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Synthesis and cytotoxic activity of 5-(1-hydroxy-2-haloethyl)-, 5-oxiranyland (E)-5-(2-iodovinyl)-2,4-dichloro (or dimethoxy) pyrimidines

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Summary — A series of 5-(1-hydroxy-2-haloethyl) 6, 7, 13, 14a, 5-oxiranyl 8, 9 and (*E*)-5-(2-iodovinyl)-2,4-dichloro(or dimethoxy)pyrimidines 11, 12 were synthesized for evaluation as cytotoxic agents. The nuclear C-2 and C-4 substituents were determinants of activity since the 2,4-dichloro compounds 6, 8 and 11 were more potent ($ED_{50} = 0.2-0.3 \mu g/ml$) than the corresponding 2,4-dimethoxypyrimidine analogues 7, 9 and 12 ($ED_{50} = 4-28 \mu g/ml$), relative to melphalan ($ED_{50} = 0.15 \mu g/ml$), in the *in vitro* L1210 screen. Within the 2,4-dichloro series of compounds 6, 8 and 11, the C-5 substituent was not a determinant of activity. In contrast, in the 2,4-dimethoxypyrimidine series, the C-5 substituents influenced activity significantly where the relative potency order was oxiranyl 9 > -CH(OH)CH₂I 7 > (E)-CH = CHI 12 > CH(OH)CH₂I 13, CH(OH)CHBr(I) 14a and CH(Br)CHOH(I) 14b. The most active compound (*E*)-5-(2-iodovinyl)-2,4-dichloropyrimidine 11 exhibited weak activity in the *in vivo* P388 screen (% T/C = 116 for a 10 mg/kg ip dose) relative to the reference drug 5-fluorouracil (% T/C = 135 for a 20 mg/kg dose).

Résumé — Synthèse et activité cytotoxique de 5-(1-hydroxy-2-haloéthyl)-, 5-oxiranyl- et (E)-5-(2-iodovinyl)-2,4-dichloro (ou diméthoxy)pyrimidines. Une série de 5-(1-hydroxy-2-haloéthyl) 6, 7, 13, 14a, 5-oxiranyl 8, 9 et (E)-5-(2-iodovinyl)-2,4-dichloro (ou diméthoxy)pyrimidines 11, 12 ont été synthétisées pour leur évaluation comme agents cytotoxiques. Les substituants sur les carbones 2 et 4 du noyau sont déterminants pour l'activité : les dérivés 2,4-dichlorés 6, 8 et 11 sont plus puissants $(ED_{50} = 0,2-0,3 \mu g(ml))$ que les analogues 2,4-diméthoxypyrimidine correspondants 7, 9 et 12 $(ED_{50} = 4-28 \mu g/ml)$, en référence au melphalan $(ED_{50} = 0,15 \mu g/ml)$ dans le test in vitro L1210. Dans la série 2,4-dichlorée des composés 6, 8 et 11, le substituant du carbone 5 n'intervient pas sur l'activité. En revanche, dans la série 2,4-diméthoxypyrimidine, les substituants du C_5 agissent de façon significative sur l'activité, C_5 ordre de puissance relative étant oxiranyl 9 > $-CH(OH)CH_2I$ 7 > (E)-CH=CHI 12 > $CH(OH)CH_2I$ 13, CH(OH)CHBR(I) 14a et CH(Br)CHOH(I) 14b. Le composé le plus actif, la (E)-5-(2-iodovinyl)-2,4-dichloropyrimidine 11 révèle une faible activité dans le test in vivo C_5 pour une dose de C_5 C_5

cytotoxic agents / pyrimidines / oxiranes / 1-hydroxy-2-haloethanes

5-Substituted-uracil analogues have been investigated extensively in the search for effective non-toxic antitumor agents [1] that may be incorporated into DNA [2–5]. These compounds may enter pyrimidine metabolic pathways following bioconversion *via* salvage reactions to 5-substituted-2'-deoxyuridines, and then to the corresponding mono-, di- and triphosphates prior to incorporation into DNA [6]. Thus, 5-bromouracil **1a** and 5-vinyluracil **1b** are incorporated into the DNA of *Escherichia coli* [3]. Alternatively, 5-substituted-2'-deoxyuridine monophosphates could act

Activated alkylating agents that react with cellular nucleophiles such as L-cysteine, glutathione or sulf-hydryl-containing enzymes, exhibit cytotoxic activity

as inhibitors of thymidylate synthetase (TS) to exhibit cytotoxic activity [7]. The antitumor activity of 5-fluorouracil 1c, which is not incorporated into DNA [3], is due to its conversion to the nucleotide 5-fluoro-2'-deoxyuridine-5'-monophosphate 2a which inhibits TS [8]. In contrast, there is evidence to suggest that the cytostatic effect of the thymidine analogue 5-hydroxymethyl-2'-deoxyuridine 2b against Ehrlich ascites carcinoma cells is due to a cross-linking of 2b, that has been incorporated in DNA, to chromatin proteins [9].

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[10]. Many oxiranyl compounds, such as the 1,3,5-triglycidyl-s-triazinetrione 3, exhibit significant antineoplastic activity [11]. It has also been demonstrated that 5-(1-hydroxyethyl)uracil 1d reacts with L-cysteine to give a mixture of two separable diastereomers 1e [12]. It is clear that 5-substituted-uracil derivatives offer the potential to exhibit cytotoxic activity *via* a number of mechanisms which are dependent upon the nature of the 5-substituent. We now describe the synthesis and cytotoxic activities of the structurally related 5-(1-hydroxy-2-haloethyl)- 6, 7, 13, 14a, 5-oxiranyl- 8, 9 and (E)-5-(2-iodovinyl)-2,4-dichloro(or dimethoxy)pyrimidines 11, 12 to acquire new structure—activity correlations for novel 5-substituted-uracil derivatives.

Chemistry

The reaction of 5-vinyluracil 1b with phosphorous oxychloride in the presence of N,N-diethylaniline yielded 5-vinyl-2,4-dichloropyrimidine 4 in 63% yield. The subsequent reaction of 4 with sodium methoxide in methanol afforded 5-vinyl-2,4-dimethoxypyrimidine 5 in 71% yield. The 5-(1hydroxy-2-iodoethyl)-2,4-dichloro(or dimethoxy)pyrimidines 6 and 7 were synthesized by reaction of 4 or 5 with iodine in the presence of the oxidizing agent iodic acid in 68 and 53% yields, respectively as illustrated in scheme 1. The ¹³C NMR spectra of 6 and 7 provided conclusive evidence for the regiospecific addition of HOI across the vinylic double bond of 4 and 5. The iodine atom is attached to a methylene carbon that exhibited a resonance at δ 11.47 and 13.19 whereas the hydroxyl substituent is attached to a methine carbon that exhibited a resonance at δ 68.04 and 67.77 for the respective compounds 6 and 7. Treatment of 6 or 7 with NaOH or Ca(OH)₂ afforded the corresponding oxiranes 8 and 9 in 58 and 94% yields, respectively. In contrast, it was demonstrated in an earlier study that the reaction of 5-(1-hydroxy-2iodoethyl)uracil **1f** with NaOMe [7], or with NaOH, Ca(OH)₂ or Na₂CO₃ under similar reaction conditions, did not afford 5-oxiranyluracil **1g**. Further attempts to synthesize 5-oxiranyluracil by oxidation of 5-vinyluracil **1b** using *m*-chloroperbenzoic acid or oxone were also unsuccessful in which case unreacted 5-vinyluracil was recovered. In a related reaction Jones *et al* observed that the oxidation of 5-vinyl-2'-deoxyuridine **2c** with *m*-chloroperbenzoic acid in watertetrahydrofuran yielded 5-(1,2-dihydroxyethyl)-2'-deoxyuridine **2d** which was characterized as its tetra-*O*-acetate derivative. The formation of **2d** was postulated to arise from reaction of the intermediate oxirane that was readily opened by nucleophiles [13].

Reaction of (E)-5-(2-iodovinyl)uracil **10** with phosphorous oxychloride yielded (E)-5-(2-iodovinyl)-2,4dichloropyrimidine 11 in 46% yield (see scheme 2). Treatment of 11 with sodium methoxide in methanol (E)-5-(2-iodovinyl)-2,4-dimethoxypyrimidine 12 in 82% yield. The reaction of 12 with iodine and potassium iodate gave 5-(1-hydroxy-2,2-diiodoethyl)-2,4-dimethoxypyrimidine 13 (66%). In contrast, the related reaction of 12 with N-bromosuccinimide gave a mixture of the regioisomers 14a and 14b in a ratio of 1:1 (76%) that could not be separated by silica gel column or TLC chromatography. The ¹H NMR spectrum (CDCl₃) for the 1:1 mixture of 14a and 14b exhibited two multiplets attributed to the CIIOH resonances at δ 5.1 and 5.14, two separate sets of closely spaced doublets assigned to the CHBr resonances at δ 5.88 and 6.0, and two closely spaced singlets for H-6 at δ 8.38 and 8.40. The ¹³C NMR spectrum provided further evidence for the assigned structures based on the resonances at δ 19.13 (CHBrI of 14a) and 50.46 (CHBr of 14b), 74.36 and 73.31 (CHOH of 14a and 14b).

Reagents: i, I₂, KIO₃, H₂O/MeCN, 5N H₂SO₄ ii, NaOH, dioxane/H₂O (3:7, v/v) or Ca(OH)₂, H₂O

Scheme 1.

Reagents: i, POCl₃, N,N-diethylaniline ii, NaOMe, MeOH iii, I₂, KIO₃, 5N H₂SO₄, MeCN, 60°C iv, N-bromosuccinimide, THF, HOAc, 25°C

Scheme 2.

Treatment of **13**, or a 1:1 mixture of **14a** and **14b**, with aqueous sodium hydroxide in dioxane yielded 5-formyl-2,4-dimethoxypyrimidine **15** in 77–80% yield, rather than the desired 5-(2-iodooxiranyl)-2,4-dimethoxypyrimidine **16** (see scheme 3). Oxidation of

Reagents: i, NaOH, dioxane/H₂O (13 or 14a-b); oxone, disodium. EDTA, NaHCO₃, H₂O (12)

Scheme 3.

(*E*)-5-(2-iodovinyl)-2,4-dimethoxypyrimidine 12 with the strong oxidizing agent oxone monopersulfate (2KHSO₅·KHSO₄·K₂SO₄) also afforded 5-formyl-2,4-dimethoxypyrimidine 15 (36%). A plausible mechanism for the formation of 15 in these reactions involves the formation of the unstable 5-(2-iodooxiranyl) intermediate 16, or bromooxiranyl intermediate for the reaction employing 14a–b, in which the oxirane ring is readily opened under the reaction conditions employed. The successive elimination of 1- and then HCHO would yield 15. This mechanism is consistent with the observation that oxiranes bearing a good leaving substituent are very unstable and highly susceptible to hydrolysis [14].

Structure-activity correlations

The 5-(1-hydroxy-2-iodoethyl) **6–7**, 5-oxiranyl **8–9**, (*E*)-5-(2-iodovinyl) **11–12**, 5-(1-hydroxy-2,2-diiodoethyl) **13**, 5-(1-hydroxy(bromo)-2-bromo(hydroxy)-2-iodoethyl **14a–b** derivatives of 2,4-dichloro (dimethoxy)pyrimidines were investigated in order to determine the effect of their nuclear and C-5 substituents upon cytotoxic activity. The cytotoxic activities acquired using the *in vitro* L1210 and KB, and the *in vivo* P388 leukemic, screens are summarized in table I.

The nuclear C-2 and C-4 substituents were significant determinants of activity since the 2,4dichloro compounds 6, 8 and 11 were more cytotoxic than the corresponding 2,4-dimethoxy analogues 7, 9 and 12. The 2,4-dichloro compounds exhibited cytotoxic activities (ED₅₀ = $0.2 - \bar{0}.3 \mu g/ml$) approaching that of the reference drug melphalan (ED_{50} = 0.15 µg/ml) in the in vitro L1210 screen. Their cytotoxicities were qualitatively similar in the in vitro KB screen (ED₅₀ = 5–10 μ g/ml) relative to the reference drug 5-fluorouridine ($ED_{50} = 0.5 \mu g/ml$). Within the 2,4-dichloro series of compounds 6, 8 and 11, the C-5 substituent was not a determinant of activity. The cytotoxic activities exhibited by 6 (R = CH(OH)- CH_2I), 8 (R = oxiranyl) and 11 [R = (E)-CH=CHI] were similar. In contrast, in the less active 2,4-dimethoxy series of compounds, the differences in cytotoxic activities were large where the relative activity profile was 9 (R = oxiranyl) > 7 [R = CH(OH)CH₂I] > $12 [R = -CH = CHI] > 13 [R = CH(OH)CHI_2]$ and 14a-b [R = CH(OH)CHBr(I), R = CH(Br)CHOH(I)]. A 2,4-dichloro 6 ($\dot{E}D_{50}$ = < 0.3 $\mu g/ml$) or 2,4-dimethoxypyrimidinyl ring 7 ($\dot{E}D_{50}$ = 15 $\mu g/ml$) also conferred superior cytotoxic activity in the L1210 screen relative to a 5-uracil ring 1f (92% cell survival at 50 µg/ml) [7].

Compounds **6**, **8**, **9** and **11** were evaluated in the P388 *in vivo* screen since drugs that exhibit an ED₅₀ < 5 μ g/ml in the *in vitro* L1210 screen are considered to

Table I. Physical and cytotoxicity data for 5-(1-hydroxy-2-iodoethyl) (6, 7), 5-oxiranyl (8, 9), (*E*)-5-(2-iodovinyl) (11,12), 5-(1-hydroxy-2,2-diiodoethyl) (13), 5-(1-hydroxy-2-bromo-2-iodo) (14a) and 5-(1-bromo-2-hydroxy-2-iodoethyl) (14b) derivatives of 2,4-dichloro (or dimethoxy)pyrimidine.

No	R	In vitro L1210 cytotoxicity ED ₅₀ , µg/ml	In vitro KB cytotoxicity ED ₅₀ , µg/ml	In vivo P388 activity % T/C, 20 mg/kga
6	Cl	< 0.3	5	105
7	OMe	15	> 10	ND^b
8	Cl	0.3	5	102
9	OMe	4	> 100	114
11	Cĺ	0.2	10	116°
12	OMe	28	> 10	ND
13	OMe	> 50	> 100	ND
14a-b	d OMe	> 50	> 100	ND
FURD ^e			0.5	ND
FU^f				135
Melphalang		0.15		

^a% T/C = treated/control; ^bND = not determined; ^cTested at a dose of 10 mg/kg ip; ^dTested as a 1:1 mixture of **14a** and **14b**; ^eFUDR = 5-fluorouridine; ^fFU = 5-fluorouracil; ^g4-[*N*-bis-2-(chloroethyl)amino]phenylalanine.

be active cytotoxic agents. In contrast to the *in vitro* test results, all compounds tested (6, 8, 9, 11) were inactive in the P388 *in vivo* screen. The most potent compound (E)-5-(2-iodovinyl)-2,4-dichloropyrimidine 11 (% T/C = 116 at a 10 mg/kg ip dose) exhibited weak activity relative to the reference drug 5-fluorouracil (% T/C = 135 at a 20 mg/kg dose).

Experimental protocols

Chemistry

Melting points were determined with a Buchi capillary apparatus and are uncorrected. Nuclear magnetic resonance (¹H NMR, ¹³C NMR) spectra were measured on a Bruker AM-300 spectrometer. ¹³C NMR (*J* modulated spin echo) spectra were determined in all instances where methyl and methine carbon resonances appear as positive peaks, and methylene and quaternary carbon resonances appear as negative peaks. Chemical shifts are given in ppm relative to TMS as internal standard. Mass spectra were recorded on a Hewlett-Packard 5995A (EI) spectrometer. Thin layer chromatography was performed using

Whatman MK6F silica gel microslides (250 µm thickness). Silica gel column chromatography was carried out using Merck 7734 silica gel (100–200 µm particle size). 5-Vinyluracil **1b** [7], 5-(1-hydroxy-2-iodoethyl)uracil **1f** [7] and (*E*)-5-(2-iodovinyl)uracil **10** [15] were prepared using literature procedures.

5-Vinyl-2,4-dichloropyrimidine 4

A mixture of 5-vinyluracil (0.24 g, 1.74 mmol), POCl₃ (4 ml) and N,N-diethylaniline (0.4 ml) was refluxed for 30 min. The volatile components were removed *in vacuo*, the oil-like residue remaining was dissolved in chloroform and washed with cold water. The chloroform extract was dried (Na₂SO₄) and the residue obtained, after removal of the solvent, was purified by silica gel column chromatography using n-hexane:ethyl acetate (98:2, v/v) as eluant. Removal of the solvent from the fractions containing the product yielded 4 as a yellowish oil (0.192 g, 63.2%) which was used immediately in subsequent reactions; ¹H NMR (Me₂SO-d₆) δ : 9.14 (s, 1H, H-6), 6.85 (dd, J_{trans} = 18 Hz, J_{cis} = 12 Hz, 1H, CH = CHH'); 6.24 (d, J_{trans} = 18 Hz, 1H, CH=CHH'); 5.74 (d, J_{cis} = 12 Hz, CH=CHH').

5-Vinyl-2,4-dimethoxypyrimidine 5

Sodium methoxide (0.97 g, 18 mmol) was added to a solution of 4 (0.095 g, 5.45 mmol) in methanol (15 ml) at 25°C with stirring. The reaction mixture was stirred at 25°C for 52 h. After neutralization to pH 7 using acidic resin (Dowex 50X 8-200) and removal of the solvent *in vacuo*, the residue obtained was purified by silica gel column chromatography using *n*-hexane:ethyl acetate (97:3, v/v) as eluant to afford 5 as a viscous oil [16] (0.065 g, 71.5% yield); ¹H NMR (Me₂SO-d₆) 8: 8.52 (s, 1H, H-6); 6.67 (dd, $J_{trans} = 18$ Hz, $J_{cis} = 12$ Hz, 1H, CH = CHH'); 5.92 (d, $J_{trans} = 18$ Hz, 1H, CH = CHH'); 5.34 (d, $J_{cis} = 12$ Hz, 1H, CH = CHH'); 4.02 and 3.94 (two s, 3H each, OMe). Mass calcd for $C_8H_{10}N_2O_2$: 166. Found: 166.

5-(1-Hydroxy-2-iodoethyl)-2,4-dichloropyrimidine **6** A mixture of 5-vinyl-2,4-dichloropyrimidine **4** (0.04 g, 0.228 mmol), iodine (0.025 g, 0.2 mmol), potassium iodate (0.015 g, 0.07 mmol), water (2 ml), acetonitrile (1 ml) and sulfuric acid (20 μl of 5 N) was stirred at 60°C for 6 h. The reaction mixture was concentrated *in vacuo* and the residue was partitioned between chloroform and water. The chloroform soluble material was then separated by silica gel column chromatography using *n*-hexane:ethyl acetate (95:5, v/v) as eluant to yield **6** (0.05 g, 68%), mp = 100–102°C; ¹H NMR (CDCl₃) δ: 8.80 (s, 1H, H-6); 5.04 (m, 1H, CHOH); 3.66 (d, J_{gem} = 10.9 Hz of d, J_{vic} = 3.0 Hz, 1H, CHHTl); 2.68 (d, $J_{CH,OH}$ = 4.1 Hz, 1H, CHOH, exchanges with deuterium oxide); ¹³C NMR (CDCl₃) δ: 159.87 (C-4); 159.55 (C-2); 159.0 (C-6); 131.81 (C-5); 68.04 (CHOH); 11.47 (CH₂I). Mass calcd for C₆H₅ ³⁵Cl₂IN₂O: 318. Found: 318. Mass calcd for C₆H₅ ³⁷Cl₂IN₂O: 322. Found: 322. Anal calcd for C₆H₅Cl₂IN₂O: C: 22.59; H: 1.58: N: 8.78. Found: C: 22.76; H: 1.54; N: 8.91.

5-(1-Hydroxy-2-iodoethyl)-2,4-dimethoxypyrimidine 7 A mixture of 5 (0.05 g, 0.3 mmol), iodine (0.038 g, 0.3 mmol), potassium iodate (0.015 g, 0.07 mmol), water (2 ml), acetonitrile (0.5 ml) and sulfuric acid (25 μ l of 5 N) was heated at 55°C with stirring for 1 h. Removal of the solvent in vacuo, extraction of the residue with chloroform, drying the chloroform extract (Na₂SO₄) and evaporation of the solvent gave a residue. Purification of this residue by silica gel column chromatography using *n*-hexane:ethyl acetate (85:15, v/v) as eluant afforded 7 as crystals (0.049 g, 53.3%), mp = 68–70°C; 1 H

NMR (CDCl₃) δ: 8.37 (s, 1H, H-6); 4.89 (m, 1H, CHOH); 4.07 and 4.04 (two s, 3H each, OMe); 3.63 (d, $J_{gem} = 10.7$ Hz of d, $J_{vic} = 3.12$ Hz, 1H, CHHT]; 3.44 (d, $J_{gem} = 10.7$ Hz of d, $J_{vic} = 7.31$ Hz, 1H, CHHT]; 2.64 (d, $J_{CHOH} = 5.5$ Hz, 1H, CHOH, exchanges with deuterium oxide); 13 C NMR (CDCl₃) δ : 167.89 (C-4); 164.97 (C-2); 156.43 (C-6); 114.64 (C-5); 67.77 (CHO). (CHOH); 54.83 (OCH₃); 54.07 (OCH₃); 13.19 (CH₂I). Mass calcd for $C_8H_{11}IN_2O_3$: 310. Found: 310. Anal calcd for $C_8H_{11}IN_2O_2$: C: 30.98; H: 3.57; N: 9.03. Found: C: 30.81; H: 3.66; N: 8.99.

5-Oxiranyl-2,4-dichloropyrimidine 8

A solution of 5-(1-hydroxy-2-iodoethyl)-2,4-dichloropyrimidine 6 (0.058 g, 0.182 mmol) in dioxane-water (3:7, v/v, 3 ml) and sodium hydroxide (0.01 g, 0.25 mmol) was stirred at 25°C for 30 min. The solvent was removed in vacuo and the residue was extracted with chloroform (25 ml). Removal of the solvent in vacuo afforded a residue which was purified by silica gel *in vacuo* arroraed a residue which was purified by silica get column chromatography. Elution with *n*-hexane:ethyl acetate (96:4, v/v) gave **8** (0.02 g, 57.6%), mp = 52–53°C; ¹H NMR (CDCl₃) δ : 8.46 (s, 1H, H-6); 4.13 (dd, J_{vic} = 3.6 Hz, J_{vic} = 2.4 Hz, 1H, oxiranyl CH); 3.30 (dd, J_{gem} = 5.4 Hz, J_{vic} = 3.6 Hz, 1H CHH'); 2.77 (dd, J_{gem} = 5.4 Hz, J_{vic} = 2.4 Hz, 1H, CHH'). Mass calcd for $C_6H_4^{35}Cl_2N_2O$: 190. Found: 190. Mass calcd for $C_6H_4^{37}Cl_2N_2O$: 194. Found: 194. Anal calcd for $C_6H_4^{47}Cl_2N_2O$: $C_6H_4^{37}Cl_2N_2O$: 195. No. 14 66. Found: $C_6H_4^{37}Cl_2N_2O$: 25. No. 14 66. Found: $C_6H_4^{37}Cl_2N_2O$: 25. No. C: 37.72; H: 2.10; N: 14.66. Found: C: 37.66; H: 2.25; N: 14.24.

5-Oxiranyl-2,4-dimethoxypyrimidine 9

Calcium hydroxide (0.44 g, 6 mmol) was added to 7 (0.112 g, 0.36 mmol) in chloroform (20 ml) and water (10 ml). The reaction was allowed to proceed with stirring at 25°C for 3 h. The chloroform layer was separated, dried (Na₂SO₄) and the solvent was removed in vacuo. Recrystallization of the residue obtained from chloroform-hexane yielded 9 (0.062 g, 94%), mp = 52° C; ¹H NMR (CDCl₃) δ : 8.06 (s, 1H, H-6); 4.02 and in $J_{c} = J_{c} + J$ 5.46; N: 14.85.

(E)-5-(2-Iodovinyl)-2,4-dichloropyrimidine 11

A mixture of **10** (0.475 g, 1.8 mmol), POCl₃ (4 ml) and N_1N_2 diethylaniline (0.4 ml) was refluxed for 30 min. Isolation and purification of the product, as described for 4, afforded 11 as light yellow crystals (0.25 g, 46%), mp = 72–73°C; ¹H NMR (CDCl₃) δ : 8.62 (s, 1H, H-6); 7.62 (d, J = 15 Hz, 1H, CH=CHI); 7.29 (d, J = 15 Hz, 1H, CH=CHI). Mass calcd for $C_6H_3^{35}Cl_2IN_2$: 300. Found: 300. Mass calcd for $C_6H_3^{37}Cl_2IN_2$: 304. Found: 304. Anal calcd for C₆H₃Cl₂IN₂: C: 23.94; H: 1.00; N: 9.31. Found: C: 24.12; H: 0.97; N: 9.09.

(E)-5-(2-Iodovinyl)-2,4-dimethoxypyrimidine 12

Reaction of 11 (0.05 g, 0.165 mmol) in methanol (20 ml) with sodium methoxide (0.24 g, 4.5 mmol), and isolation of the product as described for the preparation of 3, afforded 12 as colorless crystals (0.04 g, 82%), mp = 89–90°C; ¹H NMR (CDCl₃) δ : 8.18 (s, 1H, H-6); 7.33 (d, J = 15 Hz, 1H, CH=CHI); 7.04 (d, J = 15 Hz, 1H, CH=CHI); 4.07 and 4.04 (two s, 3H each, OMe). Mass calcd for $C_8H_0IN_2O_2$: 292. Found: 292. Anal calcd for $C_8H_0IN_2O_2$ ·1/4 H_2O : C: 32.39; H: 3.22; N: 9.44. Found: C: 32.35; H: 3.00; N: 9.09.

5-(1-Hydroxy-2,2-diiodoethyl)-2,4-dimethoxypyrimidine 13 A solution of 12 (0.05 g, 0.172 mmol), iodine (0.022 g, 0.17 mmol), potassium iodate (0.01 g, 0.047 mmol), acetonitrile (3 ml), water (2 ml) and sulfuric acid (15 μ l of 5 N) was maintained at 60°C for 4 h with stirring. Removal of the solvent in vacuo gave a residue which was purified by silica gel column chromatography using hexane:ethyl acetate (85:15, v/v) as eluant to yield 13 (0.05 g, 66.8%) as a white solid after recrystallization from hexane, mp = $118-119^{\circ}$ C; ¹H NMR (CDCl₃) δ : 8.46 (s, 1H, H-6); 5.58 (d, J = 3.6 Hz, 1H, CHI₂); 4.72 (d, $J_{\text{CH,OH}}$ = 5.0 Hz of d, $J_{\text{CH,CH}}$ = 3.6 Hz, 1H, CHOH); 4.06 and 4.04 (two s, 3H each, OMe); 2.92 (d, J = 5.0 Hz, 1H, CHOH, exchanges with deuterium oxide). Mass calcd for $C_8H_{10}I_2N_2O_3$: 436. Found: 436. Anal calcd for $C_8H_{10}I_2N_2O_3$: C: 22.03; H: 2.31; N: 6.42. Found: C: 22.42; H: 2.23; N: 7.08.

5-(1-Hydroxy-2-bromo-2-iodoethyl)-2,4-dimethoxypyrimidine 14a and 5-(1-bromo-2-hydroxy-2-iodoethyl)-2,4-dimethoxy-

pyrimidine 14b

A mixture of N-bromosuccinimide (0.03 g, 0.17 mmol), 12 (0.05 g, 0.17 mmol), water (2 ml), THF (2 ml) and glacial acetic acid (10 μ l) was allowed to stir at 25°C for 24 h. Removal of the solvent in vacuo and purification of the product by elution from a silica gel column using hexane:ethyl acetate (88:12, v/v) as eluant yielded a mixture of the two regioisomers **14a** and **14b** in a ratio of 1:1, as a white solid (0.05 g, 76%) after recrystallization form hexane, mp = 101-102°C; 1H NMR (CDCl₃) δ : 8.38 and 8.40 (two s, 2H total, H-6); 6.0 (d, J =3.68 Hz, 1H, CHBr in one regioisomer); 5.88 (d, J = 3.68 Hz, 1H, CHBr in one regioisomer); 5.14 (d, $J_{\text{CH,CH}}$ = 3.68 Hz of d, $J_{\text{CH,OH}}$ = 5 Hz, 1H, CHOH in one regioisomer); 5.1 (d, $J_{\text{CH,CH}}$ = 3.68 Hz of d, $J_{\text{CH,OH}} = 5.0$ Hz, 1H, CHOH in one regioisomer); 4.02 and 4.0 (two s, 12H total, OMe); 2.95 and 2.91 (two d, $J_{\text{CHOH}} = 5$ Hz each, 2H total, OH, exchange with deuterium oxide); ¹³C NMR (CDCl₃) δ : 19.13 (CHBrI of **14a**); 50.46 (CHBr of 14b); 55.01 and 54.30 (OCH₃ of 14a and 14b); 74.36 and 73.31 (CHOH of 14a and 14b); 112.01 (C-5); 158.07 (C-6); 167.69 (C-4). Mass calcd for $C_8H_{10}^{79}BrIN_2O_3$: 388. Found: 388. Anal calcd for $C_8H_{10}BrIN_2O_3$ -1 H_2O : C: 23.58; H: 2.94; N: 6.87. Found: C: 23.55; H: 2.73; N: 7.18.

5-Formyl-2,4-dimethoxypyrimidine 15. Method A, synthesis from 13

Sodium hydroxide (3.2 mg, 0.08 mmol) was added to a solution of **13** (0.011 g, 0.025 mmol) in dioxane (2 ml) and water (80 µl) and the reaction was allowed to proceed at 25°C for 90 min with stirring. The solvent was removed in vacuo and the residue was dissolved in methanol. Silica gel (2 g) was added to this methanol solution and the solvent was removed. This material was applied to the top of a silica gel column. Elution with ethyl acetate:hexane (5:95, v/v) and collection of the fractions containing the product afforded **15** as a white solid (3.4 mg, 80%), mp = 120–121°C (Lit [17] 123°C); ¹H NMR (CDCl₃) δ: 10.20 (s, 1H, CHO); 8.80 (s, 1H, H-6); 4.10 and 4.15 (two s, 3H each, OMe); ¹³C NMR (CDCl₃) δ: 55.61 and 54.48 (OCH₃); 112.29 (C-5); 161.37 (C-6); 167.55 (C-2); 171.11 (C-4); 186.31 (CHO). Mass calcd for C₇H₈N₂O₃: 168. Found: 168. Anal calcd for $C_7H_8N_2O_3$: C: 49.99; H: 4.79; N: 16.66. Found: C: 50.23; H: 4.65; N: 16.45.

Method B, synthesis from 14a-b

Sodium hydroxide (5 mg, 0.125 mmol) was added to a solution of 14a-b (15 mg, 0.038 mmol) in dioxane (2 ml) and water (1 ml), and the reaction was allowed to proceed at 25°C for 90 min with stirring. Removal of the solvent in vacuo and purification of the product by elution from a silica gel column using hexane:ethyl acetate (95:5, v/v) as eluant yielded 15 as a white solid (5 mg, 77%), which was identical (mp, ¹H NMR, MS) to 15 described under Method A.

Method C, synthesis from 12

Sodium bicarbonate (18 mg, 0.21 mmol) and the water (0.3 ml) was added slowly to a stirred slurry of 12 (15 mg, 0.05 mmol) in acetone (0.1 ml). A solution of oxone monopersulfate (2KHSO₅•KHSO₄•K₂SO₄) (0.03 g) in 4 x 10⁻⁴ M aqueous disodium EDTA (0.12 ml) was added dropwise during 30 min to the thick mixture with stirring. After the addition was complete, the mixture was stirred at 25°C for 2 h. Removal of the solvent in vacuo, extraction of the residue with chloroform, drying the chloroform extract (Na₂SO₄) and removal of the solvent gave a residue. Purification of this residue by silica gel column chromatography using n-hexane:ethyl acetate (96:4, v/v) as eluant afforded 15 as a white solid (3 mg, 36%), which was identical (mp, ¹H NMR, MS) to 15 described under Method A.

Pharmacology

In vitro L1210 cytotoxic screen

Mouse L1210 leukemia cells were cultivated as a suspension in Fischer's medium supplemented with 10% heat-inactivated horse serum and incubated at 37°C in a humidified 5% CO₂ atmosphere to prepare a cell stock solution. The number of cells/ml of medium was determined using a Model ZF Coulter Counter 48 h after incubation. The test compound was dissolved in ethanol:water (5:95, v/v) (7, 9) or ethanol (6, 8, 11-14), and 20 µl of this solution was added to test wells containing 2 ml of suspended L1210 cells (105 cells/ml) such that 2 ml of the cell suspension had a test compound concentration of 50, 10 and 1 µg/ml of medium, respectively. Control wells were identical, except that the test compound was absent. Melphalan, the reference compound, had the following test compound concentrations (10, 1, 0.5, 0.1 and 0.05 $\mu g/ml$ of medium). All test and controls were grown in triplicate. The % cell survival was calculated using the formula: % survival = $(T_{48}-T_0)/(C_{48}-C_0)$ x 100; where T_{48} is the number of cells/ml for each test drug concentration at 48 h, T_0 is the number for test wells at time zero (normally 10^5), C_{48} is the number for the control at 48 h, and C_0 is the number for the control at time zero (normally, $T_0 = C_0 = 10^5$ cells/ml). Compounds exhibiting an ED₅₀ $< 5 \mu g/ml$ are considered to be active in this screen.

In vitro KB cytotoxic screen

A solution (50 µl) of the test compound in aqueous ethanol (1:1, v/v) which provided final test compound concentrations of 100, 10, 5, 1, 0.5, 0.1 and 0.05 µg/ml, respectively was added to wells of a 96 well-plate. Each well was seeded with a suspension (100 µl) of KB cells (ATCC-CCL17) and the plates were incubated for 3 days at 37°C in a CO₂ incubator. Cytotoxicity toward KB cells was determined using the neutral red dye uptake assay [18]. Thus, 25 µl of a neutral red solution (0.05%) was added to each well, the plates were incubated at 37°C for 2.5 h and then washed with phosphate buffer solution. Washing buffer (50 µl) was added to each well which was incubated for 2 h and the plates were read at 540 nm using a spectrophotometer. The concentration of test compound required to reduce cell survival by 50% of the control value was determined to be the ED_{50} .

In vivo P388 lymphocytic screen

Compounds were tested for their ability to prolong the survival of B₆D₂F₁ mice (6 per group) receiving an implant of P388 mouse leukemia cells (106 cells in 0.2 ml saline) ip on day 0. The test compound was constituted in saline:ethanol (6:1, v/v) and injected (20 mg/kg ip) on days 1 through 5 (0.2 ml). The mice were observed until death, and the mean survival times were compared to controls receiving the same number of cells on day 0 and sterile saline (0.2 ml) for the five following days. Significant in vivo antineoplastic activity is ascribed to compounds exhibiting a % T/C > 125.

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