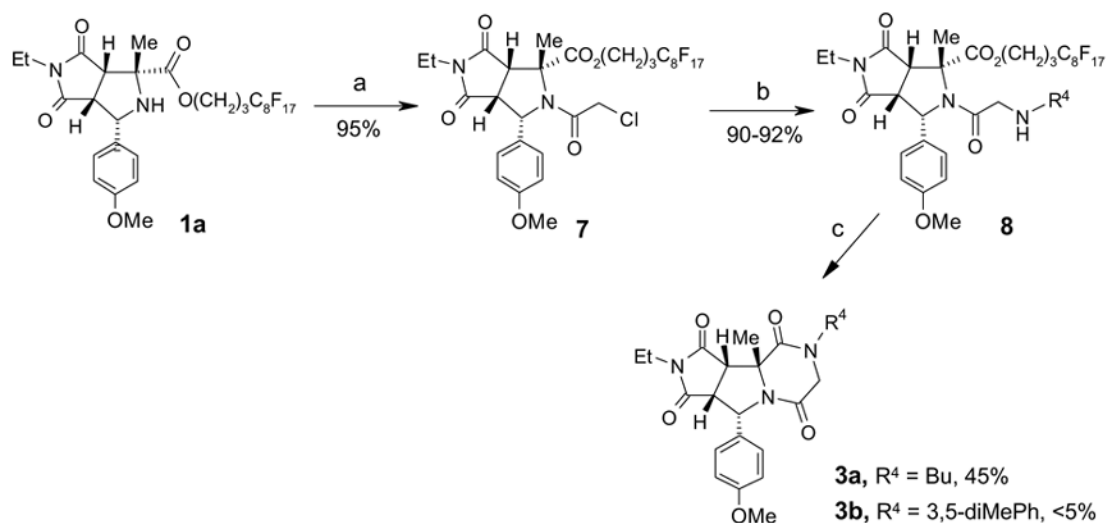
**Scheme 2.**

Synthesis of fluorinated proline derivatives by one-pot [3+2] cycloaddition of azomethine ylides.

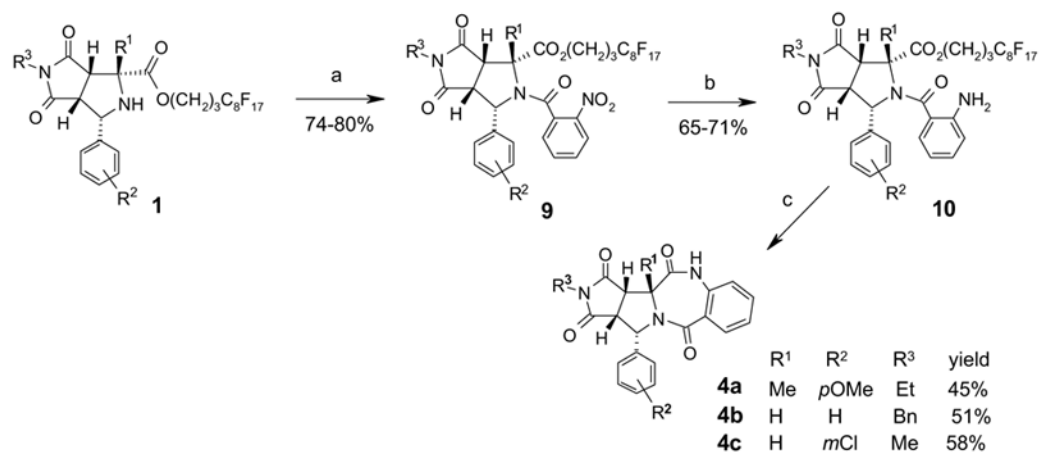
a) **5** (1 equiv), aldehyde (1.2 equiv), maleimide (1.5 equiv), Et₃N (3 equiv), DMF, μw (130 °C, 20 min), F-SPE.

**Scheme 3.**

Synthesis of hydantoin-fused tricyclic compounds **2a–d**. a) R⁴-PhNCX (5.0 equiv), DMAP (0.5 equiv), toluene, μ w (130 °C, 10 min), F-SPE. b) K₂CO₃ (2 equiv), DMF, μ w (100 °C, 5 min), F-SPE, HPLC.

**Scheme 4.**

Synthesis of piperazinedione-fused tricyclic compounds **3a–d**. a) ClCH_2COCl (1.5 equiv), Et_3N (2.5 equiv), CH_2Cl_2 , 25 °C, 30 min, F-SPE. b) R^4NH_2 (2.5 equiv), MeOH, μw (120 °C, 10 min), F-SPE. c) DBU (2 equiv), MeOH-DMF, μw (180 °C, 15 min), F-SPE, HPLC.

**Scheme 5.**

Synthesis of benzodiazepinedione-fused tetracyclic compounds **3a–d**. a) 2-nitrobenzoylchloride (3 equiv), Et₃N (2 equiv), DMF, 80 °C, 2 h, F-SPE. b) Zn dust (10 equiv), AcOH, sonication, 25 °C, 2 h, F-SPE, 65–71%. c) DBU (2 equiv), dioxane, μ w (130 °C, 5 min), F-SPE, HPLC.